# Testing the Silica Leakage Hypothesis 

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#### Abstract

The silica leakage hypothesis is one mechanism put forth to explain lower atmospheric $\mathrm{CO}_{2}$ levels during glacial times. Measuring the percent opal from sediment cores recovered in the Eastern and Western Equatorial Pacific (EEP and WEP) and plotting it against time tests the validity of the opal leakage hypothesis. During glacial times a build up of silicic acid occurs in the Southern Ocean (SO) due to low diatom productivity. The excess silicic acid is transported to low latitude waters in the Pacific Ocean via Antarctic Intermediate Water. The silica rich water is upwelled at the Equator where the silicic acid is taken up by diatoms. The opaline frustules become incorporated into the sedimentary record when the diatom dies. The percent opal of sediment samples can be determined from recovered cores. Cores TR163-22 and ME 24 from the EEP show a strong increase in the percent opal 40-60 ka however the cores from the WEP, MD98-2181 and MD06-3067 do not have this signal. There is no increase in any core during the Last Glacial Maximum (LGM). Therefore contribution by the silica leakage hypothesis in reducing $\mathrm{CO}_{2}$ during glacial times is believed to be small.


Keywords: silica leakage hypothesis, silicic acid, $\mathrm{CO}_{2}$, Eastern and Western Equatorial Pacific


#### Abstract

Atmospheric carbon dioxide $\left(\mathrm{CO}_{2}\right)$ concentrations determined from the Vostok ice core in Antarctica show average $\mathrm{CO}_{2}$ concentrations of 180-200 parts per million ( ppm ) during peak glacial times (Petit et al. 1999). Since the Last Glacial Maximum (LGM) approximately 18 ka , $\mathrm{CO}_{2}$ levels have increased by $80-100 \mathrm{ppm}$. Prior to the industrial revolution, $\mathrm{CO}_{2}$ concentrations were roughly 280 ppm ; however, anthropogenic activities have since contributed $70-90 \mathrm{ppm}$ to atmospheric $\mathrm{CO}_{2}$ concentrations (Sigman and Boyle 2000). $\mathrm{CO}_{2}$ is a natural component of the Earth's atmosphere. Though a relatively small fraction of the Earth's atmosphere, it strongly influences global temperatures due to its greenhouse effect. The role of $\mathrm{CO}_{2}$ as a greenhouse gas directly links it to past climate change and predictably future climate change.


The World Ocean is only second in the abundance of carbon to the sediment and crust of the earth. However, carbon in the deep ocean has a residence time of approximately 1000 years compared to carbon in the Earth's crust which has a residence time of 250-300 Myr (Sigman and Boyle 2000). On glacial and interglacial timescales the reservoir of carbon in the deep ocean has the ability to affect atmospheric $\mathrm{CO}_{2}$ through partial pressure relationships with aqueous $\mathrm{CO}_{2}$ in the ocean.

Many factors need to be considered to elucidate the reasons for the natural oscillations of atmospheric $\mathrm{CO}_{2}$. The ocean is the only reservoir big and reactive enough to account for $\mathrm{CO}_{2}$ oscillations on glacial/interglacial time scales (Matsumoto et al. 2002). An increase of 80-100 ppm in atmospheric $\mathrm{CO}_{2}$ needs to be explained by one or more factors. Carbon and $\mathrm{CO}_{2}$ storage, ocean temperature, ocean salinity and the ocean carbon cycle all influence atmospheric $\mathrm{CO}_{2}$. The terrestrial carbon reservoir decreases during times of glaciation due to extended ice cover on land and thus the drawdown of atmospheric $\mathrm{CO}_{2}$ diminishes. Due to buffering by the carbon reservoir in the oceans and the balance of calcium carbonate $\left(\mathrm{CaCO}_{3}\right)$ the addition of $\mathrm{CO}_{2}$ from this source is minimal. As well, the carbon reservoir on land acts as a source not a sink for $\mathrm{CO}_{2}$ during glacial periods which does not explain the decrease in $\mathrm{CO}_{2}$. The drawdown of $\mathrm{CO}_{2}$ into the ocean also depends on temperature and salinity; $\mathrm{CO}_{2}$ is less soluble in warm water then cold and increased salinity reduces $\mathrm{CO}_{2}$ solubility. During glacial times the ocean would have been colder at the surface leading to a greater $\mathrm{CO}_{2}$ drawdown into the world ocean. Due to the large quantity of ice during glaciations the world ocean contained more salt; this reduces the solubility and increases atmospheric $\mathrm{CO}_{2}$. The effect of the above factors on atmospheric $\mathrm{CO}_{2}$ concentrations is minimal at best. Therefore the influence on atmospheric $\mathrm{CO}_{2}$ attributed to the above factors is not the main reason for the $80-100 \mathrm{ppm}$ increase seen between glacial and interglacial periods (Sigman and Boyle 2000). In fact, a hypothesis is needed to explain the low levels of atmospheric $\mathrm{CO}_{2}$ on a scale that fit both the magnitude in change and the time it took for the change to occur. The opal leakage hypothesis aims to do just that.

### 1.1 Silicic acid in the World Ocean and sedimentary opal records

The distribution of nutrients in the world ocean differs with respect to location and depth. Silicic acid is concentrated in surface waters of the Southern Ocean (SO) that surround Antarctica, as well as in the Northern Pacific. Both areas of concentrated silicic acid are due to the upwelling of silica rich deep water. Figure 1.10 shows the distribution of silicic acid in surface waters of the world ocean and Figure 1.11 shows the distribution of silicic acid with respect to depth. During interglacial times the Southern Ocean is preferentially stripped of silicic acid by diatoms in surface waters. Glacial times observe an increase in iron availability equated with an increased dust flux to the SO. Iron availability limits the uptake of silicic acid to nitrate in diatoms (Brzezinski et al. 2002). A build up of silicic acid can also occur in the SO if the number of diatoms are reduced or displaced. Extended sea ice cover stops the production of diatoms in the area where the ice covers the water. If a different phytoplankton species out competes diatoms then a build up of silicic acid could also occur because it is not being taken up by diatoms. If the later is true there is no way to prove this as the phytoplankton, phaeocystis is not preserved in the sedimentary record (Matsumoto et al. 2002). The excess silicic acid is incorporated into Subantarctic Mode Water (SAMW) and transported to low latitude regions of the Eastern and Western Equatorial Pacific (EEP and WEP respectively). Opal records from the SO show a decrease in opal burial during glacial times and an increase during interglacial times. The EEP and upwelling regions of coastal Peru show the opposite; high burial rates of opal during glacial times and low burial rates during interglacial times (Kienast et al. 2006).


Figure 1.10 Distribution of silicate in the World's Ocean at the sea surface (http://iridl.ldeo.columbia.edu/SOURCES/.LEVITUS94/.ANNUAL/.)


Figure 1.11 Profile of silica distribution with depth comparing the Atlantic and Pacific Oceans. $60^{\circ} \mathrm{S}$ is representative of the Southern Ocean. Note the maximum silicic acid concentration occurs at the surface in the Southern Ocean (Sarmiento et al. 2007).

### 1.2 Opal Leakage Hypothesis

The Opal Leakage Hypothesis is one of many possibilities that accounts for the difference in $\mathrm{CO}_{2}$ levels between glacial and interglacial periods. The significant decrease during glacial times is essentially brought on by a shift in dominant organisms. Two organisms, the coccolithophorid and diatom, constitute a substantial amount of marine sediments. Coccolithophorids have calcareous shells while diatoms have siliceous. (Matsumoto et al 2002). Dust fluxes during times of glaciation bring into the oceans increased amounts of iron to the SubAntarctic and Antarctic regions. The iron affects the uptake ratio of silicic acid to nitrate. Typically the $\mathrm{Si}(\mathrm{OH})_{4}: \mathrm{NO}_{3}{ }^{-}$ratio in diatoms is $4: 1$, however; with the iron limitation in affect the ratio becomes more like 1:1; leaving the SO depleted in nitrate and oversaturated in silicic acid (Sarmiento et al 2004, Brzezinski et al. 2002). Iron deficient conditions during interglacial periods reduce nitrate uptake in diatoms; iron rich conditions during glacial periods the $\mathrm{Si}(\mathrm{OH})_{4}: \mathrm{NO}_{3}{ }^{-}$ratio lowers and SO surface waters become depleted in nitrate and oversaturated in silicic acid (Brzezinski et al. 2002). Extended sea ice cover in the Antarctic also diminishes the production of diatoms and the utilization of silicic acid (Takeda 1998). The excess silicic acid is transported via Sub-Antarctic Mode Water (SAMW) to the low latitudes of the Pacific Ocean. In regions of upwelling, mainly along the equator and off the coast of Peru, silicic acid is returned to the surface waters where it is taken up by diatoms to create their opaline structures. Usually, silicic acid is scarce in low latitude oceans and regions of upwelling (Kienast et al 2006). Glacial periods should then show an increase in the amount of biogenic opal in the sedimentary record in equatorial and upwelling regions of the Pacific Ocean.

Due to the availability of nitrate and silicate, diatoms in upwelling regions of the low latitude Pacific dominate the glacial scene. The alkalinity of sea water is balanced by the influx of sediment from continental weathering and the removal of $\mathrm{CaCO}_{3}$ through burial. Alkalinity increases during glacial periods when calcareous coccolithophorids are out stripped by siliceous diatoms. This imbalance perturbs the equilibrium that exists between alkalinity and Dissolved Inorganic Carbon (DIC); Equation 1.0 expresses the equilibrium between the two. To restore equilibrium an increase in the burial of $\mathrm{CaCO}_{3}$ must occur. Therefore the carbonate compensation depth increases in depth to allow more calcium carbonate to be buried. If this is the case then $\mathrm{CO}_{2 \text { (aq) }}$ must also be used to produce $\mathrm{HCO}_{3}{ }^{-}$. The use of oceanic $\mathrm{CO}_{2}$ effectively draws down atmospheric $\mathrm{CO}_{2}$ to restore balance (Bradtmiller et al. 2006).

$$
\begin{equation*}
\mathrm{CO}_{2(\mathrm{aq})}+\mathrm{H}_{2} \mathrm{O}+\mathrm{CO}_{3}{ }^{2-} \leftrightarrow 2 \mathrm{HCO}_{3}^{-} \tag{1.0}
\end{equation*}
$$

### 1.3 Recent studies on low-latitude Pacific sedimentary opal records and modifications of the opal leakage hypothesis

Matsumoto et al. (2002) first proposed the opal leakage hypothesis. However, this thesis uses the differing results from Bradtmiller et al. 2006 paper and Kienast et al. 2006 paper as an avenue to test the silica leakage hypothesis further. Bradtmiller et al. reject the hypothesis but restricted their work to the last 30000 years. Kienast et al. found supporting evidence and ventured back 150000 years in the sedimentary record but used cores mostly found in the EEP.

The objective of this thesis is to expand the testing of the opal leakage hypothesis in space and time which will add to the findings of the above mentioned papers. Core locations will be taken from the WEP, EEP and upwelling regions off the coast of Peru. The time range represented in the cores ranges from 30000 to 250000 years.

### 2.0 PROCEDURE

### 2.1 Study Area

The study area is located in the EEP and WEP with 2 cores in each region as shown in Figure 2.1. Cores TR163-22 and ME0005A-24JC (ME 24) are situated in the EEP. Core TR163-22 is located at $0.1^{\circ} \mathrm{N}, 86.3^{\circ} \mathrm{W}$ at a depth of 2941 m . Core ME 24 is located at $0.31^{\circ} \mathrm{N}$, $92.24^{\circ} \mathrm{W}$ at a depth of 2830 m . Cores MD06-3067 and MD98-2181 are located in the WEP. Core MD06-3067 is located at $6.3^{\circ} \mathrm{N}, 126.3 \mathrm{E}$ at a depth of 1574 m and core MD98-2181 is located at $6.3^{\circ} \mathrm{N}, 125.83^{\circ} \mathrm{E}$ at a depth of 2114 m .

### 2.2 Wet Chemistry

The following procedure outlines the steps for determining the biogenic opal in pelagic sediment modeled after Mortlock and Froelich's 1989 paper. Some steps were altered in order to accommodate the smaller sample size of 20 mg versus $25-200 \mathrm{mg}$. A smaller sample size is used to conserve the sediment for other tests. All lab ware is plastic not glass (silica) in order to reduce the potential error incurred from glassware. The procedure is broken up into 4 days.


Figure 2.1: Locations of the 4 cores used for analysis. The top map is located in the Western Equatorial Pacific and the bottom map is off the coast of South America in the Eastern Equatorial Pacific. (http://iridl.ldeo.columbia.edu/SOURCES/.LEVITUS94/.ANNUAL/.temp/)

On day one, the ground sediment samples were weighed, approximately 20 mg , and placed into centrifuge tubes. A batch of 96 sediment samples is possible which included calibration standards, internal standards and blanks.

On day two each tube received 2 millilitres $(\mathrm{ml})$ of $10 \% \mathrm{H}_{2} \mathrm{O}_{2}$ and then was immediately swirled. The $\mathrm{H}_{2} \mathrm{O}_{2}$ was added in order to oxidize the organic matter. After a $1 / 2$ hour wait, 2 ml of $10 \% \mathrm{HCl}$ were added to each tube in order to dissolve the inorganics present $\left(\mathrm{CaCO}_{3}\right) . \mathrm{Next}$, in batches of 32 , the tubes were sonicated for a $1 / 2$ hour which breaks up cell membranes. After sonication, 20 ml of Doubly Deionized Water (DDW) were added to each tube using an Eppendorf repipetor. While dispensing the DDW the sides of the tube were washed down. The tubes were then centrifuged for 10 minutes at 4000 revolutions per minute ( rpm ) and the supernatant was decanted. All tubes were placed in an oven at $50^{\circ} \mathrm{C}$ and dried overnight.

On day three a water bath was prepared at $85^{\circ} \mathrm{C}$. Before placing the tubes in the water bath, 20 ml of $2 \mathrm{M} \mathrm{Na}_{2} \mathrm{CO}_{3}$ were dispensed into each tube. After 1.5 hours in the bath each tube was vortexed. One hour later, each tube was swirled, and swirled again one hour later. The tubes were left in the bath for 1.5 hours without disturbing the sediment. Redox bottles ( 30 ml nalgene bottles) were labelled the same as the centrifuge tubes before dispensing 10 ml of DDW into each bottle with a repipetor. After a total of 5 hours in the bath, $100 \mu \mathrm{l}$ from each centrifuge tube were placed in its corresponding redox bottle. At this time a set of calibration standards of known silica concentration also have $100 \mu \mathrm{l}$ extracted and dispensed in a corresponding redox bottle. An oxidant and reductant were prepared. The oxidant was a $1: 1$ volumetric ratio of ammonium molybdate and hydrochloric acid $(\mathrm{HCl})$. The reductant was a 1:1:1 volumetric ratio of metol-sulphite, oxalic acid and sulphuric acid. 4 ml of oxidant were added to each redox bottle
at 10 second intervals. The same was done with the reductant in the amount of $6 \mathrm{ml}, 20$ minutes after the oxidant was added. Caps were then placed onto the redox bottles which were placed in a dark area overnight to allow the samples to colour.

On the fourth and final day the samples were read in the spectrophotometer at 812 nm . The spectrophotometer is first 'zeroed' using DDW meaning all sample values are relative to the DDW. A light is shone through the sample at a wavelength of 812 nm ; the relative absorbance of the wavelength to the DDW is then displayed on the computer screen. The value read on the computer is the absorption value which along with the weight of sediment is used to calculate the \% opal of each sediment sample (see below).

### 2.3 Calculations

The results for the four cores used in this thesis were obtained by taking the absorption unit recorded off the spectrophotometer and converting that into percent opal. Equation 2.3 illustrates how the absorption unit is calculated into percent opal. The absorption unit is multiplied by the inverse of the slope given by the calibration curve of known silica molarity. The weight is the sample weight of the sediment given in milligrams. The constant 2.4 is multiplied by the final result to correct for the high water content of radiolarians that are present in the sample (Mortlock and Froelich 1989).

$$
\begin{equation*}
((\mathrm{au} * 1 / \text { slope }) / \text { weight } * 56.172) * 2.4=\text { percent opal } \tag{2.3}
\end{equation*}
$$

To monitor and assess the error associated with the procedure blanks, calibration standards and internal standards were added to each batch of samples. The blanks presumably should have an absorption reading on the spectrophometer close to 0.0 absorbance units (au); in the Mortlock and Froelich 1989 paper the reading for blanks was no higher then .003 au. However, the operational blanks for this thesis recorded absorption units of much higher values. The overall au average of all the blanks was .0318 which is 12 times higher than the values given by Mortlock and Froelich. The difference in values is believed to stem from the absorption of light by the chemicals used. To test this, two solutions were prepared that had not undergone the analytical process. The first solution contained DDW and the chemicals associated with the oxidant and reductant. An absorbance unit of .0105 was recorded. The second solution contained the same as the first but $100 \mu \mathrm{l}$ of $\mathrm{Na}_{2} \mathrm{CO}_{3}$ was added. This solution gave a reading of .0285 au. Since, all the samples tested in the spectrophotometer had $\mathrm{Na}_{2} \mathrm{CO}_{3}$ added the unusually high blanks are considered normal for this procedure. Simply by subtracting the average blank au from the recorded au value for the samples solves this source of error. As well, more then the suggested 2-3 blanks were used with each batch of samples to get a collective average. In Table 2.41 the absorption units for all the blanks are given along with the overall average and standard deviation. Overall the reproducibility of the blanks was poor with a standard deviation of $\pm .0083$, however; individual standard deviations better illustrate the reproducibility of the blanks within each analytical batch.

|  | Analytical <br> Batch \# | Au values for <br> Blanks | Standard <br> Deviation | Relative <br> Standard <br> Deviation (\%) |
| :--- | :--- | :--- | :--- | :--- |
|  | 1 | 0.0219, <br> 0.0205, <br> 0.0213 | 0.0007 | 3.308 |
|  |  |  | 0.0225, <br> 0.0253, <br> 0.0327, | 0.0044 |
|  |  | 0.0254 |  | 16.48 |
|  |  | 0.0424, | 0.0029 | 6.957 |
|  |  | 0.0378, |  |  |
|  |  | 0.0406, |  |  |
| Overall <br> Average au <br> for Blanks |  | 0.0374, | .0019 | 5.290 |
| Overall <br> Standard <br> Deviation |  | 0.0344, |  |  |

Table 2.41: Demonstrates the reproducibility of the blanks within each analytical batch.

Six calibration standards in all were prepared at the beginning to allow the use of them in each batch of samples. The calibration standards are prepared by diluting the primary standard in order to get known molarities (in mM) of silica. This allows a calibration curve to be calculated; the graph allows for the calculation of the inverse of the slope. The inverse of the slope gives a value that can be compared to the values Mortlock and Froelich give as well as
used in the calculation for percent opal. A sample calibration curve is given in Figure 2.41 for the first set of samples run. In Mortlock and Froelich, the inverse of the slope for any given calibration curve is $9.2 \pm 0.2$. Table 2.42 shows the values for the four rounds of analyses with and without the blank correction applied. Without the blank corrections the inverse of the slope on average is 8.65 , however with the correction (subtracting the blank au from the sample au) the inverse of the slope has been in accordance with Mortlock and Froelich's value with an average of 9.39. As well, in each batch of samples 2 known sediment samples, JV Bulk and SN Bulk are run to ensure like values each time.

Analytical error associated with the spectrophotometer can be reduced by re-zeroing every 7-8 samples. In order to monitor the drift that occurs with the spectrophotometer samples are measured more then once during the 2-3 hour time it takes to record au values for all $90+$ samples. This helps to determine if the spectrophotometer has to be re-zeroed and to see the analytical error that is produced from the spectrophotometer. Table 2.43 represents the four times JV Bulk and SN Bulk were recorded during one session with the spectrophotometer. The average relative standard deviation between the two samples is $0.47 \%$ which is negligible compared to the RSD's calculated in Table 2.42.


Figure 2.41: Gives the calibration curve for the first analytical batch. The pink line represents the curve without corrections to account for the high au values given by the balnks. The blue line is the corrected curve.

| Batch \# | 1 | 2 | 3 | 4 |
| :--- | :--- | :--- | :--- | :--- |
| 1/slope | 8.83 | 8.86 | 8.58 | 8.34 |
| $1 /$ slope <br> corrected | 9.37 | 9.55 | 9.45 | 9.19 |

Table 2.42: Gives the values of 1/slope for the calibration curve for each analytical batch. The Mortlock and Froelich value for 1/slope is $9.2 \pm 0.2$ which corresponds to the corrected values (taking into account the blanks).

| Sample | Absorption unit |  |  |  | RSD |
| :--- | :--- | :--- | :--- | :--- | :--- |
| JV Bulk | .5053 | .5096 | .5062 | .5082 | $0.33 \%$ |
| SN Bulk | .2266 | .2271 | .2255 | .2236 | $0.60 \%$ |

Table 2.43: Demonstrates the little variation in absorption units as a result of analytical error produced by the spectrophotometer. RSD is the relative standard deviation.

The reproducibility of the procedure is calculated from running sediment from the same sample through multiple analytical batches. This allows for the error associated with the procedure to be assessed. Table 2.44 gives the percent opal values for the same sample run with different analytical batches. As well the standard deviation and relative standard deviation is given in order to compare the percent opals with differing magnitudes. The $6^{\text {th }}$ repeat for samples 67 I 90 and 67 V 90 are ignored from the calculations as they are considered to be outliers. The RSD ranges from 4.0 to $6.4 \%$ with an average of $5.1 \%$. This means that for any given value of percent opal a range of plus or minus $5.1 \%$ of the value is a more accurate representation of percent opal.

| Sample | \% Opal (repeated for same sample) |  |  |  |  | SD | RSD |  |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| 67 I 90 | 3.7 | 4.4 | 4.2 | 4.1 | 3.8 | 6.5 | 0.26 | $6.4 \%$ |
| 67 V 90 | 4.9 | 5.0 | 5.3 | 4.6 | 4.6 | 7.7 | 0.26 | $5.3 \%$ |
| JV Bulk | 11.9 | 11.1 | 12.2 |  |  |  | 0.47 | $4.6 \%$ |
| SN Bulk | 30.4 | 32.4 | 34.0 |  |  |  | 1.5 | $4.0 \%$ |

Table 2.44 The ability to reproduce results using the same sample and the error associated with the procedure. $S D$ is the standard deviation and $R S D$ is the relative standard deviation.

The results are presented in two ways: one as percent opal versus depth and the other as percent opal versus age. When interpreting both sets of data analytical error, resolution of cores and location were all taken into consideration. The analytical error determined in the procedure outlined previously is $\pm 5.1 \%$ for any given value of percent opal. The resolution of the cores differs for each one according to how many samples were used for their given depth and age. The resolutions for each core are given in Table 3.0. A higher frequency in sampling gives a more complete depth profile and therefore is considered more accurate. However, more than those factors listed above affect the percent opal preserved in a core. The dilution of opal that occurs from the photic zone to the ocean floor, the rate of dissolution and the preservation of opal all affect the outcome of percent opal. These three factors however are not considered in the results section but are explored further in the discussion.

The main focus for the graphs was to determine periods of high percent opal versus periods of low percent opal. The transition from one to the other is not a concern. In order to tell the difference between a point and a period of high or low percent opal more than five data points need to be almost equivalent. Therefore some high and low points on the graphs are ignored as only the larger signals in percent opal are acknowledged and discussed in this thesis.

|  | MD06-3067 | MD98-2181 | TR163-22 | ME 24 |
| :--- | :--- | :--- | :--- | :--- |
| Resolution | 1066 yrs | 803 yrs | 2318 yrs | 1063 yrs |

Table 3.0: The resolution of each core is calculated by dividing the number of samples taken per core into the time span the core represents in order to get the number of years one sample represents.

### 3.1 Percent Opal versus Depth

Figure 3.10 shows the distribution of percent biogenic opal for core MD06-3067 located in the WEP. The first 250 cm of the core represent a section of slightly reduced biogenic opal leading to an overall increase of $0.5 \%$ which occurs after 250 cm to an approximate depth of 1460 cm . After 1460 cm a drop of $1.5 \%$ is seen which then steadies at a percent opal of 3.1. From $0-250 \mathrm{~cm}$ the average percent opal is 4.1 . Slightly higher between 250 and 1460 cm is an average of $4.6 \%$. The last 90 cm of the core drops to an average of $3.1 \%$. The difference in percent opal for different depth ranges is very close to the analytical error of $\pm 5.1 \%$. A trend is described despite this problem due to the location of the core in the open ocean of the WEP and its resolution with respect to depth. Four outliers occur on the graph at depths of $90,120,250$ and 690 centimetres. Each sample was redone to test the validity of the suspected outliers. No specific errors were recorded for any of these samples during the first time through the procedure and therefore are included in the results, however; error may have occurred without notice.


Figure 3.10 Percent opal versus the core depth for core MD06-3067.

Core MD98-2181 was taken from an embayment close to core MD06-3067 in the WEP. The graph for percent opal versus core depth for this core is seen in Figure 3.11. No distinct trends with respect to an increase or decrease in percent opal are seen in this core. Since the core was recovered from an embayment it is assumed that the fluctuation seen in the open ocean with respect to nutrients is not as apparent in enclosed areas. Therefore the trend for the graph in Figure 3.11 is considered to have a constant percent opal. This core has the longest record of sediment with respect to length compared to the other cores. However, the 2000 cm of core only accounts for 30000 years. Therefore there is high accumulation rate of sediment at the location of core MD98-2181 presumably due to its proximity to land.

Core TR163-22 was recovered from the EEP. The graph representing percent opal versus core depth is seen in Figure 3.12. The main feature is a major increase in percent opal that occurs between depths $380-500 \mathrm{~cm}$. The average opal percent at its height is 35.0 . Values are $23.0 \%$ opal $(0-360 \mathrm{~cm})$ and $25.6 \%$ opal $(520-880 \mathrm{~cm})$ for the average of data points to either side of the high percent opal. One outlier occurs at 580 cm depth; the first result from this depth was $10.9 \%$ opal after rerunning the sample the value came out to $34.5 \%$. The increase in opal seen at 800 cm depth is noted but since the duration of the increase is not sufficient it was left out of analysis.

Core ME 24 in figure 3.13 follows the same general pattern as core TR163-22 with one main difference (data points recorded between $0-522 \mathrm{~cm}$ is taken from Kienast et al. 2006). There are two significant increases in percent opal that occur at depths of 84.5 to 224.5 cm and another increase from 592-732 cm. Again there is evidence for an additional increase in percent opal between 1188 and 1593 cm but frequency of sampling differs at this interval compared to the first increase seen in Figure 3.13. In the first increase from 84.5 to 224.5 cm samples are taken every one to three centimetres in contrast to the increase that occurred between 1188 and 1593 cm where samples were taken every twenty to thirty centimetres. During the first 522 cm of the core each sample represents 232 years but from 522 cm down each sample represents 2747 years. For this reason the apparent increase in percent opal from 1188 to 1593 cm is not described in this thesis. The increase in percent opal seen at 592 to 732 cm is accounted for due to the magnitude of the increase compared to the surrounding data points. On average, at that depth the percent opal was 38.6 a considerably increase as no other data points reach above a value of $30 \%$ opal (except for one).


Figure 3.11 Percent opal versus core depth for Core MD98-2181


Figure 3.12 Percent opal versus core depth for Core TR163-22


Figure 3.13 Percent opal versus core depth for Core ME 24

### 3.2 Percent Opal versus Age

Figure 3.20 shows the percent opal changes over time in each of the four cores. The ages of the cores are determined from oxygen isotope stratigraphy. The error for the ages given is $\pm$ 5000 years during glacial periods and $\pm$ 1000-2000 years during interglacial periods (M. Kienast pers. communication 2008). The percent opal values are lower in cores MD98-2181 and MD063067 which are found in the WEP. Most variability in the range of percent opal is found in cores ME 24 and TR163-22 indicative of the upwelling off Peru that transports silicic acid and other nutrients to the surface (Kienast et al. 2006). The most predominant increase in percent opal occurs around 50 ka in both the cores found in the EEP. Core ME 24 shows an increase in
percent opal at approximately 20 ka but core TR163-22 and MD98-2181do not seem to mimic that pattern. However, the increase is seen in core MD06-3067 (more clearly seen in Figure 3.10) occurs at the same time as core ME 24 around 20 ka . This increase lasts much longer occurring over a 180 kyrs time span versus a 10 kyrs time period. In all likely hood the increase seen in core MD06-3067 is not associated with the Last Glacial Maximum since its beginning is some 20000 years ago.

## Percent Opal Over Time



Figure 3.2 Includes all the cores in a comparison of percent opal over time.

### 4.1 Using percent opal as a paleoproxy

Two main problems arise when using percent opal as a paleoproxy for primary productivity by siliceous organisms. The first is dilution of the opal and the second is the dissolution and preservation of biogenic opal. The dilution problem is due to measuring biogenic opal as a percent. Marine sediment mainly comprises calcium carbonate, detrital sediment and biogenic opal. If one of these constituents fluctuates then the percent of each will change accordingly because all three components equal 100 percent. In an area that has a high rate of sedimentation coming from a terrestrial source the percent opal would be lower than in an area that was not experiencing increased sediment input. Cores in the open ocean would be less influenced by the input of terrestrial sediment while cores closer to land as in the case of core MD98-2181 would be more influenced.

The problem of dissolution and preservation is more complicated with many factors that need to be considered. Dissolution affects the amount of biogenic opal that reaches the ocean floor by going into solution and being recycled within the euphotic zone. On average $60 \%$ of biogenic silica produced in surface water goes into solution in the first 50 to 100 meters of the water column (Raguneau et al. 2000). That being said there are factors that contribute to the overall dissolution rate of biogenic opal. The equation for opal's dissolution rate is seen in equation 4.0 and taken from Ragueneau et al. paper 2000.
$\mathrm{V}_{\text {dis }}=\mathrm{k}\left(\left[\mathrm{Si}(\mathrm{OH})_{4}\right]_{\text {sat }}-\left[\mathrm{Si}(\mathrm{OH})_{4}\right] \mathrm{A}_{\text {sp }}\right.$
Where:
$\mathrm{V}_{\text {dis }}$ is the dissolution rate of opal in seawater
k is the first order rate constant
$\mathrm{Si}(\mathrm{OH})_{4 \text { sat }}$ is the solubility of opal
$\mathrm{Si}(\mathrm{OH})_{4}$ is the ambient silica acid concentration
$A_{\text {sp }}$ is the surface area of opal present

Temperature affects both the rate constant and the solubility of biogenic opal. If temperatures are warmer then an increase occurs in both these factors leading to an increase in the dissolution rate. The silicic acid concentration has little effect on the dissolution rate unless the opal crystal structure contains trace elements such as aluminium in place of silicon. The surface area of opal takes into consideration that different diatom species have different dissolution rates (Ragueneau et al. 2000).

Other factors affect the solubility of opal as well. The biogenic opal is covered in organic material that must be stripped before dissolution of the opal can occur. This process occurs through microbial degradation or grazing. The best way for opal to avoid dissolution is through fecal pellets or flocculation. When grazers consume diatoms their opal structures are preserved and expelled through the grazer's fecal matter. Due to the weight and the protection fecal pellets provide, biogenic opal is delivered to the ocean floor faster and with limited contact to sea water. In the same fashion, flocculation works to minimize the solubility of biogenic opal in the under saturated surface waters by quickly transporting it to the deep ocean. The weight of the aggregate of opal accomplishes this. These factors all combine to lower the amount of recycling
that occurs in the surface waters and increase the export of biogenic opal to be preserved in the sediment record.

### 4.2 Discussion of results

Figure 4.20 shows cores ME 24 and TR163-22 compared to the core results from Kienast et al. 2006 paper with respect to percent opal. The strongest correlation occurs between $40-60 \mathrm{ka}$ BP. All cores in Figure 4.2 show an increase in percent opal during this time frame (except for ME 27 because the record does not go back that far). A problem arises when looking at the data from the Last Glacial Maximum (LGM) that occurred between 19-23 ka BP. Not all the cores show an increase in percent opal during the LGM. Cores ME 24, ME 27 and TR163-31 all show an increase in percent opal. This is considered an increase due to the strong correlation from the three cores; where as in other peaks only one core exhibits a rise in percent opal. Neither Kienast et al. 2006 nor Bradtmiller at al. 2006 record an increase during the LGM. There is no strong signal in the rise of percent opal for the glacial period that occurred 125-200 ka. Cores MD98-2181 and MD06-3067 located in the WEP are affected in terms of percent opal but to much lesser extent then cores from the EEP.


Figure 4.20 Compares the results from this thesis (cores ME 24 and TR163-22) and from the Kienast et al. 2006 paper (cores ME 27, TR163-31 and TR163-19). All cores are from the Eastern Equatorial Pacific.

Water travels via the South Equatorial Current (SEC) and North Equatorial Current (NEC) to the WEP before it reaches there it upwells from the Equatorial Undercurrent (EUC) off the coast of Peru in the EEP. A main contributor to the EUC is Subantarctic Mode Water which comes directly from the Southern Ocean (Kienast et al. 2006). Nutrients get used up before travelling to the WEP causing the productivity and therefore the biogenic opal to be reduced compared to the EEP. No peaks in percent opal were observed in either core of the WEP for any given period of glaciation.

Since the only cores to show an increase in percent opal were the two recovered from the EEP and only between the contribution of the silica leakage hypothesis to atmospheric $\mathrm{CO}_{2}$
levels might be minimal. However, the strong signal seen at $40-60$ kyrs BP in cores TR163-22 and ME 24 could reflect the possibility of the silica leakage hypothesis occurring when optimal conditions present themselves. Between $40-60$ ka sea ice cover was high and the dust flux was low in the Southern Ocean (SO) (Kienast et al 2006). The extent of sea ice plus the flux in iron over the last 200 ka is better represented in Figure 4.21. Sea ice cover limits the productivity of diatom growth causing silica acid to build up in the SO (Takeda 1998).


Figure 4.21 Shows the extent of sea ice in the Antarctic and the flux of iron in the Subantarctic for the last 150000 years (Kienast et al. 2006).

### 5.0 Conclusion

The validity of the silica leakage hypothesis as a main component in the change of atmospheric $\mathrm{CO}_{2}$ over glacial/interglacial time periods is debatable. While some cores show promising results with respect to peak percent opal values occurring at glacial times more then one problem arises. First of all not every period of glaciation appears in the sedimentary record with respect to an increase in percent opal. With that said, it is proposed that optimal conditions such as extensive sea ice cover and iron flux must be present in order to evoke the silica leakage hypothesis. Secondly, only the cores from the EEP show any strong signal at all that corresponds with a glacial period. Water that upwells off the coast of Peru and at the equator does reach cores found in the EEP first before circulating to the west; accounting for the low percent opal found in WEP cores compared to the EEP cores. Although the hypothesis does fit with the EEP cores between $40-60 \mathrm{ka}$, a different hypothesis for the lower $\mathrm{CO}_{2}$ levels during glacial times needs to be called upon. There must be a different mechanism for the low levels of $\mathrm{CO}_{2}$; the silica leakage plays a contributing role when conditions are optimal.

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Appendix
Analytical Batch \#1

| Sample \# | Weight (mg) | au | corrected au | Si mM | \%Si (opal) | \% opal | Depth (cm) | Age (ka) |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 67 I 12-13 | 19.75 | 0.0631 | 0.0419 | 0.393 | 1.117 | 2.680 | 12 | 3.997 |
| 67 I 20-21 | 20.83 | 0.0839 | 0.0627 | 0.587 | 1.584 | 3.802 | 20 | 4.184 |
| 67 I 30-31 | 21.22 | 0.0938 | 0.0726 | 0.680 | 1.801 | 4.322 | 30 | 4.434 |
| 67 I 40-41 | 19.64 | 0.0845 | 0.0633 | 0.593 | 1.696 | 4.071 | 40 | 4.711 |
| 67 I 50-51 | 19.79 | 0.0833 | 0.0621 | 0.582 | 1.652 | 3.964 | 50 | 5.038 |
| 67 I 60-61 | 20.01 | 0.0901 | 0.0689 | 0.646 | 1.812 | 4.350 | 60 | 5.409 |
| 67 I 70-71 | 22.18 | 0.0795 | 0.0583 | 0.546 | 1.383 | 3.320 | 70 | 5.875 |
| $67190-91$ | 21.35 | 0.0841 | 0.0629 | 0.589 | 1.551 | 3.722 | 90 | 7.092 |
| 67 I 100-101 | 19.86 | 0.0847 | 0.0635 | 0.595 | 1.683 | 4.039 | 100 | 7.802 |
| 67 I 110-111 | 19.19 | 0.1015 | 0.0803 | 0.752 | 2.202 | 5.286 | 110 | 8.544 |
| 67 I 120-121 | 21.42 | 0.0513 | 0.0301 | 0.282 | 0.740 | 1.775 | 120 | 9.309 |
| 67 I 130-131 | 20.61 | 0.0823 | 0.0611 | 0.573 | 1.560 | 3.745 | 130 | 10.104 |
| 67 I 140-141 | 18.56 | 0.0838 | 0.0626 | 0.587 | 1.775 | 4.261 | 140 | 10.911 |
| 67 II 0-1 | 22.07 | 0.0812 | 0.06 | 0.562 | 1.431 | 3.434 | 150 | 11.729 |
| 67 II 10-11 | 20.05 | 0.0627 | 0.0415 | 0.389 | 1.089 | 2.615 | 160 | 12.563 |
| 67 II 20-21 | 19.96 | 0.0819 | 0.0607 | 0.569 | 1.601 | 3.841 | 170 | 13.403 |
| 67 II 30-31 | 20.93 | 0.0762 | 0.055 | 0.515 | 1.383 | 3.319 | 180 | 14.249 |
| 67 II 40-41 | 18.5 | 0.0777 | 0.0565 | 0.529 | 1.607 | 3.858 | 190 | 15.104 |
| 67 II 60-61 | 22.1 | 0.0821 | 0.0609 | 0.571 | 1.450 | 3.481 | 210 | 16.824 |
| 67 II 70-71 | 20.16 | 0.0773 | 0.0561 | 0.526 | 1.465 | 3.515 | 220 | 17.690 |
| 67 II 90-91 | 20.86 | 0.0887 | 0.0675 | 0.632 | 1.703 | 4.088 | 240 | 19.432 |
| 67 II 100-101 | 20.36 | 0.1996 | 0.1784 | 1.672 | 4.612 | 11.068 | 250 | 20.307 |
| 67 II 110-111 | 20.74 | 0.0862 | 0.065 | 0.609 | 1.650 | 3.959 | 260 | 21.184 |
| 67 II 120-121 | 20.13 | 0.0928 | 0.0716 | 0.671 | 1.872 | 4.493 | 270 | 22.067 |
| 67 II 130-131 | 19.66 | 0.0925 | 0.0713 | 0.668 | 1.909 | 4.581 | 280 | 22.949 |
| 67 II 140-141 | 22.58 | 0.0978 | 0.0766 | 0.718 | 1.786 | 4.285 | 290 | 23.832 |
| 67 III 0-1 | 19.1 | 0.0907 | 0.0695 | 0.651 | 1.915 | 4.596 | 300 | 24.714 |
| 67 III 10-11 | 21.98 | 0.1026 | 0.0814 | 0.763 | 1.949 | 4.678 | 310 | 25.597 |
| 67 III 20-21 | 20.66 | 0.09 | 0.0688 | 0.645 | 1.753 | 4.207 | 320 | 26.480 |
| 67 III 30-31 | 20.22 | 0.0901 | 0.0689 | 0.646 | 1.793 | 4.304 | 330 | 27.361 |
| 67 III 40-41 | 19.82 | 0.0929 | 0.0717 | 0.672 | 1.904 | 4.570 | 340 | 28.244 |
| 67 III 50-51 | 21.03 | 0.1071 | 0.0859 | 0.805 | 2.150 | 5.160 | 350 | 29.127 |
| 67 III 60-61 | 21.95 | 0.1047 | 0.0835 | 0.782 | 2.002 | 4.805 | 360 | 30.010 |
| 67 III 70-71 | 21.45 | 0.0964 | 0.0752 | 0.705 | 1.845 | 4.429 | 370 | 30.892 |
| 67 III 80-81 | 19.66 | 0.1005 | 0.0793 | 0.743 | 2.123 | 5.095 | 380 | 31.775 |
| 67 III 90-91 | 20.44 | 0.0992 | 0.078 | 0.731 | 2.009 | 4.820 | 390 | 32.657 |
| 67 III 100-101 | 21.85 | 0.1002 | 0.079 | 0.740 | 1.903 | 4.567 | 400 | 33.540 |
| 67 III 110-111 | 19.85 | 0.0942 | 0.073 | 0.684 | 1.936 | 4.646 | 410 | 34.422 |
| 67 III 120-121 | 21.28 | 0.0932 | 0.072 | 0.675 | 1.781 | 4.274 | 420 | 35.305 |
| 67 III 130-131 | 19.76 | 0.0897 | 0.0685 | 0.642 | 1.825 | 4.379 | 430 | 36.188 |
| 67 III 140-141 | 20.67 | 0.0907 | 0.0695 | 0.651 | 1.770 | 4.247 | 440 | 37.071 |
| 67 IV 0-1 | 22.47 | 0.1024 | 0.0812 | 0.761 | 1.902 | 4.565 | 450 | 37.953 |
| 67 IV 10-11 | 20.26 | 0.0936 | 0.0724 | 0.678 | 1.881 | 4.514 | 460 | 38.835 |


| Sample \# | Weight (mg) | au | corrected au | Si mM | \%Si (opal) | \% opal | Depth (cm) | Age (ka) |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 67 IV 20-21 | 19.22 | 0.0911 | 0.0699 | 0.655 | 1.914 | 4.594 | 470 | 39.718 |
| 67 IV 30-31 | 21.12 | 0.121 | 0.0998 | 0.935 | 2.487 | 5.969 | 480 | 40.600 |
| 67 IV 40-41 | 19.06 | 0.1016 | 0.0804 | 0.753 | 2.220 | 5.328 | 490 | 41.483 |
| 67 IV 50-51 | 20.87 | 0.1038 | 0.0826 | 0.774 | 2.083 | 5.000 | 500 | 42.366 |
| 67 IV 80-81 | 21.2 | 0.1007 | 0.0795 | 0.745 | 1.974 | 4.737 | 530 | 45.013 |
| 67 IV 90-91 | 20.3 | 0.0656 | 0.0444 | 0.416 | 1.151 | 2.763 | 540 | 45.895 |
| 67 IV 100-101 | 22.77 | 0.105 | 0.0838 | 0.785 | 1.937 | 4.649 | 550 | 46.777 |
| 67 IV 110-111 | 20.47 | 0.1024 | 0.0812 | 0.761 | 2.088 | 5.011 | 560 | 47.652 |
| 67 IV 120-121 | 19.8 | 0.1116 | 0.0904 | 0.847 | 2.403 | 5.767 | 570 | 48.521 |
| 67 IV 130-131 | 20.83 | 0.1053 | 0.0841 | 0.788 | 2.125 | 5.100 | 580 | 49.388 |
| 67 IV 140-141 | 19.57 | 0.0906 | 0.0694 | 0.650 | 1.867 | 4.480 | 590 | 50.255 |
| 67 V 0-1 | 23.51 | 0.1113 | 0.0901 | 0.844 | 2.017 | 4.841 | 600 | 51.121 |
| 67 V 10-11 | 23.69 | 0.1098 | 0.0886 | 0.830 | 1.968 | 4.724 | 610 | 51.987 |
| 67 V 20-21 | 19.75 | 0.0929 | 0.0717 | 0.672 | 1.911 | 4.586 | 620 | 52.854 |
| 67 V 30-31 | 20.08 | 0.096 | 0.0748 | 0.701 | 1.961 | 4.706 | 630 | 53.720 |
| 67 V 50-51 | 20.66 | 0.0948 | 0.0736 | 0.690 | 1.875 | 4.500 | 640 | 54.587 |
| 67 V 60-61 | 20.74 | 0.0884 | 0.0672 | 0.630 | 1.705 | 4.093 | 660 | 56.320 |
| 67 V 70-71 | 21.1 | 0.1034 | 0.0822 | 0.770 | 2.050 | 4.921 | 670 | 57.186 |
| 67 V 80-81 | 19.38 | 0.0996 | 0.0784 | 0.735 | 2.129 | 5.110 | 680 | 58.053 |
| 67 V 90-91 | 22.38 | 0.1088 | 0.0876 | 0.821 | 2.060 | 4.944 | 690 | 58.919 |
| 67 V 100-101 | 18.96 | 0.0856 | 0.0644 | 0.603 | 1.788 | 4.291 | 700 | 59.786 |
| 67 V 110-111 | 20.14 | 0.1001 | 0.0789 | 0.739 | 2.062 | 4.949 | 710 | 60.652 |
| 67 V 120-121 | 20.28 | 0.0975 | 0.0763 | 0.715 | 1.980 | 4.753 | 720 | 61.535 |
| 67 V 130-131 | 20.69 | 0.0925 | 0.0713 | 0.668 | 1.814 | 4.353 | 730 | 62.477 |
| 67 V 140-141 | 19.55 | 0.0951 | 0.0739 | 0.692 | 1.990 | 4.775 | 740 | 63.499 |
| 67 VI 0-1 | 20.84 | 0.1035 | 0.0823 | 0.771 | 2.079 | 4.989 | 750 | 64.587 |
| 67 VI 10-11 | 21.7 | 0.1132 | 0.092 | 0.862 | 2.231 | 5.355 | 760 | 65.684 |
| 67 VI 20-21 | 19.44 | 0.1034 | 0.0822 | 0.770 | 2.226 | 5.341 | 770 | 66.781 |
| 67 VI 30-31 | 19.42 | 0.1064 | 0.0852 | 0.798 | 2.309 | 5.542 | 780 | 67.879 |
| 67 VI 40-41 | 18.42 | 0.1036 | 0.0824 | 0.772 | 2.354 | 5.651 | 790 | 68.976 |
| 67 VI 50-51 | 22.26 | 0.1118 | 0.0906 | 0.849 | 2.142 | 5.141 | 800 | 70.073 |
| 67 VI 60-61 | 19.59 | 0.1006 | 0.0794 | 0.744 | 2.133 | 5.120 | 810 | 71.171 |
| SN Bulk | 20.22 | 0.5073 | 0.4861 | 4.555 | 12.653 | 30.368 |  |  |
| JV Bulk | 21.7 | 0.2257 | 0.2045 | 1.916 | 4.960 | 11.904 |  |  |

Analytical Batch \#2

| Sample \# | Weight mg | au | corrected au | Si mM | \% Si (opal) | \% opal | Core Dept | Age (ka) |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 67 I 120-121 | 20.86 | 0.0895 | 0.063 | 0.602 | 1.62 | 3.888 | 120 | 9.309 |
| 67 II 100-101 | 19.46 | 0.0964 | 0.0699 | 0.668 | 1.927 | 4.625 | 250 | 20.307 |
| 67 IV 90-91 | 23.83 | 0.0876 | 0.0611 | 0.584 | 1.375 | 3.301 | 540 | 45.895 |
| 67 I 90-91, 1 | 24.02 | 0.1053 | 0.0788 | 0.753 | 1.76 | 4.224 | 90 | 7.092 |
| 68 I 90-91, 2 | 20.12 | 0.0899 | 0.0634 | 0.605 | 1.69 | 4.057 | 90 | 7.092 |
| 69 I 90-91, 3 | 20.57 | 0.0895 | 0.063 | 0.602 | 1.643 | 3.943 | 90 | 7.092 |
| 67 V 90-91, 1 | 20.36 | 0.1024 | 0.0759 | 0.725 | 2 | 4.8 | 690 | 58.919 |
| 67 V 90-91, 2 | 19.23 | 0.103 | 0.0765 | 0.731 | 2.134 | 5.122 | 690 | 58.919 |
| 67 V 90-91, 3 | 21.54 | 0.1016 | 0.0751 | 0.717 | 1.87 | 4.489 | 690 | 58.919 |


| Sample \# | Weight (mg) | au | corrected au | Si mM | \%Si (opal) | \% opal | Depth (cm) | Age (ka) |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 67 VI 70-71 | 19.27 | 0.0842 | 0.0577 | 0.551 | 1.606 | 3.855 | 820 | 72.268 |
| 67 VI 80-81 | 23.06 | 0.0841 | 0.0576 | 0.55 | 1.34 | 3.216 | 830 | 73.366 |
| 67 VI 90-91 | 19.84 | 0.089 | 0.0625 | 0.597 | 1.69 | 4.056 | 840 | 74.463 |
| 67 VI 100-110 | 19.58 | 0.0952 | 0.0687 | 0.656 | 1.882 | 4.517 | 850 | 75.561 |
| 67 VI 110-111 | 22.76 | 0.1063 | 0.0798 | 0.762 | 1.881 | 4.514 | 860 | 76.658 |
| $67 . V^{\prime} 120-121$ | 20.18 | 0.0911 | 0.0646 | 0.617 | 1.717 | 4.121 | 870 | 77.756 |
| 67 VI 130-131 | 19.24 | 0.0776 | 0.0511 | 0.488 | 1.425 | 3.419 | 880 | 78.853 |
| 67 VI 140-141 | 19.52 | 0.0796 | 0.0531 | 0.507 | 1.459 | 3.502 | 890 | 79.950 |
| 67 VII 0-1 | 19.77 | 0.0939 | 0.0674 | 0.644 | 1.829 | 4.389 | 900 | 81.048 |
| 67 VII 10-11 | 21.29 | 0.1087 | 0.0822 | 0.785 | 2.071 | 4.971 | 910 | 82.145 |
| 67 VII 20-21 | 22.67 | 0.1045 | 0.078 | 0.745 | 1.846 | 4.43 | 920 | 83.251 |
| 67 VII 30-31 | 20.61 | 0.1027 | 0.0762 | 0.728 | 1.983 | 4.76 | 930 | 84.402 |
| 67 VII 40-41 | 18.11 | 0.0867 | 0.0602 | 0.575 | 1.783 | 4.28 | 940 | 85.703 |
| 67 VII 50-51 | 20.27 | 0.0936 | 0.0671 | 0.641 | 1.776 | 4.262 | 950 | 87.223 |
| 67 VII 60-61 | 22.27 | 0.115 | 0.0885 | 0.845 | 2.132 | 5.116 | 960 | 88.823 |
| 67 VII 70-71 | 20.58 | 0.1131 | 0.0866 | 0.827 | 2.257 | 5.418 | 970 | 90.423 |
| 67 VII 80-81 | 21.22 | 0.1108 | 0.0843 | 0.805 | 2.131 | 5.115 | 980 | 92.023 |
| 67 VII 90-91 | 22.18 | 0.1293 | 0.1028 | 0.982 | 2.486 | 5.967 | 990 | 93.623 |
| 67 VII 100-101 | 18.41 | 0.1107 | 0.0842 | 0.804 | 2.453 | 5.888 | 1000 | 95.220 |
| 67 VII 110-111 | 20.26 | 0.0923 | 0.0658 | 0.628 | 1.742 | 4.181 | 1010 | 96.817 |
| 67 VII 120-121 | 20.37 | 0.0963 | 0.0698 | 0.667 | 1.838 | 4.412 | 1020 | 98.417 |
| 67 VII 130-131 | 24.2 | 0.1091 | 0.0826 | 0.789 | 1.831 | 4.394 | 1030 | 100.014 |
| 67 VII 140-141 | 21.92 | 0.0889 | 0.0624 | 0.596 | 1.527 | 3.665 | 1040 | 101.615 |
| 67 VIII 0-1 | 23.46 | 0.1122 | 0.0857 | 0.818 | 1.96 | 4.703 | 1050 | 103.220 |
| 67 VIII 10-11 | 21.9 | 0.1176 | 0.0911 | 0.87 | 2.232 | 5.356 | 1060 | 104.818 |
| 67 VIII 20-21 | 19.97 | 0.0879 | 0.0614 | 0.586 | 1.649 | 3.958 | 1070 | 106.414 |
| 67 VIII 30-31 | 21.98 | 0.0885 | 0.062 | 0.592 | 1.513 | 3.632 | 1080 | 107.909 |
| 67 VIII 40-41 | 20.8 | 0.0987 | 0.0722 | 0.69 | 1.862 | 4.469 | 1090 | 109.229 |
| 67 VIII 50-51 | 24.41 | 0.0972 | 0.0707 | 0.675 | 1.554 | 3.729 | 1100 | 110.378 |
| 67 VIII 60-61 | 19.04 | 0.0878 | 0.0613 | 0.585 | 1.727 | 4.145 | 1110 | 111.463 |
| 67 VIII 70-71 | 22.2 | 0.0967 | 0.0702 | 0.67 | 1.696 | 4.071 | 1120 | 112.543 |
| 67 VIII 80-81 | 22.44 | 0.0842 | 0.0577 | 0.551 | 1.379 | 3.31 | 1130 | 113.617 |
| 67 VIII 90-91 | 18.78 | 0.095 | 0.0685 | 0.654 | 1.957 | 4.696 | 1140 | 114.695 |
| 67 VIII 100-101 | 19.85 | 0.102 | 0.0755 | 0.721 | 2.04 | 4.897 | 1150 | 115.776 |
| 67 VIII 110-111 | 21.97 | 0.0878 | 0.0613 | 0.585 | 1.497 | 3.592 | 1160 | 116.849 |
| 67 VIII 120-121 | 21.59 | 0.0834 | 0.0569 | 0.543 | 1.414 | 3.393 | 1170 | 117.927 |
| 67 VIII 130-131 | 20.32 | 0.0826 | 0.0561 | 0.536 | 1.481 | 3.554 | 1180 | 119.008 |
| 67 VIII 140-141 | 20.44 | 0.0806 | 0.0541 | 0.517 | 1.42 | 3.408 | 1190 | 120.089 |
| 67 IX 0-1 | 19.89 | 0.0919 | 0.0654 | 0.625 | 1.764 | 4.233 | 1200 | 121.170 |
| 67 IX 10-11 | 21.99 | 0.096 | 0.0695 | 0.664 | 1.695 | 4.069 | 1210 | 122.246 |
| 67 IX 20-21 | 21.68 | 0.0951 | 0.0686 | 0.655 | 1.697 | 4.074 | 1220 | 123.319 |
| 67 IX 30-31 | 21.28 | 0.1017 | 0.0752 | 0.718 | 1.896 | 4.55 | 1230 | 124.396 |
| 67 IX 40-41 | 24.64 | 0.0929 | 0.0664 | 0.634 | 1.446 | 3.469 | 1240 | 125.476 |
| 67 IX 50-51 | 21.38 | 0.106 | 0.0795 | 0.759 | 1.995 | 4.787 | 1250 | 126.556 |
| 67 IX 60-61 | 19.67 | 0.0732 | 0.0467 | 0.446 | 1.274 | 3.057 | 1260 | 127.636 |
| 67 IX 70-71 | 22.63 | 0.1212 | 0.0947 | 0.904 | 2.245 | 5.388 | 1270 | 128.709 |


| Sample \# | Weight (mg) | au | corrected au | Si mM | \%Si (opal) | \% opal | Depth (cm) | Age (ka) |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 67 IX 80-81 | 20.07 | 0.1045 | 0.078 | 0.745 | 2.085 | 5.004 | 1280 | 129.786 |
| 67 IX 90-91 | 18.57 | 0.1033 | 0.0768 | 0.733 | 2.219 | 5.325 | 1290 | 130.866 |
| 67 IX 100-101 | 18.54 | 0.0926 | 0.0661 | 0.631 | 1.913 | 4.59 | 1300 | 131.946 |
| 67 IX 110-111 | 20.28 | 0.1027 | 0.0762 | 0.728 | 2.016 | 4.838 | 1310 | 133.019 |
| 67 IX 120-121 | 18.75 | 0.0827 | 0.0562 | 0.537 | 1.608 | 3.859 | 1320 | 134.096 |
| 67 IX 130-131 | 21.82 | 0.089 | 0.0625 | 0.597 | 1.537 | 3.688 | 1330 | 135.176 |
| 67 IX 140-141 | 20.73 | 0.0944 | 0.0679 | 0.648 | 1.757 | 4.217 | 1340 | 136.254 |
| $67 \times 0-1$ | 19.43 | 0.1022 | 0.0757 | 0.723 | 2.09 | 5.016 | 1350 | 137.330 |
| $67 \times 10-11$ | 19.39 | 0.095 | 0.0685 | 0.654 | 1.895 | 4.548 | 1360 | 138.406 |
| $67 \times 20-21$ | 22.37 | 0.1058 | 0.0793 | 0.757 | 1.902 | 4.564 | 1370 | 139.486 |
| $67 \times 30-31$ | 23.46 | 0.1037 | 0.0772 | 0.737 | 1.765 | 4.237 | 1380 | 140.566 |
| $67 \times 40-41$ | 24.25 | 0.1058 | 0.0793 | 0.757 | 1.754 | 4.21 | 1390 | 141.639 |
| 67 X 50-51 | 23.3 | 0.1109 | 0.0844 | 0.806 | 1.943 | 4.664 | 1400 | 142.716 |
| 67 X 60-61 | 20.14 | 0.101 | 0.0745 | 0.711 | 1.984 | 4.762 | 1410 | 143.796 |
| 67 X 70-71 | 18.05 | 0.1046 | 0.0781 | 0.746 | 2.321 | 5.571 | 1420 | 144.876 |
| $67 \times 80-81$ | 23.06 | 0.115 | 0.0885 | 0.845 | 2.059 | 4.941 | 1430 | 145.956 |
| $67 \times 90-91$ | 21.43 | 0.1067 | 0.0802 | 0.766 | 2.008 | 4.818 | 1440 | 147.029 |
| $67 \times 100-101$ | 20.38 | 0.1001 | 0.0736 | 0.703 | 1.937 | 4.65 | 1450 | 148.106 |
| $67 \times 110-111$ | 18.88 | 0.1024 | 0.0759 | 0.725 | 2.157 | 5.176 | 1460 | 149.186 |
| $67 \times 120-121$ | 21.5 | 0.0997 | 0.0732 | 0.699 | 1.826 | 4.383 | 1470 | 150.266 |
| $67 \times 130-131$ | 19.63 | 0.0806 | 0.0541 | 0.517 | 1.478 | 3.548 | 1480 | 151.339 |
| 67 X 140-141 | 23.09 | 0.0894 | 0.0629 | 0.601 | 1.461 | 3.507 | 1490 | 152.418 |
| 67 XI 0-1 | 20.37 | 0.0674 | 0.0409 | 0.391 | 1.077 | 2.585 | 1500 | 153.500 |
| 67 XI 10-11 | 24.14 | 0.0718 | 0.0453 | 0.433 | 1.007 | 2.416 | 1510 | 154.576 |
| 67 XI 20-21 | 23.22 | 0.0928 | 0.0663 | 0.633 | 1.532 | 3.676 | 1520 | 155.649 |
| 67 XI 30-31 | 24.62 | 0.0865 | 0.06 | 0.573 | 1.307 | 3.138 | 1530 | 156.726 |
| 67 XI 40-41 | 21.37 | 0.0778 | 0.0513 | 0.49 | 1.288 | 3.091 | 1540 | 157.806 |
| 67 XI 50-51 | 23.26 | 0.0757 | 0.0492 | 0.47 | 1.135 | 2.723 | 1550 | 158.886 |
| SN Bulk | 21.66 | 0.5715 | 0.545 | 5.205 | 13.498 | 32.395 |  |  |
| JV Bulk | 21.64 | 0.2125 | 0.186 | 1.776 | 4.611 | 11.066 |  |  |

Analytical Batch \#3

| Sample \# | Weight mg | au | corrected au | Si mM | \%Si (opal) | \% opal Depth (cm) | Age (ka) |  |
| :--- | ---: | ---: | ---: | ---: | ---: | ---: | ---: | ---: |
| TR 20 | 20.3 | 0.3991 | 0.3576 | 3.379 | 9.351 | 22.442 | 20 | 3.533 |
| TR 40 | 20.78 | 0.3972 | 0.3557 | 3.361 | 9.086 | 21.807 | 40 | 6.156 |
| TR 60 | 21.39 | 0.3906 | 0.3491 | 3.299 | 8.663 | 20.792 | 60 | 8.749 |
| TR 80 | 19.62 | 0.3875 | 0.346 | 3.270 | 9.361 | 22.467 | 80 | 11.336 |
| TR 100 | 20.64 | 0.4323 | 0.3908 | 3.693 | 10.051 | 24.122 | 100 | 13.466 |
| TR 120 | 21.78 | 0.4249 | 0.3834 | 3.623 | 9.344 | 22.426 | 120 | 15.222 |
| TR 140 | 21.68 | 0.4127 | 0.3712 | 3.508 | 9.089 | 21.813 | 140 | 16.988 |
| TR 160 | 19.68 | 0.386 | 0.3445 | 3.256 | 9.292 | 22.301 | 160 | 18.950 |
| TR 180 | 18.25 | 0.4082 | 0.3667 | 3.465 | 10.666 | 25.598 | 180 | 20.912 |
| TR 200 | 19.46 | 0.4275 | 0.386 | 3.648 | 10.529 | 25.270 | 200 | 22.867 |
| TR 220 | 20.35 | 0.3912 | 0.3497 | 3.305 | 9.122 | 21.892 | 220 | 24.815 |
| TR 240 | 22.48 | 0.5073 | 0.4658 | 4.402 | 10.999 | 26.398 | 240 | 26.763 |
| TR 260 | 19.37 | 0.4039 | 0.3624 | 3.425 | 9.931 | 23.835 | 260 | 29.106 |


| Sample \# | Weight (mg) | au | corrected au | Si mM | \%Si (opal) | \% opal | Depth (cm) | Age (ka) |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| TR 280 | 19.36 | 0.43 | 0.3885 | 3.671 | 10.652 | 25.565 | 280 | 31.772 |
| TR 300 | 22.15 | 0.4112 | 0.3697 | 3.494 | 8.860 | 21.264 | 300 | 34.438 |
| TR 320 | 21.49 | 0.4536 | 0.4121 | 3.894 | 10.179 | 24.430 | 320 | 37.086 |
| TR 340 | 20.67 | 0.3559 | 0.3144 | 2.971 | 8.074 | 19.378 | 340 | 39.403 |
| TR 360 | 20.07 | 0.3966 | 0.3551 | 3.356 | 9.392 | 22.541 | 360 | 41.721 |
| TR 380 | 19.46 | 0.5406 | 0.4991 | 4.716 | 13.614 | 32.674 | 380 | 44.038 |
| TR 400 | 19.42 | 0.5475 | 0.506 | 4.782 | 13.831 | 33.194 | 400 | 46.356 |
| TR 420 | 20.06 | 0.5826 | 0.5411 | 5.113 | 14.319 | 34.364 | 420 | 48.673 |
| TR 440 | 20.44 | 0.5884 | 0.5469 | 5.168 | 14.203 | 34.087 | 440 | 50.991 |
| TR 460 | 19.76 | 0.581 | 0.5395 | 5.098 | 14.493 | 34.783 | 460 | 53.308 |
| TR 480 | 20.65 | 0.658 | 0.6165 | 5.826 | 15.848 | 38.034 | 480 | 55.626 |
| TR 500 | 21.37 | 0.6766 | 0.6351 | 6.002 | 15.776 | 37.862 | 500 | 57.943 |
| TR 520 | 20.85 | 0.552 | 0.5105 | 4.824 | 12.997 | 31.193 | 520 | 60.261 |
| TR 540 | 22.51 | 0.5394 | 0.4979 | 4.705 | 11.741 | 28.179 | 540 | 62.578 |
| TR 560 | 22.5 | 0.5191 | 0.4776 | 4.513 | 11.268 | 27.042 | 560 | 64.896 |
| TR 580 | 21.37 | 0.2246 | 0.1831 | 1.730 | 4.548 | 10.916 | 580 | 67.213 |
| TR 600 | 21.75 | 0.4629 | 0.4214 | 3.982 | 10.285 | 24.683 | 600 | 69.531 |
| TR 622 | 22.36 | 0.53 | 0.4885 | 4.616 | 11.597 | 27.833 | 620 | 71.848 |
| TR 640 | 23.06 | 0.5382 | 0.4967 | 4.694 | 11.434 | 27.441 | 640 | 74.166 |
| TR 660 | 23.06 | 0.4915 | 0.45 | 4.253 | 10.359 | 24.861 | 660 | 76.483 |
| TR 680 | 21.27 | 0.4776 | 0.4361 | 4.121 | 10.884 | 26.121 | 680 | 78.801 |
| TR 700 | 21.27 | 0.4251 | 0.3836 | 3.625 | 9.573 | 22.976 | 700 | 81.118 |
| TR 720 | 22.53 | 0.4301 | 0.3886 | 3.672 | 9.156 | 21.974 | 720 | 83.436 |
| TR 740 | 22.03 | 0.4172 | 0.3757 | 3.550 | 9.053 | 21.726 | 740 | 85.753 |
| TR 760 | 20.21 | 0.4513 | 0.4098 | 3.873 | 10.764 | 25.833 | 760 | 88.071 |
| TR 780 | 21.57 | 0.4481 | 0.4066 | 3.842 | 10.006 | 24.015 | 780 | 90.388 |
| TR 800 | 22.15 | 0.6295 | 0.588 | 5.557 | 14.091 | 33.819 | 800 | 92.706 |
| TR 820 | 22.15 | 0.553 | 0.5115 | 4.834 | 12.258 | 29.419 | 820 | 95.023 |
| TR 841 | 22.47 | 0.5016 | 0.4601 | 4.348 | 10.869 | 26.086 | 840 | 97.341 |
| TR 860 | 22.81 | 0.5635 | 0.522 | 4.933 | 12.148 | 29.155 | 860 | 99.658 |
| TR 880 | 24.05 | 0.4945 | 0.453 | 4.281 | 9.998 | 23.996 | 880 | 101.976 |
| MD98-2181, 4 | 22.49 | 0.2182 | 0.1767 | 1.670 | 4.171 | 10.009 | 4 | 0.041 |
| 50 | 21.92 | 0.2064 | 0.1649 | 1.558 | 3.993 | 9.584 | 50 | 0.518 |
| 94 | 20.85 | 0.1949 | 0.1534 | 1.450 | 3.905 | 9.373 | 94 | 0.974 |
| 133 | 19.73 | 0.1794 | 0.1379 | 1.303 | 3.710 | 8.904 | 133 | 1.378 |
| 179 | 22.17 | 0.1744 | 0.1329 | 1.256 | 3.182 | 7.637 | 179 | 1.855 |
| 226 | 21.94 | 0.2176 | 0.1761 | 1.664 | 4.261 | 10.226 | 226 | 2.342 |
| 278 | 20.47 | 0.0073 | -0.0342 | -0.323 | -0.887 | -2.128 | 278 | 2.881 |
| 333 | 21.8 | 0.1932 | 0.1517 | 1.434 | 3.694 | 8.865 | 333 | 3.450 |
| 393 | 22.89 | 0.1952 | 0.1537 | 1.452 | 3.564 | 8.554 | 393 | 4.072 |
| 452 | 19.96 | 0.1614 | 0.1199 | 1.133 | 3.189 | 7.653 | 452 | 4.683 |
| 498 | 21.39 | 0.1785 | 0.137 | 1.295 | 3.400 | 8.160 | 498 | 5.160 |
| 545 | 22.1 | 0.1809 | 0.1394 | 1.317 | 3.348 | 8.036 | 545 | 5.647 |
| 590 | 21.84 | 0.17 | 0.1285 | 1.214 | 3.123 | 7.496 | 590 | 6.113 |
| 635 | 22.42 | 0.1947 | 0.1532 | 1.448 | 3.627 | 8.705 | 635 | 6.580 |


| Sample \# | Weight (mg) | au | corrected au | Si mM | \%Si (opal) | \% opal | Depth (cm) | Age (ka) |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 680 | 21.06 | 0.1939 | 0.1524 | 1.440 | 3.841 | 9.219 | 680 | 7.046 |
| 725 | 20.88 | 0.1815 | 0.14 | 1.323 | 3.559 | 8.542 | 725 | 7.688 |
| 771 | 20.54 | 0.1904 | 0.1489 | 1.407 | 3.848 | 9.235 | 771 | 8.372 |
| 823 | 21.77 | 0.1624 | 0.1209 | 1.143 | 2.948 | 7.075 | 823 | 9.355 |
| 871 | 20.38 | 0.1837 | 0.1422 | 1.344 | 3.704 | 8.889 | 871 | 10.489 |
| 905 | 22.18 | 0.175 | 0.1335 | 1.262 | 3.195 | 7.668 | 905 | 11.288 |
| 947 | 20.39 | 0.1479 | 0.1064 | 1.005 | 2.770 | 6.648 | 947 | 12.251 |
| 989 | 23.62 | 0.1844 | 0.1429 | 1.350 | 3.211 | 7.708 | 989 | 13.215 |
| 1045 | 24.09 | 0.0829 | 0.0414 | 0.391 | 0.912 | 2.189 | 1045 | 14.499 |
| 1095 | 21.42 | 0.1661 | 0.1246 | 1.177 | 3.088 | 7.411 | 1095 | 15.617 |
| 1142 | 20.32 | 0.157 | 0.1155 | 1.091 | 3.017 | 7.241 | 1142 | 16.543 |
| 1189 | 22.6 | 0.1639 | 0.1224 | 1.157 | 2.875 | 6.900 | 1189 | 17.468 |
| 1236 | 19.76 | 0.1704 | 0.1289 | 1.218 | 3.463 | 8.311 | 1236 | 18.375 |
| 1282 | 22.1 | 0.1901 | 0.1486 | 1.404 | 3.569 | 8.566 | 1282 | 19.263 |
| 1330 | 20.18 | 0.1722 | 0.1307 | 1.235 | 3.438 | 8.251 | 1330 | 20.222 |
| 1376 | 22.97 | 0.1869 | 0.1454 | 1.374 | 3.360 | 8.064 | 1376 | 21.144 |
| 1423 | 21.07 | 0.1805 | 0.139 | 1.314 | 3.502 | 8.405 | 1423 | 22.301 |
| 1469 | 20.21 | 0.1922 | 0.1507 | 1.424 | 3.958 | 9.500 | 1469 | 23.490 |
| 1514 | 20.97 | 0.1813 | 0.1398 | 1.321 | 3.539 | 8.493 | 1514 | 24.503 |
| 1561 | 21.06 | 0.1755 | 0.134 | 1.266 | 3.378 | 8.106 | 1561 | 25.465 |
| 1606 | 18.64 | 0.159 | 0.1175 | 1.110 | 3.346 | 8.031 | 1606 | 26.386 |
| 1654 | 22.61 | 0.1953 | 0.1538 | 1.453 | 3.611 | 8.666 | 1654 | 27.369 |
| 1710 | 22.97 | 0.2 | 0.1585 | 1.498 | 3.663 | 8.791 | 1710 | 28.515 |
| 1766 | 20.42 | 0.1823 | 0.1408 | 1.331 | 3.660 | 8.784 | 1766 | 29.661 |
| 1828 | 23.11 | 0.1957 | 0.1542 | 1.457 | 3.542 | 8.501 | 1828 | 30.930 |
| 1888 | 22.2 | 0.177 | 0.1355 | 1.280 | 3.240 | 7.776 | 1888 | 32.158 |
| JV Bulk | 23.57 | 0.2678 | 0.2263 | 2.139 | 5.097 | 12.232 |  |  |
| SN Bulk | 22.41 | 0.6387 | 0.5972 | 5.644 | 14.146 | 33.950 |  |  |
| 67 I 90-91 | 22.1 | 0.1077 | 0.0662 | 0.626 | 1.590 | 3.816 | 90 | 7.092 |
| 67 V 90-91 | 20 | 0.1143 | 0.0728 | 0.688 | 1.932 | 4.637 | 690 | 58.919 |

Analytical Batch \#4

| Sample \# | Weight mg | au | corrected au | Si mM | \%Si (opal) | \% opal | Depth (cm) | Age (ka) |
| :--- | ---: | ---: | ---: | ---: | ---: | ---: | ---: | ---: |
| ME-24, 500 | 22.87 | 0.3813 | 0.3457 | 3.177 | 7.803 | 18.728 | 500 | 36.107 |
| $\mathbf{5 1 6}$ | 23.06 | 0.3931 | 0.3575 | 3.285 | 8.003 | 19.207 | 516 | 36.250 |
| $\mathbf{5 3 6}$ | 21.01 | 0.3964 | 0.3608 | 3.316 | 8.865 | 21.276 | 536 | 36.471 |
| $\mathbf{5 7 6}$ | 24.13 | 0.5941 | 0.5585 | 5.133 | 11.948 | 28.676 | 576 | 37.117 |
| $\mathbf{5 9 2}$ | 19.29 | 0.7045 | 0.6689 | 6.147 | 17.900 | 42.961 | 592 | 37.450 |
| $\mathbf{6 1 2}$ | 20.55 | 0.6265 | 0.5909 | 5.430 | 14.844 | 35.625 | 612 | 37.950 |
| $\mathbf{6 2 8}$ | 19.4 | 0.6864 | 0.6508 | 5.981 | 17.317 | 41.562 | 628 | 41.238 |
| $\mathbf{6 4 8}$ | 21.26 | 0.663 | 0.6274 | 5.766 | 15.234 | 36.562 | 648 | 43.613 |
| $\mathbf{6 6 0}$ | 21.96 | 0.7894 | 0.7538 | 6.927 | 17.720 | 42.528 | 660 | 44.760 |
| $\mathbf{6 8 0}$ | 22.06 | 0.6911 | 0.6555 | 6.024 | 15.339 | 36.814 | 680 | 46.100 |
| $\mathbf{6 9 6}$ | 21.63 | 0.7408 | 0.7052 | 6.481 | 16.830 | 40.393 | 696 | 47.038 |
| $\mathbf{7 1 2}$ | 22.47 | 0.7291 | 0.6935 | 6.373 | 15.932 | 38.238 | 712 | 48.637 |
| $\mathbf{7 3 2}$ | 21.62 | 0.6043 | 0.5687 | 5.226 | 13.579 | 32.589 | 732 | 51.719 |


| Sample \# | Weight (mg) | au | corrected au | Si mM | \%Si (opal) | \% opal | Depth (cm) | Age (ka) |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 748 | 22.01 | 0.5187 | 0.4831 | 4.440 | 11.331 | 27.193 | 748 | 54.000 |
| 768 | 23.58 | 0.5896 | 0.554 | 5.091 | 12.128 | 29.108 | 768 | 56.800 |
| 784 | 22.01 | 0.5132 | 0.4776 | 4.389 | 11.202 | 26.884 | 784 | 58.567 |
| 804 | 22.42 | 0.4557 | 0.4201 | 3.861 | 9.673 | 23.215 | 804 | 60.567 |
| 820 | 24.51 | 0.4251 | 0.3895 | 3.580 | 8.204 | 19.688 | 820 | 61.633 |
| 840 | 20.96 | 0.5896 | 0.554 | 5.091 | 13.644 | 32.747 | 840 | 62.873 |
| 856 | 23.32 | 0.5106 | 0.475 | 4.365 | 10.515 | 25.235 | 856 | 63.663 |
| 876 | 20.76 | 0.396 | 0.3604 | 3.312 | 8.962 | 21.508 | 876 | 64.700 |
| 892 | 23.42 | 0.4933 | 0.4577 | 4.206 | 10.089 | 24.213 | 892 | 66.184 |
| 912 | 21.25 | 0.4379 | 0.4023 | 3.697 | 9.773 | 23.455 | 912 | 67.500 |
| 928 | 23.46 | 0.5319 | 0.4963 | 4.561 | 10.921 | 26.210 | 928 | 68.786 |
| 944 | 22.7 | 0.3728 | 0.3372 | 3.099 | 7.668 | 18.404 | 944 | 70.800 |
| 964 | 23 | 0.3542 | 0.3186 | 2.928 | 7.151 | 17.162 | 964 | 73.393 |
| 980 | 22.97 | 0.3033 | 0.2677 | 2.460 | 6.016 | 14.439 | 980 | 80.583 |
| 996 | 22.8 | 0.3832 | 0.3476 | 3.194 | 7.870 | 18.888 | 996 | 83.671 |
| 1016 | 23.09 | 0.3425 | 0.3069 | 2.820 | 6.861 | 16.467 | 1016 | 87.050 |
| 1032 | 20.37 | 0.3404 | 0.3048 | 2.801 | 7.724 | 18.538 | 1032 | 90.743 |
| 1062 | 21.32 | 0.346 | 0.3104 | 2.853 | 7.516 | 18.038 | 1062 | 94.893 |
| 1072 | 20.26 | 0.3233 | 0.2877 | 2.644 | 7.331 | 17.593 | 1072 | 95.979 |
| 1092 | 21.56 | 0.3478 | 0.3122 | 2.869 | 7.475 | 17.940 | 1092 | 98.308 |
| 1108 | 22.57 | 0.372 | 0.3364 | 3.092 | 7.694 | 18.466 | 1108 | 100.046 |
| 1128 | 20.23 | 0.345 | 0.3094 | 2.843 | 7.895 | 18.948 | 1128 | 102.416 |
| 1144 | 21.02 | 0.3478 | 0.3122 | 2.869 | 7.667 | 18.401 | 1144 | 104.429 |
| 1172 | 23.01 | 0.4148 | 0.3792 | 3.485 | 8.507 | 20.417 | 1172 | 108.371 |
| 1188 | 20.59 | 0.4945 | 0.4589 | 4.217 | 11.505 | 27.613 | 1188 | 110.137 |
| 1208 | 21.5 | 0.4983 | 0.4627 | 4.252 | 11.110 | 26.663 | 1208 | 112.091 |
| 1228 | 21.52 | 0.4684 | 0.4328 | 3.977 | 10.382 | 24.917 | 1228 | 114.438 |
| 1248 | 19.96 | 0.3501 | 0.3145 | 2.890 | 8.134 | 19.521 | 1248 | 117.217 |
| 1264 | 24.75 | 0.4917 | 0.4561 | 4.192 | 9.513 | 22.831 | 1264 | 118.900 |
| 1284 | 22.35 | 0.4314 | 0.3958 | 3.637 | 9.142 | 21.940 | 1284 | 123.344 |
| 1300 | 23.11 | 0.4717 | 0.4361 | 4.008 | 9.741 | 23.379 | 1300 | 127.135 |
| 1320 | 21.47 | 0.4181 | 0.3825 | 3.515 | 9.197 | 22.072 | 1320 | 129.760 |
| 1336 | 20.89 | 0.4013 | 0.3657 | 3.361 | 9.037 | 21.689 | 1336 | 131.800 |
| 1356 | 20.93 | 0.3743 | 0.3387 | 3.113 | 8.354 | 20.049 | 1356 | 133.077 |
| 1372 | 20.85 | 0.4245 | 0.3889 | 3.574 | 9.629 | 23.109 | 1372 | 133.781 |
| 1392 | 21.33 | 0.4367 | 0.4011 | 3.686 | 9.707 | 23.297 | 1392 | 134.734 |
| 1408 | 21.84 | 0.4153 | 0.3797 | 3.489 | 8.975 | 21.539 | 1408 | 135.543 |
| 1428 | 22.29 | 0.4284 | 0.3928 | 3.610 | 9.097 | 21.833 | 1428 | 136.889 |
| 1444 | 21.86 | 0.415 | 0.3794 | 3.487 | 8.959 | 21.503 | 1444 | 139.123 |
| 1464 | 22.01 | 0.4155 | 0.3799 | 3.491 | 8.910 | 21.384 | 1464 | 141.912 |
| 1480 | 21.56 | 0.4207 | 0.3851 | 3.539 | 9.221 | 22.129 | 1480 | 144.135 |
| 1504 | 23.19 | 0.4551 | 0.4195 | 3.855 | 9.338 | 22.412 | 1504 | 147.933 |
| 1519 | 21.48 | 0.4256 | 0.39 | 3.584 | 9.373 | 22.495 | 1519 | 150.916 |
| 1536 | 24.61 | 0.5754 | 0.5398 | 4.961 | 11.323 | 27.175 | 1536 | 155.000 |
| 1556 | 21.99 | 0.4491 | 0.4135 | 3.800 | 9.707 | 23.297 | 1556 | 159.333 |
| 1576 | 21.82 | 0.4951 | 0.4595 | 4.223 | 10.871 | 26.090 | 1576 | 163.500 |


| Sample \# | Weight (mg | au | corrected au | Si mM | \%Si (opal) | \% opal\|Depth (cm) | Age (ka) |  |
| :--- | ---: | ---: | ---: | ---: | ---: | ---: | ---: | ---: |
| $\mathbf{1 5 9 3}$ | 21.01 | 0.4244 | 0.3888 | 3.573 | 9.553 | 22.927 | 1593 | 167.750 |
| $\mathbf{1 6 0 8}$ | 23.23 | 0.3377 | 0.3021 | 2.776 | 6.713 | 16.112 | 1608 | 171.500 |
| $\mathbf{1 6 2 3}$ | 20.96 | 0.3488 | 0.3132 | 2.878 | 7.714 | 18.513 | 1623 | 175.250 |
| $\mathbf{1 6 4 2}$ | 23.43 | 0.3992 | 0.3636 | 3.341 | 8.011 | 19.226 | 1642 | 180.000 |
| $\mathbf{1 6 5 4}$ | 22.25 | 0.4034 | 0.3678 | 3.380 | 8.533 | 20.480 | 1654 | 183.000 |
| $\mathbf{1 6 7 2}$ | 21.25 | 0.344 | 0.3084 | 2.834 | 7.492 | 17.981 | 1672 | 187.500 |
| $\mathbf{1 6 8 8}$ | 21.92 | 0.4201 | 0.3845 | 3.534 | 9.055 | 21.732 | 1688 | 191.500 |
| $\mathbf{1 7 0 8}$ | 20.52 | 0.319 | 0.2834 | 2.604 | 7.129 | 17.111 | 1708 | 196.500 |
| $\mathbf{1 7 2 2}$ | 21.61 | 0.2749 | 0.2393 | 2.199 | 5.716 | 13.719 | 1722 | 200.000 |
| $\mathbf{1 7 4 2}$ | 21.28 | 0.2623 | 0.2267 | 2.083 | 5.499 | 13.199 | 1742 | 205.000 |
| $\mathbf{1 7 5 8}$ | 21.53 | 0.2957 | 0.2601 | 2.390 | 6.236 | 14.967 | 1758 | 209.000 |
| $\mathbf{1 7 7 8}$ | 23.4 | 0.2946 | 0.259 | 2.380 | 5.714 | 13.713 | 1778 | 214.000 |
| $\mathbf{1 7 9 2}$ | 19.95 | 0.2931 | 0.2575 | 2.366 | 6.663 | 15.991 | 1792 | 217.500 |
| $\mathbf{1 8 0 6}$ | 23.36 | 0.3316 | 0.296 | 2.720 | 6.541 | 15.699 | 1806 | 221.000 |
| $\mathbf{1 8 1 9}$ | 22.22 | 0.3034 | 0.2678 | 2.461 | 6.222 | 14.932 | 1819 | 224.250 |
| $\mathbf{1 8 3 6}$ | 19.49 | 0.2817 | 0.2461 | 2.262 | 6.518 | 15.644 | 1836 | 228.500 |
| $\mathbf{1 8 5 5}$ | 20.8 | 0.3921 | 0.3565 | 3.276 | 8.848 | 21.235 | 1855 | 233.250 |
| $\mathbf{1 8 7 3}$ | 19.98 | 0.4289 | 0.3933 | 3.614 | 10.162 | 24.388 | 1873 | 237.750 |
| $\mathbf{1 8 9 3}$ | 22.99 | 0.4268 | 0.3912 | 3.595 | 8.784 | 21.082 | 1893 | 242.750 |
| $\mathbf{1 9 1 3}$ | 22.12 | 0.3526 | 0.317 | 2.913 | 7.398 | 17.755 | 1913 | 247.750 |
| JV Bulk | 20.33 | 0.2694 | 0.2338 | 2.149 | 5.937 | 14.248 |  |  |
| $\mathbf{6 7 ~ I ~ 9 0 - 9 1}$ | 22.14 | 0.1522 | 0.1166 | 1.072 | 2.719 | 6.525 | 90 | 7.092 |
| $\mathbf{6 7 ~ V 9 0 - 9 1}$ | 19.46 | 0.1572 | 0.1216 | 1.118 | 3.226 | 7.742 | 690 | 58.919 |
| TR 580 | 22.66 | 0.6665 | 0.6309 | 5.798 | 14.373 | 34.494 | 710 | 67.213 |
| MD98 278 | 21.52 | 0.2296 | 0.194 | 1.783 | 4.654 | 11.169 | 278 | 2.881 |
| MD98 1045 | 19.98 | 0.1639 | 0.1283 | 1.179 | 3.315 | 7.956 | 1045 | 14.499 |
| $\mathbf{Y}$ |  |  |  |  |  |  |  |  |

