# GEOCHEMICAL DISCRIMINATION OF THE PERALUMINOUS DEVONIAN-CARBONIFEROUS GRANITOIDS OF NOVA SCOTIA AND MOROCCO 

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#### Abstract

The peraluminous granites of Nova Scotia and Morocco are compared to assess the viability of the Moroccan Model that postulates that the Meguma Zone (a suspect terrane of the northern Appalachians) was derived from north-western Africa. In addition, the relationship between the so-called northern and southern plutons of Nova Scotia is examined. A geochemical database of over 1300 analyses of peraluminous granites from Nova Scotia, Morocco, Iberia (another potential source area) and, Australia (a presumably unrelated belt) was carefully compiled for this study. The geochemical data were examined using both a traditional approach and discriminant function analysis (a multivariate statistical technique).

Results obtained by traditional methods of comparison indicate that the northern and southern plutons of Nova Scotia, although different, could not be clearly separated into two groups. In addition, the granites of Nova Scotia and Morocco appear indistinguishable.

Various statistical models demonstrate the applicability of discriminant analysis to the geochemical compositional granitic data by succesfully analyzing bimodal and skewed populations, uneven sample groups, and compositional data.

Several statistical models compare the geochemical populations of northern and southern Nova Scotia; Nova Scotia and Morocco; Nova Scotia, Morocco, and Iberia and; the Atlantic and Australian granites. Results indicate that the northern and southern plutons of Nova Scotia appear geochemically distinct; the Nova Scotia and Morocco populations show some similarities; Nova Scotia, Morocco (Zaer pluton of the Central Massif) and Iberia are equally similar; and, the Atlantic and Australian granites are clearly distinct.

Comparison of discriminant function coefficients obtained on the local (north-south Nova Scotia), regional (Nova Scotia-Morocco and Nova Scotia-Morocco-Iberia) and orogen-scale (Atlantic and Australian granites) show that at each scale of reference a characteristic suite of elements can be defined as good discriminators.

Results obtained thoughout this study indicate that discriminant function analysis is more useful and revealing than traditional methods of comparison. Although no clear evidence was found to confirm the Moroccan Model, results suggest that Morocco cannot be excluded as a potential source area for the Meguma Terrane.


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## CHAPTER 1

## INTRODUCTION AND GEOLOGICAL SETTING

### 1.1 Introduction

Granites form an integral part of orogenic belts and may be components of suspect terranes. A suspect terrane is an area of unknown origin that has a geological history distinct from, and unrelated to, neighbouring terranes. As such, the study of a granite population, when present in a suspect terrane, can be useful in determining probable provenance. In order to make correlations between igneous rocks of a suspect terrane and those of a possible source region using geochemical populations, it is necessary to distinguish between locally induced chemical variations, and those chemical characteristics that are indigenous to the region of study. Given the complexity of granitic systems, and the difficulties encountered in isolating the effects of the many different processes that affect granitic rocks, the simple traditional approach which examines the interrelationships among only a few components may not be satisfactory. Instead, a multicomponent approach is suggested as a more appropriate way of solving this complex problem.

In this study, the granitic rocks of the Meguma Zone, an apparently suspect terrane of the northern Appalachians, are examined and compared with rocks of similar composition and age from both Morocco (NW Africa), and Iberia. Australian rocks from a presumably unrelated orogenic belt, but of similar composition and age are considered to provide an independent datum for comparison.

The results of this investigation should provide further insight into the origin of the Meguma Terrane. In addition, some understanding may be gained about the variation of granites from a local scale to the orogenic scale.

The Meguma Zone, located in the southern part of Nova Scotia, consists mainly of Cambro-Ordovician clastic metasediments intruded by Devono-Carboniferous peraluminous granitoid rocks, and overlain by a late Devonian to Carboniferous sequence of red beds, clastic sediments, carbonates and evaporites.

The stratigraphy of the Meguma Terrane is summarized in Figure 1.1, where the lithologies are grouped with reference to the Acadian orogenic event. Pre-orogenic lithologies include the Goldenville, Halifax, White Rock, Kentville and Torbrook Formations, whereas the Horton Group, Windsor Group and Supra-Windsor Carboniferous Units postdate the Acadian Orogeny.

The Cambro-Ordovician Meguma Group is the oldest exposed sequence in the region and includes the Goldenville (sandy high density turbidites) and Halifax (silty low density turbidites) Formations. This succession, approximately 18 Km thick, has been interpreted as a proximal and distal turbidite sequence (Schenk 1970, 1973; Schenk and Lane 1982). Scarce fossil assemblages indicate a Tremadocian age for the upper part of the Meguma Group (Schenk, 1983). Radiometric ages on detrital muscovite grains from the Goldenville Formation, yield Tremadocian to Arenigian ages (Poole, 1971). Based on lithological and paleocurrent studies, Schenk (1970) suggested that the sediments were derived from a landmass located to the present southeast. Schenk (1983) further suggested a low-lying, deeply eroded source area of predominantly granodioritic composition and probably Precambrian in age. Clarke and Halliday (1985) using $\mathrm{Sm} / \mathrm{Nd}$ isotopes were able to determine a mean crustal residence time of $\mathrm{T}_{\mathrm{DM}}=1773 \pm 95 \mathrm{Ma}$ for the Meguma Group sediments. Krogh and Keppie (1986) using the U-Pb isotopic technique on detrital zircons from the greenschist facies of the Goldenville Formation were able to identify two different source ages, one of 600 Ma and the other of $2000-2700 \mathrm{Ma}$. These provinces correspond to the Hoggar Area of West Africa.


Figure 1.1. General stratigraphy of the Meguma Terrane. See text for source references.

The White Rock Formation ( 270 m ), conformably overlies the Meguma Group and is mainly composed of quartzites, siltstones and minor felsic-mafic volcanics. It is considered to be a paralic sequence with volcanics (Schenk, 1983). Fossils in the White Rock Formation suggest an age range from Late Ordovician (post Caradocian) to Devonian (Lane, 1975, 1979).

The Kentville Formation, which conformably overlies the White Rock Formation, consists of black graptolitic slates, diamictites and felsites (Schenk, 1983). It has been interpreted as a neritic sequence with a thickness of approximately 1700 m . A late Silurian (Ludlovian) age is assigned to the Kentville Formation (Taylor, 1969).

The Torbrook Formation, a 1500 m thick sequence of quartzites, siltstones and shales, are the youngest pre-orogenic rocks in the Meguma Zone. This formation is an Early Devonian (Gedinnian to Emsian) (Jensen, 1976) inner shelf to estuarine sequence.

The late Devonian Acadian orogeny resulted in the folding, metamorphism and granitization of the pre-orogenic lithologies. Elias (1987) placed the minimum age of the Acadian orogeny at approximately 390 Ma . Structural studies carried out by various authors (Fyson, 1966; Poole, 1967; Taylor, 1967; Taylor, 1969; Keppie, 1977; Keppie, 1982; Keppie, 1984) have revealed a complex deformational history for these rocks. The main phase of deformation resulted in the formation of open to isoclinal folds, generally trending SW-NE.

Several episodes of metamorphism have been identified by Muecke (1984). Two high-grade metamorphic zones have been identified by Keppie and Muecke (1979) and Muecke (1984). The first high, located in the southwestern corner of the terrane, forms a concentric pattern interpreted by Muecke (1984) as a broad thermal doming event formed prior to granite emplacement. The second high grade zone forms a linear belt subparallel to the Glooscap shear zone and seems to be controlled by granite generation (Muecke, 1984). Based on observed
metamorphic assemblages a low to medium pressure and high temperature regime is inferred, characteristic of Pyreneean-type metamorphic belts (Clarke et al., 1980; Muecke, 1984).

Peraluminous granites intrude the pre-orogenic lithologies. The South Mountain batholith and the northern satellite plutons were emplaced during the middle to late Devonian ( $372-361 \mathrm{Ma}$ ) (Clarke and Halliday, 1985; Reynolds et al., 1981). ${ }^{40} \mathrm{Ar}-39 \mathrm{Ar}$ dating of the Meguma granites by Elias (1987) revealed a complex thermal history for the socalled southern satellite plutons. In a study of these results Reynolds et al., 1987 suggested that even though the argon dates on the southern plutons span a wider range of time they were generally emplaced at the same time as the northern plutons. The younger ages obtained for the southern plutons are attributed to a resetting event resulting from a Hercynian ( $=$ Variscan) tectono-thermal event around $300-320 \mathrm{Ma}$. A second reheating event of lesser impact affecting the southern plutons occurred around $220-230 \mathrm{Ma}$, and is believed to be related to the initial rifting of the Atlantic (Reynolds et al., 1987). The field relations of the Meguma granites are discussed in Chapter 2.

Upper Devonian to Permian red continental sediments were deposited unconformably upon the orogen. The Windsor Group limestones, evaporites and clastics were deposited in local marine basins which developed on the Meguma platform in the late Mississipian time (Poole, 1967; Schenk, 1978). Deformation which occurred late in the Carboniferous, is known as the Maritime Disturbance, and as also been referred to as the Hercynian deformation (Poole, 1967).

Insight into the place of origin of the Meguma suspect terrane has been sought by many authors (Schenk, 1970, 1971, 1983; Hollard and Schaer 1973; Clarke and Halliday, 1985; Krogh and Keppie, 1986). As a result of these investigations which have centered around the Meguma Group metasediments, various potential source areas have been suggested, including: western Europe, northwestrn Africa, Colombia, and western south America. This study provides further insight into the
origin of the Meguma Zone by comparing its peraluminous granites with rocks of similar composition from two of its postulated source area, Morocco (NW Africa) and Iberia. Before beginning the study of the granitic rocks the general stratigraphy of Morocco will be given below.

### 1.3 General Moroccan Geology

In contrast to the Meguma Zone, Moroccan geology is complex and varied, and spans a wider range of time (oldest continental crust 3 Ga , Michard, 1976 p. 315). A number of orogenic events have affected Moroccan rocks throughout its history (Table 1.1). Generally the geology of Morocco is discussed in reference to structural domains defined by Michard (1976).

Several structural domains may be defined depending on the scale of reference (Table 1.2). Peraluminous granites in Morocco occur within Paleozoic massifs of the Meseta and Atlas domains (Figure 1.2). The general stratigraphy of these massifs is considered below.

### 1.3.1 Stratigraphy of Moroccan Paleozoic Massifs

The stratigraphy of the Jebilet, Rehamna and Central massifs of the Meseta domain as well as the Atlas domain is presented in this section. Only the general characteristics are given as detailed consideration of the Moroccan stratigraphy is beyond the scope of this thesis. The stratigraphy of each area is summarized and presented separately because many of the stratigraphic relationships within and between the massifs are unclear, and no succinct stratigraphic column exists for the Moroccan rocks. This general approach will also allow the reader to appreciate how variable the stratigraphy can be within and between the massifs.

In many instances the ages of the host rocks have been assigned on the bases of lithology rather than actual dating. For example, a host flysch will be given a Visean age simply because Visean flysch appear

Table 1.1. Moroccan orogenic cycles as reported by Michard, 1976 (simplified).

| OROGENIC <br> CYCLE | AGE |  |
| :--- | :--- | :--- |
| Alpine | Triassic to Pliocene | $(180-1 \mathrm{Ma})$ |
| Caledono-Hercynian | Infracambrian to Permian | $(530 ?-250 \mathrm{Ma})$ |
| Pan-African | Eocambrian (?) | $(1100 ?-550 ? \mathrm{Ma})$ |
| Eburnean | Ancient Precambrian | $(1800-2000 \mathrm{Ma}) ?$ |
| Archean | Relic in ancient Precambrian | $(3000 ? \mathrm{Ma})$ |

Table 1.2. Main stuctural domains of Morocco as defined by Michard, 1976. Based on the Alpine Orogeny.

| Rif | Part of the broader Alpine orogenic belt <br> extending north with the Betique mountain <br> chain and east with the Tellian and Kabyle <br> mountain chains. |
| :--- | :--- | :--- |

## Morocco



Figure 1.2. Geological map of Morocco.
to be most common. Therefore the following Figures should be considered with caution.

The Jebilet Massif (Figure 1.2) of the southern Meseta is located approximately 7 km north of the city of Marrakech. Three main structural units have been defined within the Jebilet Massif; they are 1) the western Jebilet, 2) the Bou-Gader and Skhirat units and 3) central and eastern Jebilet (Michard, 1976; Huvelin, 1977; Hollard et al., 1977; Pique et al. 1983). They are characterized by the nature of the terrain and style of Hercynian deformation. The stratigraphy of each unit is given in Figure 1.3.

The Rehamna Massif (Figure 1.2), located approximately half way between Casablanca and Marrakech, offers many lithological similarities with the Jebilet Massif. A summary of the stratigraphy of four structural domains as defined by Michard (1982) is presented in Figure 1.4. They are the Mechra-Ben-Abbou (northern Rehamna), the western, the eastern and the central Rehamna domains (Michard, 1976, 1982; Hollard et al., 1982; Destombes et al., 1982).

The stratigraphy of the Central massif of the northern Moroccan Meseta is given in Figure 1.5. A summary stratigraphic column is also given for Paleozoic rocks of the Atlas domain from the Tichka Massif (Figure 1.5).

The Paleozoic history of Morocco can be summarized as follows (Michard, 1976, 1978, 1982; Huvelin, 1977; Destombes J. et al., 1985; Schenk, in prep.). An Infracambrian transgression over the western African platform resulted in the deposition of sedimentary carbonate rocks (Anti-Atlas). Thick siliciclastic sequences were deposited (from the S.E.) throughout Morocco in the Middle Cambrian and were supplied continuously thru the Lower Cambrian from the NW Craton. Also prominent throughout Morocco during the Middle Cambrian are thick complexes of trachyandesites, basalts, andesites, breccias and tuffites. A generalized regression occurred during the middle to late

## Jebilet Massif



Figure 1.3. Schematic representation of the Cambrian to Carboniferous stratigraphy of the Jebilet Massif, Morocco. See text for source references.

## Rehamna Massif


Fm. $=$ formation carb. = carbonate
sd. $=$ sandstone congl. $=$ conglomerate Hiatus
qtz. $=$ quartzite gwck $=$ graywacke ? ? ? ?? Unknown extent
$\ldots . . .$. Relationship with adjacent lithology unclear

Figure 1.4. Schematic representation of the Cambrian to Carboniferous stratigraphy of the Rehamna Massif, Morocco. See text for source references.

## Central Massif \& High-Atlas

Central High-Atlas (Tichka)

|  | Sandstone and conglomerate felsic and mafic magmatism <br> Flysch, carbonate sandy flysch ??? ? ? ? ? ? ? ? ? ? ? ? ? ? ? | Red conglomerate, sandstone |
| :---: | :---: | :---: |
|  |  | Flysch, intraformational breccia |
|  |  |  |
| a0000 | Shale and sandstone |  |
|  | Sandstone, flysch shale, carbonate minor conglomerate | Carbonate and red conglomerate |
| Silurian | Shale, carbonate nodules | Shale, limestone |
| $\begin{aligned} & \text { 感 } \\ & \text { 荡 } \end{aligned}$ | Quartzite, conglomerate bioturbate sandstone argillite. Bou Regreg flysch. <br> ????????????????? | Sandy shale, carbonate lenses, black shale, turbidite, micaceous sandy clay <br>  |
|  |  |  |
|  | Graywacke, sandstone quartzite, carbonate ????????????????? | Sandstone, volcanicsedimentary complexes (andesitic) minor carbonate. ????????????????? |

Figure 1.5. Schematic representation of the Cambrian to Carboniferous stratigraphy of the Central Massif and High Atlas, Morocco. See text for source references.

Cambrian, earlier in the Anti-Atlas region and later in central Morocco, terminating with the deposition of sandstones and minor volcanic facies throughout Morocco (with pyroclastics in the western high Atlas, Jebilet and Rehamna).

No upper Cambrian or early Ordovician (Tremadoc) rocks have been identified in northern and central Morocco (Tremadoc deposits occur within the Anti-Atlas), and this time is considered an epeirogenic period during which pre-existing lithologies may have been weakly folded.

A transgression occurred early in the Ordovician (Arenig). Little is known in detail about the Ordovician north of the Anti-Atlas, however, it generally consists of predominantly argillaceous sediments with minor amounts of sandstones. A continental influence appears several times during the Ordovician and is suspected to represent a Caledonian event in western Morocco during this period. The Ordovician ended with a glaciation from the SE resulting in the deposition of glaciomarine sediments during the Late Ashgill.

A transgression occurred early in the Silurian which peaked during the middle to upper Llandoverian. Platy sandstones, graptolitic shales and siltstones were deposited in the Anti-Atlas region and dark graptolitic shales in northern Morocco (Hollard, 1970). Thin basaltic flows are intercalated in western Morocco with the lower Silurian series (Cornee et al., 1985).

Lower to Middle Devonian epicontinental, sometimes intertidal, and even subaerial deposits, occur throughout Morocco. Frasnian rocks are absent in most places. Local deformation in the Late Devonian indicates the beginning of the Hercynian deformation.

Upper Visean rocks discordantly overlie the previous discussed lithologies. A Late Visean transgression is evident throughout the
area. Felsic and mafic magmatism are also associated with these deposits.

Syn-, but mostly, post- tectonic peraluminous granites intrude the "pre-orogenic" lithologies discussed above, and are found within the Meseta and Atlas Paleozoic massifs. These granites are the main thrust of this thesis and are considered in detail in Chapter 2.

### 1.4 Comparing the Nova Scotian and Moroccan Stratigraphy

Based on the lithologies of the different massifs and work done by Schenk (in prep.), a summary table of the stratigraphy of Morocco is compared with the Meguma Terrane in order to demonstrate the lithologic similarities between these areas, and provides justification for comparing Morocco with the Meguma Terrane (Figure 1.6).

### 1.5 Purpose of This Study

The objectives of this thesis are:

1) to assess the significance of the chemical variations observed within plutons of the Meguma Terrane;
2) to investigate the geochemical characteristics of the Moroccan granites sampled during this study, as well as those studied by previous workers, to characterize the geochemical nature of the Moroccan granites;
3) to evaluate the applicability of multicomponent analysis (discriminant function analysis) to granitic rocks and, in particular, to assess the reliability of the application of such analysis when performed on compositional data (geochemical data);
4) to examine the relationship between the Nova Scotian, Moroccan and Iberian granites using discriminant function analysis, to assess the

Nova Scotia-Morocco
Nova Scotia
Morocco

[IIIII] Hiatus ????? Unknown extent
Figure 1.6. Schematic comparison of the Cambrian to Carboniferous stratigraphy of Nova Scotia and Morocco. See text for source references.
degree of similarity between these Hercynian rocks. These rocks are in turn compared with Australian rocks presumably of an unrelated orogenic belt but of similar composition and age, providing an independent datum by which to assess the validity of observed variations;
5) to compare and discuss results obtained using both the traditional geochemical and the multicomponent approaches, in order to evaluate the degree of information obtained by each method;
6) to discuss and evaluate the viability of the "Moroccan model" in light of the results obtained during this study.

## CHAPTER 2

## FIELD RELATIONS, PETROGRAPHY AND GEOCHRONOLOGY

### 2.1 Introduction

In this Chapter the general field relations, petrography and, geochronology of peraluminous granites of Nova Scotia and Morocco are discussed in order to familiarize the reader with the granitic rocks from these two areas. A broad regional approach is adopted compatible with the approach to assess variation only on a regional scale.

### 2.2 Nova Scotian Granites

### 2.2.1 Introduction

Differences in composition, age and, emplacement between the northern and southern peraluminous plutons of the Meguma Zone of Nova Scotia have been recognized by workers in the past. A selected division based on some of their observations was drawn between both areas and is shown in Figure 2.1. The general characteristics of each area is given below. Emphasis has been placed on their differences.

### 2.2.2 The Northern Plutons

The northern plutons include the South Mountain batholith (McKenzie, 1974; McKenzie and Clarke, 1975; Clarke and Muecke, 1987; Richardson, in prep.), the Musquodoboit pluton (MacDonald, 1981; MacDonald and Clarke, 1985), the Liscomb pluton (Cameron, 1985), the Sherbrooke pluton (Smith et al., 1987, Alizay, 1981), the Bull Ridge pluton (Bernadette, 198?), the Ellison Lake pluton (Allan, 198?), the Kinsac pluton (Coolen, 1974), the Mulgrave pluton (Dwyer, 1975), the Halfway Cove and Queensport plutons (Ham, in prep.), the Sangster Lake and Larry's River plutons (O'Reilly, in prep.) and the Canso pluton (Hill, 1986). Generally all of these plutons range in composition from

## Nova Scotia



## Northern Grpanites

1 White Haven
2 Queensport/ Halfway cove
3 Larry's River
4 Sangster Lake
5 Forest Hill
6 Bull Ridge
7 Sherbrooke
8 Liscomb
9 Mulgrave
10 Musquodoboit
11 Kinsac
12 South Mountain
13 Ellison Lake

## Southern (hranites

14 Brenton
15 Wedgeport
16 Barrington Passage
17 Bald Mountain
18 Beach Hill
19 Shelburne
20 Lyons Bay/ Seal Island/
Western Granite
21 Port Mouton
22 Eastern Head/
Moose Point

Figure 2.1. Geological map of the Meguma Terrane.
granodiorite, to monzogranite and may include some late stage leucomonzogranites. The only exception, the Canso pluton, is reported to contain some tonalites (Hill, 1986).

### 2.2.3 The Southern Plutons

The southern plutons include the Barrington Passage pluton (Smith, 1979; Rogers, 1985), the Lyons Bay, Seal Island and Western Granite plutons (Rogers, 1985), the Wedgeport pluton (Reynolds et al. 1981; Keppie et al., 1983; Wolfson, 1983; Chatterjee et al., 1985), the Shelburne pluton (Rogers, 1985), the Brenton pluton (O'Reilly, 1976; Clarke et al., 1979), the Bald Mountain pluton (Rogers, 1985), the Moose Point pluton (Weagle, 1983) and the Port Mouton pluton (Alburquerque, 1977; Douma, 1988).

In contrast to the northern plutons, compositions within the southern plutons are more varied, showing a composition range from diorite, norite, hornblende and biotite tonalites, trondhjemites, granodiorites, monzogranites and, leucomonzogranites. In addition, Douma (1988) has reported lamprophyres in the Port Mounton pluton.

### 2.2.4 Geochronology of the Nova Scotian Granites

Average ${ }^{40} \mathrm{Ar} /{ }^{39} \mathrm{Ar}$ radiometric ages within the northern plutons range from $366.7 \pm 11 \mathrm{Ma}$ for the SMB , to $365.5 \pm 4.3 \mathrm{Ma}$ for the remaining northern plutons (Reynolds et al., 1981). These ages contrasts markedly with $337 \pm 16 \mathrm{Ma}$ (Elias, 1987) reported for the southern plutons, with some older ages obtained within the Bald Mountain and Barrington Passage plutons ( $370-386 \mathrm{Ma}$ ). As discussed in Chapter 1 these younger ages are believed to be the result of a tectono-thermal event within the area around $300-320 \mathrm{Ma}$ (Reynolds et al., 1987). Such younger ages are not exclusive to the southern plutons, mineralization within the northern plutons, particularly the SMB, have been reported at around 330 Ma (e.g. East Kemptville, SMB). In addition, distinctly younger ages of about 280 Ma are reported
within other metasomatized and altered granites of the northern plutons (Reynolds et al., 1981; Zentilli and Reynolds, 1985), as well as the southern plutons (Elias, 1987).

In summary, the intrusions of granites throughout the Meguma Zone occurred between $386-360 \mathrm{Ma}$ (Elias, 1987). Evidence suggests that the northern plutons were generally unaffected by the Hercynian event except for some localized areas; however the southern plutons were pervasively affected by this tectono-thermal event. This suggests a difference in the Hercynian tectono-thermal history of both areas.

### 2.2.5 Emplacement of the Nova Scotian Granite

The regional metamorphic grade of the northern country rocks is predominantly of the greenshist facies. In contrast the metamorphic grade is observed to increase southward, reaching an amphibolite facies near the Shelburne pluton (Muecke, 1984). This suggests that the southern plutons may have been emplaced at greater depths than the northern plutons. Some southern plutons (Brenton, Barrington Passage and Shelburne) show pre- and syn- tectonic emplacement as opposed to the northern plutons which are reported to be post-tectonic.

### 2.3 Moroccan Granites

### 2.3.1 The Jebilet Massif

### 2.3.1.1 Introduction

Peraluminous granites occur within the central part of the Jebilet massif of Morocco (Figure 1.2) and intrude rocks folded during upper Visean (possibly Namurian?) time. They are the Oulad Ouaslam batholith, Tabouchennt-Bamega pluton, Bramram greisen and, Sidi Bou Othmane pegmatites.

### 2.3.1.2 Oulad Ouaslam Batholith

The Oulad Ouaslam batholith (OOB), the most extensive of all of the Jebilet granites, crops out over an area of approximately $300 \mathrm{Km}^{2}$ and is located north of Marrakech and east of the main highway leading to Casablanca. Forty-eight samples collected from the Oulad Ouaslam batholith were slabbed, stained with sodium cobaltinitrite, and pointcounted using a binocular microscope (details are presented in Appendix A). Sample locations are shown in Figure 2.2. Individual results for each specimen are presented in Table 2.2 and classified using the terminology of Streckeisen (1976) (Figure 2.3).

It should be pointed out that the sample population was intentionally biased by selecting those specimens which were not associated with mineralisation or with any form of extensive alteration. These "secondary" processess were excluded to ensure the comparability of the populations. Given time constraints and the considerable area covered, sampling was not carried out in any detail. Instead, the objective of the sampling was to obtain a good regional coverage of each pluton.

Compositions within the batholith range from granodiorite to monzogranite, and plot as one coherent population on a Streckeisen diagram (Figure 2.3), indicating that individual units can not be defined within the batholith based on modal mineralogical proportions. Mrini (1985) reported a classification scheme determined by Rose (in prep) which he suggested could be applied to all of the granites within the Jebilet Massif. This scheme essentially divides the Jebilet granites into two main rock types; the granodiorites (or calc-alkaline biotite granites) and leucogranites (or two-mica granites).

Granodiorite (referred to as monzogranites and granodiorites using the terminology of Streckeisen), is the dominant facies within the Jebilet granites. According to Mrini (1985) the granodiorites can be

## Oulad Ouaslam Batholith



Figure 2.2. Location of samples collected from the Oulad Ouaslam batholith, Jebilet Massif. Note: The prefixes JBL and JUB are used to identify samples from the West and the East respectively.

Table 2.1. Modal analysis of samples from the Oulad Ouaslam batholith (Jebilet Massif).
Based on stained slabs counting, 1000-2000 points. See Figure 2.3 for plot of QAP diagram.

| SAMPLE | JBL1B | JBL2A | JBL3 | JBL4 | JBL6 | JBL8 | JBL9 | JBL10 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Quartz | 30.3 | 32.4 | 32.1 | 29.8 | 37.8 | 33.1 | 22.6 | 27.4 |
| Plagioclase | 33.2 | 37.6 | 36.3 | 38.6 | 38.7 | 39.5 | 41.4 | 41.4 |
| K-Feldspar | 34.0 | 16.0 | 20.4 | 18.4 | 10.0 | 16.1 | 21.4 | 14.6 |
| Biotite | 1.3 | 9.2 | 7.2 | 10.3 | 10.9 | 10.3 | 13.5 | 13.3 |
| Muscovite | 1.3 | 4.7 | 4.0 | 2.9 | 2.7 | 1.0 | 1.1 | 3.3 |
| Rock name | Mg | Gd | Mg | Gd | Gd | Gd | Gd | Gd |
| SAMPLE | JBL11 | JBL12 | JBL13 | JBL14 | JBL15 | JBL17 | JBL19 | JBL21 |
| Quartz | 34.8 | 27.9 | 32.0 | 33.5 | 29.1 | 28.2 | 29.9 | 32.0 |
| Plagioclase | 36.7 | 39.8 | 33.2 | 39.1 | 34.4 | 34.3 | 41.3 | 35.0 |
| K-Feldspar | 20.9 | 20.4 | 27.2 | 15.3 | 31.9 | 23.2 | 15.0 | 14.0 |
| Biotite | 4.7 | 10.3 | 5.9 | 11.2 | 3.7 | 10.1 | 11.5 | 15.4 |
| Muscovite | 2.9 | 1.6 | 1.8 | 0.8 | 0.9 | 4.2 | 2.4 | 3.7 |
| Rock name | Mg | Gd | Mg | Gd | Mg | Mg | Gd | Gd |
| SAMPLE | JBL22 | JBL23 | JBL24 | JBL25 | JBL26 | JBL27 | JBL28 | JUB1 |
| Quartz | 27.8 | 28.5 | 30.2 | 28.6 | 30.4 | 28.6 | 28.0 | 24.1 |
| Plagioclase | 41.3 | 42.5 | 40.4 | 37.3 | 34.2 | 38.4 | 48.9 | 39.3 |
| K-Feldspar | 18.0 | 15.7 | 19.0 | 22.5 | 26.6 | 17.0 | 7.4 | 27.3 |
| Biotite | 11.8 | 11.0 | 8.0 | 11.2 | 6.4 | 12.7 | 13.9 | 8.2 |
| Muscovite | 1.1 | 2.3 | 2.5 | 0.3 | 2.4 | 3.4 | 1.8 | 1.1 |
| Rock name | Gd | Gd | Gd | Mg | Mg | Gd | Gd | Mg |
| SAMPLE | JUB2 | JUB3 | JUB4 | JUB5 | JUB6 | JUB7 | JUB8 | JUB9 |
| Quartz | 38.2 | 29.8 | 29.2 | 25.1 | 32.9 | 25.1 | 27.9 | 33.5 |
| Plagioclase | 29.6 | 38.3 | 35.9 | 40.3 | 44.1 | 42.6 | 39.0 | 36.4 |
| K-Feldspar | 18.4 | 20.1 | 19.9 | 18.4 | 17.8 | 26.4 | 22.6 | 19.5 |
| Biotite | 12.1 | 10.6 | 13.6 | 13.7 | 4.9 | 5.8 | 9.0 | 8.8 |
| Muscovite | 1.7 | 1.3 | 1.3 | 2.4 | 0.4 | 0.1 | 1.5 | 1.7 |
| Rock name | Mg | Gd | Mg | Gd | Gd | Mg | Mg | Gd |

$\mathrm{Mg}=$ monzogranite $\quad \mathrm{Gd}=$ granodiorite

Table 2.1 (cont.). Modal analysis of samples from the Oulad Ouaslam batholith (Jebilet Massif). Based on stained slabs counting, 1000-2000 points. See Figure 2.3 for plot of QAP diagram.

| SAMPLE | JUB10 | JUB11 | JUB12 | JU13 | JUB14 | JUB15 | JUB16 | JUB17 |
| :--- | ---: | ---: | ---: | ---: | ---: | ---: | ---: | ---: |
| Quartz | 32.1 | 33.6 | 26.1 | 32.8 | 28.3 | 30.2 | 28.7 | 30.5 |
| Plagioclase | 36.9 | 40.5 | 45.5 | 40.2 | 33.2 | 43.7 | 37.2 | 41.5 |
| K-Feldspar | 19.3 | 15.6 | 21.7 | 14.8 | 27.9 | 20.0 | 20.5 | 19.3 |
| Biotite | 9.9 | 6.0 | 5.9 | 10.4 | 8.9 | 5.1 | 12.3 | 7.3 |
| Muscovite | 1.8 | 4.3 | 0.9 | 2.3 | 1.8 | 1.1 | 1.3 | 1.3 |
| Rock name | Gd | Gd | Gd | Gd | Mg | Gd | Mg | Gd |
| SAMPLE | JUB18 | JUB19 | JUB20 | JUB21 | JUB22 | JUB23 | JUB24 | JUB25 |
| Quartz | 31.2 | 33.5 | 36.1 | 32.8 | 31.7 | 26.3 | 33.7 | 36.7 |
| Plagioclase | 36.1 | 35.7 | 34.4 | 30.9 | 35.1 | 38.7 | 32.8 | 35.3 |
| K-Feldspar | 24.0 | 20.4 | 23.8 | 29.2 | 21.4 | 26.1 | 24.8 | 21.4 |
| Biotite | 7.5 | 9.4 | 4.2 | 5.5 | 10.0 | 7.8 | 6.7 | 5.3 |
| Muscovite | 1.2 | 1.0 | 1.6 | 1.6 | 1.8 | 1.1 | 2.0 | 1.3 |
| Rock name | Mg | Mg | Mg | Mg | Mg | Mg | Mg | Mg |

$$
\mathrm{Mg}=\text { monzogranite } \quad \mathrm{Gd}=\text { granodiorite }
$$

## Oulad Ouaslam Batholith


$\mathrm{Sg}=$ syenogranite $\mathrm{Mg}=$ monzogranite $\mathrm{Gd}=$ granodiorite

Figure 2.3. QAP (Quartz- Alkali Feldspar- Plagioclase) plot of modal analysis from the Oulad Ouaslam batholith (Jebilet Massif, Morocco). (A) western part, (B) eastern part of the OOB. See Table 2.1 for modal analysis and Figure 2.2 for sample locations. Classification after Streckeisen, 1976.
further divided into two groups depending on whether or not cordierite is present in the assemblage.

The monzogranites and granodiorites are variable in grain size and contain quartz, phenocrysts of K-feldspar, plagioclase, and abundant biotite ( $5-15 \%$ ). The latter may be found as individual crystal or as agglomerations. Rare muscovite, garnet, cordierite, sillimanite, and megacrysts of andalusite are found within the granite. Accessory minerals include apatite, zircon, corundum, and tourmaline.

Andalusite megacrysts appear to have not precipitated directly from the magma. This is evident in their higher concentration near the contact zone, the presence of chiastolite and, the development of reaction rims. Based on this evidence andalusite megacrysts were not included in whole rock analysis (see Chapter 3).

The OOB (and in general the Jebilet granites) contain most of the characteristic peraluminous minerals (Clarke, 1981), i.e biotite, muscovite, aluminosilicates, cordierite, garnet, tourmaline, spinel and, corundum. However, in most cases the origin of these minerals remains problematic, and will need to be resolved by future workers.

The leucogranites of the OOB were not sampled during this study. Although some leucogranites have been reported within the batholith (Huvelin, 1977) they are mainly found within the Bramram greisen (not discussed in this thesis because of their high degree of alteration).

Aplites with abundant microcline, albite, oligoclase, muscovite, rare biotite, and accessory brown tourmaline occur as dykes or sheets cutting the monzogranites and granodiorites. Aplitic dykes have also been observed crosscutting hornfels west of the OOB. An aplite dyke was sampled during this study (JBL1B) and was classified as a monzogranite on a Streckeisen QAP diagram (Figure 2.3). It should be pointed out that even though this sample showed evidence of
considerable alteration (sericitization, chloritization) it was
selected for chemical analyses because it was the only aplite available.

Pegmatitic dykes occur almost exclusively in the country rock and very rarely are they found intruding the granites (Huvelin, 1977). They were not sampled during this study.

### 2.3.1.3 Tabouchennt-Bamega Pluton

The Tabouchennt-Bamega pluton (TBP) is located north of Marrakech and west of the main highway leading to Casablanca. The pluton crops out over an area of approximately $50 \mathrm{~km}^{2}$. Unfortunately Quaternary deposits cover a large area of the pluton and have made sampling of the granite difficult. Sixteen samples were collected from the Tabouchennt-Bamega pluton (Figure 2.4). Point-counting and classification results determined using on a Streckeisen QAP diagram are given in Table 2.2 and Figure 2.5. Compositions within the Tabouchennt-Bamega pluton are essentially monzogranitic, except for specimen BRR10a, which is a granodiorite.

The TBP presents many similarities with the OOB and descriptions given of the OOB generally apply for the TBP. Unlike the OOB no megacryst of andalusite and few enclaves have been observed in the TBP. The TBP has a higher concentration of alkali feldspars indicating that the TBP may represent a more evolved member in the differentiation sequence than the OOB.

### 2.3.2 Rehamna Massif

### 2.3.2.1 Introduction

The Rehamna massif is located approximately 50 km north of the Jebilet massif (Figure 1.2). Granitoids of weakly peraluminous composition occur in the southern part of the massif and intrude rocks

## Tabouchennt-Bamega Pluton




- Sample Location
$\ldots$ Non Granitic Lithologies


Figure 2.4. Location of samples collected from the TabouchenntBamega pluton.

Table 2.2. Modal analysis of samples from the Tabouchennt-Bamega pluton (Jebilet Massif). Based on stained slabs counting, 1000-2000 points. See Figure 2.5 for plot of QAP diagram.

| SAMPLE | BRR7 | BRR8 | BRR10 | BRR11 | BRR12 | BRR13 | BRR14 | BRR15 |
| :--- | ---: | ---: | ---: | ---: | ---: | ---: | ---: | ---: |
| Quartz | 29.9 | 33.7 | 33.3 | 37.0 | 30.7 | 31.3 | 35.4 | 26.8 |
| Plagioclase | 34.6 | 28.2 | 37.0 | 25.8 | 32.4 | 26.9 | 25.0 | 27.0 |
| K-Feldspar | 28.8 | 31.7 | 19.4 | 31.0 | 26.5 | 32.6 | 33.5 | 39.1 |
| Biotite | 5.8 | 4.7 | 9.0 | 5.2 | 8.7 | 8.4 | 5.1 | 5.8 |
| Muscovite | 1.0 | 1.8 | 1.4 | 1.0 | 1.7 | 0.9 | 1.1 | 1.4 |
| Rock name | Mg | Mg | Gd | Mg | Mg | Mg | Mg | Mg |
| SAMPLE | BRR16 | BRR17 | BRR18 | BRR19 | BRR20 | BRR21 | BRR22A BRR22B |  |
| Quartz | 28.6 | 26.2 | 24.0 | 30.8 | 35.1 | 34.9 | 33.3 | 34.2 |
| Plagioclase | 27.8 | 28.3 | 28.7 | 26.9 | 33.2 | 32.2 | 36.3 | 36.4 |
| K-Feldspar | 32.6 | 35.1 | 42.6 | 35.0 | 21.4 | 26.3 | 20.6 | 20.3 |
| Biotite | 9.3 | 9.3 | 3.8 | 6.5 | 8.9 | 4.6 | 8.2 | 8.0 |
| Muscovite | 1.7 | 1.1 | 0.9 | 0.8 | 1.4 | 2.1 | 1.6 | 1.1 |
| Rock name | Mg | Mg | Mg | Mg | Mg | Mg | Mg | Mg |

$$
\mathrm{Mg}=\text { monzogranite } \quad \mathrm{Gd}=\text { granodiorite }
$$

## Tabouchennt-Bamega Pluton


$\mathrm{Sg}=$ syenogranite $\mathrm{Mg}=$ monzogranite $\mathrm{Gd}=$ granodiorite

Figure 2.5. QAP (Quartz- Alkali Feldspar- Plagioclase) plot of modal analysis from the Tabouchennt-Bamega pluton (Jebilet Massif, Morocco). See Table 2.2 for modal analysis and Figure 2.4 for sample locations. Classification after Streckeisen, 1976.
ranging from Cambrian? to upper Visean? in age. The country rock is reported to be of the highest metamorphic grade encountered in the Meseta (Michard, 1976, 1982). The Rehamna granites are also believed to correspond to the apex of a larger batholith occurring beneath the micashists.

Although many studies have been undertaken on the Rehamna massif, very little is known about the granitic rocks which comprise this massif. Granites of the Sebt de Brikiine batholith and the Ajar el Bark stock were sampled.

### 2.3.2.2 Sebt de Brikiine Batholith

Sixteen samples were collected from the Sebt de Brikiine batholith (also referred to as the Si-Mohamed-Jerari batholith) (Figure 2.6). Point counting and classification results according to Streckeisen (1976) are given in Table 2.3 and Figure 2.7. Compositions within the Sebt de Brikiine batholith are essentially monzogranitic, with one exception SDB7 which was classified as a syenogranite (Figure 2.6), all of which generally cluster as one coherent population. Mrini (1985) reported that the granite may be classified as a calc-alkaline or a differentiated alkaline suite of the monzonitic series of Lameyre (1982). This can not be confirmed given the limited number of samples on Figure 2.6. If her observation is correct this would suggest a difference in the crystallisation history of the Rehamna and Jebilet granites (which are clearly of the calc-alcaline series) (Figure 2.2 and 2.4).

The SDB is a fine- to coarse-grained assemblage of quartz, K-feldspar, plagioclase, sparse biotite (more abundant in the east) and muscovite, and rare tourmaline (Mrini, 1985). Microscopic sphene and magnetite have also been reported (Gigout, 1951). Idir E.H. (pers. comm.) suggested that these minerals appear to be related to a much later faulting event and would not have crystallized directly from the magma. Mrini (1985) reported that the granite does not contain any

## Sebt De Brikiine Batholith



Figure 2.6. Location of samples collected from the Sebt de Brikiine batholith.

Table 2.3. Modal analysis of samples from the Sebt de Brikiine batholith (Rehamna Massif). Based on stained slabs counting, 1000-2000 points. See Figure 2.7 for plot of QAP diagram.

| SAMPLE | SDB1 | SDB2A | SDB2B | SDB3 | SDB4 | SDB5 | SDB6A | SDB6B |
| :--- | ---: | ---: | ---: | ---: | ---: | ---: | ---: | ---: |
| Quartz | 27.6 | 34.0 | 30.0 | 33.9 | 21.6 | 30.1 | 24.0 | 30.9 |
| Plagioclase | 33.4 | 29.9 | 27.4 | 33.6 | 36.8 | 30.5 | 19.6 | 28.6 |
| K-Feldspar | 38.5 | 33.8 | 42.0 | 30.4 | 40.7 | 35.0 | 28.5 | 37.9 |
| Biotite | 0.4 | 2.3 | 0.6 | 2.0 | 1.0 | 4.1 | 27.9 | 2.7 |
| Muscovite | 0.2 | tr. | tr. | tr. | tr. | 0.3 | tr. | tr. |
| Rock name | Mg | Mg | Mg | Mg | Mg | Mg | Mg | Mg |
| SAMPLE | SDB7 | SDB8 | SDB9 | SDB10 | SDB11 | SDB12 | SDB13 | SDB14 |
| Quartz | 30.5 | 24.1 | 30.4 | 28.8 | 31.2 | 20.9 | 30.9 | 25.3 |
| Plarioclase | 23.2 | 30.5 | 31.9 | 31.0 | 30.4 | 43.1 | 30.6 | 30.8 |
| K-Feldspar | 44.7 | 44.4 | 37.2 | 38.6 | 35.7 | 32.1 | 36.9 | 43.0 |
| Biotite | 1.6 | 0.9 | 0.6 | 1.6 | 2.6 | 3.6 | 1.6 | 0.9 |
| Muscovite | tr. | tr. | tr. | tr. | tr. | 0.3 | tr. | tr |
| Rock name | Sg | Mg | Mg | Mg | Mg | Mg | Mg | Mg |

$$
\begin{gathered}
\mathrm{Mg}=\text { monzogranite } \quad \mathrm{Sg}=\text { syenogranite } \\
\operatorname{tr} .=\text { trace }
\end{gathered}
$$

## Sebt De Brikiine Batholith


$\mathrm{Sg}=$ syenogranite $\mathrm{Mg}=$ monzogranite $\mathrm{Gd}=$ granodiorite

Figure 2.7. QAP (Quartz- Alkali Feldspar- Plagioclase) plot of modal analysis from the Sebt de Brikiine batholith (Rehamna Massif, Morocco). See Table 2.3 for modal analysis and Figure 2.6 for sample locations. Classification after Streckeisen, 1976.
enclaves from the country rock; however, Gigout (1951) noted the presence of enclaves near the contact zone in the Douar Souala vicinity. No xenolithic enclaves were obseryed by this author, however clots of biotite were found which may represent assimilated xenolithic material, this is still unclear. Perhaps the most striking features of the SDB are its pink color, which contrasts markedly with the usual grey-white color of other Moroccan granites, and the absence of peraluminous minerals, observed in the Jebilet granites and characteristic of many of the Nova Scotian granites (Clarke, 1981; MacDonald and Clarke, 1985).

### 2.3.2.3 Ajar El Bark Stock

The Ajar El Bark stock, once used as a quarry, covers an area of approximately $1 \mathrm{Km}^{2}$ and is located SE of Ben Guerir. Six samples were collected from the Ajar El Bark Stock (Figure 2.8). Point counting and classification results are given in Table 2.4 and Figure 2.9. Specimens are monzogranitic in composition and varied in grain size. They are identical to SDB both in color and mineralogical composition.

### 2.3.2.4 Ras El Abiod Pluton

The Ras El Abiod pluton (REA) is a peraluminous muscovite-granite of variable grain size (Morin, 1951; Gigout, 1951) and crops out over an area of approximately $8 \mathrm{Km}^{2}$ (Figure 2.6). It is composed of quartz, K-feldspar, sodic plagioclase, rare biotite (most often chloritized) and, abundant muscovite (Hoepffner, 1982). The granite clearly shows a pneumatolitic tendency with the development of quartz-rich greisens and muscovite rosettes (Chauris and Huvelin, 1964). The granite will often contain tourmaline, and may include "minuscule" garnets (Gigout, 1951). It is gray to pink in colour, and may sometimes be red, where fluid interaction was at its greatest.

Aplitic dykes and small bodies occur throughout the pluton and commonly contain tourmaline. Pegmatites within the pluton usually

## Ajar El Bark Stock



Figure 2.8. Location of samples collected from the Ajar El Bark stock.

Table 2.4. Modal analysis of samples from the Ajar El Bark stock (Rehamna Massif). Based on stained slabs counting, 1000-2000 points. See Figure 2.9 for plot of QAP diagram.

| SAMPLE | QR1 | QR2 | QR3 | QR4 | QR5 | QR6 |
| :--- | ---: | ---: | ---: | ---: | ---: | ---: |
| Quartz | 30.7 | 34.0 | 34.9 | 30.1 | 27.2 | 36.1 |
| Plagioclase | 32.0 | 29.5 | 28.1 | 27.6 | 32.3 | 28.9 |
| K-Feldspar | 34.8 | 33.5 | 32.7 | 39.9 | 38.8 | 32.4 |
| Biotite | 2.4 | 2.4 | 4.1 | 2.4 | 1.8 | 2.5 |
| Muscovite | 0.1 | 0.6 | 0.2 | tr. | tr. | 0.2 |
| Rock name | Mg | Mg | Mg | Mg | Mg | Mg |

$\mathrm{Mg}=$ monzogranite

$$
\operatorname{tr} .=\text { trace }
$$

## Ajar El Bark Stock


$\mathrm{Sg}=$ syenogranite $\mathrm{Mg}=$ monzogranite $\mathrm{Gd}=$ granodiorite

Figure 2.9. QAP (Quartz- Alkali Feldspar- Plagioclase) plot of modal analysis from the Ajar el Bark stock (Rehamna Massif, Morocco). See Table 2.4 for modal analysis and Figure 2.8 for sample locations. Classification after Streckeisen, 1976.
occur as small pockets (or miarolitic cavities) and contain tourmaline. A pegmatite containing crystals of beryl has also been reported intruding the country rock (Chauris and Huvelin, 1964).

Although the granite was sampled in some detail (48 specimens) its altered state rendered it "unsuitable" for this study.

### 2.3.3 Central Massif

### 2.3.3.1 Introduction

The Central massif is the northermost massif considered in this study. The characteristics of the Zaer, Ment and Oulmes plutons as described by previous authors are given below.

### 2.3.3.2 Zaer Pluton

The Zaer pluton, probably one of the better-known granitic bodies of Morocco has been the subject of many recent studies (Mrini, 1985; Guiliani, 1982; Guiliani and Sonet, 1982; Mahmood, 1980; 1984; 1985). The pluton is principally composed of a two-mica monzogranite (two-mica granite) that intrudes a biotite granodiorite (biotite granite). More detailed divisions have been suggested (Guiliani, 1982, Mahmood,1980), however these constitute more refinement than is needed for this study.

Modal mineralogical studies of the Zaer pluton conducted by Mahmood (1980) and Guiliani (1982) revealed that compositions within the granite generally range from tonalite to syenogranite on a Streckeisen diagram, with some quartz diorites. Both the biotite granite and the two-mica granite belong to the calc-alkaline series as defined by Lameyre (1980). The granite of varied grain size is mainly an assemblage of quartz, plagioclase, K-feldspar, biotite and muscovite. Accessory minerals include ilmenite, rutile, apatite, zircon, rare sphene and monazite (only found in the most mafic facies). Very rare garnet and andalusite have also been reported in the
muscovite-bearing facies of the granite (Mahmood and Bennani, 1984). Detailed mineralogical work caried out by Guiliani (1982) has revealed some "significant" differences in the evolution of the biotite granites and two-mica granites. This is also supported by chemical evidence. Minor facies within the pluton include a muscovite leucogranite and some aplitic bodies.

### 2.3.3.3 Ment Pluton

The Ment pluton located north of Aguelmous can be divided into two rock types: biotite granite and leucogranites (Boushaba, 1984). The biotite granites (or monzogranites and syenogranites on a Streckeisen diagram) account for most of the exposed area and can be further subdivided into a porphyritic and a non-porphyritic subfacies. The latter is characterized by the absence of phenocrysts of K -feldspar and enclaves. The biotite granites generally contain quartz, K-feldspar, plagioclase, biotite with minor amounts of zircon, apatite, topaz and rare hematite inclusions in the biotite. The biotite granites are comparable to the biotite granite of the Zaer pluton (Boushara, 1984). A second minor facies of the pluton, the leucogranites, are essentially found in the north-western part of the pluton. They include quartz, plagioclase, K-feldspar, muscovite and may contain siderophyllite topaz, zinnwaldite and very rare biotite (Boushara, 1984). As a result of intense tourmalinisation, tourmaline can be found throughout the pluton.

### 2.3.3.4 Oulmes Pluton

The Oulmes massif located some 100 Km east of Rabat, covers an area of approximately $30 \mathrm{Km}^{2}$. Although many different facies have been described for the Oulmes pluton (Termier et al., 1950) it can generally be divided into two main rock types: two-mica granites and muscovite granites (Mahmood, 1980; Mahmood and Bennani, 1984; Mrini, 1985). Both can be classified as monzogranites on a Streickeisen diagram (Mahmood, 1980). In addition to the usual quartz, K-feldspar, plagioclase,
muscovite and rare biotite, minerals such as garnet, andalusite, cordierite, tourmaline, accessory zircon, apatite and, ilmenite have been reported in the Oulmes pluton. High fluid activity is evident with the presence of primary tourmaline as a main phase in some parts of the granite. Aplites, pegmatites and quartz veins are also reported thoughout the pluton (Termier et al., 1950). The Oulmes monzogranites are reported to be very similar to the two-mica facies of the Zaer pluton (Mahmood, 1980)

### 2.3.4 Atlas Domain

### 2.3.4.1 Introduction

The granites of the Atlas domain are the southernmost and easternmost plutons considered in this study. They include the Tichka, Azegour, Aouli and Bou Mia plutons and are described below.

### 2.3.4.2 Tichka Complex

The Tichka complex located some 100 km SW of Marrakech, covers an area of approximately $320 \mathrm{~km}^{2}$. With a vertical relief of about 2000 m the massif represents one of the better exposed plutons of Morocco and, as such, has been the subject of many recent studies (Termier et al., 1971; Vogel and Walker, 1975; Vogel et al., 1976; Scott and Vogel, 1980; Termier, 1982; Lagarde and Roddaz, 1983; Mrini, 1985; Gasquet et al, in prep.). The Tichka rocks were emplaced during the Westphalian into mostly Cambrian rocks (Termier et al. 1972). According to Gasquet et al. in prep.) the Tichka massif can be divided into 4 distinct groups: 1) gabbros, 2) diorites, 3) granites and 4) late stage aplites and pegmatites.

The gabbros can be further subdivided into olivine and pyroxene bearing facies and include, cumulate plagioclase, green and/or brown hornblende and locally orthopyroxene. The diorite may be fine grained or porphyritic and has been described as hornblende diorites and quartz
diorites (Lagarde and Roddaz, 1983). These rocks are somewhat heterogenous in composition and are reported to contain smaller amounts of gabbro and granodiorite (Vogel et al. 1976).

The granitoids are generally discussed with reference to two groups (Vogel et al. 1976; Scott and Vogel, 1980; Lagarde and Roddaz, 1983): the southern and northern granites. The granites in the southern portion of the massif include monzogranites, granodiorites and tonalites. They are porphyritic and contain biotite, amphibole, sphene, apatite and allanite. The northern granites are non-porphyritic and sometimes almost pegmatitic. The latter are characterized by the absence of hornblende and enclaves of quartz diorite. Hornblende is rarely present in the non-porphyritic variety and not reported in the "pegmatitic" granite.

The late stage dykes present a duality in composition and can either be felsic (aplites and pegmatites) or mafic (microgabbros and dolerites).

The duality of magmatism in the Tichika complex and in particular the presence of a possible mantle component will be readressed in Chapter 3.

### 2.3.4.3 Azegour, Aouli and Bou Mia Plutons

The Azegour pluton of the High-Atlas and the Aouli and Bou Mia plutons of the Middle-Atlas are of minimal importance to this thesis and therefore require only a brief mention (Figure 1.3).

The Azegour pluton located east of the Tichka massif has been described as an evolved monzogranite and syenogranite belonging to the calc-alkaline monzonitic series of Lameyre (1982) (Mrini, 1985).

The Aouli and Bou Mia plutons of the Middle Atlas (east of the Meseta) contain diorites, granodiorites and monzogranites) and, calc-alkaline granites respectively (Tisserant, 1977; Mrini, 1985).

### 2.3.5 Emplacement of the Moroccan Granites

The granites of the Jebilet Massif intrude a sequence of volcano-sedimentary rocks, folded during Late Visean-Namurian time. This suggests a post-Visean age for their emplacement. Although these granites are generally considered to be post-tectonic, some evidence suggests at least a partial syn-tectonic history for these rocks (Huvelin, 1977; Lagarde and Choukroune, 1982). Contact metamorphic minerals such as chlorite, biotite and andalusite are stretched and deformed in the plane of the schistosity. The trajectory of the schistosity is reported in some areas to affect the plutons (south of the TBP and Bramram). Megazones of ductile shearing of S-SE direction and sinistrial displacement offer further evidence for partial syn-tectonic emplacement for these plutons. Abundant hornfels and the preservation of contact minerals (such as post-tectonic equant biotite) in rocks affected by moderately low grade regional metamorphism indicate that the granites remained active after the effects of the Hercynian deformation had ceased (Huvelin, 1977).

The Rehamna granites were emplaced in micaschist of the mesozone (garnet straurolite and kyanite assemblages). The presence of pegmatitic dykes of the Ras El Abiod pluton, which cut the final phase of the Hercynian deformation (Jenny, 1974) and the development of contact metamorphic minerals which clearly postdate the regional metamorphism (the latter occasionaly retrograded in the contact aureoles of the granites) indicate a post-tectonic emplacement for the granites of the Rehamna massif (Pique, 1972; Hoepffner, 1974; Jenny, 1974; Huvelin, 1977; Michard, 1982).

The granites of the Central massif are generaly considered to be syn- to post-tectonic (all display a contact aureole). The Zaer pluton
intrudes a sequence of schist, quartzites and carbonates ranging from Cambro-Ordovician to Devonian in age and of low-grade regional metamorphism. The Ment pluton intrudes sandstones and carbonates of upper Visean age, flysch of Namurian age, and Cambro-Ordovician schist. Foliation of contact aureole minerals in the Cambro-Ordovician rocks suggest a partial syn-tectonic (Hercynian) emplacement for the Oulmes granite (Boushaba, 1985; Pique, 1976)

The Tichka massif of the Atlas domain was emplaced in a series essentially Cambrian in age. The massif is syn to post-Hercynian, with a development of a contact aureole in a terrane characterized by epizonal metamorphism.

### 2.3.6 Geochronology of the Moroccan Granites

Geochronological data for the Moroccan granites determined using the $\mathrm{Rb}-\mathrm{Sr}$ method by Mrini (1985) and Tisserant (1976) are summarized in Figure 2.10. Magmatism in Morocco ranges from the Visean ( 340 Ma ) to the Permian ( 245 Ma ). No clear regional patterns or zonations of these ages are evident in the area.
${ }^{87} \mathrm{Rb} /{ }^{86} \mathrm{Sr}$ isotopic analyses of the Oulad Ouaslam batholith, Tabouchennt-Bamega pluton and, Bramram greisen suggest that magmatism lasted at least 10 million years and occurred between 330 and 340 Ma ago in the Jebilet Massif (Mrini, 1985).

The Sebt de Brikiine batholith and Ajar El Bark stock were dated at $268 \pm 6 \mathrm{Ma}$ (both massifs were used for the age determination) (Mrini, 1985). An approximate age of 265 Ma was determined for the leucogranites of the Ras El Abiod pluton (Mrini, 1985).
${ }^{87} \mathrm{Rb} /{ }^{86} \mathrm{Sr}$ analyses of the Zaer pluton indicate an age of $303 \pm 13$ Ma for the biotite granite (tonalite-granodiorite) and $279 \pm 11 \mathrm{Ma}$ for the two-mica granite (monzogranite) (Mrini, 1985). The biotite granite facies of the Ment pluton was emplaced some $279 \pm 6 \mathrm{Ma}$ and the

Mrini, 1985 Tisserant, 1977 X Approximate age

Figure 2.10. Sr ages for the Moroccan granites data from Mrini, 1985 and Tisserant, 1977.
leucogranite facies was dated at $270 \pm 3 \mathrm{Ma}$. The age of the Oulmes pluton was estimated at 298 Ma (Mrini, 1985).

In the Atlas domain an approximate age of 330 Ma was reported for the emplacement of the Tichka complex (Mrini, 1985). The Azegour pluton is reported to have been emplaced some $271 \pm 3 \mathrm{Ma}$ ago (Mrini, 1985). The Aouli and Bou Mia granites of the Middle Atlas as determined by Tisserant (1977) are reported to range from $270-330 \mathrm{Ma}$ in age.

### 2.3.7 Summary of the Moroccan Granites

Magmatism in Paleozoic terranes of the Meseta and Atlas domains of Morocco cover a wide range of time from approximately 330 Ma to 245 Ma . Compositions within and between the plutons are also quite varied. The granites intrude rocks of varied composition and age (from Cambrian to Upper Visean). The granites were intruded into sequences which have undergone different degrees of metamorphism. Evidence also suggests that these were emplaced at different crustal levels. All of these factors will of course contribute complexity to the problem of characterizing Moroccan magmatism.

### 2.4 Comparison of the Geochronology of the Nova Scotian and Moroccan Granites

Perhaps one of the most contentious issues in this study is the significance of the difference in timing between magmatism in Morocco and Nova Scotia. A basic assumption of this thesis is that granite compositions have more to do with the source and the sum of processes which affect the granite throughout its evolution than with time itself, and even though the magmatism in Nova Scotia and Morocco is different, their granite populations may show some geochemical similarities (Clarke and Richard, 1986).

When the granites of Nova Scotia were first dated the southern plutons appeared to be younger than the northern plutons. Upon further detailed analyses of the granites it became evident that the southern plutons were in fact generally emplaced at the same time as the northern plutons. The younger ages obtained for the southern plutons were the result of a tectono-thermal reset event. It is not unreasonable to expect that the granites of Morocco may have been similarly affected by such an event and perhaps future workers may discover that the Moroccan granites are in effect older than first believed.

## CHAPTER 3

## GEOCHEMISTRY

### 3.1 Introduction

The advent of abundant and high-quality geochemical data opens new avenues of investigation and interpretation of geochemical data. In particular, as geochemical databases increase in size, multivariate statistical analyses may now be applied. In this thesis the geochemical data from Nova Scotia and Morocco are compared using both the traditional (this chapter) and a multivariate statistical approach (Chapter 5). The applicability of such methods to the geochemical data will be assessed in this study (Chapter 4).

In this chapter, the geochemical data are investigated using the traditional approach. The regional characteristics of the Nova Scotian granites are first reviewed. The characteristics of granitic bodies in Morocco are then reviewed. Finally the Nova Scotian and Moroccan granites are compared in order to further assess the viability of the "Moroccan Model".

### 3.2 Nova Scotian Granites

### 3.2.1 Introduction

Differences in composition, age and emplacement between the northern and southern peraluminous plutons of the Meguma Zone were outlined in Chapter 2. However, no attempts were made by previous workers to evaluate the scope of these differences from a geochemical point of view. In this section the geochemical data are studied to determine whether a true distinction can be made between the northern and southern plutons.

### 3.2.2 Major and Trace Elements

A geochemical database of 400 analyses from the Nova Scotian granites was compiled during this study. A complete listing of the data as well as their source references is given in Appendix D.

The basic statistics of the northern ( N ) and southern ( S ) plutons of Nova Scotia are presented in Table 3.1. The more varied compositions of the southern plutons relative to the northern plutons are evident when the minimum and maximum values of the major and trace element data are compared (Table 3.1). Although some differences are apparent in the major and trace element contents of these two groups, a clear distinction can not be drawn between the groups simply on the basis of their range, as a great deal of overlap exists between both groups. This is illustrated in Figure 3.1, where the frequency distributions of the various elements were plotted for each group.

Differences between the two populations include a unimodal distribution of $\mathrm{K}_{2} \mathrm{O}$ in the N compared with bimodal in the S , and the distributions of $\mathrm{CaO}, \mathrm{Ba}, \mathrm{Rb}, \mathrm{Sr}, \mathrm{Zr}$ and V . However, the ranges for the N and S populations overlap considerably and separation of the two groups using frequency distributions is not possible.

The visual analysis of histogram data, although of limited use, does show some apparent differences in the distribution of elements within both areas, suggesting that further analyses of these populations are warranted using more "sophisticated" methods, such as basic bivariate statistics and ultimately multivariate statistics.

The mean and standard deviation values of major and trace elements from the northern and southern plutons from Table 3.1 were examined. To aid in the visualisation of the variation, the average and standard deviation values of the northern and southern plutons were plotted on a spider diagram (Figure 3.2). Normalizing values for spidergrams used thoughout this Chapter were drawn from Taylor, 1980 except for

Table 3.1. Basic statistics for major and trace element data from the northern and southern plutons of Nova Scotia.

|  | North South | North South | North South | North South |
| :---: | :---: | :---: | :---: | :---: |
|  | $\mathrm{SiO}_{2}$ | $\mathrm{TiO}_{2}$ | $\mathrm{Al}_{2} \mathrm{O}_{3}$ | $\mathrm{Fe}_{2} \mathrm{O}_{3 t}$ |
| Minimum | $63.93 \quad 59.75$ | $0.02 \quad 0.02$ | 11.3912 .14 | $0.17 \quad 0.47$ |
| Maximum | $78.13 \quad 76.70$ | $0.82 \quad 1.07$ | $17.70 \quad 20.18$ | $6.10 \quad 6.74$ |
| Mean | $72.13 \quad 70.79$ | $0.24 \quad 0.38$ | $14.44 \quad 15.18$ | 1.902 .58 |
| St.Deviation | $2.57 \quad 3.68$ | $0.17 \quad 0.24$ | $0.90 \quad 1.35$ | $1.15 \quad 1.39$ |
| Number | 230153 | 230153 | 230153 | 230153 |
|  | MnO | MgO | CaO | $\mathrm{Na}_{2} \mathrm{O}$ |
| Minimum | $0.01 \quad 0.02$ | $0.02 \quad 0.07$ | $0.28 \quad 0.19$ | $2.21 \quad 1.52$ |
| Maximum | $0.60 \quad 0.30$ | $1.72 \quad 3.62$ | $2.31 \quad 5.19$ | $5.02 \quad 5.82$ |
| Mean | $0.06 \quad 0.06$ | $0.44 \quad 0.95$ | 0.791 .83 | $3.58 \quad 3.74$ |
| St.Deviation | $0.06 \quad 0.04$ | $0.37 \quad 0.76$ | $0.52 \quad 1.18$ | 0.450 .60 |
| Number | $230 \quad 152$ | $230 \quad 153$ | $230 \quad 153$ | $230 \quad 153$ |
|  | $\mathrm{K}_{2} \mathrm{O}$ | $\mathrm{P}_{2} \mathrm{O}_{5}$ | A/CNK | D.I. |
| Minimum | $3.15 \quad 1.15$ | 0.020 .03 | 1.020 .93 | 73.9259 .39 |
| Maximum | 5.98 6.40 | $0.87 \quad 0.82$ | 1.391 .46 | $96.69 \quad 95.37$ |
| Mean | 4.43 3.52 | $0.24 \quad 0.21$ | $1.19 \quad 1.15$ | $89.54 \quad 83.20$ |
| St.Deviation | $0.51 \quad 1.24$ | $0.12 \quad 0.10$ | $0.07 \quad 0.09$ | 5.088 .68 |
| Number | $230 \quad 153$ | $230 \quad 153$ | $230 \quad 153$ | $230 \quad 153$ |
|  | Ba | $\mathbf{R b}$ | Sr | Y |
| Minimum | 24 | 123 36 | $2 \quad 3$ | $3 \quad 7$ |
| Maximum | 9631200 | 931360 | 248720 | 5035 |
| Mean | 282535 | 314138 | 71191 | 1517 |
| St.Deviation | 237222 | 15561 | 57135 | 105 |
| Number | $232 \quad 134$ | $239 \quad 147$ | $226 \quad 146$ | $87 \quad 107$ |
|  | $\mathbf{Z r}$ | Nb | Th | Pb |
| Minimum | $15 \quad 32$ | 13 | $1.0 \quad 0.4$ | 4 4 |
| Maximum | 267389 | $31 \quad 29$ | $34.0 \quad 44.0$ | $74 \quad 36$ |
| Mean | 84141 | $12 \quad 11$ | $9.9 \quad 7.1$ | $25 \quad 19$ |
| St.Deviation | 4763 | 45 | 6.67 .6 | $11 \quad 7$ |
| Number | 203125 | 134114 | $153-119$ | $128 \quad 128$ |

Table 3.1 (cont.). Basic statistics for major and trace element data from the northern and southern plutons of Nova Scotia.

|  | North | South | North | South | North | South | North | South |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | Ga |  | Zn |  | Cu |  | Ni |  |
| Minimum | 17 | 12 | 5 | 5 | 1 | 1 | 1 | 1 |
| Maximum | 27 | 26 | 195 | 91 | 104 | 19 | 95 | 47 |
| Mean | 22 | 20 |  | 53 | 10 | 6 | 11 | 10 |
| St.Deviation | 3 | 3 | 24 | 18 | 13 | 4 | 14 | 7 |
| Number | 45 | 107 | 197 | 129 | 115 | 42 | 117 | 110 |
|  | V |  | Cr |  | Hf |  | Cs |  |
| Minimum | 1 | 1 | 2 | 3 | 1.4 |  | 5.0 |  |
| Maximum | 73 | 136 | 404 | 111 | 4.2 |  | 44.2 | - |
| Mean | 14 | 38 |  | 38 | 2.7 |  | 20.5 | - |
| St.Deviation Number | 13 | 34 | 57 | 27 | 0.8 |  | 10.5 | - |
|  | 74 | 104 | 99 | 114 | 15 | - | 19 | . |
|  | Sc |  | Ta |  | Co |  | Li |  |
| Minimum | 1 | - | 0.7 | - | 1 | - | 36 | 19 |
| Maximum | 7 | - | 9.7 | - | 5 | - | 801 | 262 |
| Mean | 5 | - | 5.2 | - | 2 | - | 184 | 73 |
| St.Deviation | 1 | - | 2.7 | - | 1 | - | 143 | 40 |
| Number | 26 | - | 15 | - | 27 | - | 128 | 86 |
|  | Be |  | B |  | F |  | Cl |  |
| Minimum | 2.0 | 0.5 | 3 | 3 | 210 | 60 | 50 |  |
| Maximum | 205.0 | 20.0 | 150 | 50 | 2400 | 1300 | 175 | - |
| Mean | 9.8 | 4.2 | 22 | 15 | 643 | 428 | 81 | - |
| St.Deviation | 22.8 | 3.4 | 18 | 13 | 381 | 218 | 62 | - |
| Number | 83 | 60 | 99 | 15 |  | 85 |  | - |
|  | U |  | W |  | Sn |  | Mo |  |
| Minimum | 1.6 | 0.8 | 1 | 4 | 1 | 1 | 0.6 | 0.8 |
| Maximum | 35.0 | 9.9 | 19 | 34 | 52 | 31 | 4.0 | 10.0 |
| Mean | 8.8 | 3.6 | 4 | 17 | 14 | 9 | 1.5 | 2.6 |
| St.Deviation | 6.7 | 2.0 | 3 | 12 | 12 | 6 | 0.7 | 2.8 |
| Number | 122 | 75 | 67 | 6 | 194 | 77 | 130 | 14 |



Figure 3.1. Frequency distributions of major and trace element data from Northern and Southern plutons of Nova Scotia. Major oxides in wt \% and trace elements in ppm.


Figure 3.1 (cont.). Frequency distributions of major and trace element data from the Northern and Southern plutons of Nova Scotia.


Figure 3.1 (cont.). Frequency distributions of major and trace element data from Northern and Southern plutons of Nova Scotia.

## Nova Scotia (North-South)



Figure 3.2. Spider diagram comparing the Northern and Southern geochemical populations of Nova Scotia. Each range represents the mean $\pm 1 \sigma$.

## Nova Scotia (North-South)



Figure 3.3. REE plots comparing the Northern (A) and Southern (B) plutons of Nova Scotia.
phosphorus which was taken from Allegre and Michard, 1973. The normalizing values used for each element are shown on the diagrams. Both groups display a considerable amount of overlap. Some variation is evident in the generally lower Rb , and higher $\mathrm{Ba}, \mathrm{Sr}, \mathrm{Fe}, \mathrm{Mg}$, and V values of the southern group relative to the north. A spidergram in this case does not appear to be a useful tool in separating the two areas.

Although the mean values for some oxides and trace elements in Table 3.1 and Figure 3.1 suggest some apparent differences between the southern and northern plutons, their variances are generally large. As a result, a great deal of overlap exists between the various populations. Consequently, more detailed analysis of the means may be misleading and was not pursued.

### 3.2.3 Rare Earth Elements

All available rare earth data from the northern and southern plutons was plotted on Figure 3.2. The limited amount of data within each group has made an adequate comparison difficult. The patterns and values for both groups are very similar, and within analytical error and the uncertainty associated to the fact that these are probably not representative sample populations, they can not be separated.

### 3.2.4 Isotopic Data

Both oxygen (Longstaffe et al., 1979) and sulfur isotopes (Kubilius, 1983) suggest some difference between the northern and southern compositions. Although the northern and southern plutons appear to have been derived by anatexis of clastic metasedimentary rocks, the southern plutons also appear to contain a more mafic (mantle?) component. This is evident in the lower sample $\delta^{18} \mathrm{O}$ and significantly lower $\delta{ }^{34} \mathrm{~S}$ values within the southern plutons when compared with the northern plutons.

Nd and Sr isotopic data are only available for the northern plutons. These data will be discussed in Section 4.5 .1 in which Nova Scotia is compared with Morocco.

### 3.3 Moroccan granites

### 3.3.1 Introduction

In this section results of geochemical analysis of samples collected during this study are investigated. The major, trace and rare earth element contents of the Jebilet and Rehamna granites are examined and compared with reported values for the Central massif granites as well as the Atlas domain granites in order to characterize the behaviour of the geochemical elements in the Moroccan granites.

### 3.3.2 The Moroccan Data

The geochemistry of twenty-seven samples from the Jebilet Massif and seventeen samples from the Rehamna Massif is presented and discussed in this section (see Figures 2.1, 2.3, 2.5 and 2.7 for sample locations). This includes eight samples from the Tabouchennt-Bamega Pluton (TBP) and nineteen are from the Oulad Ouaslam Batholith (OOB) of the Jebilet Massif, and fourteen samples from the Sebt de Brikiine batholith (SDB) and three from the Ajar El Bark stock (AEB) of the Rehamna Massif. Samples were analysed for major and trace element (Ba, $\mathrm{Rb}, \mathrm{Sr}, \mathrm{Y}, \mathrm{Zr}, \mathrm{Nb}, \mathrm{Th}, \mathrm{Pb}, \mathrm{Ga}, \mathrm{Zn}, \mathrm{Cu}, \mathrm{Ni}, \mathrm{V}, \mathrm{Cr}$ ) content by X-ray fluorescence. Four samples from the Oulad Ouaslam Batholith, three from the Tabouchennt-Bamega Pluton, four from the Sebt De Brikiine batholith and 2 from the Ajar El Bark stock were analysed by INAA for REE ( $\mathrm{Ce}, \mathrm{Nd}, \mathrm{Sm}, \mathrm{Eu}, \mathrm{Tb}, \mathrm{Yb}, \mathrm{Lu}$ ) content. The precision and accuracy of the analytical techniques are discussed in Appendix B.

Additional Moroccan data from the Central and Tichka Massifs were coilected from various bibliographic sources. A complete listing of the data is given in Appendix D.

### 3.3.3 Major and Trace Elements

The minimum, maximum, mean and standard deviation values for the Jebilet, Rehamna, Central and Tichka granites are compared in Table 3.1. Some differences between the massifs are evident in Table 3.2. Particularly averages for the Rehamna granites contrast clearly with those of the other massifs, markably in its higher $\mathrm{SiO}_{2}$, and Y values as well as its lower $\mathrm{TiO}_{2}, \mathrm{Al}_{2} \mathrm{O}_{3}, \mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}, \mathrm{MgO}, \mathrm{CaO}, \mathrm{Ba}, \mathrm{Sr}$ and Nb values. A spidergram of the Central, Rehamna and Jebilet granites is presented in Figure 3.4. The Tichka granite could not be included because of the limited amount of data. The ranges for each massif were determined by calculating the mean $\pm 1$ from Table 3.2. The contrast between the Rehamna and Jebilet granites is evident in this diagram. The trace elements of the Rehamna granites display more variation than the Jebilet granites, however the Central massif show the most variation with values covering the range defined by the two other massifs.

### 3.3.4 Rare Earth Elements

Rare earth element analysis performed on four samples from the OOB and three from the TBP, four from the SDB and two from the AEB are plotted onto a chondrite normalized diagram in Figure 3.5.

The Jebilet and Rehamna granites display similar REE patterns, and differ only in their Eu values which is probably related to the crystallisation of feldspars. As for the trace element data the REE patterns for the Rehamna show greater variation than the Jebilet granites. The limited amount of data make the characterization of Rare Earths within the Moroccan granites difficult.

### 3.3.5 $\mathrm{Rb}-\mathrm{Sr}$ and $\mathrm{Sm}-\mathrm{Nd}$ Isotopes

$\mathrm{Rb}-\mathrm{Sr}$ and $\mathrm{Sm}-\mathrm{Nd}$ isotopic values as determined by Mrini (1985) and Tisserant (1977) are plotted on Figure 3.6. Values were plotted by

Table 3.2. Basic statistics for major and trace element data from the Central, Rehamna, Jebilet and Tichka Massifs of Morocco.

|  | Central | Rehamna | Jebilet | Tichka | Central | Rehamna | Jebilet | Tichka |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\mathrm{SiO}_{2}$ |  |  |  | $\mathrm{TiO}_{2}$ |  |  |  |
| Minimum | 60.48 | 74.74 | 62.80 | 67.32 | 0.01 | 0.02 | 0.06 | 0.14 |
| Maximum | 77.61 | 77.93 | 75.66 | 75.08 | 1.13 | 0.20 | 0.96 | 0.51 |
| Mean | 71.89 | 76.58 | 68.25 | 70.41 | 0.25 | 0.09 | 0.58 | 0.40 |
| St.Deviation | 3.30 | 0.90 | 2.67 | 2.60 | 0.22 | 0.05 | 0.17 | 0.14 |
| Number | 216 | 17 | 31 | 6 | 204 | 17 | 31 | 6 |
|  | $\mathrm{Al}_{2} \mathrm{O}_{3}$ |  |  |  | $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ |  |  |  |
| Minimum | 12.19 | 12.25 | 13.45 | 13.14 | 0.10 | 0.40 | 0.42 | 0.95 |
| Maximum | 18.90 | 13.56 | 16.55 | 15.34 | 6.55 | 1.49 | 6.59 | 3.15 |
| Mean | 15.05 | 12.80 | 15.35 | 14.39 | 1.88 | 0.99 | 4.13 | 2.27 |
| St.Deviation | 1.16 | 0.39 | 0.80 | 0.81 | 1.24 | 0.29 | 1.11 | 0.77 |
| Number | 216 | 17 | 31 | 6 | 216 | 17 | 31 | 6 |
|  | MnO |  |  |  | MgO |  |  |  |
| Minimum | 0.01 | 0.01 | 0.02 | 0.02 | 0.01 | 0.01 | 0.06 | 0.80 |
| Maximum | 1.07 | 0.04 | 0.10 | 0.10 | 3.17 | 0.21 | 2.34 | 2.14 |
| Mean | 0.05 | 0.02 | 0.06 | 0.06 | 0.63 | 0.06 | 1.24 | 1.56 |
| St.Deviation | 0.08 | 0.01 | 0.02 | 0.03 | 0.65 | 0.06 | 0.52 | 0.46 |
| Number | 183 | 17 | 31 | 6 | 199 | 17 | 31 | 6 |
|  | CaO |  |  |  | $\mathrm{Na}_{2} \mathrm{O}$ |  |  |  |
| Minimum | 0.01 | 0.33 | 0.37 | 0.75 | 0.20 | 3.05 | 1.80 | 3.39 |
| Maximum | 4.49 | 1.92 | 3.37 | 2.16 | 4.80 | 4.23 | 3.86 | 4.72 |
| Mean | 1.04 | 0.75 | 1.90 | 1.65 | 3.16 | 3.64 | 2.86 | 4.32 |
| St.Deviation | 0.93 | 0.41 | 0.63 | 0.56 | 0.63 | 0.29 | 0.36 | 0.48 |
| Number | 203 | 17 | 31 | 6 | 216 | 17 | 31 | 6 |
|  | $\mathrm{K}_{2} \mathrm{O}$ |  |  |  | $\mathrm{P}_{2} \mathrm{O}_{5}$ |  |  |  |
| Minimum | 1.15 | 4.32 | 3.65 | 3.19 | 0.02 | 0.01 | 0.01 | 0.10 |
| Maximum | 7.32 | 5.19 | 6.06 | 5.00 | 1.45 | 0.10 | 0.46 | 0.15 |
| Mean | 4.43 | 4.79 | 4.47 | 3.71 | 0.30 | 0.03 | 0.17 | 0.13 |
| St.Deviation | 0.94 | 0.21 | 0.61 | 0.68 | 0.27 | 0.02 | 0.07 | 0.01 |
| Number | 216 | 17 | 31 | 6 | 69 | 17 | 31 | 5 |

Table 3.2 (cont.). Basic statistics for major and trace element data from the Central, Rehamna, Jebilet and Tichka Massifs of Morocco.

|  | Central | Rehamna | Jebilet | Tichka | Central | Rehamna | Jebilet | Tichka |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | A/CNK |  |  |  | D.I. |  |  |  |
| Minimum | 0.94 | 0.86 | 1.03 | 0.99 | 61.70 | 92.96 | 65.69 | 79.01 |
| Maximum | 2.83 | 1.13 | 1.41 | 1.06 | 96.60 | 96.93 | 94.95 | 90.04 |
| Mean | 1.30 | 1.02 | 1.19 | 1.02 | 87.22 | 94.83 | 79.50 | 83.33 |
| St.Deviation | 0.20 | 0.07 | 0.08 | 0.03 | 6.91 | 1.22 | 5.87 | 3.97 |
| Number | 216 | 17 | 31 | 6 | 216 | 17 | 31 | 6 |
|  | Ba |  |  |  | Rb |  |  |  |
| Minimum | 7 | 4 | 217 | - | 5 | 150 | 130 | - |
| Maximum | 1540 | 234 | 778 | - | 1473 | 368 | 286 | - |
| Mean | 317 | 91 | 545 | - | 345 | 248 | 193 | - |
| St.Deviation | 244 | 75 | 121 | - | 214 | 68 | 46 | - |
| Number | 190 | 17 | 27 | - | 195 | 17 | 27 | - |
|  | Sr |  |  |  | Y |  |  |  |
| Minimum | 10 | 4 | 74 | - | 8 | 19 | 17 | - |
| Maximum | 681 | 75 | 631 | - | 51 | 89 | 46 | - |
| Mean | 150 | 28 | 208 | - | 25 | 40 | 35 | - |
| St.Deviation | 142 | 22 | 124 | - | 12 | 20 | 7 | - |
| Number | 192 | 17 | 27 | - | 27 | 17 | 27 | - |
|  | Zr |  |  |  | No |  |  |  |
| Minimum | 32 | 43 | 40 | - | 7 | 12 | 6 | - |
| Maximum | 271 | 138 | 238 | - | 37 | 67 | 17 | - |
| Mean | 135 | 94 | 196 | - | 17 | 31 | 14 | - |
| St.Deviation | 75 | 26 | 39 | - | 7 | 13 | 2 | - |
| Number | 27 | 17 | 27 | - | 27 | 17 | 27 | - |
|  | Th |  |  |  | Pb |  |  |  |
| Minimum | 1.0 | 15.0 | 2.0 | - | 2 | 8 | 18 | - |
| Maximum | 38.0 | 55.0 | 27.0 | - | 37 | 28 | 72 | - |
| Mean | 12.8 | 38.6 | 15.8 | - | 22 | 18 | 26 | - |
| St.Deviation | 9.3 | 10.4 | 5.9 | - | 8 | 6 | 11 | - |
| Number | 27 | 17 | 27 | - | 27 | 17 | 27 | - |

Table 3.2 (cont.). Basic statistics for major and trace element data from the Central, Rehamna, Jebilet and Tichka Massifs of Morocco.

|  | Central | Rehamr | Jebilet | Tichka | Central | Reham | Jebilet | Tichka |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | Ga |  |  |  | Zn |  |  |  |
| Minimum | 16 | 17 | 15 | - | 1 | 11 | 15 | - |
| Maximum | 36 | 23 | 23 | - | 122 | 27 | 157 | - |
| Mean | 23 | 19 | 20 | - | 54 | 17 | 61 | - |
| St.Deviation | 5 | 2 | 2 | - | 23 | 4 | 28 | - |
| Number | 27 | 17 | 27 | - | 137 | 17 | 27 | - |
|  |  | Cu |  |  |  | N |  |  |
| Minimum | 1 | 1 | 3 | - | 4 | 11 | 17 | - |
| Maximum | 112 | 9 | 23 | - | 94 | 47 | 32 | - |
| Mean | 11 | 3 | 12 | - | 17 | 27 | 27 | - |
| St.Deviation | 15 | 3 | 6 | - | 14 | 11 | 4 | - |
| Number | 121 | 11 | 26 | - | 65 | 17 | 27 | - |
|  |  | V |  |  |  | C |  |  |
| Minimum | 1 | 1 | 1 | - | 6 | 3 | 10 | - |
| Maximum | 209 | 11 | 117 | - | 265 | 22 | 60 | - |
| Mean | 42 | 4 | 68 | - | 34 | 8 | 41 | - |
| St.Deviation | 44 | 3 | 21 | - | 43 | 4 | 12 | - |
| Number | 82 | 15 | 27 | - | 65 | 17 | 27 | - |
|  |  | C |  |  |  | Li |  |  |
| Minimum | 10 | - | - | - | 23 | - | - | - |
| Maximum | 62 | - | - | - | 1451 | - | - | - |
| Mean | 12 | - | - | - | 140 | - | - | - |
| St.Deviation | 9 | - | - | - | 161 | - | - | - |
| Number | 38 | $-$ | - | - | 130 | - | - | - |
|  |  | B |  |  |  | F |  |  |
| Minimum | 20 | - | - | - | 10 | - | - | - |
| Maximum | 26 | - | - | - | 1000 | - | - | - |
| Mean | 21 | - | - | - | 337 | - | - | - |
| St.Deviation | 2 | - | - | - | 260 | - | - | - |
| Number | 16 | - | - | - | 16 | - | - | - |

Table 3.2 (cont.). Basic statistics for major and trace element data from the Central, Rehamna, Jebilet and Tichka Massifs of Morocco.

|  | Central | Rehamna | Jebilet | Tichka | Central | Rehamna | Jebilet | Tichka |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | U |  |  |  | W |  |  |  |
| Minimum | 1.0 | - | - | - | 0.3 | - | - | - |
| Maximum | 3.1 | - | - | - | 104.0 | - | - | - |
| Mean | 2.2 | - | - | - | 16.8 | - | - | - |
| St.Deviation | 0.7 | - | - | - | 22.9 | - | - | - |
| Number | 9 | - | - | - | 27 | - | - | - |
|  | Sn |  |  |  | Mo |  |  |  |
| Minimum | 2 | - | - | - | 0.1 | - | - | - |
| Maximum | 930 | - | - | - | 0.5 | - | - | - |
| Mean | 30 | - | - | - | 0.2 | - | - | - |
| St.Deviation | 99 | - | - | - | 0.1 | - | - | - |
| Number | 87 | - | - | - | 16 | - | - | . |

Morocco




Figure 3.4. Spider diagram comparing geochemical populations of the Jebilet, Rehamna and Central Massifs. Each range represents the mean $\pm 1 \sigma$.

## Jebilet and Rehamna Granites



Figure 3.5. REE plots comparing the Jebilet (A) and Rehamna (B) granites of Morocco.
massif in which the southernmost massif is found at the bottom of the diagram and the northernmost at the top. A broad regional pattern is evident in the $\mathrm{Rb}-\mathrm{Sr}$ isotopes of Figure 3.6a. Mantle values were detected in the Tichka massif and the Azegour pluton displays low values which are almost mantle-like. Further north in the Jebilet and Rehamna massifs, intermediate values become apparent. In the Central massif and the Middle Atlas (which are at similar latitudes) distinct crustal values and intermediate values are measured.

This regional variation in the $\mathrm{Rb}-\mathrm{Sr}$ isotopes is confirmed by the $\mathrm{Sm}-\mathrm{Nd}$ isotopes (Figure 3.6b), with higher values in the south and lower, crustal values in the north.

The geochronological data for the Moroccan granites is shown in Figure 3.6c. Provided that the age obtained for the Azegour pluton is correct, there does not seem to be a relationship between the age of the pluton and the initial isotopic ratios. This may indicate that the variation in the isotopes did not occur through time, but rather geographically. A bimodality in the magmatism (mantle and crustal values) is suggested by the isotopic data. A dominant mantle component to the south probably mixed with a crustal component in the intermediate plutons, while the northern plutons show less influence by a mantle component.

### 3.4 Comparing the Nova Scotian and Moroccan Granites

### 3.4.1 Introduction

In this section the geochemical data from Nova Scotia and Morocco are compared in order to determine the degree of similarity between the regions. The Nova Scotian granites are considered as one coherent population in this section, that is no distinction is made between the northern and southern granites of Nova Scotia. Equally all of the Massifs of Morocco are considered to be one population.


Nd Isotopes


Age

|  | 340 |  | 290 | 240 |
| :---: | :---: | :---: | :---: | :---: |
| Central <br> Massif | Zaer (biot.gr.) <br> (2 mica) <br> Oulmes  <br> Ment (biot.gr.) <br>  (2 mica) |  |  |  |
| Middle Atlas | Aouli (gran.) <br> (leucogr.) <br> Bou Mia (Calc-alkaline) <br>  (graphic struc.) | 20000 | $\square$ |  |
| Rehamna Massif | Sebt De Brikiine |  |  |  |
| Jebilet Massif | Bramram Tabouchennt-Bamega Oulad Ouaslam | $\frac{x}{x}$ |  |  |
| High Atlas | Azegour Tichka | x |  |  |

*) Mrini, 1985
Tisserant, 1977

Figure 3.6. Sr, Nd initial isotopic ratios and ages for the Moroccan granites data from Mrini, 1985 and Tisserant, 1977.

### 3.4.2 Major and Trace elements

The minimum, maximum, mean and standard deviation of values from the Nova Scotian and Moroccan granites are compared in Table 3.3. Frequency distributions are shown in Figure 3.7. Generally values for both areas display a considerable amount of overlap. Some minor differences are evident between the two areas particularly in the higher $\mathrm{CaO}, \mathrm{Sr}$, and V values in Morocco. As for the northern and southern granites of Nova Scotia, statistical analysis of the means was not pursued because of the large variances measured in the different populations. The degree of similarity between the two areas is also evident in the spidergram of Figure 3.8, in which the mean $\pm 1$ of each population is presented.

### 3.4.3 Rare Earth Elements

All available rare earth data from Nova Scotia and Morocco are plotted onto Figure 3.9. Because each population includes so few plutons it is doubtful that the sample populations are truly representative. Therefore very little can be said about Figure 3.9 except that both sample populations present some overlap in their values.

### 3.4.4 Isotopes

Nd and Sr isotopic data from Nova Scotia and Morocco are presented in Figure 3.10. Again interpretations are constrained by the unrepresentative nature of the data, in this case Nova Scotia from which Nd isotopic data is only available for the South Mountain batholith.

Data from Australia (McCulloch and Chappell, 1982), the Sierra Nevada Peninsular ranges (Allegre and Othman, 1980; DePaolo, 1980), the french Hercynian (Allegre and Othman, 1980), and the Caledonian

Table 3.3. Basic statistics for major and trace element data from Nova Scotia and Morocco.

|  | N.S. | Mor. | N.S. | Mor. | N.S. | Mor. | N.S. | Mor. |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\mathrm{SiO}_{2}$ |  | $\mathrm{TiO}_{2}$ |  | $\mathrm{Al}_{2} \mathrm{O}_{3}$ |  | $\mathrm{Fe}_{2} \mathrm{O}_{3 \mathrm{t}}$ |  |
| Minimum | 59.75 | 60.48 | 0.02 | 0.01 | 11.39 | 12.19 | 0.17 | 0.10 |
| Maximum | 78.13 | 77.93 | 1.07 | 1.13 | 20.18 | 18.90 | 6.74 | 6.59 |
| Mean | 71.96 | 71.73 | 0.30 | 0.28 | 14.74 | 14.93 | 2.17 | 2.09 |
| St.Deviation | 3.20 | 3.55 | 0.21 | 0.24 | 1.16 | 1.22 | 1.29 | 1.41 |
| Number | 383 | 270 | 383 | 258 | 383 | 270 | 383 | 270 |
|  |  | 0 |  | g0 | Ca |  |  | $\mathrm{a}_{2} \mathrm{O}$ |
| Minimum | 0.01 | 0.01 | 0.02 | 0.01 | 0.19 |  | 1.52 | 0.20 |
| Maximum | 0.60 | 1.07 | 3.62 | 3.17 | 5.19 | 3.17 | 5.82 | 4.80 |
| Mean | 0.06 | 0.05 | 0.64 | 0.69 | 1.21 | 0.69 | 3.65 | 3.18 |
| St.Deviation | 0.05 | 0.07 | 0.61 | 0.67 | 0.99 | 0.67 | 0.52 | 0.63 |
| Number | 382 | 237 | 383 | 253 | 383 | 253 | 383 | 270 |
|  |  | 0 |  | $\mathrm{O}_{5}$ |  | CNK |  | I. |
| Minimum | 1.15 | 1.15 | 0.02 | 0.01 | 0.93 | 0.86 | 59.39 | 61.70 |
| Maximum | 6.40 | 7.32 | 0.87 | 1.45 | 1.46 | 2.83 | 96.69 | 96.93 |
| Mean | 4.07 | 4.44 | 0.23 | 0.22 | 1.18 | 1.27 | 87.00 | 86.73 |
| St.Deviation | 0.98 | 0.89 | 0.12 | 0.23 | 0.08 | 0.20 | 7.42 | 7.28 |
| Number | 383 | 270 | 383 | 122 | 383 | 270 | 383 | 270 |
|  | B |  | R |  | Sr |  |  |  |
| Minimum | 2 | 4 |  | 5 |  |  | 3 | 8 |
| Maximum | 1200 | 1540 | 931 | 1473 | 720 | 1473 | 50 | 89 |
| Mean | 375 | 327 | 247 | 321 | 118 | 321 | 16 | 33 |
| St.Deviation | 262 | 245 | 154 | 202 | 112 | 202 | 8 | 14 |
| Number | 366 | 234 | 386 | 236 | 372 | 239 | 194 | 71 |
|  | Z |  | N |  | T | h |  |  |
| Minimum | 15 | 32 | 1 | 6 | 0.4 | 1.0 | 4 | 2 |
| Maximum | 389 | 271 | 31 | 67 | 44.0 | 55.0 | 74 | 72 |
| Mean | 105 | 148 | 12 | 19 | 8.7 | 20.1 | 22 | 22 |
| St.Deviation | 61 | 67 | 4 | 10 | 7.2 | 13.4 | 9 | 9 |
| Number | 328 | 71 | 248 | 71 | 272 | 71 | 256 | 71 |

N.S. $=$ Nova Scotia Mor. $=$ Morocco

Table 3.3 (cont.). Basic statistics for major and trace element data from Nova Scotia and Morocco.

|  | N.S. | Mor. | N.S. | Mor. | N.S. | Mor. | N.S. | Mor. |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | Ga |  | Zn |  | Cu |  | Ni |  |
| Minimum | 12 | 15 | 5 | 1 | 1 | 1 | 1 | 4 |
| Maximum | 27 | 36 | 195 | 157 | 104 | 112 | 104 | 94 |
| Mean | 20 | 21 | 51 | 52 | 9 | 52 | 9 | 21 |
| St.Deviation | 3 | 4 | 22 | 25 | 11 | 25 | 11 | 13 |
| Number | 152 | 71 | 326 | 181 | 157 | 181 | 157 | 109 |
|  | V |  | Cr |  | Hf |  | Cs |  |
| Minimum | 1 | 1 | 2 | 3 | 1.4 | - | 5.0 | - |
| Maximum | 136 | 209 | 404 | 265 | 4.2 | - | 44.2 | - |
| Mean | 28 | 43 | 38 | 32 | 2.7 | - | 20.5 | - |
| St.Deviation | 30 | 41 | 43 | 26 | 0.8 | - | 10.5 | - |
| Number | 178 | 124 | 213 | 109 | 15 | - | 19 | - |
|  | Sc |  | Ta |  | Co |  | Li |  |
| Minimum | 1 | - | 0.7 | - | 1 | 10 | 19 | 23 |
| Maximum | 7 | - | 9.7 | - | 5 | 62 | 801 | 451 |
| Mean | 5 | - | 5.2 | - | 2 | 12 | 140 | 140 |
| St.Deviation | 1 | - | 2.7 | - | 1 | 9 | 125 | 161 |
| Number | 26 | - | 15 | - | 27 | 38 | 214 | 130 |
|  | Be |  | B |  | F |  | Cl |  |
| Minimum | 0.5 | - | 3 | 20 | 60 | 10 | 50 | - |
| Maximum | 205.0 | - | 150 | 26 | 2400 | 1000 | 175 | - |
| Mean | 7.5 | - | 21 | 21 | 546 | 337 | 81 | - |
| St.Deviation | 17.7 | - | 17 | 2 | 334 | 260 | 62 | - |
| Number | 143 | - | 114 | 16 | 188 | 16 | 4 | - |
|  | U |  | W |  | Sn |  | Mo |  |
| Minimum | 0.8 | 1.0 | 1 | 0.3 | 1 | 1 | 0.6 | 0.1 |
| Maximum | 35.0 | 3.1 | 34 | 104.0 | 52 | 930 | 10.0 | 0.5 |
| Mean | 6.8 | 2.2 | 5 | 16.8 | 12 | 30 | 1.6 | 0.2 |
| St.Deviation | 6.0 | 0.7 | 6 | 22.9 | 11 | 99 | 0.1 | 0.1 |
| Number | 197 | 9 | 73 | 27 | 271 | 87 | 144 | 16 |

N.S. $=$ Nova Scotia Mor. $=$ Morocco


Figure 3.7. Frequency distributions of major and trace element data from Nova Scotia and Morocco.Major oxided in wt\%, trace elements in ppm.


Figure 3.7 (cont.). Frequency distributions of major and trace element data from Nova Scotia and Morocco.


Figure 3.7 (cont.). Frequency distributions of major and trace element data from Nova Scotia and Morocco.

Nova Scotia-Morocco



Figure 3.8. Spider diagram comparing geochemical populations of Nova Scotia and Morocco. Each range represents the mean $\pm 1 \sigma$.

## Nova Scotia - Morocco



Figure 3.9. REE plots comparing geochemical populations of Nova Scotia (A) and Morocco (B).


## (Morocco, Nova Scotia)

| - Central Granites | $\circ$ | Jebilet Granites |
| :--- | :--- | :--- |
| - Rehamna Granites | + | Tichka Granites |

Figure 3.10. Plot comparing the initial Nd-Sr data from Morocco and Nova Scotia. See text for source references.
granites of Scotland (Hamilton et al., 1980) are included on Figure 3.10 for comparison.

Generally it can be said that the SMB is isotopically similar to the Central Massif of Morocco (particularly the Ment pluton), and to the French Hercynian granites. The SMB is definitely different from the Sierra Nevada Peninsular ranges granites. However without a more complete and representative database, similarity with the other granite groups cannot be ruled out on the basis of these isotopic values.

### 3.5 Summary

Comparison of the major, trace, rare earth and isotopic data using traditional methods of investigation has revealed that:

1- The geochemical populations of the northern and southern plutons of Nova Scotia, although apparently different, can not be clearly separated into two groups.

2- The Nova Scotian and Moroccan geochemical populations could not be separated and, in effect, appear indistinguishable.

3- In addition, limited Nd and Sr isotopic data also suggest some similarity between both areas.

## CHAPTER 4

## STATISTICAL INTERPRETATION OF GEOCHEMICAL DATA

### 4.1 Introduction

In petrology, petrogenetic interpretations of a geochemical data set are often made on the basis of correlations displayed within a subset of two or three of the variables. In recent years there has been a growing debate on the validity of some of these correlations, and the interpretations, because of the closed nature of geochemical data, i.e. each chemical analysis of a rock will sum to a constant $(100 \%)$. Such a data set is said to be closed, and as such this problem has commonly been refered to as the closure or constant sum problem.

The problems outlined above can be overcome by using multivariate statistical techniques. However, to realize the full potential of these new techniques, many geologists will have to overcome their fears of both numbers and statistics, neither of which were part of a traditionally descriptive science. General concepts of statistics, and descriptions of multivariate techniques are presented in this chapter. In addition various theoretical data sets were generated in order to assess discriminant function analysis, a multivariate technique used for the modelling of the peraluminous granite database. Results and discussion on the data processing follow in Chapter 5.

### 4.2 Nature of the Geochemical Data

### 4.2.1 Introduction

In this section the nature of the geochemical data is examined from a mathematical point of view. Some of the basic principles relating to descriptions of populations are also reviewed.

### 4.2.2 Principal Types of Variables

Geological observations are made on definite quantitative variables that can be continuous or discrete (non-continuous), and predetermined or random (Le Maitre, 1982; Guillaume, 1977). Mathematically the values of predetermined variables are predictable and can often be defined by a function (eg. $\mathrm{y}=\cos \mathrm{x}$ ). On the other hand, the variation of random variables is not predetermined, but can be described statistically (Guillaume, 1977). Discrete or noncontinuous variables can only have certain values, for example the number of plutons in a massif. Continuous variables can, however, have any value. Geochemical events are continuous because they have an infinite number of possible outcomes. Although the range of possible outcomes is actually finite and may, in fact, be limited, the exact results that may be obtained remain unpredictable (Davis, 1973) (e.g major element data which is constrained to lie between the limits of 0 and 100). Geochemical events are, therefore, continuous random variables which may be described statistically, even though they may appear to be discrete variables when they are rounded to a specific number of decimal places because of analytical precision (eg. ppm and $\mathrm{wt} \%$ ). Geochemical data are also dimensionless, as they are expressed in weight per unit weight or volume per unit volume (e.g. wt\% and ppm) (Le Maitre, 1982).

### 4.2.3 Specimen, Population and Sample

A specimen is the object on which the observations or measurements are made. A population is a set of measurements (not the objects) of a specific property made on a group of objects (Till, 1974). A population may be infinite or bounded. Geochemical populations can be considered infinite though they are in fact finite (Guillaume, 1977). For example, the alumina content of every feldspar in a granitic body can be thought of as an infinite population. Generally, it is not possible or practical to analyse every feldspar in a pluton, so we take
a small sub-set of the population, a sample, to study its properties. Then we need to determine if the sample was well selected, and if its properties are representative of the population (Guillaume, 1977). The reader will note that the term sample is used to represent a number of observations from a population rather than one object as it is commonly used in geology. The term specimen will be used to designate the geologist's sample.

### 4.2.4 Specimen Selection Procedure

Ideally, to avoid bias, a sampling campaign should be carried out randomly, so that a priori each individual of a population has the same chance of being selected (Le Maitre,1982). Descriptions and discussion on appropriate selection procedures are beyond the scope of this thesis, therefore, the reader is referred for details to Le Maitre (1982), Guillaume (1977), Till (1974), and Davis (1972). It is, however, important that the reader realize that bias could be introduced at various stages of selection procedure: target outcrop, specimen location on the outcrop, hand specimen for analysis, portion of the crushed rock, and portion of powder for analyses (Le Maitre, 1982).

Therefore, bias could be introduced in up to 5 distinct stages of sampling, again assuming that the intent of the selection procedure was to collect an unbiased population. Clearly, the nature of some studies results in sample bias, such as those which include more evolved members of a granitic suite because of their association to mineralization. As some of the data were drawn from studies such as these (e.g. Charest, 1976; Farley, 1979; Giuliani, 1982) it is expected that the database will show some bias. It is not unreasonable to expect that all populations (Morocco, Iberia and Nova Scotia) used in this study may have been treated in a similar fashion, suggesting that such a bias may not be a problem.

### 4.2.5 Distribution

As part of a geochemical study we may want to examine the variation of silica within a granitic body. This variation in the silica content is a characteristic of the body, and is the final result of all physico-chemical processes which have affected the pluton throughout its evolutionary history. When we make a series of measurements on different hand specimens from the pluton, we are in effect determining the distribution of silica within the granitic body.

A histogram can be produced in order to visualize the frequency distribution of a measured property. Many different forms have been described by statisticians; however, only normal and lognormal distributions are observed in geochemical data (Le Maitre, 1982) (Figure 4.1). It is assumed that the reader is familiar with the different parameters used to described distributions (i.e. mean, maximum, standard deviation, variance). For details the reader is refered to Davis, (1973); Guillaume, (1982); Till, (1979); Le Maitre, (1982).

Frequency distributions may have either a symmetrical or asymmetrical distribution about a central value (Figure 4.2). Asymmetrical distributions are either positively or negatively skewed when tailing of the distribution are to the right and left respectively of the maximum value.

Depending on the number of observed maxima, a distribution may also be unimodal, bimodal or even trimodal (Figure 4.2). Generally, statisticians regard multimodality as representing more than one population; however, this is not always the case in geology. For example, in the silica distribution of a zoned granitic pluton, a bimodal distribution in an otherwise determined comagmatic sequence may be regarded as one population.


Figure 4.1. (A) Normal and (B) lognormal distribution curves.


Figure 4.2. (A) Symmetrical (B) rightward skewed (C) leftward skewed (D) bimodal and ( E ) trimodal distributions.

Determining the distributions of all the variables is central to any study which attemps to utilize statistical techniques, as these are based on the normal distribution theory (Nie et al., 1975; Le Maitre, 1982). Transformations may be necessary if the data are not normally distributed. For example, trace element data which commonly display lognormal distributions may be made normal by effecting a $\log$ transformation, either as $z=\log (x)$ or $z=\log (x+w)$ where $w$ is a constant. Transformations as they apply to the geochemical data used in this study will be discussed further in Section 4.4.2.

### 4.2.6 The Closure Problem

One of the fundamental problems plaguing geochemical data is that of the constant sum or closure. Simply stated, closure occurs when the sum of all components (variables) measured during the analysis of an object is constant. In such a case an increase in one component would result in the decrease of at least one other. Moreover, in a closed data set of 4 components $\left(X^{(d+1)}\right)$, for example, only 3 components are necessary to determine the fourth, the remaining component being one minus the sum of the first three:

$$
\left(X^{d+1}=1-E X_{i}\right)
$$

Individual components in such a data vector (analysis) cannot vary independently.

Many authors have analysed and discussed the effects of closure on a data set Chayes (1962), Chayes and Krustal (1966), Connor and Mossiman (1969), Chayes (1971), Aitchinson (1981), Aitchinson (1984a, 1984b), Butler and Woronow (1986). Summarized below are some of the problems as outlined by these authors.

To illustrate the problem from a petrological point of view the following example is considered. In a plot of MgO and $\mathrm{SiO}_{2}$ values from a typical granitic pluton, the decrease in MgO with increasing $\mathrm{SiO}_{2}$ might be interpreted three ways according to Butler and Woronow (1986): 1) some process such as fractional crystallization of biotite may
account for the decrease in MgO and the increase in $\mathrm{SiO}_{2}$ in the residual liquid; 2) the negative correlation could be interpreted as purely a numeric response, in which a decrease in one component induces an increase in the other components, because of the constant sum restriction; 3) a combination of both one and two, with some physicochemical response and a numerical response being responsible for the variation.

The effects of closure are demonstrated using a randomlygenerated, open data set (or basis of Aitchinson (1986)), which was then closed by recalculating each row (analysis) to the sum of 100 (or composition of Aitchinson (1986)). Statistical descriptors (means, variances, and correlation coefficients) were also computed and are presented in Figure 4.3 (Data from Butler and Woronow, 1986).

The following observations can be made: The rank order for sample means are $\mathrm{B}<\mathrm{A}<\mathrm{C}$ in the basis and $\mathrm{B}<\mathrm{C}<\mathrm{A}$ in the composition; percentage formations have reversed the rank order of variables $A$ and $C$ showing the largest sample means; the rank order for the variances of the basis ( $\mathrm{A}<\mathrm{B}<\mathrm{C}$ ) and the composition ( $\mathrm{B}<\mathrm{C}<\mathrm{A}$ ) have also been modified.

Percentage formations will generally result in either expansion or contraction of the variability of the components when compared with their original form in the basis. Similarly, the skewness may be modified by percentage manipulations with, for example, positively skewed distributions made to be symmetrical or even negatively skewed (Chayes, 1972).

Correlation coefficients may also be adversely affected by percentage formation. For example, two basis components (variables) which were originally noncorrelated may become correlated in the composition. Generally, basis components with large variances will show a negative correlation for the composition, and basis components with small variances will have positive correlations. If a pair of variables in the basis is correlated (not independent) there will still

## RAW DATA

|  | A | B | C | Sum |
| :--- | :--- | :--- | :---: | ---: |
| 1 | 3.5 | 2.0 | 1.3 | 6.8 |
| 2 | 4.7 | 0.8 | 1.2 | 6.7 |
| 3 | 2.3 | 4.7 | 8.9 | 15.9 |


|  | A | B | C | Sum |
| :--- | :--- | :--- | :--- | :--- |
| 1 | 51.47 | 29.41 | 19.12 | 100.00 |
| 2 | 70.15 | 11.94 | 17.91 | 100.00 |
| 3 | 14.47 | 27.56 | 55.97 | 100.00 |

## BASIC STATISTICS

|  | A | B | C |
| :--- | :---: | :--- | :---: |
| Minimum | 2.3 | 0.8 | 1.2 |
| Maximum | 4.7 | 4.7 | 8.9 |
| Mean | 3.5 | 2.5 | 3.8 |
| Variance | 1.44 | 3.99 | 19.51 |


|  | A | B | C |
| :--- | :--- | :--- | :--- |
| Minimum | 14.47 | 11.94 | 17.91 |
| Maximum | 70.15 | 29.56 | 55.97 |
| Mean | 45.36 | 23.64 | 31.00 |
| Variance | 830.16 | 102.62 | 468.14 |

## CORRELATION COEFFICIENTS

|  | A | B |
| :--- | :--- | :--- |
| B | -0.976 |  |
| C | -0.872 | 0.957 |


|  | A | B |
| :--- | :--- | :--- |
| B | -0.762 |  |
| C | -0.953 | 0.530 |

Figure 4.3. Basic statistics and Pearson correlation coefficients for Basis and Composition (Data from Butler and Woronow, 1986).
be some change in the value of the correlation coefficients as a result of percentage formation. In the case of basis variables with large variances, the correlation coefficients are expected to change in the negative direction (i.e. towards the -1.0 value).

Correlation coefficients of Figure 4.3 show significant variations from basis to composition, with shifting occurring towards more negative values.

Aitchinson $(1981,1982)$ suggested two non-linear transformations which would essentially allow us to circumvent the closure problem

1) the $\log$ ratio transformation
where $\mathrm{Y}_{\mathrm{i}}=\log \left(\mathrm{X}_{\mathrm{i}} / \mathrm{X}_{\mathrm{j}}\right)$ with $\mathrm{i}=1,2 \ldots . \mathrm{D}$
2) the $\log$ centering transformation
where $\mathrm{Z}_{\mathrm{i}}=\log \left(\mathrm{X}_{\mathrm{i}}\right)-\mathrm{g}(\mathrm{X})$ with $\mathrm{i}=1,2 \ldots . . \mathrm{D}$
$(\mathrm{g}(\mathrm{X})$ the geometric mean is the sum of the logarithms of the components in an analysis divided by the number of components in the analysis.)

Both these transformations were applied to the basis and compositional data from Figure 4.3, and the results are given in Figure 4.4. The basis and composition for each separate transformation have matrices of equal value, suggesting that these transformations may be a solution in the analysis of compositions such as geochemical data.

In further analysis of these methods some drawbacks become apparent. For instance in the log centering tranformation each individual case will sum to the constant 0 , essentially recreating closure. The question is how useful is a transformation which eliminates closure to the constant (1) to create closure to the constant (0). Should another transformation be effected to eliminate

## BASIS

## COMLPOSITION

## LOG CENTER

|  | A | B | C |
| :--- | :--- | :--- | :--- |
| 1 | 0.517 | -0.043 | -0.474 |
| 2 | 1.045 | -0.725 | -0.320 |
| 3 | -0.689 | 0.025 | 0.664 |


|  | A | B | C |
| :--- | :--- | :--- | :--- |
| 1 | 0.517 | -0.043 | -0.474 |
| 2 | 1.045 | -0.725 | -0.320 |
| 3 | -0.689 | 0.025 | 0.664 |

LOG RATIO

|  | $\mathrm{A} / \mathrm{C}$ | $\mathrm{B} / \mathrm{C}$ |
| :---: | :---: | :---: |
| 1 | 0.430 | 0.187 |
| 2 | 0.593 | -0.176 |
| 3 | 0.949 | -0.277 |


|  | $\mathrm{A} / \mathrm{C}$ | $\mathrm{B} / \mathrm{C}$ |
| :---: | :---: | :---: |
| 1 | 0.430 | 0.187 |
| 2 | 0.593 | -0.176 |
| 3 | 0.949 | -0.277 |

Figure 4.4. Log center and log ratio transformation applied to the basis and compositional data of Figure 4.3.
the second closure problem (this is of course ridiculous), or is it one at all? This question remains unanswered in the literature.

The second method, the $\log$ ratio transformation also presents a problem, namely the fact that when we divide a component $\left(X_{i}\right)$ by another component $\left(X_{j}\right)$ variability in the $X_{i} / X_{j}$ ratio may occur. Let us consider the following example of two compositions:

| Component | A | B | C | SUM |
| :--- | :--- | :--- | :--- | ---: |
| Case 1 | .30 | .60 | .10 | 1 |
| Case 2 | .30 | .25 | .45 | 1 |

after transformation become:

| Component | $\mathrm{A} / \mathrm{B}$ | $\log \mathrm{A} / \mathrm{B}$ | $\log \mathrm{A} / \mathrm{C}$ |
| :--- | ---: | :---: | :---: |
| Case 1 | 0.5 | -0.693 | 1.098 |
| Case 2 | 1.2 | 0.182 | -0.405 |

Such a transformation is highly dependent on the denominator selected, and the question arises as to which denominator is the most appropriate, in order to approximate the interrelationships displayed in the original unknown basis. This question cannot yet be answered.

Although analysis of the data now appear possible using the log ratio and $\log$ centering transformation, it is still unclear whether or not such transformations represent a final solution to the closure problem. The problem of proving variable independence remains. A procedure to test for independence has been suggested by Aitchinson (1984). Although this procedure represents an interesting contribution towards solving the problem, much remains to be done, and further consideration was deemed beyond the scope of this thesis.

### 4.2.7 Graphical Representation of the Geochemical Data

In their studies of geochemical data, petrologists commonly resort to the use of graphical representations to portray the quantitative relationships within their data sets. This stems from the belief that graphical representations will promote better comprehension of the interrelationships between variables. However, these relationships may be adversely modified by the manipulations which we effect in our graphical representations. These distortions of the data may result in trends which are more numerical than petrogenetic in origin. The advantages and disadvantages of plotting geochemical data using binary and ternary diagrams are reviewed briefly in this section.

### 4.2.7.2 Binary Diagrams

Bivariate plots provide information on the interrelationships between only two components, and to rely on these partial analyses of a complete data set to make petrological interpretations may be misleading (Chayes, 1962; Baker, 1978; Butler, 1979; Aitchinson, 1984a, 1984b, 1984c). (See discussion below in Section 4.3.1).

Ratios in scatter diagrams are also commonly used in geology. In a study of such diagrams Skala (1979) found that in some circumstances two distinct populations may appear as a single highly correlated population by applying the appropriate ratios, and lead to possible misinterpretation of the data.

### 4.2.7.1 Ternary Diagrams

A common misconception in petrology is that ternary diagrams are more revealing than binary representations because they allow the user to illustrate the variation of more components in a given diagram. Such a diagram may be more misleading than useful because of the percentage transformation which we must effect in order to plot the data onto these triangular diagrams.

A great deal of variation in the means, variances and correlation coefficients is expected in the closing of the subset of 3 components used in a ternary plot. These changes in mean, variance and correlation values will be reflected in the distribution of sample points in a ternary plot (Skala, 1979; Butler, 1979). In order to ensure that no unreasonable amount of distortion is created as a result of plotting the data onto these diagrams, descriptors of variables (particularly correlation coefficients) before and after the transformations should accompany ternary diagrams when they are used to present the data. Perhaps instead of the ternary diagrams geologists should be moving towards three dimensional plots for these components, aided by computer graphics.

Suffice it to say that the solution appears to lie in the cautious application of the methods mentioned above in conjunction with multivariate statistical methods, as will be demonstrated below.

### 4.3 Multiple Discriminant Analysis

### 4.3.1 Introduction

Geologists have always preferred to use simple binary, ternary and tetrahedral diagrams to represent variations within their data set, no doubt related to the inability of the human eye to visualize more than three dimensions of space at one time. Commonly geologists will select from a compositon of major oxides certain variables such as $\mathrm{CaO}, \mathrm{Na}_{2} \mathrm{O}$, $\mathrm{K}_{2} \mathrm{O}$ and rescale to define a new subcomposition which can then be represented in a ternary CNK diagram (Aitchison, 1984a, 1984b, 1984c). The use of such subcompositional analyses has long been critized by geostatisticians (Chayes, 1962; Baker, 1978; Butler, 1979; Aitchison, 1984a, 1984b, 1984c) who contend that these partial analyses are subject to substantial loss of information which, in turn, may lead to serious misinterpretation of the data. Thus, a subcompositional analysis will retain only some of the variability displayed in the complete data set. If the goal during an analysis is to study the true
variability of the data set, then new methods such as multivariate analysis need to be adopted which will allow the user to effect more realistic and objective analysis of the data set.

Having determined that multivariate analysis is a useful and even an essential part of a geochemical study an appropriate multivariate technique must then be selected to satisfy the goals set for the analysis. For instance, principal component analysis would allow the user to visualize the variation within a data set, a pluton for example. On the other hand multiple discriminant analysis could be computed to investigate the variation between known groups of data, two plutons for example. Cluster analysis allows the user to classify a set of data into separate, relatively distinct, and homogeneous groups (or clusters). A classification does not depend on any prior knowledge of grouping as does discrimination. Thus, clustering defines an unpredetermined number of "natural" groups, which may then be assigned a geological meaning (if any). Discriminant analysis, however, will produce a number of linear discriminant functions, which are dependent on the number of defined groups. Discriminant function analysis will enable the user to quantify the differences between predetermined groups and may also be used to allocate new specimens of unknown origin to one of the initially defined groups.

The main objective of this study is to investigate the variation between granitic rocks of various geographic regions and, therefore, multiple discriminant analysis was selected as an appropriate technique to achieve this goal.

### 4.3.2 Geometric Interpretation

The geometric interpretation of discriminant analysis using two groups, A and B is presented in Figure 4.5. Probability contours for both groups illustrate their distribution in bivariate space (Le Maitre, 1982). A discriminant analysis in this case would produce a single linear function, or eigenvector, normal to the "hyperplane" with

## Discriminant function



Figure 4.5. Geometric interpretation of discriminant function analysis. Dark shaded areas represent the amount of overlap between both populations (light shaded areas). Note that the best separation is obtained when the populations are projected onto the discriminant function. (From Le Maitre, 1982).
the largest ratio of between-group to within group variance. In other words, the hyperplane is the orientation which best separates the two groups (largest between-group variance) while simultaneously ensuring that each group has the least inflation (to minimize within group variance) (Davis, 1973).

The dark shaded areas in Figure 4.5 represent the amount of overlap between the two groups and serves as an indication of the efficiency of the separation. The best separation is obtained when the original data are projected onto the discriminant function. If the data were projected in any other direction, such as $X_{1}$ or $X_{2}$ then the amount of overlap would increase and the discriminating power would decrease.

Multiple discriminant analysis is based on the assumption that all variables are normally distributed, and that all groups have equal dispersion (variance-covariance) matrices i.e. all groups would have probability contours of similar shape and orientation (Le Maitre, 1982). Multiple discriminant analysis is believed to be "robust" (Nie et al., 1975) with respect to the form of the distribution, and therefore if departures from normality are small, they may be ignored. This problem will be evaluated in Section 4.3.6. On the other hand, unequal dispersion matrices may be of some consequence, as they theoretically cause the surface which best separates the groups to become curved. This may be solved either by using quadratic discriminant functions, or by transforming the data so that all dispersion matrices become similar.

To assume that "simple" linear discriminant function(s) will best separate groups of multivariate data is only a first approximation. Given the limited understanding of both the geochemical data and quadratic discriminant functions any further transformation of the data and use of more "sophisticated" methods might be premature.

Additional information on Discriminant function analysis is given in Appendix F.

### 4.3.3 The Curse of High Dimensionality

Statistical books in geology commonly mention a problem in multivariate statistics which they refer to as the curse of high dimensionality, and is related to the probability of separation of groups which increases with the number of variates.

To illustrate the curse of high dimensionality in discriminant analysis an example from Foley (1971) is discussed in this section. Two data sets of 28 variables and 110 cases were randomly generated with the following criterion: Each class has an equal a priori probability of occurence, the variables are independent and have uniform probability distributions in the range of 0 to 1 .

In a first run 10 random specimens were randomly selected from each population (group) and the Fisher discriminant analysis was performed on the data. The data were projected onto the discriminant plane using the method described by Sammon (1970). The discriminant plane is, simply stated, an $x-y$ plot of the two orthogonal vectors which maximize the discrimination between the two groups under different constraints. The x axis of the optimal plane is the Fisher discriminant vector and the $y$ axis, while selected to insure maximum discrimination between the two groups, must also be orthogonal to the Fisher direction (Sammon, 1970). In Figure 4.6a a "perfect" separation between the two groups is evident. When all 110 samples are used in a second run (Figure 4.6b) discrimination between the two classes is not possible. This example suggests that if too few specimens are used in an analysis, then the discriminant function may tend to over separate groups (Foley, 1971; Howarth, 1983). This is the curse of high dimensionality.

## The Curse of High Dimensionality




Figure 4.6. Illustration of the curse of High dimensionality. (A) Shows a "perfect" separation between two groups of 10 specimens. (B) Shows that no separation is possible between the two groups when complete and representative populations are used in the analysis. (Note: Scales are different).

Foley (1971), suggests that the ratio of the number of cases (specimens) to the number of parameters (variables such as $\mathrm{SiO}_{2}, \mathrm{TiO}_{2}$ etc.) should be at least 3 to 1 . Most other workers suggest a ratio of 10 to 1 (Le Maitre, 1982; Guillaume 1977; Davis, 1973). If the number of cases is not sufficient, then the number of variables will have to be reduced, using only those variables which are the best discriminators.

### 4.3.4 Procedure for Discriminant Analysis

The application of discriminant function analysis begins with the selection of the variables to be used in the analysis. Ideally, all numerical parameters should be utilized to capture all of the variability in the data set. However, the number of elements determined in one study may differ from those in another investigation. Thus, a database formed from a collection of geochemical studies will show an uneven number of measured parameters. A general rule of discriminant analysis states that there should be at least ten times more case numbers than there are variables (Le Maitre, 1983). To meet this requirement, certain elements must be omitted from the analysis. Exactly which elements were used in the analysis will be discussed as the results of the analysis on the database are presented in Section 4.5.

The next step consists of determining the number of groups to be used in the analysis. Assigning group membership to each set of observations, must be done using a property (or criterion) which is "independent" of variables used in the computation of the discriminant function in order to avoid pre-judging the results (Le Maitre, 1982). For instance group membership may be assigned on the basis of geographic location, a variable not used in the subsequent calculations of discriminant functions for the geochemical data.

The statistical validity of the derived discriminant function(s) may be tested by spliting the group populations and using the first
data set (analysis sample) to calculate the discriminant function, and the second set (the holdout sample) to test the discriminant function. This is done because the classification test will be upwardly biased, that is groups will appear more distinct than in reality, when the same individuals are used for both computing the discriminant analysis and developing the classification matrix. However, once again we are faced with a dilemma of adequate sample size versus valid classification for the discriminant function. Comparison of classification matrices produced both ways, shows very little difference ( $1-3 \%$ ) between these two methods. As discussed in Section 4.3.3, the problem of high dimensionality (which tends to over separate small groups) is much more serious than a misclassification of a few percentage points. Therefore sample splits should only be done when the number of cases is sufficient, and done randomly to avoid introducing bias. Ideally, the procedure should be repeated several times, so that greater confidence can be placed on the validity of the function.

The discriminant function(s) may be computed two ways, with the direct (simultaneous) method or the stepwise method (Nie et al, 1975). The direct method derives the discriminant function(s) using the entire set of variables, regardless of the discriminating power of each. The stepwise method involves entering one variable at a time into the discriminant function(s). The variable with the greatest discriminating power will be selected at each step of the analysis. The discriminating power of the remaining variables is recalculated after each step. Some variables previously selected may be removed if the information they contained is available in a combination of other variables. The process will cease when the remaining variables (if any) no longer contribute significantly to the discriminating power of the function(s) (Nie et al. 1975).

It is difficult to determine which of the two methods is the best to use. Nie et al. (1975) suggest using the direct method to evaluate the discriminating power of the different variables and then employing the stepwise method with only the most discriminating variables. Other
statisticians warn users against attempting the stepwise method (Johnson, 1982). As neither method is universally accepted, and particularly given that some doubt exists about the applicability of the stepwise method, only the direct method will be used for computations. Discriminant function analysis was calculated using the statistical package SPSS (Version IX) by Nie et al. (1987).

Once the discriminant functions have been computed, their statistical significance can be evaluated using Chi-square $\chi^{2}$ (Nie et al, 1975). Unfortunately, this statistic is sometimes misleading. For instance, with large sample sizes, the $\chi^{2}$ values for various groups may be significantly different even though their group means (or centroids) are almost identical.

Classification matrices may also be used to evaluate the number of specimens correctly classified. As previously discussed, this may be done two ways either by re-running the data through the classification procedure or by using the holdout sample.

Classification of a specimen in discriminant analysis is based on the laws of probability. As such, the percentage of specimens that could be correctly classified by chance warrants some consideration. For example in a two-group analysis where the sample sizes are equal, the chance classification statistic (C) equals 1.0 divided by the number of groups. If the sample sizes were different then C would be based on the sample size of the largest group; this is known as the maximum chance criterion. In such a case, the discriminant function will defy the odds when it classifies a specimen into the smaller group (Morrison, 1969). Because populations used in this study are generally of unequal size, it is important to evaluate the impact of uneven populations on the discriminant models. This will be done in Section 4.3.7.

Comparison of classification results from two separate models could be done using the standard error:
$S_{e}=\sqrt{\frac{r / n^{*}(1-r / n)}{n}}$
where $\mathrm{r}=$ total number of correctly classified specimens (all groups) $\mathrm{n}=$ total number of specimens considered in the model

For example, the classification results from two different models can be compared by adding and subtracting their respective $2 \mathrm{~S}_{\mathrm{e}}$ values. If the results do not overlap than the two models can be qualified as significantly different. Again, the reader is reminded that because the same specimens were used in both the calculation of the discriminant function, the classification matrix results will be upwardly biased.

Once the discriminant functions are determined to be statistically significant, and the classification results are acceptable, then the interpretation of the results may begin. Usually the discriminant function will be analyzed to determine the relative importance of each of the variables. The problem of the independence of the compositional data remains an obstacle, as it casts some doubt as to the reliability of the coefficients when considered in isolation.

### 4.3.5 Discriminant Model

The effects of closure on a data set were discussed in some detail in Section 4.2 .6 and it was concluded that the analysis of the data now appeared possible with the use of the $\log$ ratio and $\log$ centering transformations. However, the problem of the independence of the variables still remains.

To examine the restrictions that such a problem places on the analysis and interpretation of discriminant functions, a theoretical, normally distributed data set of 3 variables was randomly generated using the normal density function:
$\mathrm{Y}=\frac{1}{\sigma \sqrt{2 \pi}} \exp \left[-(\mathrm{x}-\mu)^{2} / 2 \sigma^{2}\right]$ ) (Le Maitre, 1982)
The distribution of the variables from each group was controlled by preselecting their mean and standard deviation values.

The parameters for each variable and group are presented in Figure 4.7. The distribution of variable A from both groups was made to show some overlap, with different means, but identical variances. Variable B is identical for both groups, and C shows no overlap with different means and standard deviations.

The compositions for the data set were recalculated to $100 \%$. The parameters for each variable and group are presented in Figure 4.7.

Four separate runs of discriminant analysis were done on the basis, composition, $\log$ ratio and $\log$ centering transformations. The results of the discriminant function coefficients and classification matrices are presented in Figure 4.8.

As a whole, the variation within the data set is essentially maintained thoughout all of the transformations, the measured few percent misclassification are probably related to the inaccuracy of the technique. The problem occurs when we try to interpret the discriminant function coefficients. Usually the magnitude of the absolute values of each coefficient represents the discriminant power of that individual variable, in which the greater the value, the greater the discriminating power. However, if we compare the results of the basis with those results obtained from the $\log$ ratio and $\log$ centering transformations, it becomes very difficult to put any faith in the coefficients obtained as they differ significantly in magnitude (e.g. variable B , basis $=0.01422$, logcenter $=1.23041$, logratio $(B / A)=$ 1.51513).

In a study of distortion induced by closure of data, Skala (1977) found that although distortion of the variates will occur in a closed

## Basis Group 1

|  | A | B | C |
| :--- | :--- | :--- | :--- |
| Minimum | 1.01 | 0.60 | 2.03 |
| Maximum | 4.99 | 0.99 | 5.99 |
| Mean | 3.00 | 0.79 | 4.00 |
| St. Deviation | 0.93 | 0.09 | 0.93 |

B


## Basis Group 2

|  | A | B | C |
| :--- | :--- | :--- | :--- |
| Minimum | 4.01 | 0.60 | 8.00 |
| Maximum | 7.99 | 0.99 | 9.99 |
| Mean | 5.99 | 0.80 | 9.00 |
| St. Deviation | 0.93 | 0.09 | 0.46 |

B

| 0.05 | B |
| :---: | :---: |
| 0.03 | 0.10 |

## Composition Group 1

|  | $\dot{A}$ | B | C |
| :--- | ---: | ---: | ---: |
| Minimum | 13.78 | 5.51 | 27.38 |
| Maximum | 60.76 | 18.02 | 75.85 |
| Mean | 38.11 | 10.54 | 51.35 |
| St. Deviation | 8.69 | 2.28 | 8.41 |



## Composition Group 2

|  | A | B | C |
| :--- | ---: | :--- | ---: |
| Minimum | 28.64 | 3.64 | 48.89 |
| Maximum | 47.45 | 6.80 | 66.64 |
| Mean | 37.80 | 5.06 | 57.15 |
| St. Deviation | 3.81 | 0.63 | 3.59 |


|  |  | A |  |  |
| :---: | :---: | :---: | :---: | :---: |
|  | -0.42 | B |  |  |
|  | -0.27 |  |  |  |
|  | -0.99 | 0.27 |  |  |
|  |  |  |  |  |

Figure 4.7. Basic statistics and Pearson correlation coefficents for Model data.

Basis

| Actual <br> Group | No. of <br> Cases | Predicted Group |  |
| :---: | :---: | :---: | :---: |
|  |  | 1 | 2 |
| 1 | 312 | 312 | 0 |
|  |  | 100.0 | 0 |
| 2 | 312 | 0 | 312 |
|  |  | 0 | 100.0 |

Percent correctly classified $\quad 100.0$

| Discriminant function |  |  |
| :---: | :---: | :---: |
| Variable |  | Coefficient |
| A |  | 0.39714 |
| B |  | 0.01422 |
| C |  | 0.90129 |


| C Log center |
| :--- |
| Actual <br> Group No. of <br> Cases Predicted Group  <br>  1 2  <br> 1 312 300 <br> 96.2 12 <br> 3.8 <br> 2 312 0 312 <br> 0    |

Percent correctly classified $\quad \mathbf{9 8 . 0 8}$

| Discriminant function |  |  |
| :---: | :---: | :---: |
| Variable |  | Coefficient |
|  |  | 0.49925 |
| B |  | 1.23041 |

B

| Actual <br> Group | No. of <br> Cases | Predicted Group |  |
| :---: | :---: | :---: | :---: |
|  |  | 1 | 2 |
| 1 | 312 | 287 <br> 92.0 | 25 <br> 8.0 |
| 2 | 312 | 0 | 312 |
| 0 | 100.0 |  |  |

Percent correctly classified $\quad \mathbf{9 5 . 9 9}$

| Discriminant function |  |  |
| :---: | :---: | :---: |
| Variable |  | Coefficient |
| A |  | 0.29126 |
| B |  | 1.03776 |


| Actual <br> Group | No. of <br> Cases | Predicted Group |  |
| :---: | :---: | :---: | :---: |
|  |  | 1 | 2 |
| 1 | 312 | 300 <br> 96.2 | 12 <br> 3.8 |
| 2 | 312 | 0 <br> 0 | 312 |

Percent correctly classified $\quad 98.08$

| Discriminant function |  |  |
| :---: | :---: | :---: |
| Variable |  | Coefficient |
| B |  | 1.51513 |
| C |  | -1.21977 |

Figure 4.8. Results from discriminant function analyses done on (a) basis (b) composition (c) log center and (d) log ratio.
system, it is significantly reduced in a system of seven or more variables, to the point where distortion becomes negligible. In order to evaluate this idea another data set of 8 variables (A-H) and 1839 cases was generated in the same manner as discussed above. Parameters and discriminant analysis results are presented in Figures 4.9 and 4.10. In this example, discriminant coefficients for the transformed data better approximate values obtained during the analysis of the basis (e.g variable C basis $=.91423$ logcenter $=.91024 \operatorname{logratio}(\mathrm{C} / \mathrm{H})=$ 1.044417). Thus in a rock of essentially 90 chemical components, one would expect that the distortion would be minimal. Even though we do not measure all 90 components in our analysis those that we do measure constitute over $99 \%$ of the total volume of the rock, and the remaining space is not sufficient to induce any markable distortion. Therefore the discriminant coefficients will be interpreted in the analysis of the geochemical database.

### 4.3.6 The Granite Model

Discriminant function analysis is based on the normal distribution theory. Generally the technique is believed to be reasonably robust and will tolerate some departure from normality. The question is, how much will it tolerate?

Generally individual elements in the geochemical database (Appendix D) will show either normal, lognormal (skewed) or bimodal distributions. Therefore, it is of some interest to evaluate how these distributions will affect the discriminant function. In order to investigate these effects, a theoretical granite database was generated using the linear relationships between various major element oxides $\left(\mathrm{Al}_{2} \mathrm{O}_{3}, \mathrm{TiO}_{2}, \mathrm{Fe}_{2} \mathrm{O}_{3}, \mathrm{MnO}, \mathrm{MgO}, \mathrm{CaO}, \mathrm{Na}_{2} \mathrm{O}, \mathrm{K}_{2} \mathrm{O}\right)$ and $\mathrm{SiO}_{2}$. Their equations were determined by examining the relationship between the oxides and $\mathrm{SiO}_{2}$ in the peraluminous granites of Nova Scotia (Figure 4.11).

## Basis Group 1

|  | A | B | C | D | E | F | G | H |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| Minimum | 1.01 | 0.60 | 2.03 | 2.14 | 2.24 | 4.61 | 4.70 | 5.01 |
| Maximum | 4.96 | 1.00 | 5.98 | 6.08 | 6.18 | 8.58 | 8.68 | 8.98 |
| Mean | 3.00 | 0.80 | 4.00 | 4.10 | 4.20 | 6.60 | 6.70 | 7.00 |
| Variance | 0.66 | 0.01 | 0.66 | 0.66 | 0.66 | 0.66 | 0.66 | 0.66 |


| A |  | B | C | D | E | F |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| B | 0.016 |  |  |  |  |  |  |
| C | 0.021 | -0.080 |  |  |  |  |  |
| D | 0.034 | -0.017 | 0.208 |  |  |  |  |
| E | 0.061 | 0.002 | 0.174 | 0.194 |  |  |  |
| F | -0.030 | -0.007 | 0.013 | -0.037 | 0.034 |  |  |
| G | $-0.043$ | -0.039 | -0.009 | -0.048 | -0.056 | 0.313 | G |
| H | -0.096 | -0.068 | -0.046 | 0.004 | -0.041 | 0.037 | 0.040 |

## Basis Group 2

|  | A | B | C | D | E | F | G | H |
| :--- | :--- | :--- | ---: | :--- | :--- | :--- | :--- | :--- |
| Minimum | 4.01 | 0.60 | 8.01 | 2.14 | 2.24 | 4.61 | 4.70 | 5.01 |
| Maximum | 7.98 | 1.00 | 10.00 | 6.08 | 6.18 | 8.58 | 8.68 | 8.98 |
| Mean | 6.00 | 0.80 | 9.00 | 4.10 | 4.20 | 6.60 | 6.70 | 7.00 |
| Variance | 0.66 | 0.01 | 0.16 | 0.66 | 0.66 | 0.66 | 0.66 | 0.66 |


| A |  | B | C | D | E | F |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| B | 0.066 |  |  |  |  |  |  |
| C | 0.005 | -0.015 |  |  |  |  |  |
| D | -0.014 | 0.013 | $-0.018$ |  |  |  |  |
| E | -0.026 | -0.073 | -0.025 | -0.038 |  |  |  |
| F | 0.107 | -0.020 | 0.024 | -0.008 | 0.098 |  |  |
| G | 0.068 | -0.074 | 0.030 | -0.021 | 0.071 | 0.263 | G |
| H | 0.039 | -0.062 | 0.025 | 0.042 | -0.012 | 0.209 | 0.208 |

Figure 4.9. Basic statistics and pearson correlation coefficents for Model data.

## Composition Group 1

|  | A | B | C | D | E | F | G | H |
| :--- | ---: | ---: | ---: | ---: | ---: | ---: | ---: | ---: |
| Minimum | 2.47 | 1.40 | 5.86 | 6.06 | 5.89 | 12.27 | 12.40 | 12.27 |
| Maximum | 14.32 | 4.00 | 17.78 | 17.21 | 17.81 | 25.39 | 25.98 | 28.75 |
| Mean | 8.23 | 2.20 | 10.96 | 11.24 | 11.51 | 18.14 | 18.42 | 19.26 |
| Variance | 4.55 | 0.08 | 4.05 | 4.02 | 4.01 | 4.02 | 4.33 | 4.92 |



## Composition Group 2

|  | A | B | C | D | E | F | G | H |
| :--- | ---: | ---: | ---: | ---: | ---: | ---: | ---: | ---: |
| Minimum | 8.48 | 1.21 | 15.28 | 5.39 | 5.95 | 10.61 | 11.35 | 11.42 |
| Maximum | 18.52 | 2.61 | 25.14 | 14.60 | 15.01 | 19.20 | 20.68 | 21.19 |
| Mean | 13.18 | 1.76 | 19.82 | 9.44 | 10.10 | 14.93 | 15.15 | 15.59 |
| Variance | 2.66 | 0.04 | 1.52 | 2.88 | 2.79 | 2.19 | 2.27 | 2.38 |


| A |  | B | C | D | E |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| B | 0.058 |  |  |  |  |  |  |
| C | -0.087 | 0.324 |  |  |  |  |  |
| D | -0.168 | -0.012 | $-0.104$ |  |  |  |  |
| E | -0.202 | -0.098 | -0.133 | -0.178 |  |  |  |
| F | -0.170 | -0.098 | -0.208 | -0.253 | -0.141 | F' |  |
| G | -0.207 | -0.142 | -0.172 | $-0.253$ | -0.165 | $-0.056$ | G |
| H | -0.227 | -0.099 | $-0.124$ | -0.156 | -0.257 | -0.114 | -0.090 |

Figure 4.9 (cont.). Basic statistics and pearson correlation coefficents for Model data.

## Basis

## Composition

| Actual <br> Group | No. of <br> Case | Predicted Group |  |
| :---: | :---: | :---: | :---: |
|  |  | 1 | 2 |
| 1 | 918 | 915 | 3 |
|  |  | 99.7 | 0.3 |
| 2 | 918 | 0 | 918 |
|  |  | 0 | 100.0 |

Total percent correctly classified 99.84

| Variable |  | Coefficient |
| :---: | :---: | :---: |
| A |  | .49366 |
| B |  | -.28712 |
| C | .87173 |  |
| D | -.07001 |  |
| E | -.01771 |  |
| F | -.00746 |  |
| G | . .0209 |  |
| H |  |  |

Log Ratio *

| Actual <br> Group | No. of <br> Cases | Predicted Group |  |
| :---: | :---: | :---: | :---: |
|  |  | 918 | 898 <br> 98.5 |
| 1 |  | 14 |  |
| 2 | 918 | 0 <br> 0 | 913 |

Total percent correctly classified 99.23

| Variable |  | Coefficient |
| :---: | :---: | :---: |
| A |  | .63932 |
| B |  | -.12623 |
| C |  | .91024 |
| D | .02912 |  |
| E | .07042 |  |
| F | .01746 |  |
| G |  | $*$ |

Variable Coefficient

| A | .50495 |
| :--- | ---: |
| B | -.36970 |
| C | 1.04417 |
| D | -.22973 |
| E | -.17134 |
| F | -.16496 |
| G | -.15869 |

* Variable failed the tolerance test
\# Variables are Log(variable/H)
© Variables are Log centered
Figure 4.10. Results from Discriminant function analysis of basis, composition, log center and, log ratio of Model 8.


## Granite Model



Figure 4.11. Plot and equations of values used in the granite model of Section 4.3.6.

Normally distributed values for $\mathrm{SiO}_{2}$ were randomly generated using the normal density function. $\mathrm{SiO}_{2}$ values were then substituted in the equation of Figure 4.11 to determine the values of the other major element oxides. All values were then adjusted to a total of $100 \%$. In this way, a normally distributed comagmatic peraluminous granite database of 304 cases was produced.

The first test consisted of evaluating the effect of bimodal distribution on the discriminant analysis. Three separate bimodal data sets were produced from the original data, by randomly removing midcase values to produce bimodal distributions of 254,194 and 124 cases, thus increasing the amount of separation between the groups. The frequency distributions for $\mathrm{SiO}_{2}$ are presented in Figure $4.12 \mathrm{a}-\mathrm{c}$, and the other oxides, all of which are a function of $\mathrm{SiO}_{2}$, display similar distributions.

Three separate runs of discriminant analysis were done in which the original data set ( 304 cases) representing group 1 , was compared to each of the three bimodal sets. The data were transformed before discriminant function analysis to eliminate the closure problem by using $\mathrm{X}_{\mathrm{i}}=\ln \left(\mathrm{X}_{\mathrm{i}} / \mathrm{SiO}_{2}\right)$, and $\mathrm{SiO}_{2}$ was removed from the analysis. In each of these runs, the discriminant analysis could not discriminate between the three pairs of groups (of approximately the same mean and standard deviation).

A second test consisted of evaluating the effect of skewed distributions on the discriminant function. Two data sets were created by randomly removing 50 cases greater than and smaller than $70 \% \mathrm{SiO}_{2}$ from the original data set, thus producing positive and negative skewed distributions. The frequency distributions for $\mathrm{SiO}_{2}$ are presented in figure $4.12 \mathrm{~d}-\mathrm{e}$. The other dependent oxides, not shown, display similar distributions.

Two separate runs of discriminant analysis were done in which the original data set ( 304 cases) representing Group 1 was compared to each of the skewed sets. The data were transformed as discussed above.

## Granite model, bimodal and skewed distributions



Figure 4.12. Frequency distributions of Bimodal ( $\mathrm{A}, \mathrm{B}$, and C ) and skewed ( $D$ and $E$ ) populations used in the granite model.

Once again the discriminant analysis could not discriminate between both groups even though the mean and standard deviation of both groups are different.

These results are very encouraging. For example a bimodal distribution as might occur in a zoned granitic pluton, or a skewed distribution as might occur from sampling bias, can be correctly analysed by discriminant function analysis. Further tests showed that when the degree of skewness is increased to the point where it becomes similar to the lognormal form (i.e. mean values are significantly changed), results of the discriminant analysis are affected.

### 4.3.7 The Effect of Group Sample Size on the Maximum Chance Criterion

Differences in group size during discriminant analysis was identified as a potential problem in Section 4.3.4. The geochemical populations compared in this thesis are generally of unequal size. In order to validate these analysis a test model was run to determine the impact of uneven population sizes on the discriminant analysis.

A normaly distributed population of 2430 granite analysis was generated in the same manner as discussed in Section 4.3.6. From this 2430 case population (hereby known as Group 1) six separate groups of $1215,607,405,304,243$ and 162 cases were randomly taken. Each of these 6 populations (Group 2) were then compared with the original population (Group 1) using discriminant analysis. Ideally no separation between the groups should occur given that each of the separated populations were taken from Group 1 and therefore all groups should be identical. The results of the 6 runs are presented in Figure 4.13. The reader will note that significant deviations from expected results occur, in figure 4.13, when the difference in size between the two groups is equal to or greater than 10:1.

In reality two factors are being considered in this mode: The first is the effect of group size on the maximum chance criterion and a
second unescapable factor which is the representativity of the population within the sample group. In other words, are the randomly separated groups truly representative of the original population?

It can be argued that Model 6 (Figure 4.13) suggests that believable results can be obtained when the group size difference is $1: 15$. The deviations from expected values in Models 4 and 5 can be related to the probability of obtaining a representative sample group from the original population. When a sample group was taken from the original population a random number generator was used to determine whether or not a case (granite analysis) should be included in the sample group. If a number less than 0.5 was obtained by the random numbers generated the case was not included in the sample group if a value greater than 0.5 was obtained it was included. This process was repeated for each case until the target total number of cases was obtained. It is at this stage that some bias could have been introduced in the sample group causing the observed deviations in the results in the Models of Figure 4.17.

In sum it can be concluded that if the difference in group size is less than 10:1, no significant effect on the discriminant results should occur.

### 4.4 Summary

In this chapter it was demonstrated that multivariate analysis of compositional data could be done using the log ratio transformation to eliminate the constant-sum problem. The applicability of discriminant analysis to the geochemical data (granite analyses) was also demonstrated using various test models. Bimodal and skewed populations and, uneven sample groups (up to 10:1) can all be successfully analysed using discriminant function analysis. It was also concluded that the discriminant function coefficients were reliable, and that the classification matrix represented the best method of determining the efficiency of the separation during the analysis.

## Maximum Chance Criterion

| No. of cases in Group 1 | No. of cases in Group 2 | Group 2 as \% of Group 1 | Classification Matrix |  |  |  | Total percent correctly classified |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 2430 | 125 | $50 \%(1 / 2)$ | Actual | No. of | Predicte | Group | 51.85 |
|  |  |  | Group | Cases | 1 | 2 |  |
|  |  |  | 1 | 2430 | $\begin{gathered} 1336 \\ 55.0 \end{gathered}$ | $\begin{gathered} 1094 \\ 45.0 \end{gathered}$ |  |
|  |  |  | 2 | 1215 | $\begin{aligned} & 661 \\ & 54.4 \end{aligned}$ | ${ }_{\substack{554 \\ 45.6 \\ \hline}}$ |  |
| 2430 | 607 | $25 \%(1 / 4)$ | $\begin{aligned} & \text { Actual } \\ & \text { Group } \end{aligned}$ | $\begin{aligned} & \text { No. of } \\ & \text { Cases } \end{aligned}$ | $\begin{array}{\|c\|} \hline \text { Predicted Group } \\ \hline 1 \\ \hline \end{array}$ |  | 47.02 |
|  |  |  | 1 | 2430 | $\begin{gathered} 1081 \\ 44.5 \end{gathered}$ | $\underset{\substack{1349 \\ 5.5}}{ }$ |  |
|  |  |  | 2 | 607 | ${ }_{420}^{260}$ | 347 <br> 57.2 |  |
| 2430 | 405 | $17 \%(1 / 6)$ | $\begin{array}{\|l\|} \hline \begin{array}{l} \text { Actual } \\ \text { Group } \end{array} \end{array}$ | $\begin{aligned} & \mathrm{No} \text { o. of } \\ & \text { Cases } \end{aligned}$ | $$ |  | 42.12 |
|  |  |  | 1 | 2430 | $\begin{aligned} & 912 \\ & \hline 37.5 \\ & \hline \end{aligned}$ | $\begin{gathered} 1518 \\ 62.5 \\ \hline \end{gathered}$ |  |
|  |  |  | 2 | 405 | $\begin{gathered} 123 \\ 30.4 \\ \hline \end{gathered}$ | $\begin{gathered} 282 \\ 69.6 \\ \hline \end{gathered}$ |  |
| 2430 | 304 | $12 \%(1 / 8)$ | Actual <br> Group No. of <br> Cases <br>   | $\begin{gathered} \text { No. of } \\ \text { Cases } \end{gathered}$ | $\begin{gathered} \text { Predicted Groue } \\ \hline 1 \end{gathered}$ |  | 40.45 |
|  |  |  | 1 | 2430 | $\begin{gathered} 902 \\ 37.1 \end{gathered}$ | $\begin{gathered} 1528 \\ 62.9 \\ \hline \end{gathered}$ |  |
|  |  |  | 2 | 304 | $\begin{gathered} 100 \\ 32.9 \\ \hline \end{gathered}$ | $\underset{\substack{204 \\ 67.1 \\ \hline}}{ }$ |  |
| 2430 | 243 | $10 \%(1 / 10)$ |  | $\begin{array}{\|l} \begin{array}{l} \text { No. of } \\ \text { Cases } \end{array} \\ \hline \end{array}$ | Predicted Group |  | 39.13 |
|  |  |  |  |  |  |  |  |
|  |  |  | 1 | 2430 | $\begin{gathered} 869 \\ 35.8 \end{gathered}$ | $\begin{gathered} 1561 \\ 64.2 \\ \hline \end{gathered}$ |  |
|  |  |  | 2 | 243 | $\begin{aligned} & 66 \\ & 26 \end{aligned}$ | $\begin{gathered} 177 \\ 72.8 \\ \hline \end{gathered}$ |  |
| 2430 | 162 | $7 \%$ (1/15) | $\begin{aligned} & \text { Actual } \\ & \text { Group } \\ & \hline \end{aligned}$ | $\begin{array}{\|l\|} \hline \begin{array}{l} \text { No. of } \\ \text { Cases } \\ \hline \end{array} \\ \hline \end{array}$ | $\begin{gathered} \text { Predicted Group } \\ \hline 1 \quad 2 \\ \hline \end{gathered}$ |  | 47.15 |
|  |  |  |  |  |  |  |  |
|  |  |  | 1 | 2430 | $\begin{gathered} 1129 \\ 46.5 \end{gathered}$ | $\begin{gathered} 1301 \\ 53.5 \\ \hline \end{gathered}$ |  |
|  |  |  | 2 | 162 | $\begin{aligned} & 69 \\ & 42.6 \end{aligned}$ | $\begin{aligned} & 93 \\ & 57.4 \\ & \hline \end{aligned}$ |  |

Figure 4.13. Discriminant analysis model results testing the effect of group size on the maximum chance criterion.

## CHAPTER 5

## STATISTICAL MODELLING OF THE GEOCHEMICAL DATABASE

### 5.1 Introduction

Discriminant analyses of peraluminous granites from Nova Scotia, Morocco and Iberia are presented in this chapter. Data from Australia are also included in some of the statistical tests for comparative purposes. These analysis were undertaken in an attempt to answer the following questions:

1) Can a discriminant function be determined to distinguish clearly between the northern and southern plutons of Nova Scotia?
2) To what extent are the Nova Scotian and Moroccan granites similar?
3) Do the Nova Scotian granites show more geochemical affinity with the Moroccan or the Iberian granites?
4) Are variations measured locally (within Nova Scotia), regionally (between Nova Scotia and Morocco), within an orogenic belt (Nova Scotia, Morocco, Iberia) and between orogenic belts (when compared with Australia) characteristic of the scale of measurement?

Results on the processing of the geochemical data are presented in this chapter. Discussion on the results follow in Chapter 6.

### 5.2 Data Preparation

Before attempting multivariate analysis on the geochemical data certain steps need to be taken to ensure that such analysis can be applied to the data. These include both geological and mathematical considerations, all of which are summarized below. It can not be
overemphasized how important these considerations are to a proper, correct and believable analysis of the data.
A) Carefully define the scope of the problem to be resolved. We are all familiar with the phrase "garbage in, garbage out", and if data not applicable to the problem are included in the analyses, incorrect or misleading results may be obtained. In this case the intent of the analysis is to evaluate the degree of variation/similarity between granitic rock from different geographic areas, and it was decided that granitic rocks which showed clear evidence of secondary processes should not be included in the database, as such processes can not always be related to the granite in question. Their inclusion would add an undesirable uncertainty to the reliability of the results.

Another geological consideration is that only peraluminous granites are found within the Meguma Zone, therefore only peraluminous granites should be included in comparisons. They should also intrude Paleozoic rocks, as this is characteristic of the Meguma granites. In addition to the data from Nova Scotia and Morocco some data from Iberia were also used in certain analyses. Details on provenance are given in Appendix E. It should be pointed out that the Iberian granites are peraluminous, intrusive within Paleozoic terranes, and Hercynian in age. Details on Australian data are also given in Appendix E.
B) The distribution for each parameter used in statistical analysis must be examined to ensure that all the data are normally distributed. If the data are not normally distributed, then transformation may be necessary.

Transformation of the major element data are done to eliminate the constant sum problem. Two transformations may be applied: the log ratio or the $\log$ centering tranformation. As discussed in Chapter 4 $\log$ centering effectively recreates closure, thus the $\log$ ratio transformation is preferred. All the major element data used in the
discriminant analysis have been transformed using $\mathrm{Z}_{\mathrm{i}}=\ln \left(\mathrm{X}_{\mathrm{i}} / \mathrm{X}_{\mathrm{j}}\right)$ with $\mathrm{i}=1,2 \ldots \mathrm{D}$ where i does not equal j .

Trace element data are generally not lognormally distributed in the strictest sense of the word (Appendix E). However most minorelement distributions are far more similar to the lognormal than the normal. Although the statistical methods used in this study are supposedly robust with respect to the form of the distribution departures of most trace elements from normality are so large that the use of normal probability theory on untransformed data could be risky. It is preferable to effect $\log$ transformation before processing the data. Therefore trace element data were transformed using $\mathrm{Z}=\ln (\mathrm{x})$.

Although certain major element populations display bimodal distributions (Appendix B), they are still considered to represent only one population.
C) Critical review of results obtained during the analysis is essential. Statistical results must be supported by geological evidence to be acceptable and, should confirm what is already suspected and if unexpected results are obtained they should be scrutinized thoroughly.

### 5.3 Basic Assumptions

As with any attempt to model data, these statistical models depend on underlying assumptions. These assumptions are given below.

1- The Meguma granites were emplaced before the docking of the terrane onto eastern North America, and therefore should correlate with some other area (geological evidence; deformation of the granites along the suture)

2- The granites do not show any significant facies differences across the wide Scotian shelf.

3- Restrictions placed on which granites should be included in the database are correct (i.e. composition, age and association).

4- Analytical errors and interlaboratory differences are of no serious consequence (i.e. they do not make the data appear either more similar or more different).

5- Applied transformations to the geochemical data (log ratio for the majors and $\log$ for the traces) are correct.

6- Granite compositions have more to do with the source and the sum of processes which affect the granite throughout its evolution than with time itself.

### 5.4 Database Limitations

The geochemical database used in this study consists of both major and trace element analysis, mostly collected from bibliographic sources. Intuitively, trace elements are expected to be better discriminators than major elements. This is due to the fact that the granitic classification restricts major element variation, but not trace elements.

The frequency of measurements for each parameter in the database are shown in Figure 5.1. Ideally, all available variates should be used in a discriminant analysis, however, for many trace elements, the number of cases is simply not sufficient to ensure that the ratio of the number of measurements to variates is at least $10: 1$. This, of course, means that possibly some of the better discrimators (trace elements) can not be included because of database restrictions.


Figure 5.1. Frequency distribution of measured parameters for each granite population used in the geochemical models.
5.5 Northern and Southern Plutons of Nova Scotia

### 5.5.1 Introduction

Three separate runs of discriminant analysis on the northern and southern plutons of Nova Scotia are presented in this Section. These were selected as the most informative from a considerable number analyses done during this study. The purpose behind these analyses was to determine whether the northern and southern plutons could be successfully distinguished. The reader will recall that previous analysis presented in Chapter 3 suggested that some differences existed between both areas, however, no binary or ternary diagram could be succesfully developed to distinguish clearly geochemically between both areas.

### 5.5.2 Model 1 (North-South, Nova Scotia)

5.5.2.1 Results of Model 1 (North-South, N.S.)

A first model was run using $\mathrm{SiO}_{2}-\mathrm{TiO}_{2}-\mathrm{Al}_{2} \mathrm{O}_{3}-\mathrm{Fe}_{2} \mathrm{O}_{3} t-\mathrm{MgO}-$ $\mathrm{CaO}-\mathrm{Na}_{2} \mathrm{O}-\mathrm{K}_{2} \mathrm{O}-\mathrm{P}_{2} \mathrm{O}_{5}-\mathrm{Ba}-\mathrm{Rb}-$ Sr. Variables were selected by examining the frequency of measurements for each parameter in Figure 5.1. Making certain that the ratio of variates to number of measurements was at least $10: 1$. The reader is reminded that all major oxides have been normalized to $\mathrm{SiO}_{2}$ (oxide $/ \mathrm{SiO}_{2}$ ).

Figure 5.2 illustrates the frequency distribution of the discriminant scores for the two groups. The discriminant function and classification matrix are presented in Figure 5.3. A detailed listing of the classification matrix for each pluton is also presented in Figure 5.3.
$\mathrm{Rb}, \mathrm{Ba}, \mathrm{CaO}, \mathrm{Sr}$, and $\mathrm{TiO}_{2}$ with absolute discriminant coefficients greater than 0.4 are considered important discriminators. $\mathrm{K}_{2} \mathrm{O}, \mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ and $\mathrm{P}_{2} \mathrm{O}_{5}$ with coefficient values between 0.4 and 0.3 are in this

## Model 1 - Nova Scotia (North-South)

North


South

$\square$ Northern plutons $\square$ Misclassified

Figure 5.2. Frequency plot of discriminant scores from Model 1.

## Model 1 - Nova Scotia (North-South) Detailed Classification Matrix

| $\underbrace{C}_{B}$ |  | GRANITE | $\begin{aligned} & \text { CORRECTLY } \\ & \text { CLASSIFIED } \end{aligned}$ |  | MISCLASSIFIED |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | SMB Charest | 17 | 100.0\% | 0 | 0.0\% |
|  |  | SMB McKenzie | 16 | 88.9\% | 2 | 11.1\% |
|  |  | SMB Smith | 32 | 100.0\% | 0 | 0.0\% |
|  |  | SMB Richardson | 8 | 100.0\% | 0 | 0.0\% |
|  |  | Musquodoboit | 9 | 90.0\% | 1 | 10.0\% |
|  |  | Liscomb | 8 | 88.9\% | 1 | 11.1\% |
|  |  | Sherbrooke | 17 | 94.4\% | 1 | 5.6\% |
|  |  | Bull Ridge | 10 | 83.3\% | 2 | 16.7\% |
|  |  | Sangster Lake \& Larry's River | 29 | 96.7\% | 1 | 3.3\% |
|  |  | Queensport \& Halfway Cove | 36 | 100.0\% | 0 | 0.0\% |
|  |  | Ellison Lake | 13 | 92.9\% | 1 | 7.1\% |
|  |  | Barrington Passage | 31 | 100.0\% | 0 | 0.0\% |
| 'Z |  | Shelburne | 22 | 81.5\% | 5 | 18.5\% |
|  |  | Bald Mountain | 4 | 100.0\% | 0 | 0.0\% |
|  | 5 | Port Mouton | 50 | 96.1\% | 2 | 3.9\% |
|  | $\hat{O}$ | Moose Point <br> Lyons Bay | 6 | 75.0\% | 2 | 25.0\% |
|  |  | Seal Island Western Granite | 6 | 60.0\% | 4 | 40.0\% |

Discriminant Function

| Rb | 0.94774 |
| :--- | ---: |
| Ba | -0.69987 |
| CaO | -0.62054 |
| Sr | 0.50335 |
| $\mathrm{TiO}_{2}$ | 0.50259 |
| $\mathrm{~K}_{2} \mathrm{O}$ | 0.39666 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3 \mathrm{t}}$ | 0.31488 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | -0.30121 |
| $\mathrm{MgO}^{2}$ | 0.11960 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 0.03353 |

Classification Matrix


Total Correct 93.47\%

Figure 5.3. Discriminant function, detailed and summary classification matrix for Model 1.
function of lesser importance. The reader will note that $\mathrm{Al}_{2} \mathrm{O}_{3}$ is not found in the discriminant function, i.e. $\mathrm{Al}_{2} \mathrm{O}_{3}$ failed the minimum tolerance test and was excluded from the analysis (for more information on the minimum tolerance test see Appendix F).

The classification matrix of Figure 5.3 indicates that $93.47 \%$ of all cases were correctly classified during the analyses. This suggests that the northern and southern plutons of Nova Scotia are in effect geochemically different.

### 5.5.2.2 Discussion of Model 1 (North-South, N.S.)

Analysis of the northern and southern granites of Nova Scotia was undertaken in an attempt to characterise variation within this area. A dividing line (Figure 2.1) was drawn based on geological evidence which indicated some difference in the evolution of these granites (see Chapter 2 for detailed discussion). Group membership for the Nova Scotian Models (including subsequent Models 2 and 3) were determined based on this division. As a result of these model contraints the following conclusions can be drawn from this analyses. The northern and southern plutons are in effect statistically distinct, and membership to each group can be defined by the discriminant function. Based on this equation (discriminant function) a new chemical analysis from the Nova Scotian granites could be correctly classified $93.47 \%$ of the time, i.e. its provenance from either the northern or southern plutons could be correctly predicted in $93.47 \%$ cases. The standard error $\left(\mathrm{S}_{\mathrm{e}}\right)$ for Model 1 is $1.35 \%$.

The discriminant scores of individual data vectors were examined to investigate which analyses were not correctly classified, particularly to decipher any existing misclassification patterns (Figure 5.3). Generally, the more "mafic" members of the northern plutons (eg. granodiorites from the South Mountain Batholith and Liscomb Pluton) as well as the more "felsic" members of the southern plutons (monzogranites from Moose point, Shelburne, Lyons Bay etc...)
are misclassified by the analyses. This observation is difficult to interpret. It may simply reflect that given the restricted number and the nature of variates used in the analyses, differences relating to the degree of mafic or felsic component are being emphasised. Another possible explanation is that the extreme compositions within both groups are simply not that different. Alternatively perhaps no meaning should be read into these misclassified cases, given the very low percentage of misclassified samples.

### 5.5.3 Model 2 (North-South, Nova Scotia)

### 5.5.3.1 Results of Model 2 (North-South, N.S.)

Potentially, with the addition of more trace elements, discrimination between the two groups should improve, therefore, to better constrain the modelling, some trace elements were added in the following analyses. However, because of the limitations imposed by the number of measurements, some parameters previously used in Model 1 had to be omitted from further analysis. Variates retained in the analyses were selected on the basis of their discriminating power in the function of Model 1.

A second run (Model 2) was done using $\mathrm{SiO}_{2}-\mathrm{TiO}_{2}-\mathrm{CaO}-\mathrm{K}_{2} \mathrm{O}-\mathrm{Ba}$ $-\mathrm{Rb}-\mathrm{Zr}-\mathrm{Zn}-\mathrm{Th}$ (again major oxides were normalized to $\mathrm{SiO}_{2}$ ).
Figure 5.4 illustrates the frequency distribution of the discriminant scores for each group. The discriminant function and classification matrix are presented in Figure 5.5. A detailed listing of the classification matrix for each pluton is also shown in Figure 5.5.
$\mathrm{Rb}, \mathrm{TiO}_{2}, \mathrm{Ba}, \mathrm{Sr}$, are important discriminators in this function, and $\mathrm{Th}, \mathrm{Zr}, \mathrm{CaO}, \mathrm{K}_{2} \mathrm{O}$ and Zn appear to be of lesser significance. The classification matrix of Figure 5.5 indicates that $91.51 \%$ of all cases were correctly classified during the analysis with a $\mathrm{S}_{\mathrm{e}}=1.91 \%$. Once again the northern and southern plutons appear to be geochemically different.

## Model 2 - Nova Scotia (North-South)

North

South

$\square$ Northern plutons $\square$ Southern plutons
$\square$ Misclassified

Figure 5.4. Frequency plot of discriminant scores from Model 2.

## Model 2 - Nova Scotia (North-South)

Detailed Classification Matrix

| $\frac{4}{8}$ | $\begin{gathered} \frac{G}{2} \\ \frac{0}{0} \\ \frac{2}{2} \end{gathered}$ | GRANITE | $\begin{aligned} & \text { CORRECTLY } \\ & \text { CLASSIFIED } \end{aligned}$ |  | MISCLASSIFIED |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | SMB Charest | - | - \% | - | - \% |
|  |  | SMB McKenzie | 3 | 37.5\% | 5 | 62.5\% |
|  |  | SMB Smith | - | - \% | - | - \% |
|  |  | SMB Richardson | 6 | 100.0\% | 0 | 0.0\% |
|  |  | Musquodoboit | 9 | 100.0\% | 0 | 0.0\% |
|  |  | Liscomb | 5 | 62.5\% | 3 | 37.5\% |
|  |  | Sherbrooke | 13 | 100.0\% | 0 | 0.0\% |
|  |  | Bull Ridge | - | - \% | - | - \% |
|  |  | Sangster Lake \& Larry's River | 27 | 100.0\% | 0 | 0.0\% |
|  |  | Queensport \& Halfway Cove | 36 | 100.0\% | 0 | 0.0\% |
|  |  | Ellison Lake | . | - \% | - | - \% |
|  |  | Barrington Passage | 26 | 0.0\% | 0 | 0.0\% |
|  |  | Shelburne | 22 | 95.7\% | 1 | 4.3\% |
|  |  | Bald Mountain | 0 | 0.0\% | 3 | 100.0\% |
|  | E | Port Mouton | 38 | 88.4\% | 5 | 11.6\% |
|  | \%) | Moose Point <br> Lyons Bay | - | - \% |  | - \% |
|  |  | Seal Island <br> Western Granite | 9 | 90.0\% | 1 | 10.0\% |

## Discriminant Function

| Rb | 0.90684 |
| :--- | ---: |
| TiO | 0.79808 |
| Ba | -0.61060 |
| Sr | 0.57276 |
| Th | 0.32050 |
| Zr | -0.31923 |
| CaO | -0.28093 |
| $\mathrm{~K}_{2} \mathrm{O}$ | 0.26254 |
| Zn | -0.22088 |

Classification Matrix

|  |  |  | Predicted Group |  |
| :---: | :---: | :---: | :---: | :---: |
|  |  |  | North | South |
|  |  | 107 | 99 | 8 |
|  |  | 92.5\% | 7.5\% |
| $\frac{3}{4}$ |  |  | 105 | 10 | 95 |
|  |  | 9.5\% |  | 90.5\% |

Figure 5.5. Discriminant function, detailed and summary classification matrix for Model 2.
5.5.3.2 Discussion of Model 2 (North-South, N.S.)

Obviously the addition of new trace elements in this model have not improved the degree of separation between the two groups (from $93.47 \% \pm 2.70$ (at $2 \mathrm{~S}_{\mathrm{e}}$ ) in Model 1 to $91.51 \% \pm 3.82$ (at $2 \mathrm{~S}_{\mathrm{e}}$ ) in this model). These results overlap considerably and cannot be considered statisticaly different. It is difficult to predict whether the inclusion of the major elements used in Model 1 would contribute measurably to the discrimination, and this question can not be answered until the number of variates and cases are increased in the database.

Examination of misclassified samples in Figure 5.5 reveal that extreme compositions (i.e. granodiorites of SMB from McKenzie and the Liscomb pluton, and monzogranites from the Port Mouton pluton) are still being misclassified. All three samples from the Bald Mountain pluton are also being misclassified during the analyses, all of these are monzogranites. Again it is difficult to assess the significance of the misclassifications, as so few cases are being misclassified.

The reader will note the limited amount of data used from the South Mountain Batholith, and especially the decrease in the overall number of cases as well as number of plutons used in the analyses. This may result in a decrease in the variation between populations and account for the sligthly lower discrimination.

### 5.5.4 Model 3 (North-South, Nova Scotia)

### 5.5.4.1 Results of Model 3 (North-South, N.S.)

In a third run of the Nova Scotian granites Pb was included in the list of variables, in addition some of the better discriminators of Models 1 and 2 were also selected for the analyses, therefore Model 3 includes the following variables: $\mathrm{SiO}_{2}-\mathrm{TiO}_{2}-\mathrm{CaO}-\mathrm{K}_{2} \mathrm{O}-\mathrm{Ba}-\mathrm{Rb}-$ $\mathrm{Sr}-\mathrm{Zr}-\mathrm{Zn}-\mathrm{Th}-\mathrm{Pb}$.

## Model 3 - Nova Scotia (North-South)

North

South

$\square$ Northern plutons ए画 Southern plutons
$\square$ Misclassified

Figure 5.6. Frequency plot of discriminant scores from Model 3.

## Model 3 - Nova Scotia (North-South)

## Detailed Classification Matrix



Discriminant Function
Classification Matrix

| Sr | 1.32740 |
| :--- | ---: |
| Rb | 1.24913 |
| Ba | -1.02254 |
| Pb | 0.63290 |
| $\mathrm{TiO}_{2}$ | 0.59358 |
| CaO | -0.41234 |
| Th | 0.27591 |
| Zr | -0.20109 |
| Zn | -0.16912 |
| $\mathrm{~K}_{2} \mathrm{O}$ | -0.07839 |



Figure 5.7. Discriminant function, detailed and summary classification matrix for Model 3.

The frequency distribution of the discriminant scores for each group is shown in Figure 5.6. The discriminant function and classification matrix are presented in Figure 5.7. A detailed listing of the classification matrix for each pluton is also presented in Figure 5.7.
$\mathrm{Sr}, \mathrm{Rb}, \mathrm{Ba}, \mathrm{Pb}, \mathrm{TiO}_{2}, \mathrm{CaO}$ with coefficient values greater than 0.4 are considered important discriminators. $\mathrm{Th}, \mathrm{Zr}, \mathrm{Zn}$ and $\mathrm{K}_{2} \mathrm{O}$ contribute to a lesser degree. The classification matrix of Figure 5.7 indicates that $96.41 \%$ (with a $\mathrm{S}_{\mathrm{e}}$ of $1.33 \%$ ) of all cases were correctly classified during the analysis, and once again confirms the geochemical distinction between the northern and southern plutons.

### 5.5.4.2 Discussion of Model 3 (North-South, N.S.)

Model 3 with a number of correctly classified cases at $96.41 \% \pm$ 2.66 (at $2 \mathrm{~S}_{\mathrm{e}}$ ), represents the best separation obtained for the Nova Scotian granites in this study and with such a high value the Meguma granites can be characterized as geochemically distinct. Upon further scrutiny of the results some weaknesses become apparent in the Model. For instance, the analyses includes very little data from the SMB and none from the Musquodoboit batholith, as they account for a high proportion of the granite outcrop within Nova Scotia perhaps, their exclusion may have lead to slight modification in the results. This would occur if the absence of data from these two important granitic bodies induced some changes in the characteristics of the northern populations, and these distortions may account for the few percent differences with previous models.

Generally similar observations presented for previous Nova Scotian Models appear to apply here. Some granodioritic samples from the Liscomb pluton are still being misclassified.

### 5.5.5 Discussion of the Nova Scotian Models

The following observations can be made about the three models presented in this section:

1) The northern and southern plutons of Nova scotia can be successfully distinguished geochemicaly, even with the limited number of variates used in the models.
2) The few percent misclassified samples seem to be of extreme composition. That is, the more mafic members of the northern plutons and the more felsic members of the southern plutons are being misclassified. This may be the result of the nature and number of variates used in the models. This may also indicate that the intermediate compositions are similar for both groups. Alternatively because of their small numbers, their misclassification may simply mean nothing.
3) The importance of an element in the discriminant function (i.e. the value of the coefficient) may vary slightly between models. Even though many different functions can be defined to effectively separate the two groups, some elements are consistently excellent discriminators in all of the models considered in this study. This is illustrated in Figure 5.8 where generally $\mathrm{Rb}, \mathrm{Sr}, \mathrm{Ba}, \mathrm{TiO}_{2}, \mathrm{CaO}$ and Pb are considered the most significant discriminators in Nova Scotia.

### 5.6 Nova Scotia and Morocco

### 5.6.1 Introduction

Three separate runs of discriminant analysis on the granites of Nova Scotia (north and south are combined) and Morocco are presented in this Section. As in the previous section these were selected as the most informative from a number of analysis carried out during the course of this study. These analysis were done in order to examine the

## Nova Scotia (North-South) Models

| Absolute Coefficient Value | Model <br> 1 | Model <br> 2 | Model <br> 3 |
| :---: | :---: | :---: | :---: |
| C $>0.8$ | Rb | Rb | Sr <br> Rb <br> Ba |
| $0.8>\mathrm{C}>0.6$ | $\begin{aligned} & \mathrm{Ba} \\ & \mathrm{CaO} \end{aligned}$ | $\begin{aligned} & \mathrm{TiO}_{2} \\ & \mathrm{Ba} \end{aligned}$ | Pb |
| $0.6>\mathrm{C}>0.4$ | $\begin{aligned} & \mathrm{Sr} \\ & \mathrm{TiO}_{2} \end{aligned}$ | Sr | $\begin{aligned} & \mathrm{TiO}_{2} \\ & \mathrm{CaO} \end{aligned}$ |
| $0.4>\mathrm{C}>0.2$ | $\begin{aligned} & \mathrm{K}_{2} \mathrm{O} \\ & \mathrm{Fe}_{2} \mathrm{O}_{3 \mathrm{t}} \\ & \mathrm{P}_{2} \mathrm{O}_{5} \end{aligned}$ | $\begin{array}{ll} \mathrm{Th} & \mathrm{~K}_{2} \mathrm{O} \\ \mathrm{Zr} & \mathrm{Zn} \\ \mathrm{CaO} \end{array}$ | $\begin{aligned} & \mathrm{Th} \\ & \mathrm{Zr} \end{aligned}$ |
| $0.2>\mathrm{C}>0.0$ | $\begin{aligned} & \mathrm{MgO} \\ & \mathrm{Na}_{2} \mathrm{O} \end{aligned}$ |  | $\begin{aligned} & \mathrm{Zn} \\ & \mathrm{~K}_{2} \mathrm{O} \end{aligned}$ |

Figure 5.8. Generally $\mathrm{Rb}, \mathrm{Sr}, \mathrm{Ba}, \mathrm{TiO}_{2}, \mathrm{CaO}$, and Pb are the most significant discriminators within Nova Scotia.
geochemical relationship between the Nova Scotian and Moroccan granites.

### 5.6.2 Model 4 (Nova Scotia - Morocco)

### 5.6.2.1 Results of Model 4 (Nova Scotia-Morocco)

Model 4 was run using $\mathrm{SiO}_{2}-\mathrm{TiO}_{2}-\mathrm{Al}_{2} \mathrm{O}_{3}-\mathrm{Fe}_{2} \mathrm{O}_{3 \mathrm{t}}-\mathrm{MgO}-\mathrm{CaO}-$ $\mathrm{Na}_{2} \mathrm{O}-\mathrm{K}_{2} \mathrm{O}-\mathrm{Ba}-\mathrm{Rb}-\mathrm{Sr}$. Figure 5.9 illustrates the frequency distribution of the discriminant scores for each group. The discriminant function and classification matrix are presented in Figure 5.10. A detailed listing of the classification matrix for each pluton is also presented in Figure 5.10.
$\mathrm{Sr}, \mathrm{MgO}, \mathrm{Rb}, \mathrm{CaO}, \mathrm{Na}_{2} \mathrm{O}, \mathrm{Al}_{2} \mathrm{O}_{3}$ and Ba with coefficient values greater than 0.4 are important discriminators; $\mathrm{K}_{2} \mathrm{O}, \mathrm{TiO}_{2}$ and $\mathrm{Fe}_{2} \mathrm{O}_{3}$ contribute to a lesser degree. With a classification value of $80.84 \%$ ( $\mathrm{S}_{\mathrm{e}}=1.72 \%$ ), Nova Scotia and Morocco can not be characterized as geochemically distinct, such a value indicates some similarity between both areas.

### 5.6.2.2 Discussion of Model 4 (Nova Scotia - Morocco)

Variates which were considered important discriminators within Nova Scotia (i.e. locally), particularly $\mathrm{TiO}_{2}$ and, to a lesser extent Ba , are no longer important regionally (between Nova Scotia and Morocco). Conversely, good regional discriminators such as $\mathrm{MgO}, \mathrm{Na}_{2} \mathrm{O}$ and $\mathrm{Al}_{2} \mathrm{O}_{3}$ are of little consequence locally.

The classification matrix shows that $80.84 \%$ of all cases were correctly classified during the analyses, indicating a certain amount of similarity between both areas. The detailed listing of the classification matrix in Figure 5.10 reveals a great deal of similarity between the Central and Rehamna Massifs of Morocco, and Nova Scotia (a considerable number of cases from both massifs are misclassified).

## Model 4 - Nova Scotia - Morocco




| Nova Scotia $\quad \square$ Morocco |  |
| :--- | :--- |
| $\square$ | Misclassified |

Figure 5.9. Frequency plot of discriminant scores from Model 4.

## Model 4 - Nova Scotia - Morocco <br> Detailed Classification Matrix



Discriminant Function Classification Matrix

| Sr | 1.23845 |
| :--- | ---: |
| MgO | -1.06681 |
| Rb | 0.75486 |
| CaO | 0.74029 |
| $\mathrm{Na}_{2} \mathrm{O}$ | -0.70413 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 0.47063 |
| Ba | -0.44260 |
| $\mathrm{~K}_{2} \mathrm{O}$ | 0.16886 |
| $\mathrm{TiO}_{2}$ | -0.15365 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3 t}$ | 0.14997 |


|  |  |  | Predicted Group |  |
| :---: | :---: | :---: | :---: | :---: |
|  |  |  | N.S | Morocco |
|  | 安 | 337 | $\begin{aligned} & 283 \\ & 84.0 \% \end{aligned}$ | 54 <br> $16.0 \%$ |
|  | 苞 | 185 | $\begin{aligned} & 46 \\ & 24.9 \% \end{aligned}$ | $\begin{aligned} & 139 \\ & 75.1 \% \end{aligned}$ |

Total Correctly Classified $80.84 \%$

Figure 5.10. Discriminant function, detailed and summary classification matrix for Model 4.

Misclassification of samples is also occuring within Nova Scotia. Particularly from the Ellison lake pluton of southern Nova Scotia with $100 \%$ being misclassified into the Moroccan group. In addition samples from the Sherbrooke pluton (north), and the Moose Point pluton (south) show remarkable similarity with the Moroccan population.

All samples from the Jebilet Massif of Morocco are correctly classified in the model suggesting that the Jebilet granites are geochemically distinct from the Nova Scotian granites. The misclassification of the Rehamna granites is interesting. Geological evidence seems to suggest (see Chapter 2) that of all the Moroccan granites they appear to be the least similar to Nova Scotia. In particular most characteristic peraluminous minerals are absent in the plutons, and their Permian age represents the widest age gap with the Nova Scotian granites (this may support the assumption that age is of minimal consequence). It is difficult to attribute a meaning to their misclassification because of the limited number of variates used in the calculation. This problem will be readressed later in this section as more elements are added to subsequent Models.

### 5.6.3 Model 5 (Nova Scotia - Morocco)

### 5.6.3.1 Results of Model 5 (Nova Scotia - Morocco)

In a second run of the Nova Scotian and Moroccan granites (Model 5) $\mathrm{P}_{2} \mathrm{O}_{5}$ was added to the list of variables from Model 4. Figure 5.11 illustrates the frequency distribution of the discriminant scores for the two groups. The discriminant function and classification matrix are presented in Table 5.12. A detailed listing of the classification matrix for each pluton is also presented in Figure 5.12.
$\mathrm{MgO}, \mathrm{Sr}, \mathrm{Rb}, \mathrm{Na}_{2} \mathrm{O}, \mathrm{Al}_{2} \mathrm{O}_{3}, \mathrm{P}_{2} \mathrm{O}_{5}$, and CaO have absolute coefficient values greater than 0.4 and are considered important discriminators in this function; $\mathrm{Ba}, \mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}, \mathrm{TiO}_{2}$, and $\mathrm{K}_{2} \mathrm{O}$ are of lesser importance. The

## Model 5 - Nova Scotia - Morocco



Figure 5.11. Frequency plot of discriminant scores from Model 5.

## Model 5 - Nova Scotia - Morocco Detailed Classification Matrix

|  | $\begin{aligned} & \text { 恖 } \\ & 0 \\ & \text { Z } \end{aligned}$ | GRANITE | $\begin{aligned} & \text { CORRECTLY } \\ & \text { CLASSIFIED } \end{aligned}$ |  | MISCLASSIFIED |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | SMB Charest | 16 | 94.1\% | 1 | 5.9\% |
|  |  | SMB McKenzie | 13 | 72.2\% | 5 | 27.8\% |
|  |  | SMB Smith | 26 | 81.3\% | 6 | 18.7\% |
|  |  | SMB Richardson | 4 | 50.0\% | 4 | 50.0\% |
|  |  | Musquodoboit | 9 | 90.0\% | 1 | 10.0\% |
|  |  | Liscomb | 9 | 100.0\% | 0 | 0.0\% |
|  |  | Sherbrooke | 11 | 64.7\% | 6 | 35.3\% |
|  |  | Bull Ridge | 12 | 100.0\% | 0 | 0.0\% |
|  |  | Sangster Lake \& Larry's River | 30 | 93.8\% | 2 | 6.2\% |
|  |  | Queensport \& Halfway Cove | 28 | 77.8\% | 8 | 22.2\% |
|  |  | Ellison Lake | 2 | 14.3\% | 12 | 85.7\% |
|  | $\begin{aligned} & \text { 筲 } \\ & 0 \\ & \text { n } \end{aligned}$ | Barrington Passage | 27 | 87.1\% | 4 | 12.9\% |
|  |  | Shelburne | 26 | 96.3\% | 1 | 3.7\% |
|  |  | Bald Mountain | 4 | 100.0\% | 0 | 0.0\% |
|  |  | Port Mouton | 44 | 84.6\% | 8 | 15.4\% |
|  |  | Moose Point | 8 | 100.0\% | 0 | 0.0\% |
|  |  | Seal Island Western Granite | 10 | 100.0\% | 0 | 0.0\% |
| $\begin{aligned} & 8 \\ & 8 \\ & 8 \\ & 8 \\ & i \end{aligned}$ |  | Zaer | 21 | 60.0\% | 14 | 40.0\% |
|  |  | Ment | 3 | 100.0\% | 0 | 0.0\% |
|  |  | Oulmes | - | - \% | - | - \% |
|  |  | Sebt de Brikiine | 13 | 92.9\% | 1 | 7.1\% |
|  |  | Ajar El Bark | 3 | 100.0\% | 0 | 0.0\% |
|  |  | Tabouchennt-Bamega | 8 | 100.0\% | 0 | 0.0\% |
|  |  | Oulad Ouaslam | 18 | 100.0\% | 0 | 0.0\% |
|  |  | Tichka | - | - \% | $-$ | - \% |

## Discriminant Function Classification Matrix

| MgO | -1.31727 |
| :--- | ---: |
| Sr | 1.16580 |
| Rb | 0.75380 |
| $\mathrm{Na}_{2} \mathrm{O}$ | -0.61486 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 0.58987 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | -0.49626 |
| CaO | 0.43934 |
| Ba | -0.38912 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3 \mathrm{l}}$ | 0.29879 |
| $\mathrm{TiO}_{2}$ | 0.03925 |
| $\mathrm{~K}_{2} \mathrm{O}$ | 0.01124 |


|  |  |  | Predicted Group |  |
| :---: | :---: | :---: | :---: | :---: |
|  |  |  | N.S | Morocco |
|  | $\stackrel{\text { ¢ }}{\text { ¢ }}$ | 337 | $\begin{aligned} & 277 \\ & 82.2 \% \end{aligned}$ | $\begin{aligned} & 60 \\ & 17.8 \% \end{aligned}$ |
|  |  | 81 | $\begin{aligned} & 15 \\ & 18.5 \% \end{aligned}$ | $\begin{aligned} & 66 \\ & 81.5 \% \end{aligned}$ |

Total Correctly Classified $82.06 \%$

Figure 5.12. Discriminant function, detailed and summary classification matrix for Model 5.
classification matrix indicates that $82.06 \%\left(\mathrm{~S}_{\mathrm{e}}=1.88 \%\right)$ of all cases are correctly classified during the analysis.

### 5.6.3.2 Discussion of Model 5 (Nova Scotia - Morocco)

The addition of $\mathrm{P}_{2} \mathrm{O}_{5}$ may account for the slight improvement in the discrimination between both areas. As in Model 4, some elements appear to be particularly good regional discriminators (i.e. $\mathrm{MgO}, \mathrm{Na}_{2} \mathrm{O}$ and $\mathrm{Al}_{2} \mathrm{O}_{3}$ ). Samples from the Rehamna Massif are no longer misclassified in the calculations evidently as a result of the inclusion of $\mathrm{P}_{2} \mathrm{O}_{5}$ in the analysis. These results indicate that $\mathrm{P}_{2} \mathrm{O}_{5}$ populations are different in both areas. $\mathrm{P}_{2} \mathrm{O}_{5}$ populations are also variable within Morocco i.e. the Rehamna granites are no longer being misclassified while the Zaer pluton is still being misclassified.

Generally, significant amounts of samples from the Ellison Lake, SMB, and Sherbrooke plutons of Nova Scotia are misclassified into the Moroccan group. In Morocco, only the Zaer granites (Central Massif) appear similar to the Nova Scotian granites.

### 5.6.4 Model 6 (Nova Scotia - Morocco)

### 5.6.4.1 Results of Model 6 (Nova Scotia - Morocco)

A third run (Model 6) was done with $\mathrm{SiO}_{2}-\mathrm{Al}_{2} \mathrm{O}_{3}-\mathrm{MgO}-\mathrm{CaO}-$ $\mathrm{Na}_{2} \mathrm{O}-\mathrm{Ba}-\mathrm{Rb}-\mathrm{Sr}-\mathrm{V}$. The frequency distributions of the discriminant scores are shown in Figure 5.13. The discriminant function and classification matrix are presented in Figure 5.14. A detailed listing of the classification matrix for each pluton is presented in Figure 5.14.
$\mathrm{Sr}, \mathrm{MgO}, \mathrm{Rb}, \mathrm{CaO}, \mathrm{V}, \mathrm{Na}_{2} \mathrm{O}$ are important discriminators in the function. Ba and $\mathrm{Al}_{2} \mathrm{O}_{3}$ are of lesser importance. The classification matrix shows that 87.45 ( $\mathrm{S}_{\mathrm{e}}=2.11 \%$ ) of all cases have been correctly classified.

## Model 6 - Nova Scotia - Morocco



Figure 5.13. Frequency plot of discriminant scores from Model 6.

## Model 6 －Nova Scotia－Morocco <br> Detailed Classification Matrix

|  | $\begin{aligned} & \text { 采 } \\ & 0 \\ & 0 \\ & \hline 2 \end{aligned}$ | GRANITE | $\begin{aligned} & \text { CORRECTLY } \\ & \text { CLASSIFIED } \end{aligned}$ |  | MISCLASSIFIED |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | SMB Charest | － | －\％ | － | －\％ |
|  |  | SMB McKenzie | － | －\％ | － | －\％ |
|  |  | SMB $\quad$ Smith | 24 | 75．0\％ | 8 | 25．0\％ |
|  |  | SMB Richardson | 3 | 75．0\％ | 1 | 25．0\％ |
|  |  | Musquodoboit | － | －\％ | － | －\％ |
|  |  | Liscomb | 8 | 100．0\％ | 0 | 0．0\％ |
|  |  | Sherbrooke | 4 | 80．0\％ | 1 | 20．0\％ |
|  |  | Bull Ridge | － | －\％ | － | －\％ |
|  |  | Sangster Lake \＆Larry＇s River | － | －\％ | － | －\％ |
|  |  | Queensport \＆Halfway CoveElison Lake | 20 | 95．2\％ | 1 | 4．8\％ |
|  |  |  | － | －\％ | － | －\％ |
|  | $\begin{aligned} & \text { H } \\ & 8 \\ & 8 \end{aligned}$ | Barrington Passage Shelburne | 20 | 86．9\％ | 3 | 13．1\％ |
|  |  |  | 18 | 85．7\％ | 3 | 14．3\％ |
|  |  | Bald Mountain | 3 | 100．0\％ | 0 | 0．0\％ |
|  |  | Port Mouton | 42 | 95．4\％ | 2 | 4．5\％ |
|  |  | Moose Point | － | －\％ | － | －\％ |
|  |  | Lyons Bay |  |  |  |  |
|  |  | Seal Island Western Granite | 9 | 100．0\％ | 0 | 0．0\％ |
|  |  | Zaer | 30 | 85．7\％ | 5 | 14．3\％ |
|  |  | Ment | 4 | 80．0\％ | 1 | 20．0\％ |
|  |  | Oulmes | － | －\％ | － | －\％ |
|  |  | Sebt de Brikiine | 5 | 55．6\％ | 4 | 44．4\％ |
|  |  | Ajar El Bark | － | －\％ | － | －\％ |
|  |  | Tabouchennt－Bamega | 8 | 100．0\％ | 0 | 0．0\％ |
|  |  | Oulad OuaslamTichka | 17 | 100．0\％ | 0 | 0．0\％ |
|  |  |  | － | －\％ | － | －\％ |

Discriminant Function Classification Matrix

| Sr | 1.53573 |
| :--- | ---: |
| MgO | -1.26787 |
| Rb | 1.07485 |
| CaO | 0.71022 |
| V | 0.62819 |
| $\mathrm{Na}_{2} \mathrm{O}$ | -0.59992 |
| Ba | -0.51128 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 0.03239 |


|  |  |  | Predicted Group |  |
| :---: | :---: | :---: | :---: | :---: |
|  |  |  | N．S | Morocco |
| 喿 | $\stackrel{\text { ¢ }}{\text { ¢ }}$ | 170 | $\begin{aligned} & 151 \\ & 88.8 \% \end{aligned}$ | $\begin{aligned} & 19 \\ & 11.2 \% \end{aligned}$ |
|  | 器 | 77 | $\begin{aligned} & 12 \\ & 15.6 \% \end{aligned}$ | 65 $84.4 \%$ |

Total Correctly Classified $87.45 \%$

Figure 5．14．Discriminant function，detailed and summary classification matrix for Model 6.
5.6.4.2 Discussion of Model 6 (Nova Scotia - Morocco)

Although the number of correctly classified samples appears high when the results are compared with other models (Model $4=80.84 \pm$ 3.14, Model $5=82.06 \pm 3.76$ and this Model $6=87.45 \pm 4.27$ ) overlap does exist therefore these Models cannot be considered significantly different.

Some data vectors from the Central and Rehamna Massifs are being misclassified during the analyses. However, as demonstrated in Model 5 , the addition of $\mathrm{P}_{2} \mathrm{O}_{5}$ to the variable list, will allow discrimination between these two massifs. $\mathrm{P}_{2} \mathrm{O}_{5}$ was not included in the analyses because of restrictions resulting from the limited number of cases.

### 5.6.5 Discussion of the Nova Scotia- Morocco Models

The following observations can be made on the three models presented in this section:

1) The Nova Scotian and Moroccan granites can not be characterized as geochemicaly distinct based on the limited number of variates used in the analysis.
2) $\mathrm{Sr}, \mathrm{MgO}, \mathrm{Rb}, \mathrm{CaO}, \mathrm{Na}_{2} \mathrm{O}, \mathrm{V}, \mathrm{P}_{2} \mathrm{O}_{5}, \mathrm{Al}_{2} \mathrm{O}_{3}$, and Ba show the most variation between Nova Scotia and Morocco (Figure 5.15).
3) The Zaer pluton of the Central Massif appears to present more geochemical similarities with the Nova Scotian granites than any other granite within Morocco. It is particularly interesting to note that the Zaer pluton was intruded within Ordovician Flysch, like the Nova Scotian granites.
4) Some elements can be defined as good regional discriminators (Nova Scotia- Morocco) while others appear more variable at the local scale (North-South, Nova Scotia).

### 5.7.1 Introduction

Results of discriminant analysis on the Nova Scotian and Moroccan granites suggested that these granites could not be characterized as clearly distinct and seemingly present some geochemical similarities. The significance of these results needs to be assessed. To do this another potential area for correlation needed to be investigated. Consequently Iberian peraluminous granites were included in the analysis to determine whether the results are characteristic of the Nova Scotian and Moroccan granites or if similar results might be obtained when another Hercynian population is considered.

### 5.7.2 Model 7 Nova Scotia - Morocco - Iberia

5.7.2.1 Results of Model 7 (Nova Scotia, Morocco, Iberia)

Model 7 was run using $\mathrm{SiO}_{2}-\mathrm{TiO}_{2}-\mathrm{Al}_{2} \mathrm{O}_{3}-\mathrm{Fe}_{2} \mathrm{O}_{3}$ t $-\mathrm{MgO}-\mathrm{CaO}-$ $\mathrm{Na}_{2} \mathrm{O}-\mathrm{K}_{2} \mathrm{O}-\mathrm{Ba}-\mathrm{Rb}-\mathrm{Sr}$. A scatter plot of the discriminant scores is given in Figure 5.16. A territorial map delimiting the various group domains was drawn on Figure 5.16. The two discriminant functions and the classification matrix are presented in Figure 5.17. Function 1 accounts for $60.79 \%$ of the total variation and function $2,30.21 \%$. These values indicate the importance of each function in the total discrimination. Analysis of the discriminant functions shows that MgO , $\mathrm{Sr}, \mathrm{Rb}, \mathrm{Na}_{2} \mathrm{O}, \mathrm{Fe}_{2} \mathrm{O}_{3}$, and CaO are important discriminators. $\mathrm{TiO}_{2}, \mathrm{Ba}$, $\mathrm{Al}_{2} \mathrm{O}_{3}, \mathrm{~K}_{2} \mathrm{O}$ are of lesser importance. The classification matrix shows that $60.69 \%$ of all cases were correctly classified (with a $\mathrm{S}_{\mathrm{e}}=1.74$ $\%$ ). The reader will note that in these 3 group models identical populations are indicated by a correct classification result of $33 \%$ as opposed to $50 \%$ in the previous 2 group models.

## Nova Scotia - Morocco Models

| Absolute Coefficient Value | Model <br> 4 | Model <br> 5 | Model <br> 6 |
| :---: | :---: | :---: | :---: |
| $\mathrm{C}>0.8$ | $\begin{aligned} & \mathrm{Sr} \\ & \mathrm{MgO} \end{aligned}$ | $\begin{aligned} & \mathrm{MgO} \\ & \mathrm{Sr} \end{aligned}$ | Sr <br> MgO <br> Rb |
| $0.8>\mathrm{C}>0.6$ | Rb <br> CaO <br> $\mathrm{Na}_{2} \mathrm{O}$ | Rb $\mathrm{Na}_{2} \mathrm{O}$ | $\begin{aligned} & \mathrm{CaO} \\ & \mathrm{~V} \end{aligned}$ |
| $0.6>\mathrm{C}>0.4$ | $\begin{aligned} & \mathrm{Al}_{2} \mathrm{O}_{3} \\ & \mathrm{Ba} \end{aligned}$ | $\begin{aligned} & \mathrm{Al}_{2} \mathrm{O}_{3} \\ & \mathrm{P}_{2} \mathrm{O}_{5} \\ & \mathrm{CaO} \end{aligned}$ | $\begin{aligned} & \mathrm{Na}_{2} \mathrm{O} \\ & \mathrm{Ba} \end{aligned}$ |
| $0.4>\mathrm{C}>0.2$ |  | Ba $\mathrm{Fe}_{2} \mathrm{O}_{3 \mathrm{t}}$ |  |
| $0.2>\mathrm{C}>0.0$ | $\begin{aligned} & \mathrm{K}_{2} \mathrm{O} \\ & \mathrm{TiO}_{2} \\ & \mathrm{Fe}_{2} \mathrm{O}_{3 \mathrm{t}} \\ & \hline \end{aligned}$ | $\begin{aligned} & \mathrm{TiO}_{2} \\ & \mathrm{~K}_{2} \mathrm{O} \end{aligned}$ | $\mathrm{Al}_{2} \mathrm{O}_{3}$ |

Figure 5.15. Generally $\mathrm{Sr}, \mathrm{MgO}, \mathrm{Rb}, \mathrm{CaO}, \mathrm{Na}_{2} \mathrm{O}, \mathrm{V}$, $\mathrm{Al}_{2} \mathrm{O}_{3}$, and Ba are the most significant discriminators between Nova Scotia and Morocco.

Model 7 - Nova Scotia -Morocco - Iberia


Figure 5.16. Scatter plot and territorial map of discriminant scores from Model 7.

## Model 7 - Nova Scotia - Morocco - Iberia

## Discriminant Functions

|  | Function 1 | Function 2 |
| :--- | :---: | :---: |
| Sr | 1.37301 | -0.17214 |
| MgO | -1.16372 | 1.51257 |
| Rb | 0.84938 | 0.68103 |
| Na | O | -0.56679 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3 \mathrm{t}}$ | 0.55137 | -0.44581 |
| CaO | 0.40303 | -0.36974 |
| TiO | -0.38145 | 0.14793 |
| Ba | -0.32686 | -0.06713 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 0.32610 | -0.20805 |
| $\mathrm{~K}_{2} \mathrm{O}$ | 0.12770 | -0.20307 |


|  | Eigenvalue | Percent of <br> Variance |
| :--- | :--- | :--- |
| Function 1 | 0.83614 | 71.09 |
| Function 2 | 0.21222 | 22.00 |

Classification Matrix

|  |  |  | Predicted Group |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  | N.S | Morocco | Iberia |
| $\begin{aligned} & \text { B } \\ & 0 \\ & 0 \\ & \text { B } \\ & \text { In } \\ & \text { U } \\ & \text { U } \end{aligned}$ | $\stackrel{\sim}{2}$ | 337 | $\begin{aligned} & 210 \\ & 62.3 \% \end{aligned}$ | $\begin{aligned} & 43 \\ & 12.8 \% \end{aligned}$ | $\begin{aligned} & 84 \\ & 24.9 \% \end{aligned}$ |
|  | \% | 185 | $\begin{aligned} & 40 \\ & 21.6 \% \end{aligned}$ | $\begin{aligned} & 111 \\ & 60.0 \% \end{aligned}$ | $\begin{aligned} & 34 \\ & 18.4 \% \end{aligned}$ |
|  | 欴 | 264 | $\begin{aligned} & 62 \\ & 23.5 \% \end{aligned}$ | $\begin{aligned} & 46 \\ & 17.4 \% \end{aligned}$ | $\begin{gathered} 156 \\ 59.1 \% \end{gathered}$ |

Total Correctly Classified $60.69 \%$

Figure 5.17. Discriminant functions, eigenvalues and classification matrix for Model 7.

### 5.7.2.2 Discussion of Model 7 (Nova Scotia, Morocco, Iberia)

The classification matrix in Model 7 indicates that the Nova Scotian population is being misclassified into the Iberian populations twice as often as into the Moroccan populations. This suggests that the Nova Scotian granites are more similar to the Iberian granites than the Moroccan granites. Upon scrutuny of the detailed classification matrix (not shown here) it becomes evident that the distinctive nature of the granites from the southern part of Morocco are the reason why the Moroccan granites are not as well correlated with the Nova Scotian granites. In effect the northern granites of Morocco (Central and Rehamna Massifs) and the Iberian granites are equally similar to the Nova Scotian granites.

### 5.7.3 Model 8 Nova Scotia - Morocco - Iberia

5.7.3.1 Results of Model 8 (Nova Scotia, Morocco, Iberia)

Model 8 was run using $\mathrm{SiO}_{2}-\mathrm{TiO}_{2}-\mathrm{Al}_{2} \mathrm{O}_{3}-\mathrm{Fe}_{2} \mathrm{O}_{3} t-\mathrm{MgO}-\mathrm{CaO}-$ $\mathrm{Na}_{2} \mathrm{O}-\mathrm{K}_{2} \mathrm{O}-\mathrm{P}_{2} \mathrm{O}_{5}-\mathrm{Ba}-\mathrm{Rb}-\mathrm{Sr}$. A scatter plot of the discriminant score results and a territorial map of each group is shown on Figure 5.18. The discriminant functions and classification results are given in Figure 5.19. The classification matrix shows that $68.12 \%$ of all cases were correctly classified with a $\mathrm{S}_{\mathrm{e}}$ of $1.81 \%$.

### 5.7.3.2 Discussion of Model 8 (Nova Scotia, Morocco, Iberia)

As in Model 7 the Nova Scotian population is being misclassified into the Iberian population twice as often as into the Moroccan populations. The addition of $\mathrm{P}_{2} \mathrm{O}_{5}$ in the Model has eliminated the Rehamna in the misclassified samples. In this model the northern granites of Morocco (Central Massif, Zaer pluton in particular) and the Iberian granites are equally similar to Nova Scotia.


Figure 5.18. Scatter plot and territorial map of discriminant scores from Model 8.

## Model 8 - Nova Scotia - Morocco - Iberia

Discriminant Functions

|  | Function 1 | Function 2 |
| :--- | ---: | ---: |
| Rb | 1.13963 | 0.48297 |
| MgO | 1.10553 | -1.75698 |
| $\mathrm{Na}_{2} \mathrm{O}$ | -0.64409 | -0.37710 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3 \mathrm{t}}$ | -0.57660 | 0.88458 |
| Sr | 0.41701 | 0.99108 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | -0.35913 | -0.37510 |
| Ba | -0.17339 | -0.14015 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 0.14148 | 0.19968 |
| $\mathrm{~K}_{2} \mathrm{O}$ | -0.09712 | 0.10375 |
| CaO | -0.06603 | 0.35598 |
| $\mathrm{TiO}_{2}$ | 0.03662 | -0.00511 |


|  | Eigenvalue | Percent of <br> Variance |
| :--- | :--- | :--- |
| Function 1 | 0.40805 | 56.79 |
| Function2 | 0.31044 | 43.21 |

## Classification Matrix

|  |  |  | Predicted Group |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  | N.S | Morocco | Iberia |
|  | $\stackrel{\sim}{2}$ | 337 | $\begin{aligned} & 228 \\ & 67.7 \% \end{aligned}$ | $\begin{aligned} & 44 \\ & 13.1 \% \end{aligned}$ | $\begin{aligned} & 65 \\ & 19.3 \% \end{aligned}$ |
|  | 苞 | 81 | $\begin{aligned} & 12 \\ & 14.8 \% \end{aligned}$ | $\begin{aligned} & 51 \\ & 63.0 \% \end{aligned}$ | $\begin{aligned} & 18 \\ & 22.2 \% \end{aligned}$ |
|  |  | 247 | $\begin{aligned} & 61 \\ & 24.7 \% \end{aligned}$ | $\begin{gathered} 12 \\ 4.9 \% \end{gathered}$ | $\begin{aligned} & 174 \\ & 70.4 \% \end{aligned}$ |

Total Correctly Classified $68.12 \%$

Figure 5.19. Discriminant functions, eigenvalues and classification matrix for Model 8.

### 5.8 Nova Scotia-Morocco-Iberia-Australia

### 5.8.1 Introduction

Previous statistical models presented in this chapter suggested some geochemical similarities between the Acadian/Hercynian granites of Nova Scotia, Morocco and Iberia. A question arises as to the significance of the analysis. Particularly are all peraluminous granites the same worldwide. To answer this question the Nova Scotian, Moroccan and Iberian granites were compared with peraluminous granites Australia of similar age and composition, but of an unrelated orogenic belt.

### 5.8.2 Model 9 (Nova Scotia - Morocco - Iberia - Australia)

### 5.8.2.1 Results of Model 9 (Nova Scotia, Morocco, Iberia, Australia)

A first model was run using $\mathrm{SiO}_{2}-\mathrm{TiO}_{2}-\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}-\mathrm{MgO}-\mathrm{CaO}-$ $\mathrm{Na}_{2} \mathrm{O}-\mathrm{K}_{2} \mathrm{O}-\mathrm{P}_{2} \mathrm{O}_{5}$. The frequency distribution of the discriminant scores is shown in Figure 5.20. The discriminant function and classification matrix are given in Figure 5.21.
$\mathrm{CaO}, \mathrm{Al}_{2} \mathrm{O}_{3}, \mathrm{Na}_{2} \mathrm{O}, \mathrm{TiO}_{2}$ are important discriminators in the function and MgO to a lesser degree. With a percentage of correctly classified cases at $92.08 \%$, the two populations do in fact appear to be geochemicaly different ( $\mathrm{S}_{\mathrm{e}}=0.88 \%$ ).

### 5.8.2.2 Discussion of Model 9 (Nova Scotia, Morocco, Iberia, Australia)

The importance of CaO in the discriminant function of Model 9 reflects the significantly higher concentration of CaO in the Australian granites relative to the Acadian/Hercynian granites. The Australian granites are also characterised by their slightly lower $\mathrm{Al}_{2} \mathrm{O}_{3}, \mathrm{Na}_{2} \mathrm{O}$ and higher $\mathrm{TiO}_{2}$ values.

## Model 9 - Atlantic - Australia

Atlantic

Australia

$\square$ Atlantic
$\square$ Australia
Misclassified

Figure 5.20. Frequency plot of discriminant scores from Model 9.

## Model 9 - Nova Scotia - Morocco Iberia - Australia

## Discriminant Function

| CaO | -1.40157 |
| :--- | ---: |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 0.95117 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 0.84547 |
| $\mathrm{THO}_{2}$ | 0.44212 |
| MgO | -0.25003 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3 \mathrm{t}}$ | 0.12494 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | -0.09584 |
| $\mathrm{~K}_{2} \mathrm{O}$ | 0.00133 |

## Classification Matrix

|  |  |  | Predicted Group |  |
| :---: | :---: | :---: | :---: | :---: |
|  |  |  | Herc. | Austr. |
|  | 尊 | 787 | $\begin{aligned} & 728 \\ & 92.5 \% \end{aligned}$ | $\begin{aligned} & 59 \\ & 7.5 \% \end{aligned}$ |
|  | 涼 | 102 | $\begin{aligned} & 7 \\ & 6.9 \% \end{aligned}$ | 95 $93.1 \%$ |

Total Correctly Classified $92.58 \%$
Figure 5.21. Discriminant functions and classification matrix for Model 9.

### 5.8.3 Models 10 and 11 Nova Scotia - Morocco - Iberia - Australia


#### Abstract

5.8.3.1 Results and Discussion of Models 10 and 11 (Nova Scotia, Morocco, Iberia, Australia)


The discriminant function and classification results of two additional models are presented in Figure 5.22. Although these two runs do not present the appropriate number of cases in the Australian group, they do however indicate that the addition of $\mathrm{Ba}, \mathrm{Rb}$ and Sr to these models do not contribute significantly to the discrimination.

### 5.8.4 Model 12 Nova Scotia - Morocco - Iberia - Australia

5.8.4.1 Results of Model 12 (Nova Scotia, Morocco, Iberia, Australia)

In this model the four populations were evaluated separately in the Model. This was done to determine whether the Hercynian granites were equally different from the Australia population. Model 12 included the following variates: $\mathrm{TiO}_{2}, \mathrm{Al}_{2} \mathrm{O}_{3}, \mathrm{Fe}_{2} \mathrm{O}_{3}$ t, $\mathrm{MgO}, \mathrm{CaO}, \mathrm{Na}_{2} \mathrm{O}$, $\mathrm{K}_{2} \mathrm{O}$, and $\mathrm{P}_{2} \mathrm{O}_{5}$. Again $\mathrm{CaO}, \mathrm{Al}_{2} \mathrm{O}_{3}, \mathrm{Na}_{2} \mathrm{O}$ and $\mathrm{Ti}_{2} \mathrm{O}$ are important discriminators.

### 5.8.4.2 Discussion of Model 12 (Nova Scotia, Morocco, Iberia, Australia)

The classification matrix of Figure 5.24 clearly indicates the difference between the Atlantic and Australian populations. Although the Atlantic granites present similar classification values amongst themselves (correctly classified between $54-61 \%$ ) they are very seldom misclassified into the Australian population. The Australian population are distinct with a value of $92.2 \%$ for correctly classified cases $\left(\mathrm{S}_{\mathrm{e}}=1.66 \%\right)$. The Australian granites are also rarely misclassified into the Atlantic populations. Therefore, the Hercynian/Acadian granites are clearly different from the Australian granites.

## Model 10 and 11 －Nova Scotia＋Morocco＋ Iberia－Australia

## MIoclel 10

Discriminant Function

| CaO | -1.38553 |
| :--- | ---: |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 0.91585 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 0.82037 |
| MgO | -0.55969 |
| $\mathrm{TiO}_{2}$ | 0.38933 |
| $\mathrm{Rb}^{2}$ | -0.21136 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3 \mathrm{t}}$ | 0.18877 |
| Ba | 0.18227 |

## Classification Matrix

|  |  |  | Predicted Group |  |
| :---: | :---: | :---: | :---: | :---: |
|  |  |  | Atla． | Austr． |
|  | 苞 | 800 | $\begin{aligned} & 761 \\ & 95.1 \% \end{aligned}$ | $\begin{aligned} & 39 \\ & 4.9 \% \end{aligned}$ |
|  | 窝 | 80 | $\begin{aligned} & 5 \\ & 6.3 \% \end{aligned}$ | $\begin{aligned} & 75 \\ & 93.8 \% \end{aligned}$ |

Total Correctly Classified $95.00 \%$

## Model 11

Discriminant Function

## Classification Matrix

| CaO | -1.51801 |
| :--- | ---: |
| $\mathrm{Na}_{2} \mathrm{O}$ | 0.92519 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 0.89607 |
| $\mathrm{TiO}_{2}$ | 0.4002 |
| MgO | -0.38492 |
| Rb | -0.34015 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3 \mathrm{l}}$ | 0.21389 |
| Ba | 0.17591 |
| Sr | -0.14569 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | -0.08779 |
| $\mathrm{~K}_{2} \mathrm{O}$ | 0.01576 |


|  |  |  | Predicted Group |  |
| :---: | :---: | :---: | :---: | :---: |
|  |  |  | Atla． | Austr． |
|  | 嵳 | 665 | $\begin{aligned} & 634 \\ & 95.3 \% \end{aligned}$ | $31$ $4.7 \%$ |
|  | 鸷 | 80 | $\begin{aligned} & 3 \\ & 3.7 \% \end{aligned}$ | $\begin{aligned} & 77 \\ & 96.2 \% \end{aligned}$ |

Total Correctly Classified $95.44 \%$

Figure 5．22．Discriminant functions and classification matrices for Models 10 and 11.

## Model 12 - Nova Scotia -Morocco - Iberia Australia



Figure 5.23. Scatter plot and territorial map of discriminant scores from Model 12.

## Model 12 －Nova Scotia＋Morocco＋ Iberia－Australia

Discriminant Functions

|  | Function 1 | Function 2 |  | Function 3 |
| :--- | :---: | :---: | :---: | :---: |
| CaO | -1.42977 | 0.58000 | 0.62038 |  |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 0.88204 | -0.26101 | 0.78786 |  |
| $\mathrm{Na}_{2} \mathrm{O}$ | 0.83798 | 0.47216 | -0.27583 |  |
| $\mathrm{TiO}_{2}$ | 0.38086 | -0.46879 | 0.55568 |  |
| $\mathrm{Fe}_{2} \mathrm{O}_{3 \mathrm{l}}$ | 0.15473 | 0.86051 | -0.3552 |  |
| MgO | -0.13146 | -1.02501 | -1.60567 |  |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | -0.05000 | 0.28654 | -0.40104 |  |
| $\mathrm{~K}_{2} \mathrm{O}$ | -0.03950 | -0.07006 | 0.41490 |  |


|  | Eigenvalue | Percent of <br> Variance |
| :--- | :--- | :---: |
| Function 1 | 0.83614 | 71.09 |
| Function 2 | 0.24133 | 20.52 |
| Function 3 | 0.09869 | 8.39 |

Classification Matrix

|  |  |  | Predicted Group |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  | N．S | Morocco | Iberia | Austr． |
| $\begin{aligned} & \text { 歌 } \\ & \text { 淢 } \end{aligned}$ | 号 | 333 | $\begin{aligned} & 209 \\ & 54.6 \% \end{aligned}$ | $\begin{aligned} & 66 \\ & 22.5 \% \end{aligned}$ | $\begin{aligned} & 77 \\ & 20.1 \% \end{aligned}$ | $\begin{aligned} & 11 \\ & 2.9 \% \end{aligned}$ |
|  |  | 107 | $\begin{aligned} & 22 \\ & 21.6 \% \end{aligned}$ | $\begin{aligned} & 58 \\ & 54.2 \% \end{aligned}$ | $\begin{aligned} & 17 \\ & 15.9 \% \end{aligned}$ | $\begin{aligned} & 10 \\ & 9.3 \% \end{aligned}$ |
|  | $\begin{array}{\|l\|l} \stackrel{a}{⿷ 匚 ⿳ ⿻ コ 一 冖 巾 刂 ~} \\ \hline \end{array}$ | 297 | $\begin{aligned} & 51 \\ & 17.2 \% \end{aligned}$ | $\begin{aligned} & 47 \\ & 15.8 \% \end{aligned}$ | $\begin{aligned} & 181 \\ & 60.9 \% \end{aligned}$ | $\begin{aligned} & 18 \\ & 6.1 \% \end{aligned}$ |
|  | 苞 | 102 | $\begin{aligned} & 0 \\ & 0.0 \% \end{aligned}$ | $\begin{aligned} & 6 \\ & 5.9 \% \end{aligned}$ | $\begin{aligned} & 2 \\ & 2.0 \% \end{aligned}$ | $\begin{aligned} & 94 \\ & 92.2 \% \end{aligned}$ |

Total Correctly Classified 60．97\％

Figure 5．24．Discriminant functions，eigenvalues and classification matrix for Model 12.

### 5.8.5 Model 13 and 14 Nova Scotia - Morocco - Iberia - Australia

5.8.5.1 Results and discussion of Models 13 and 14 (Nova Scotia, Morocco, Iberia, Australia)

The discriminant function and classification results of two additional models are presented in Figure 5.25. As for Models 10 and 11 these two runs do not present the appropriate number of cases in the Australian group. There is an apparent increase in the importance of $\mathrm{Ba}, \mathrm{Rb}$ and Sr as relative to models 10 and 11 however, it is difficult to characterise there importance in the discrimination.

### 5.8.6 Discussion of the Nova Scotian - Moroccan - Iberian - Australian Models

The following observations can be made on the models presented in this section:

1) The Atlantic granites and the Australian granites are in effect geochemically distinct.
2) Some elements appear to be good discriminators on the orogenic scale.
3) These models confirm that all peraluminous granites worldwide are not the same.

# Model 13 - Nova Scotia + Morocco + Iberia - Australia 

Discriminant Functions

|  | Function 1 |  | Function 2 |
| :---: | :---: | :---: | :---: |
| CaO | -1.30471 |  | -0.40407 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 0.88246 |  | 0.51205 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 0.83159 |  | -0.33081 |
| MgO | -0.69648 |  | 0.53493 |
| $\mathrm{TiO}_{2}$ | 0.36526 |  | 0.12714 |
| Rb | -0.30668 |  | 0.88576 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3 \mathrm{t}}$ | 0.26151 |  | 0.53493 |
| Ba | 0.18190 |  | 0.14639 |
|  |  | Eigenvalue | Percent of Variance |
|  | Function 1 | 0.77751 | 69.81 |
|  | Function 2 | 0.17710 | 15.90 |
|  | Function 3 | 0.15909 | 14.28 |

Classification Matrix

|  |  |  | Predicted Group |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  | N.S | Morocco | Iberia | Austr. |
|  | $\stackrel{\text { ¢ }}{\text { ¢ }}$ | 350 | $\begin{aligned} & 165 \\ & 52.9 \% \end{aligned}$ | $\begin{aligned} & 58 \\ & 19.4 \% \end{aligned}$ | $\begin{aligned} & 92 \\ & 26.3 \% \end{aligned}$ | $\begin{aligned} & 5 \\ & 1.4 \% \end{aligned}$ |
|  |  | 186 | $\begin{aligned} & 43 \\ & 23.1 \% \end{aligned}$ | $\begin{aligned} & 94 \\ & 50.5 \% \end{aligned}$ | $\begin{aligned} & 42 \\ & 22.6 \% \end{aligned}$ | $\begin{aligned} & 7 \\ & 3.8 \% \end{aligned}$ |
|  | $\begin{array}{\|l\|l} \hline \text { 㫊 } \\ \text { a } \end{array}$ | 264 | $\begin{aligned} & 48 \\ & 18.2 \% \end{aligned}$ | $\begin{aligned} & 49 \\ & 18.6 \% \end{aligned}$ | $\begin{gathered} 148 \\ 56.1 \% \end{gathered}$ | $19$ |
|  | 蝺 | 80 | $\begin{aligned} & 2 \\ & 2.5 \% \end{aligned}$ | $\begin{aligned} & 2 \\ & 2.5 \% \end{aligned}$ | $\begin{aligned} & 1 \\ & 1.2 \% \end{aligned}$ | 75 $93.8 \%$ |

Total Correctly Classified 57.05\%

Figure 5.25. Discriminant functions, eigenvalues and classification matrix for Model 13.

## Model 14 - Nova Scotia + Morocco + Iberia - Australia

Discriminant Functions

|  | Function $1 \quad \mathrm{~F}$ |  | Function 2 | Function 3 |
| :---: | :---: | :---: | :---: | :---: |
| CaO | 1.30884 |  | -0.80106 | -0.58002 |
| $\mathrm{Na}_{2} \mathrm{O}$ | -0.98728 |  | 0.04278 | -0.23219 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | -0.70974 |  | 0.80922 | 0.71967 |
| Rb | 0.68321 |  | 0.49561 | 0.90047 |
| MgO | 0.50836 |  | 1.80734 | -0.93220 |
| Sr | $0.34969-0.3$ |  | -0.31349 | 0.96984 |
| $\mathrm{TiO}_{2}$ | -0.33645 |  | 0.17361 | 0.24143 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3 \mathrm{t}}$ | -0.28737 |  | -0.93500 | 0.46932 |
| Ba | -0.21457 |  | 0.02731 | -0.12014 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | -0.07576 |  | -0.09295 | -0.57064 |
| $\mathrm{K}_{2} \mathrm{O}$ | -0.03238 |  | -0.12651 | 0.03337 |
|  |  | Eigenvalue | Perce Vari |  |
|  | Function 1 | 1.03990 | 63.35 |  |
|  | Function 2 | 0.32747 | 1.9 .95 |  |
|  | Function 3 | 0.27417 | 16.70 |  |

Classification Matrix

|  |  |  | Predicted Group |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  | N.S | Morocco | Iberia | Austr. |
|  | $\stackrel{\sim}{\square}$ | 337 | $\begin{aligned} & 226 \\ & 67.1 \% \end{aligned}$ | $\begin{aligned} & 44 \\ & 13.1 \% \end{aligned}$ | $\begin{aligned} & 63 \\ & 18.7 \% \end{aligned}$ | $\begin{aligned} & 4 \\ & 1.2 \% \end{aligned}$ |
|  | - | 81 | $\begin{aligned} & 11 \\ & 13.6 \% \end{aligned}$ | $\begin{aligned} & 50 \\ & 61.7 \% \end{aligned}$ | $\begin{aligned} & 15 \\ & 18.5 \% \end{aligned}$ | $\begin{aligned} & 5 \\ & 6.2 \% \end{aligned}$ |
|  | 㛯 | 247 | $\begin{aligned} & 59 \\ & 23.9 \% \end{aligned}$ | $\begin{gathered} 12 \\ 4.9 \% \end{gathered}$ | $\begin{gathered} 163 \\ 66.0 \% \end{gathered}$ | $\begin{array}{r} 13 \\ 5.3 \% \end{array}$ |
|  | 菬 | 80 | $\begin{aligned} & 1 \\ & 1.2 \% \end{aligned}$ | $\begin{aligned} & 3 \\ & 3.7 \% \\ & \hline \end{aligned}$ | $\begin{aligned} & 1 \\ & 1.2 \% \end{aligned}$ | $\begin{aligned} & 75 \\ & 93.8 \% \\ & \hline \end{aligned}$ |

Total Correctly Classified $68.99 \%$

Figure 5.26. Discriminant functions, eigenvalues and classification matrix for Model 14.

## CHAPTER 6

CONCLUSIONS AND RECOMMENDATIONS FOR FUTURE WORK

### 6.1 Conclusions

Comparison of the stratigraphy of the Meguma Terrane of Nova Scotia and Paleozoic Massifs of Morocco shows similarity in their Cambrian to Carboniferous successions, and provides justification for this study.

A comparison of the major, trace, and rare earth element data using traditional approaches has revealed that:

1- The geochemical populations of the northern and southern plutons of Nova Scotia, although apparently different, cannot be clearly separated into two groups.

2- The Nova Scotian and Moroccan geochemical populations could not be separated and, in effect, appear inclistinguishable.

3- In addition, limited Nd and Sr isotopic data also suggest some similarity between both areas.

The applicability of discriminant analysis to granitic data was demonstrated using various test models. In particular, results show that:

1- Discriminant analysis of compositional geochemical data can be successfully applied using the log ratio transformation to eliminate the closure problem.

2- Bimodal and moderately skewed populations and uneven sample groups (up to 10:1) can be successfully analysed using discriminant function analysis.

3- The reliability of the discriminant coefficients was also demonstrated indicating that the contribution of the different variates (elements, oxides) to the discrimination could be interpreted.

Based on the statistical models the following conclusions can be made:

1- The results confirm geological observations made by previous workers that suggested that the granites from Northern and Southern Nova Scotia were different. In addition, the discriminant analysis also allows the classification of samples using the discriminant function. In the models some elements are consistently excellent discriminators within Nova Scotia ( $\mathrm{Rb}, \mathrm{Sr}, \mathrm{Ba}, \mathrm{TiO}_{2}, \mathrm{CaO}$ and, Pb ).

2- The Nova Scotian and Moroccan models suggest that the two groups can not be characterized as geochemically distinct. The Zaer pluton of the Central Massif appears to show more geochemical similarities with the Nova Scotian granites than any other granite within Morocco. All Ellison Lake specimens from Nova Scotia are misclassified into the Moroccan population. Variables such as $\mathrm{Sr}, \mathrm{MgO}, \mathrm{Rb}, \mathrm{CaO}, \mathrm{Na}_{2} \mathrm{O}, \mathrm{V}$, $\mathrm{P}_{2} \mathrm{O}_{5}, \mathrm{Al}_{2} \mathrm{O}_{3}$ and, Ba show the most the most variation between Nova Scotia and Morocco.

3- The Nova Scotian, Moroccan and Iberian models indicate that the granites of Morocco (Central Massif, Zaer pluton in particular) and the Iberian granites are equally similar to Nova Scotia. As in the Nova Scotian- Moroccan models, $\mathrm{Sr}, \mathrm{MgO}, \mathrm{Rb}, \mathrm{Na}_{2} \mathrm{O}, \mathrm{CaO}$, and $\mathrm{P}_{2} \mathrm{O} 5$ are good discriminators, however, $\mathrm{Fe}_{2} \mathrm{O}_{3 \mathrm{t}}$ is much more important in the Nova Scotia-Morocco-Iberia models than in the Nova Scotia-Morocco ones.

4- The models comparing the Atlantic and Australian granites indicate that the two groups are in effect geochemically distinct. Variables such as $\mathrm{CaO}, \mathrm{Al}_{2} \mathrm{O}_{3}, \mathrm{Na}_{2} \mathrm{O}, \mathrm{MgO}$, and $\mathrm{TiO}_{2}$ show the most variation between these two orogenic belts.

5- Comparison of results obtained on the local (north-south Nova Scotia), regional (Nova Scotia-Morocco and Nova Scotia-Morocco-Iberia) and the orogenic scale (Atlantic and Australian granites) show that different suites of elements can be characterised as good discriminators depending on the scale of reference.

Therefore, discriminant function analyses are applicable to granitic data and appear more useful and more revealing than traditional methods of comparisons.

The answer to the question :Is the Moroccan model that correlates the Meguma terrane to north-western Africa correct?, is possibly. Although no clear evidence was found to confirm the Moroccan Model for the origin of the Meguma zone, it can not be ruled out as a potential source area.

### 6.2 Recommendations for Future Work

Throughout the thesis different questions remaining to be resolved were outlined the following points are recommended for possible future consideration:

1- Increase the number of variates and cases (analyses) in the database to allow more elaborate comparisons of the granite populations. In particular the inclusion of REE and other immobile elements could possibly improve the modelling. Systematic isotopic analyses (e.g. $\mathrm{Pb}, \mathrm{Nd}, \mathrm{Sr}$ and O ) are also needed to improve the comparison between the granites.

2- Include other potential areas for correlation, as well as those areas which are not correlated to allow better comparisons (e.g. Columbia).
3- Study the geochemical data using other parametric statistical methods. In addition the applicability of non-parametric multivariate statistics to the geochemical data needs to be assessed.

4- Many of the plutons considered in this study were not sampled systematically, therefore more comprehensive detailed work is needed to better constrain each geochemical population.

5- Study the effects of analytical error and interlaboratory differences on the results.

6- Investigate the origin of the Meguma terrane using other geological evidence.

## APPENDIX A

## MODAL ANALYSIS

Modal analysis were carried out on the Moroccan granites using the following stain and point counting method:

1- Slabs of granite were cut using a diamond rock saw to a thickness of $0.5-1.5 \mathrm{~cm}$. All alteration surfaces were removed at this stage using the saw.

2- A clean surface of each slab was then immersed in hydrofluoric acid for approximately 1 minute.

3- The slabs were then rinsed in a bath of tap water for about 15-30 seconds.

4- Slabs were then immersed in a saturated solution of sodium cobaltinitrite for approximately 45 seconds -2 minutes (the length of this stage is highly dependent on the state of the solution i.e. the number of samples previously stained, and the desired intensity of the stain).

5- Slabs were then rinsed ( $2-3$ seconds) in a bath of tap water to remove excess solution.

6- Slabs were then dried using acetone.
7- The stained slabs were point counted using a binocular microscope. The slabs are mounted on a piece of modelling clay to insure that the surface being point counted is horizontal for easy focussing. A grid in placed onto this surface and point counting is done at 1 mm intervals ( $>1000$ points, generally $1500-2000$ points are counted). K-feldspar (dark yellow) plagioclase (light yellow to light gray), quartz (glassy gray), biotite and muscovite were counted.

8- A model 100 computer was used during the point counting. A basic program was written (not included in this thesis) to compute the modal \% and determine rock classifications using Steckeisens (1976) terminology.

## APPENDIX B

B. 1 Petrographic Description of Analytical Samples
B.1.1 Definition of Terms Used in the Petrographic Descriptions
B.1.1.1 Quality of Outcrop (Modified $\mathrm{M}^{\mathrm{C} k e n z i e, ~}$ ..... 1974)
Excellent: Large outcrop with abundant fresh surfaces.
Good: Large to intermediate-sized outcrop with some fresh surfaces.
Fair: Intermediate to small outcrop with few fresh surfaces.
Poor: Weathered outcrop with rare fresh surfaces.
B.1.1.2 Degree of Alteration (Modified $\mathrm{M}^{\mathrm{C} K e n z i e, ~ 1974) ~}$
Fresh: Minor alteration of plagioclase, alkali feldspar and biotiteshowing minor amounts of secondary sericite, kaolinite and chlorite.Moderate: Moderate feldspar kaolinization and muscovite alteration.Extensive: Alteration of feldspar intense, biotite extensively orcompletely muscovitized and/or chloritized.
B.1.1.3 Grain Size (Same as $\mathrm{M}^{\text {C }}$ kenzie, ..... 1974)
Coarse > 5 mm
Medium 1.5 mm
Fine $<1 \mathrm{~mm}$
B.1.2 Petrographic Description of Chemically Analyzed Samples
B.1.2.1 Oulad Ouaslam Batholith
JBL-1B
Quality of outcrop: Poor
Degree of alteration: Moderate - Extensive
Fine-grained aplite (classified as a monzogranite) with quartz,seritised and muscovitised plagioclase, K-feldspar (microclinetwinning), minor sligthly chloritized biotite and moderatly kinkedmuscovite.
JBL-2A
Quality of outcrop: Fair
Degree of alteration: Moderate

Fine- to medium- grained granodiorite with quartz, plagioclase, Kfeldspar (found in clots and generally includes plagioclase) (feldspar are saussuritized), biotite is slightly chloritized and includes opaques and zircons. Abundant secondary muscovite.

## JBL-3

Quality of outcrop: Good
Degree of alteration: Moderate
Fine- to medium- grained monzogranite with quartz, plagioclase, Kfeldspar (found in clots and generally includes plagioclase), minor biotite (slightly chloritized) and muscovite with opaques, zircons and apatite. Minor clots of biotite, muscovite, quartz, opaques and apatite can be found throughout the rock.

JBL-6
Quality of outcrop: Good
Degree of alteration: Moderate
Fine- to medium-grained granodiorite with quartz, plagioclase, Kfeldspar (in individual small crystals and as phenocrysts) feldspars are seritised, biotite and muscovite. Small veins (microscopic) of opaques are observed cross-cutting the rock. The rock is slightly foliated (defined by biotite, muscovite and feldspars).

JBL-8
Quality of outcrop: Good
Degree of alteration: Fresh
Mediun- to coarse grained granodiorite with quartz, seritised plagioclase and K-feldspar (as phenocrysts), abundant biotite (includes opaques, zircon and apatite), and minor muscovite and zircon. Clots of biotite, muscovite, quartz minor garnet and sillimanite are present in hand specimen (not included in the analysis).

JBL-10
Quality of outcrop: Excellent
Degree of alteration: Fresh
Medium- to coarse grained granodiorite with quartz, seritised plagioclase and K-feldspar (as phenocrysts), chloritized biotite (include opaques, zircon and apatite), and muscovite. Accessory phases are apatite, cordierite and sillimanite.

## JBL-12

Quality of outcrop: Excellent
Degree of alteration: Fresh to moderate
Fine-grained granodiorite with quartz, seritised plagioclase and K-feldspar, minor chloritized biotite and, muscovite. Accessory phases are apatite, cordierite and garnet. Inclusions of biotite muscovite and cordierite in hand specimen (not included in the analysis).

JBL-13
Quality of outcrop: Good
Degree of alteration: Moderate
Medium-grained monzogranite with quartz, seritised plagioclase and K-feldspar (as clots), chloritized biotite (inclusions of apatite and opaques), and muscovite. Accessory phase is apatite. Biotite and muscovite are generally found as clots (shadow inclusions?).

JBL-21
Quality of outcrop: Excellent
Degree of alteration: Moderate to fresh
Medium- to coarse grained granodiorite with quartz, plagioclase, K-feldspar (phenocrysts), abundant biotite (slightky chloritized) and, muscovite. Accessory phases are garnet, cordierite, sillimanite? and apatite.

JBL-23
Quality of outcrop: Good
Degree of alteration: Fresh
Medium-grained granodiorite with quartz, plagioclase, K-feldspar, biotite and muscovite. Accessory phases are sillimanite?, cordierite, and apatite. Shadow enclaves? are present in hand specimen (not included in the analysis).

JBL-24
Quality of outcrop: Excellent
Degree of alteration: Moderate to fresh
Fine-grained granodiorite with quartz, plagioclase, K-feldspar, chloritized biotite and muscovite. Minor clots of biotite, muscovite, quartz, opaques can be found throughout the rock.

JBL-26
Quality of outcrop: Good Degree of alteration: Moderate

Medium- to coarse- grained monzogranite with quartz, plagioclase, K-feldspar (as phenocrysts and interstially), chloritized biotite and, muscovite.

## JUB-4

Quality of outcrop: Good to excellent
Degree of alteration: Fresh to moderate
Medium-grained monzogranite with quartz, plagioclase, K-feldspar (as phenocrysts) chloritized biotite and, muscovite. Accessory phases are cordierite, sillimanite?, and garnet. Shadow enclaves? are observed in hand specimen (not included in analysis).

## JUB-6

Quality of outcrop: Good to excellent
Degree of alteration: Fresh to moderate
Fine-grained granodiorite with quartz plagioclase (as phenocrysts) and K-feldspar, slightly chloritized biotite and minor muscovite. Accessory phases are opaques.

## JUB-9

Quality of outcrop: Good Degree of alteration: Moderate to fresh

Fine- to medium-grained granodiorite with quartz plagioclase, Kfeldspar (as phenocrysts and interstially), chloritized biotite (opaque inclusions) and, muscovite. In hand specimen shadow enclaves? are evident with cordierite and garnet (not included in analysis).

## JUB-18

Quality of outcrop: Good to excellent
Degree of alteration: Moderate to fresh
Fine-grained monzogranite with quartz plagioclase, K-feldspar (as phenocrysts and interstially), chloritized biotite (inclusion of apatite and zircon), and muscovite. Accessory phase is apatite. In hand specimen cordierite and biotite clots are visible and may represent shadow enclaves?.

## JUB-19

Quality of outcrop: Excellent
Degree of alteration: Fresh
Fine- to medium grained granodiorite with quartz, plagioclase, Kfeldspar (as clusters), slightly chloritized biotite (inclusions of apatite and zircon), and muscovite. Accessory phase is andalusite (in thin section). Clots of biotite, muscovite and plagioclase may represent shadow enclaves.

## JUB-21

Quality of outcrop: Good to excellent
Degree of alteration: Moderate to fresh
Fine-grained monzogranite with quartz, seritised plagioclase and K-feldspar, biotite (inclusions of apatite, opaques, and zircon), and muscovite. Accessory phase is cordierite (shadow enclave?).

## JUB-24

Quality of outcrop: Excellent
Degree of alteration: Moderate to fresh
Fine-grained monzogranite with quartz, seritised plagioclase and K-feldspar. Minor phases are chloritized biotite (inclusions opaques and zircon), muscovite, cordierite and opaques.

## B.1.2.2 Tabouchennt-Bamega Pluton

BRR-7C
Quality of outcrop: Excellent
Degree of alteration: Fresh
Fine- to medium-grained monzogranite with quartz, seritised plagioclase and K-feldspar. The plagioclase is observed mantling the K-feldspar to form phenocrysts. Minor phases are chloritized biotite (inclusions opaques and zircon) and muscovite.

BRR-11A
Quality of outcrop: Good
Degree of alteration: Fresh
Coarse-grained phophyritic monzogranite with quartz, seritised plagioclase (phenocrysts) and K-feldspar (large phenocrysts), and minor chloritized biotite (inclusions opaques and zircon) and muscovite.

BRR-12A
Quality of outcrop: Good to excellent
Degree of alteration: moderate
Medium-grained monzogranite with quartz, seritised plagioclase and K-feldspar, biotite and muscovite. A slight foliation is defined by the biotite and K-feldspar.

## BRR-14

Quality of outcrop: Excellent
Degree of alteration: Fresh
Medium- to coarse- grained quartz, plagioclase and K-feldspar (interstially and as phenocrysts), biotite (inclusions of opaques, apatite and zircon), and minor muscovite (possibly cordierite in thin section?).

BRR-16A
Quality of outcrop: Excellent Degree of alteration: Fresh

Medium- to coarse- grained monzogranite with quartz, seritised plagioclase and K-felsdpar (interstially and as phenocrysts), chloritized biotite (inclusions of opaques, apatite and zircon), and muscovite.

BRR-18
Quality of outcrop: Excellent to good
Degree of alteration: Fresh
Medium-grained porphyritic monzogranite with quartz, seritised plagioclase and K-feldspar, biotite and muscovite. The plagioclase mantles the K-feldspar to form phenocrysts.

## BRR-22A

Quality of outcrop: Good Degree of alteration: Fresh

Fine-grained equigranular monzogranite with quartz, seritised plagioclase and K-feldspar, biotite and muscovite (no thin section for this sample).

## B.1.2.3 Sebt De Brikiine Batholith

SDB-1
Quality of outcrop: Excellent
Degree of alteration: Fresh
Fine-grained aplite (classified as a monzogranite) with quartz, plagioclase and, K-feldspar. The feldspars are pink and appear hematized. Accessory phases are opaques and rare biotite.

SDB-2A
Quality of outcrop: Good
Degree of alteration: Moderate
Medium- to coarse grained monzogranite with quartz, seritised plagioclase and K-feldspar. The feldspar are pink and appear hematized. Minor phases are biotite and opaques. Minor muscovite as an alteration product.

## SDB-3

Quality of outcrop: Excellent
Degree of alteration: Fresh to moderate
Fine-grained equigranular monzogranite with quartz, seritised plagioclase and K-feldspar. The feldspar are pink and appear hematized. Minor phases are biotite and opaques. Minor muscovite as an alteration product.

## SDB-4

Quality of outcrop: Good to excellent
Degree of alteration: Moderate
Fine-grained aplite (classified as a monzogranite) with quartz, plagioclase and, K-feldspar. The feldspars are pink and appear hematized. Accessory phases are opaques and rare biotite.

## SDB-6

Quality of outcrop: Good to excellent
Degree of alteration: Moderate
Fine-grained equigranular monzogranite with quartz, seritised plagioclase and K-feldspar. The feldspar are pink and appear hematized. Minor phases are biotite and opaques. Minor muscovite as an alteration product.

SDB-7
Quality of outcrop: Excellent
Degree of alteration: Moderate
Medium- to coarse-grained equigranular syenogranite with quartz, seritised plagioclase and K-feldspar. The feldspar are pink and appear hematized. Minor phases are biotite and opaques. Minor muscovite as an alteration product.

SDB-8
Quality of outcrop: Excellent Degree of alteration: Fresh to moderate

Fine-grained aplite (classified as a monzogranite) with quartz, plagioclase and, K-feldspar. The feldspars are pink and appear hematized. Accessory phases are opaques and rare biotite.

SDB-9
Quality of outcrop: Excellent to good Degree of alteration: moderate

Fine-to medium-grained equigranular monzogranite with quartz, seritised plagioclase and K-feldspar. The feldspar are pink and appear hematized. Minor phases are biotite and opaques. Minor muscovite as an alteration product.

SDB-10
Quality of outcrop: Good Degree of alteration: Moderate

Medium- to coarse-grained monzogranite with quartz, seritised plagioclase and K-feldspar. The feldspar are pink and appear hematized. There are three different types of phenocrysts, plagioclase, K-feldspar and plagioclase mantling the K-feldspar. Minor phases are biotite and opaques. Minor muscovite as an alteration product.

## SDB-11

Quality of outcrop: Excellent to good
Degree of alteration: Fresh to moderate
Medium-grained monzogranite with quartz, seritised plagioclase and K-feldspar. The feldspar are pink and appear hematized. Minor phases are biotite and opaques. Minor muscovite as an alteration product.

SDB-12
Quality of outcrop: Good
Degree of alteration: Moderate
Fine-grained aplite (classified as a monzogranite) with quartz, plagioclase and, K-feldspar. The feldspars are pink and appear hematized. Accessory phases are opaques and rare biotite.

## SDB-13

Quality of outcrop: Good to excellent
Degree of alteration: Moderate to fresh
Fine-grained monzogranite with quartz, seritised plagioclase and K-feldspar. The feldspar are pink and appear hematized. There are three different types of phenocrysts, plagioclase, K-feldspar(less abundant) and plagioclase mantling the K-feldspar. Minor phases are biotite and opaques. Minor muscovite as an alteration product.

## SDB-14

Quality of outcrop: Good to excellent Degree of alteration: Moderate to fresh

Fine-grained equigranular monzogranite with quartz, seritised plagioclase and K-feldspar. The feldspar are pink and appear hematized. Minor phases are biotite and opaques. Minor muscovite as an alteration product (no thin section for this sample).

## B.1.2.4 Ajar El Bark Stock

## QR-2

Quality of outcrop: Excellent
Degree of alteration: Fresh to moderate
Medium-grained equigranular monzogranite with quartz, seritised plagioclase and K-feldspar. The feldspar are pink and appear hematized. Minor phases are biotite and opaques. Minor muscovite as an alteration product.

## QR-4

Quality of outcrop: Excellent
Degree of alteration: Fresh to moderate
Medium-grained equigranular monzogranite with quartz, seritised plagioclase and K-feldspar. The feldspar are pink phenocrysts and appear hematized. Minor phases are biotite and opaques. Minor muscovite as an alteration product.

## QR-6

Quality of outcrop: Excellent
Degree of alteration: Fresh to moderate
Fine-grained equigranular monzogranite with quartz, seritised plagioclase and K-feldspar. The feldspar are pink phenocrysts and appear hematized. Minor phases are biotite and opaques. Minor muscovite as an alteration product.

## C. 1 Sample Selection, Analytical Methods and, Precision and Acuracy

## C.1.1 Sample Selection

Approximately one hundred and twenty samples were collected from the Jebilet and Rehamna Massif granites. All except those from the Ras El Abiod pluton were slabed, stained and, point counted to determine the modal quartz, K-feldspar and Plagioclase content. Thin sections of these samples were also made and examined. Based on this data, 70 representative samples were chosen for whole rock chemical analysis. Geographic location was also considered in sample selection, with the objective of obtaining a good regional coverage of the granites.

## C.1.2 Sample Pulverization

Weathered or contaminated surfaces were removed using a rock saw. The specimens were then broken down into cubes of approximately $2 \times 2 \times 2$ cm using a cut rock. Any remaining weathered or contaminated surfaces were removed at this stage. Fresh fragments were deposited into a Dayton Tow Crusher (model 4 K 731 ) which has ceramic plates. The samples were then split using a Soiltest splitter. Finally, a representative sample of approximately 200 to 300 grams was pulverized in a Siebtechnik tungsten carbide ring mill (model TS 250) until the sample was less than 100 mesh. A ceramic ring mill was used for the REE samples to avoid the tungsten contamination. All equipment were carefuly cleaned after each preparation to avoid contamination.

## C.1.3 Major and Trace Element Analysis

Whole rock trace element analysis were done at St. Mary's University by K.Cameron and major element analysis were done at McGill University by S.T. Ahmedali. Both laboratories use the same equipment and methods of sample preparation and analysis. Samples were analysed on a Philips DW 1400 sequential x-ray fluorescence spectrometer using a Phanode x-ray tube. Fused glass disks were used to determine the major elements while pressed powder pellets were used for trace elements. Loss on ignition (LOI) was determined by heating the sample for 1.5 hours at 1050 C in an electric furnace. Analytical precision as determined on replicate analyses is generally better than $5 \%$ for the major oxides and between $5-10 \%$ for trace elements.

## C.1.4 Rare Earth Element Analysis (NAA)

Ten samples were analysed for $\mathrm{Ce}, \mathrm{Nd}, \mathrm{Sm}, \mathrm{Eu}, \mathrm{Gd}, \mathrm{Tb}, \mathrm{Yb}$ and Lu . Unfortunately Lanthanum was not determined. The analyses were done at Waterloo University using neutron activation. Approximately 0.1-0.2 grams of rock powder were weighed and sealed in clean plastic containers. Two international standards and one repeat sample are considered to determine the precision and accuracy.

Table C.1. Duplicate major element analysis from the Sebt de Brikiine batholith (McGill University).

|  | SDB-4 | SDB-4D | Percent <br> Deviation |
| :--- | ---: | ---: | :---: |
| $\mathrm{SiO}_{2}$ | 76.96 | 77.55 | 0.77 |
| $\mathrm{TiO}_{2}$ | 0.07 | 0.07 | 0.00 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 12.63 | 12.58 | 0.40 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3 t}$ | 0.83 | 0.84 | 1.20 |
| MnO | 0.01 | 0.01 | 0.00 |
| MgO | 0.01 | 0.01 | 0.00 |
| CaO | 0.33 | 0.03 | 0.00 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3.93 | 3.83 | 2.54 |
| $\mathrm{~K}_{2} \mathrm{O}$ | 4.69 | 4.59 | 2.13 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | 0.02 | 0.02 | 0.00 |

Table C.3. Duplicate major element analysis from the TabouchenntBamega pluton (McGill University).

|  | BRR-16 | BRR-16D | Percent <br> Deviation |
| :--- | ---: | ---: | ---: |
| $\mathrm{SiO}_{2}$ | 70.93 | 71.66 | 1.03 |
| $\mathrm{TiO}_{2}$ | 0.43 | 0.44 | 2.33 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14.28 | 14.34 | 0.42 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3 \mathrm{t}}$ | 3.46 | 3.29 | 4.91 |
| MnO | 0.06 | 0.05 | 16.67 |
| MgO | 0.79 | 0.76 | 3.80 |
| CaO | 1.57 | 1.46 | 7.01 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 2.59 | 2.59 | 0.00 |
| $\mathrm{~K}_{2} \mathrm{O}$ | 4.88 | 4.97 | 1.84 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | 0.12 | 0.13 | 8.33 |

Table C.2. Duplicate major element analysis from the Oulad Ouaslam batholith (McGill University).

|  | JUB-6 | JUB-6D | Percent Deviation |
| :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 68.57 | 68.66 | 0.13 |
| $\mathrm{TiO}_{2}$ | 0.48 | 0.48 | 0.00 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 15.75 | 15.70 | 0.32 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3 \mathrm{t}}$ | 3.21 | 3.19 | 0.62 |
| MnO | 0.05 | 0.05 | 0.00 |
| MgO | 1.01 | 1.04 | 2.97 |
| CaO | 2.02 | 2.04 | 0.99 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3.86 | 3.84 | 0.52 |
| $\mathrm{K}_{2} \mathrm{O}$ | 3.65 | 3.63 | 0.55 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | 0.13 | 0.13 | 0.00 |

Table C.4. Duplicate major element analysis from the TabouchenntBamega pluton (McGill University).

|  | BRR-14 | BRR-14D | Percent <br> Deviation |
| :--- | ---: | ---: | :---: |
| $\mathrm{SiO}_{2}$ | 71.09 | 70.64 | 0.63 |
| $\mathrm{TiO}_{2}$ | 0.45 | 0.48 | 6.67 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14.59 | 14.69 | 0.69 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3 t}$ | 3.28 | 3.51 | 7.01 |
| MnO | 0.05 | 0.04 | 20.00 |
| MgO | 0.69 | 0.69 | 0.00 |
| CaO | 1.57 | 1.53 | 2.55 |
| Na | O | 2.65 | 2.72 |
| 2.64 |  |  |  |
| $\mathrm{~K}_{2} \mathrm{O}$ | 4.90 | 5.06 | 3.27 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | 0.12 | 0.13 | 8.33 |

Table C.5. Major element precision using the international standard G-2, granite (McGill University).

|  | Abbey, 1983 | Obtained McGill | Percent Deviation |
| :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 69.22 | 69.05 | 0.25 |
| $\mathrm{TiO}_{2}$ | 0.48 | 0.48 | 0.00 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 15.40 | 15.48 | 0.52 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3 \mathrm{t}}$ | 2.69 | 2.70 | 0.37 |
| MnO | 0.03 | 0.03 | 0.00 |
| MgO | 0.75 | 0.72 | 4.00 |
| CaO | 1.96 | 2.00 | 2.04 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 4.06 | 4.08 | 0.49 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4.46 | 4.45 | 0.22 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | 0.13 | 0.13 | 0.00 |

Table C.6. Major element precision using the international standard NIM-1, granite (McGill University).

|  | Sarm, <br> 1979 | Obtained <br> McGill | Percent <br> Deviation |
| :--- | ---: | ---: | ---: |
| $\mathrm{SiO}_{2}$ | 75.70 | 72.23 | 0.62 |
| $\mathrm{TiO}_{2}$ | 0.09 | 0.09 | 0.00 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 12.08 | 12.20 | 0.99 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3 \mathrm{t}}$ | 2.02 | 1.99 | 1.49 |
| MnO | 0.02 | 0.02 | 0.00 |
| MgO | 0.06 | 0.01 | 83.33 |
| CaO | 0.78 | 1.42 | 83.05 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3.36 | 3.25 | 3.27 |
| $\mathrm{~K}_{2} \mathrm{O}$ | 4.99 | 4.97 | 0.40 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | 0.01 | 0.01 | 0.00 |

Table C.7. Duplicate trace element analysis from the Ajar el Bark stock (St-Mary's University).

|  | $\mathrm{QR}-4$ | $\mathrm{QR}-4 \mathrm{D}$ | Percent <br> Deviation |
| :--- | ---: | :---: | ---: |
| Ba | 84 | 88 | 4.76 |
| Rb | 291 | 291 | 0.00 |
| Sr | 28 | 29 | 3.57 |
| Y | 62 | 65 | 4.84 |
| Zr | 80 | 79 | 1.25 |
| Nb | 17 | 19 | 11.76 |
| Th | 34 | 30 | 11.76 |
| Pb | 28 | 20 | 28.57 |
| Ga | 18 | 20 | 11.11 |
| Zn | 15 | 19 | 26.67 |
| Cu | 1 | 2 | 100.00 |
| Ni | 36 | 38 | 5.56 |
| $\mathrm{Ti} \mathrm{O}_{2} \mathrm{O}$ | 0.05 | 0.05 | 0.00 |
| V | 4 | 1 | 75.00 |
| Cr | 3 | 10 | 233.33 |

Table C.9. Duplicate trace element analysis from the Oulad Ouaslam batholith (St-Mary's University).

|  | $J U B-6$ | JUB-6D | Percent <br> Deviation |
| :--- | :---: | :---: | :---: |
| Ba | 778 | 770 | 1.03 |
| Rb | 130 | 128 | 1.54 |
| Sr | 631 | 631 | 0.00 |
| Y | 24 | 25 | 4.17 |
| Zr | 160 | 163 | 1.88 |
| Nb | 10 | 10 | 0.00 |
| Th | 2 | 7 | 250.00 |
| Pb | 20 | 21 | 5.00 |
| Ga | 20 | 19 | 5.00 |
| Zn | 50 | 44 | 12.00 |
| Cu | 12 | 8 | 33.33 |
| Ni | 17 | 19 | 11.76 |
| $\mathrm{Ti} \mathrm{O}_{2} \mathrm{O}$ | 0.53 | 0.52 | 1.89 |
| V | 54 | 61 | 12.96 |
| Cr | 38 | 38 | 0.00 |

Table C.8. Duplicate trace element analysis from the Sebt de Brikiine batholith (St-Mary's University).

|  | SDB-4 | SDB-4D | Percent <br> Deviation |
| :--- | ---: | ---: | :---: |
| Ba | 15 | 24 | 60.00 |
| Rb | 368 | 367 | 0.27 |
| Sr | 9 | 9 | 0.00 |
| Y | 53 | 51 | 3.77 |
| Zr | 134 | 132 | 1.49 |
| Nb | 67 | 68 | 1.49 |
| Th | 55 | 52 | 5.45 |
| Pb | 23 | 22 | 4.35 |
| Ga | 22 | 21 | 4.55 |
| Zn | 11 | 13 | 18.18 |
| Cu | 1 | 1 | 0.00 |
| Ni | 42 | 42 | 0.00 |
| $\mathrm{Ti}{ }_{2} \mathrm{O}$ | 0.06 | 0.06 | 0.00 |
| V | 0 | 3 | 100.00 |
| Cr | 9 | 5 | 44.44 |

Table C.10. Duplicate trace element analysis from the TabouchenntBamega pluton (St-Mary's University).

|  | BRR-16 | BRR-16D | Percent <br> Deviation |
| :--- | :---: | :---: | :---: |
| Ba | 360 | 344 | 4.44 |
| Rb | 259 | 258 | 0.39 |
| Sr | 74 | 76 | 2.70 |
| Y | 43 | 44 | 2.33 |
| Zr | 168 | 164 | 2.38 |
| Nb | 11 | 11 | 0.00 |
| Th | 22 | 21 | 4.55 |
| Pb | 21 | 22 | 4.76 |
| Ga | 19 | 20 | 5.26 |
| Zn | 38 | 43 | 13.16 |
| Cu | 7 | 6 | 14.29 |
| Ni | 28 | 31 | 10.71 |
| $\mathrm{Ti} \mathrm{O}_{2} \mathrm{O}$ | 0.45 | 0.45 | 0.00 |
| V | 50 | 47 | 6.00 |
| Cr | 27 | 23 | 14.81 |

Table C.11. Duplicate trace element analysis of the internal standard HFL-1 (St-Mary's University).

|  | HFL-1A | HFL-1B | Percent <br> Deviation |
| :--- | :---: | :---: | :---: |
| Ba | 892 | 903 | 1.23 |
| Rb | 217 | 215 | 0.92 |
| Sr | 207 | 206 | 0.48 |
| Y | 36 | 35 | 2.78 |
| Zr | 194 | 193 | 0.52 |
| Nb | 21 | 21 | 0.00 |
| Th | 19 | 20 | 5.26 |
| Pb | 28 | 26 | 7.14 |
| Ga | 26 | 29 | 11.54 |
| Zn | 109 | 105 | 3.67 |
| Cu | 31 | 31 | 0.00 |
| Ni | 36 | 44 | 4.35 |
| $\mathrm{Ti} \mathrm{O}_{2} \mathrm{O}$ | 1.07 | 1.07 | 0.00 |
| V | 126 | 121 | 3.97 |
| Cr | 105 | 111 | 5.71 |

Table C.12. Duplicate Rare Earth element analysis from the Tabouchennt-Bamega pluton.
(Waterloo University).

|  | BRR-7C | BRR-7CD | Percent <br> Deviation |
| :--- | ---: | ---: | :---: |
| Ce | 97.19 | 97.19 | 0.00 |
| Nd | 45.62 | 44.68 | 2.06 |
| Sm | 10.40 | 10.72 | 3.08 |
| Eu | 0.95 | 0.90 | 5.26 |
| Tb | 1.35 | 1.46 | 8.15 |
| Yb | 5.60 | 5.57 | 0.54 |
| Lu | 0.73 | 0.74 | 1.37 |

Table C.13. Rare Earth element precision using the international standard G-2, granite (Waterloo University).

|  | Abbey, <br> 1983 | Obtained <br> Waterloo | Percent <br> Deviation |
| :--- | :---: | :---: | :---: |
| Ce | 400 | 500.28 | 27.00 |
| Nd | 190 | 260.17 | 36.90 |
| Sm | 26.8 | 32.96 | 22.90 |
| Eu | 2.4 | 2.72 | 13.30 |
| Tb | 1.36 | 0.81 | 40.40 |
| Yb | 1.7 | 2.19 | 28.80 |
| Lu | - | - | - |

Table C.14. Rare Earth element precision using the international standärd NIM-1, granite (Waterloo University).

|  | Sarm, <br> 1979 | Obtained <br> Waterloo | Percent <br> Deviation |
| :--- | :---: | :---: | :---: |
| Ce | 195 | 204.26 | 4.70 |
| Nd | 72 | 80.40 | 11.20 |
| Sm | 15.8 | 15.07 | 4.60 |
| Eu | 0.35 | 0.34 | 2.80 |
| Tb | 14 | 16.16 | 15.40 |
| Yb | 3 | 4.60 | 53.30 |
| Lu | 14.2 | 14.46 | 1.80 |

## APPENDIX D

## D. 1 Source References for the Geochemical Database

## D.1.1 Nova Scotia

## Sample Prefix Author(s)

NAL De Alburquerque, 1977
NBC Bernadette, 1982
NBM
Rodgers, 1985
NBP
Smith, 1979; Rogers, 1985
$\mathrm{NCH} \quad$ Charest, 1976
NDW Dwyer, 1975
NEH
NEK
NFE
Weagle, 1983
Richardson, in prep.

NIW
Farley, 1979
,
NKA
NL.B
NLH
NLT
NLR
NMD
NMK
nMO
NOE
NPL
NPM
NPS
NSH
NSI
Nolison, 1983
Alizay, 1981
Rodgers, 1985
Ham, in prep.
Cameron, 1985
O'Reilly, 1988
McDonald, 1981; McDonald and Clarke, 1985
McKenzie, 1974; McKenzie and Clarke, 1975
Weagle, 1983
O'Reilly, 1976
Allan, 1983
De Alburquerque, 1977; Douma, 1988
Smith, 1977
Rodgers, 1985
Rodgers, 1985
NSL
O'Reilly, 1988
NTS
Smith et al., 1987
NWG
Rodgers, 1985

## D.1.2 Morocco

Sample Prefix Author(s)

MAG Mahmood, 1980
MAL Mahmood and Bennani, 1984
MAM Mahmood, 1980
MBC Analysis from D.B. Clarke (Unpubl.)
MBR This study
MHU
Huvelin, 1974
MJB
This study
MJU
This study
MME
Mahmood and Bennani, 1984

MMG
MMN
MMO
MMI
MMZ
MOU
MQR
MSD
MII

MZA

Boushaba, 1984
Boushaba, 1984
Boushaba, 1984
Boushaba, 1984
Boushaba, 1984
Mahmood and Bennani, 1984
This study
This study
Vogel and Walker, 1975; Vogel et al., 1976;
Scott and Vogel, 1980
Guiliani, 1982

## D.1.3 Iberia

Iberian analyses courtesy of J.L. Barrera.

## D.1.4 Australia

Sample Prefix Author (s)
ABB White et al., 1977

AKB
AKO
Hine et al., 1978
White et al., 1977
AMO
Chappell, 1978
AMU
ARP
ASB*
AST
AWB
TBI*

Flood and Shaw, 1977
Price and Taylor, 1977
Clemens, 1981
Philips et al., 1981
Shaw et al., 1982
Higgins et al., 1985

* Data excluded from final analysis
D. 2 Pluton Abbreviations


## D.2.1 Nova Scotia

BMOU - Bald Mountain
BPAS - Barrington Passage
BRID - Bull Ridge
BREN - Brenton
CSMB - South Mountain
DLAK - South Mountain (Davis Lake)
EHEA - Eastern Head
ELAK - Ellison Lake
HALI - South Mountain (Halifax)
HCOV - Halfway Cove
KINS - Kinsac
LBAY - Lyons Bay
IISC - Liscomb
IRIV - Larry's River

```
MLAK - Mulgrave
MUSQ - Musquodoboit
MPOI - Moose point
NROS - South Mountain (New Ross)
PMOU - Port Mouton
QUEE - Queensport
SHEL - Shelburne
SHER - Sherbrooke
SISL - Seal Island
SLAK - Sangster Lake
IURN - South Mountain (Turner)
WALK - South Mountain (Walker)
WEDG - Wedgeport
WEST - Western Granite
D.2.2 Morocoo
MENT - Ment
ZAER - Zaer
OULM - Oulmes
SDBR - Debt de Brikiine
AEBA - Ajar El Bark
TBAM - Tabouchennt-Bamega
OUOU - Oulad Ouaslam
TICH - Tichka
```


## D.2.3 Other

```
UNKN - Not specified
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## D. 3 Iithologies

```
ALAS - Alaskite
APII - Aplite
DRNI - Dyke rock minor intrusives
GRAD - Granodiorite
GRAN - Granite
IMON - Leuco monzogranite
LHON - Leuco tonalite
MONG - Monzogranite
PORP - Porphyry
SYEN - Syenogranite
TONA - Tonalite
```


## NOVA SCOTIA

|  | NAL1 BPAS TONA | NAL10 SHEL MONG | NAL 11 SHEL APLI | NAL2 BPAS TONA | NAL3 BPAS TONA | NAL6 <br> PMOU <br> GRAD | NAL7 <br> PMOU <br> GRAD | NAL8 <br> PMOU <br> MONG | NALG <br> PMOU <br> MONG |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 65.39 | 72.74 | 73. 42 | 64.13 | Б3. 93 | 72.10 | 73. 25 | 70. 12 | 70.86 |
| $\mathrm{TiO}_{2}$ | . 72 | . 19 | . 04 | . 71 | . 79 | . 23 | . 22 | . 41 | . 41 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 16.48 | 15.35 | 16. 62 | 17.75 | 17.78 | 15.15 | 15.62 | 16.44 | 16.55 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 70 | . 23 | . 35 | . 75 | . 59 | . 31 | . 24 | . 19 | . 33 |
| FeO | 3.12 | . 97 | 1.11 | 3. 12 | 3. 33 | 1.37 | 1.13 | 1.80 | 1.63 |
| MnO | . 05 | . 04 | . 28 | . 06 | . 07 | . 03 | . 03 | . 04 | . 14 |
| MgO | 1. 56 | . 28 | .10 | 1. 96 | 2. 11 | . 33 | . 42 | . 71 | . 52 |
| CaO | 3. 66 | . 82 | . 52 | 4.07 | 3. 93 | 1. 36 | 1. 33 | 1. 68 | 1.74 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3. 64 | 3. 65 | 4. 31 | 3. 94 | 3. 74 | 3. 82 | 3.79 | 3.66 | 3.73 |
| $\mathrm{K}_{2} \mathrm{O}$ | 2.18 | 3. 97 | 3.08 | 2. 19 | 2. 33 | 3. 54 | 3. 52 | 4.17 | 3.80 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 18 | . 07 | . 06 | . 25 | . 24 | . 06 | . 07 | . 13 | . 11 |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1.18 | . 77 | .40 | . 78 | 1.15 | . 83 | . 73 | . 67 | . 57 |
| $\mathrm{H}_{2} \mathrm{O}-$ | . 11 | . 04 | . 02 | .08 | .10 | . 08 | . 04 | . 05 | . 10 |
| $\mathrm{CO}_{2}$ | - | - | - | - | $-$ | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 99. 98 | 99.12 | 100.32 | 99.82 | 100.09 | 100.21 | 100. 39 | 99.97 | 100.49 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 4.16 | 1. 31 | 1.59 | 4. 21 | 4. 29 | 1.83 | 1.49 | 2. 19 | 2.14 |
| A/CNK | 1.1 | 1.3 | 1.5 | 1.1 | 1.1 | 1.3 | 1.3 | 1.2 | 1. 2 |
| DI | 69.9 | 88.8 | 89.5 | 65.9 | 65. 6 | 86. 2 | 87.3 | 83. 5 | 83.9 |
| Ba | 415. | - | - | 445. | 442. | 478. | 365. | 836. | 573. |
| Rb | 80. | - | - | 85. | 86. | 158. | 171. | 158. | 154. |
| Sr | 304. | - | - | 345. | 352. | 99. | 87. | 198. | 154. |
| Y | - | - | - | - | - | - | - | - | - |
| Zr | - | - | _ | - | - | - | - | - | - |
| Nb | - | - | - | - | - | - | - | - | - |
| Th | - | - | - | - | - | - | _ | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Go | - | - | - | - | - | - | - | - | - |
| Zn | - | - | - | - | - | - | - | - | - |
| Cu | - | - | - | - | - | - | - | - | - |
| Ni | - | - | - | - | _ | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| $V$ | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | _ | _ | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| Ta | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | - | - | - | - | - | - | - | - | - |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| $F$ | - | - | - | _ | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| Lo | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D.1. Geochemical databose

NOVA SCOTIA

|  | NMK 100A CSME DRMI | NMK102 CSMB MONG | NMK1 14 CSMB MONG | NMK116A CSMB DRMI | NMK1 19 CSMB MONG | NMK121 CSMB MONG | NMK123 CSMB DRMI | NMK124 CSMB GRAD | NMK127 CSMB GRAD |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 75.00 | 73. 70 | 74.50 | 74.10 | 74.30 | 13. 50 | 74.00 | 88.80 | 72. 20 |
| $\mathrm{TiO}_{2}$ | . 09 | . 25 | . 12 | . 09 | . 22 | . 18 | . 18 | . 58 | . 42 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14. 20 | 13.42 | 13.40 | 14.07 | 13.08 | 13.87 | 13. 20 | 14. 52 | 14.25 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 58 | . 17 | . 16 | . 23 | . 05 | 22 | 17 | 45 | 25 |
| FeO | . 62 | 1.69 | 1. 51 | . 82 | 1.77 | 1. 55 | 1.09 | 3.88 | 2.52 |
| MnO | . 03 | . 04 | . 05 | . 03 | . 05 | . 03 | . 03 | . 10 | . 09 |
| MgO | . 08 | . 28 | . 12 | . 10 | . 24 | . 24 | . 08 | 1.18 | . 75 |
| CaO | . 53 | . 58 | . 45 | . 14 | . 66 | . 60 | . 43 | 2.21 | 1.44 |
| $\mathrm{No}_{2} \mathrm{O}$ | 4.57 | 3.32 | 3. 46 | 3.84 | 3. 60 | 3.54 | 4. 00 | 3.18 | 3.31 |
| $\mathrm{K}_{2} \mathrm{O}$ | 3.80 | 4.91 | 4.59 | 4. 37 | 4. 34 | 4.72 | 4.31 | 3. 46 | 4. 04 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 28 | . 07 | . 14 | .14 | . 07 | . 15 | . 21 | . 10 | . 16 |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1.01 | . 82 | . 58 | . 79 | . 77 | . 69 | . 68 | . 91 | . 88 |
| $\mathrm{H}_{2} \mathrm{O}$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | . 14 | . 13 | . 28 | . 33 | . 10 | . 18 | . 26 | . 05 | . 06 |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 100.73 | 99.39 | 99. 35 | 99. 35 | 99.25 | 99.47 | 98.84 | 99.52 | 100.39 |
| $\mathrm{Fe}_{2} \mathrm{O} 3 \mathrm{t}$ | 1.27 | 2.05 | 1. 84 | 1.14 | 2.01 | 1.94 | 1. 38 | 4.76 | 3.06 |
| A/CNK | 1.2 | 1.1 | 1.2 | 1.2 | 1.1 | 1.2 | 1.1 | 1.1 | 1.2 |
| 01 | 94.8 | 90.8 | 91.8 | 92.8 | 90.7 | 91.0 | 93. 0 | 76.2 | 84.6 |
| Ba | 8. | 230. | 66. | 16. | 200. | 164. | 28. | 658. | 429. |
| Rb | 402. | 330. | 532. | 650. | 299. | 411. | 620. | 143. | 174. |
| Sr | - | - | - | - | - | - | - | 155. | 112. |
| Y | - | - | - | - | - | - | - | - | - |
| Zr | - | 91. | 51. | - | 76. | 64. | 20. | 210. | 168. |
| Nb | - | - | - | - | - | - | - | - | - |
| Th | - | - | 10.90 | 2.05 | 11.00 | - | 4.43 | - | 8. 33 |
| Pb | - | - | - | - | - | - | - | - | - |
| Go | - | - | - | - | - | - | - | - | - |
| Zn | 33.0 | 72.0 | 64.0 | 43.0 | 56.0 | 64. 0 | 64.0 | 71.0 | 60.0 |
| Cu | - | - | - | - | - | - | - | - | - |
| Ni | 9. 0 | 8.0 | 8.0 | 9.0 | 10.0 | 9.0 | 9.0 | 16.0 | 12.0 |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| V | - | - | - | - | - | - | - | - | - |
| Cr | 42. | 43. | 45. | 37. | 47. | 47. | 38. | 50. | 42. |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | - | - | - | - | - | - | - | - | - |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | - | - | - | - |  |  |  |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | 22.0 | 19.0 | 30.0 | 40.0 | 15.0 | 20.0 | 25.0 | 7.0 | 18.0 |
| Mo | - | - | - | - | - | - | - | - | - |
| Lo | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.


Figure D. 1 (cont.). Geochemical database.

NOVA SCOTIA

|  | NBCO10 <br> BRID <br> MONG | NBCO11 <br> BRID <br> MONG | NBCO 12 <br> BRID <br> MONG | NBM106 <br> BMOU <br> MONG | NBM148 <br> BMOU <br> MONG | NEM193 <br> BMOU MONG | NBMA52 <br> BMOU <br> MONG | NBP2J3 BPAS TONA | NBP243 <br> BPAS <br> TONA |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 73.40 | 72. 20 | 72.80 | 73. 34 | 74.11 | 73. 23 | 71.60 | 69.82 | 61.95 |
| $\mathrm{TiO}_{2}$ | . 23 | . 22 | . 18 | . 17 | . 17 | . 22 | . 20 | . 68 | . 94 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 15. 0 | 16. 30 | 15.10 | 14.84 | 14. 21 | 14.54 | 15.30 | 15. 12 | 17.83 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 30 | . 04 | . 15 | 1.30 | 1.45 | 1. 59 | 1.40 | 3. 49 | 5. 95 |
| FeO | . 90 | .90 | . 86 | - | - | - | - | - | - |
| MnO | . 02 | . 03 | . 01 | . 03 | . 04 | . 04 | . 02 | . 05 | . 12 |
| MgO | . 50 | . 39 | . 33 | . 18 | . 22 | . 29 | . 67 | . 97 | 2. 18 |
| CaO | . 5 日 | . 46 | . 60 | . 83 | . 65 | . 67 | . 68 | 2. 63 | 4.03 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3. 62 | 4. 06 | 1. 01 | 3. 60 | 3. 50 | 3. 33 | 3.94 | 3. 64 | 3.43 |
| $\mathrm{K}_{2} \mathrm{O}$ | 5. 21 | 5.14 | 4.86 | 4.49 | 4.50 | 4.49 | 4. 53 | 2. 45 | 2. 58 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 32 | . 31 | . 28 | . 33 | . 33 | . 32 | . 30 | . 13 | . 16 |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{Cl}^{2}$ | - | - | _ | _ | - | _ | - | - | - |
| F | _ | - | - | - | - | - | - | - | - |
| LOI | . 80 | . 90 | .80 | .99 | . 84 | . 86 | 1.00 | . 55 | - |
| TOTAL | 100.88 | 100.95 | 99. 98 | 99.90 | 100. 02 | 99. 58 | 99.64 | 99. 43 | 99.17 |
| $\mathrm{Fe}_{2} \mathrm{O}_{31}$ | 1.30 | 1.04 | 1.10 | 1. 30 | 1.45 | 1.59 | 1.40 | 3.49 | 5. 95 |
| A/CNK | 1.2 | 1.3 | 1.2 | 1.2 | 1.2 | 1.3 | 1. 2 | 1.1 | 1.1 |
| DI | 92.3 | 92.3 | 92.0 | 91.5 | 91.9 | 90.4 | 89.8 | 77.8 | 65.0 |
| Bo | 350. | 500. | 650. | 453. | 352. | 620. | 410. | 1108. | 1022. |
| Rb | 255. | 295. | 170. | 219. | 196. | 190. | 174. | 70. | 93. |
| Sr | 54. | 61. | 80. | 59. | 70. | 60. | 60. | 359. | 396. |
| $Y$ | - | - | - | 11. | 15. | - | 10. | 25. | 23. |
| Zr | - | - | - | 74. | 74. | - | 62. | 281. | 246. |
| Nb | - | - | - | 9. | 11. | - | 8. | 10. | 12. |
| Th | 7.80 | 4. 00 | 7.00 | 6. 30 | 6. 30 | - | 7.70 | 2.10 | 2.80 |
| Pb | 16. | 29. | 22. | 23. | 30. | - | 24. | 14. | 13. |
| Ga | - | - | - | 19. | 17. | - | 18. | 18. | 22. |
| Zn | 59.0 | 62.0 | 40.0 | 18.0 | 39.0 | - | 44.0 | 48.0 | 74.0 |
| Cu | 8.0 | 11.0 | 11.0 | - |  | - | 1 |  | - |
| Ni | - | - | 1 | 3.0 | 1.0 | - | 3.0 | 1.0 | 16.0 |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| $V$ | - | - | - | 8. | 5. | - | 2. | 52. | 135. |
| Cr | - | - | - | 47. | 39. | - | 37. | 38. | 85. |
| Hf | - | - | - | , |  | - | . |  |  |
| Cs | - | - | - | - | _ | - | _ | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 144.0 | 130.0 | 51.0 | 154.0 | 134.0 | 170.0 | 158.0 | 45.0 | 48.0 |
| Be | 5.6 | 14.0 | 8.5 | 5. 2 | 5. 2 | - | 7.4 | 2.8 | 1.0 |
| 8 | 3. 0 | 4.0 | 6. 0 | - | S. | - | . | , | , |
| $F$ | 380. | 420. | 250. | 330. | 380. | 370. | 430. | 460. | 60. |
| Cl | - | - | - | - | - | - | - | - | $\rightarrow$ |
| U | 4. 50 | 3.40 | 3.60 | 6. 10 | 5.90 | - | 5.70 | 2.80 | . 90 |
| W | , | - | - | - | - | - | - | - | - |
| Sn | 5.9 | 7.4 | 5.2 | 8.0 | 14.0 | - | 5. 0 | 7. 0 | 14.0 |
| Mo | . 60 | . 70 | 1.00 | - | - | - | - | - | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.), Geochemical databose.

NOVA SCOTIA

|  | NBP265 BPAS TONA | NBP3O2 BPAS TONA | NBP344 BPAS TONA | NBP345 BPAS TONA | NBP348 EPAS <br> TONA | NBP361 BPAS TONA | NBP364 BPAS <br> TONA | NBP386 BPAS TONA | NBP387 BPAS TONA |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 67.27 | 65. 94 | 64. 28 | 57.60 | 63. 32 | 65.01 | 68. 73 | 67.83 | 60.91 |
| $\mathrm{TiO}_{2}$ | . 73 | 76 | . 74 | 65 | 18 | 85 | . 78 | 64 | 1. 07 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 15. 45 | 15. 20 | 16.92 | 15.91 | 18.93 | 15. 57 | 15.08 | 15. 55 | 18.08 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 4.10 | 4.47 | 4. 96 | 4.50 | 3.67 | 5. 46 | 4.45 | 4.52 | 6.74 |
| FeO | - | - | - | - | - | - | - | - | - |
| MnO | . 87 | . 11 | . 06 | . 08 | . 07 | . 10 | . 06 | . 11 | . 11 |
| MgO | 2.10 | 2. 08 | 2. 39 | 2. 12 | 1.70 | 2. 30 | 1.61 | 2.10 | 2.43 |
| CoO | 3.08 | 3.59 | 3.30 | 3.42 | 4.13 | 3. 28 | 2. 36 | 3.05 | 3. 57 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3.81 | 3.98 | 3.64 | 4.09 | 4. 81 | 2. 76 | 3.11 | 3.05 | 3.36 |
| $\mathrm{K}_{2} \mathrm{O}$ | 1.96 | 2. 03 | 2. 20 | 1.98 | 2. 57 | 2. 48 | 2.13 | 1.81 | 2.74 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 16 | . 26 | . 23 | . 38 | . 82 | . 15 | . 05 | . 16 | . 23 |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | 1.61 | . 61 | 98 | . 69 | . 57 | . 81 | 90 | . 71 | . 88 |
| TOTAL | 100. 34 | 100.03 | 99. 70 | 101.43 | 101.07 | 99. 78 | 99.27 | 99.83 | 100.12 |
| $\mathrm{Fe}_{2} \mathrm{O} 3 \mathrm{t}$ | 4.10 | 4. 17 | 4. 96 | 4.50 | 3.67 | 5. 46 | 4. 46 | 4. 62 | 6.74 |
| A/CNK | 1.1 | 1.1 | 1.2 | 1.1 | 1.0 | 1. 2 | 1.3 | 1.2 | 1.2 |
| DI | 72.2 | 70. 6 | 68.5 | 73.1 | 72. 3 | 68.4 | 74.1 | 70.5 | 64.9 |
| Bo | 355. | 401. | 563. | - | 701. | 438. | 718. | 560. | 918. |
| Rb | 79. | 74. | 124. | - | 86. | 181. | 81. | 61. | 105. |
| Sr | 341. | 278. | 80. | - | 34. | 63. | 311. | 339. | 351. |
| Y | 10. | 16. | 15. | - | 33. | 12. | 10. | 18. | 26. |
| Zr | 191. | 174. | 51. | - | 148. | 88. | 191. | 198. | 245. |
| Nb | 11. | 13. | 8. | - | 14. | 8. | 11. | 10. | 12. |
| Th | 2. 50 | 3. 00 | 2. 00 | 3. 20 | 1.90 | 2. 40 | 1. 50 | . 40 | 4. 10 |
| Pb | 15. | 12. | 29. | - | 17. | 24. | 18. | 18. | 22. |
| Go | 20. | 19. | 15. | - | 20. | 21. | 18. | 18. | 23. |
| Zn | 71.0 | 69.0 | 37.0 | - | 55.0 | 77.0 | 57.0 | 57. 0 | 81.0 |
| Cu | - | - | - | - | - | - | - | - | - |
| Ni | 12.0 | 12.0 | 3.0 | - | 11.0 | 5. 0 | 8.0 | 11.0 | 22.0 |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| V | 93. | 83. | 1. | - | 54. | 9. | 74. | 83. | 136. |
| Cr | 95. | 80. | 27. | - | 44. | 37. | 79. | 56. | 92. |
| Hf | - | - | 27. | - | - | , | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 55.0 | 19.0 | 61.0 | 50.0 | 18.0 | 78. 0 | 41.0 | 37.0 | 52.0 |
| Be | 2.7 | 2.3 | . 5 | 3.1 | - | 1.1 | . 9 | 2.6 | 3.4 |
| B | - | - | - | - | - | - | - | - | - |
| F | 360. | 410. | 670. | 870. | - | 520. | 490. | 490. | 840. |
| Cl | - | - | - | - | - | - | - | - | - |
| U | 1.10 | 1.90 | 1.60 | 1. 20 | 2.00 | 1.10 | 1.70 | 1.20 | 2.00 |
| W |  | - |  | - |  | , | - | - | - |
| Sn | 5.0 | 7. 0 | 5. 0 | - | 2. 0 | 13.0 | 11.0 | 11.0 | 14.0 |
| Mo | - | - | - | - | - | - | - | - | - |
| Lo | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical databose.

NOVA SCOTIA

|  | NBP39 1 BPAS TONA | NBP403 BPAS TONA | NEP405 BPAS TONA | NBP415 BPAS TONA | NBP418 BPAS GRAD | NBP531 BPAS TONA | NBP560 BPAS TONA | NBP562 BPAS TONA | NBP594 BPAS TONA |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 70.03 | 65.59 | 65.75 | 68.95 | 72. 39 | 63.37 | 54. 42 | 85. 13 | 66.20 |
| $\mathrm{TiO}_{2}$ | . 29 | . 67 | . 81 | . 51 | . 42 | . 95 | 85 | 90 | . 71 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 16.88 | 15.83 | 15.58 | 15.02 | 14.14 | 16.52 | 16.33 | 15. 73 | 15.30 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 1.62 | 4. 51 | 4.85 | 3. 64 | 2.54 | 6. 04 | 5. 32 | 5. 81 | 4. 67 |
| FeO | 1.62 | 4. 5 | 4. | 3. 8 | 2. | 6. | 2 |  |  |
| Mno | . 04 | . 08 | . 08 | . 08 | . 07 | . 12 | 10 | . 14 | . 08 |
| MgO | . 81 | 2.20 | 2. 20 | 1. 65 | 1.16 | 2. 08 | 2. 31 | 2. 39 | 2.41 |
| CoO | 2. 98 | 2. 98 | 2. 98 | 3. 03 | 2. 38 | 2.90 | 3. 47 | 3. 27 | 3. 34 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 5. 44 | 3. 63 | 3.63 | 4. 07 | 3.55 | 3. 55 | 3. 24 | 3.19 | 3.75 |
| $\mathrm{K}_{2} \mathrm{O}$ | 1.15 | 2.67 | 2. 67 | 1.66 | 1. 94 | 2.76 | 2.70 | 2.30 | 2.32 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 06 | . 25 | . 26 | . 17 | .11 | . 47 | . 17 | . 20 | . 24 |
| $\mathrm{H}_{2} \mathrm{O}+$ | . | - | - | - | - | - | - | - |  |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | _ | _ | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - |  |  | - |
| LOI | . 49 | . 69 | . 99 | . 84 | . 87 | 1.07 | 83 | . 88 | . 51 |
| TOTAL. | 99.59 | 100.10 | 99. 81 | 99. 63 | 99. 58 | 99.83 | 99.74 | 99, 74 | 100. 63 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 1.82 | 4. 51 | 4. 86 | 3.64 | 2.54 | 6. 04 | 5. 32 | 5.61 | 4. 67 |
| A/CNK | 1.1 | 1.1 | 1.1 | 1.1 | 1.2 | 1.2 | 1.1 | 1.2 | 1.1 |
| Ol | 79.4 | 72.8 | 72.0 | 74.7 | 79.3 | 70.5 | 88.2 | 68.4 | 70. 9 |
| Ba | 156. | 270. | 545. | 255. | 348. | 690. | 843. | 735. | 423. |
| Rb | 36. | 92. | 105. | 79. | 74. | 110. | 95. | 85. | 111. |
| Sr | 720. | 250. | 250. | 282. | 226. | 306. | 390. | 339. | 274. |
| Y | 7. | 15. | 17. | 9. | 7. | 35. | 17. | 16. | 22. |
| Zr | 125. | 166. | 165. | 154. | 107. | 234. | 215. | 189. | 159. |
| Nb | 3. | 13. | 17. | 15. | 9. | 18. | 10. | 13. | 12. |
| Th | 2. 70 | 1. 60 | 4. 30 | 1. 50 | 4. 70 | 5. 30 | 1.90 | 4. 30 | 3. 50 |
| Pb | 11. | 18. | 16. | 12. | 17. | 23. | 17. | 11. | 13. |
| Ga | 21. | 19. | 20. | 19. | 15. | 21. | 19. | 18. | 23. |
| Zn | 44.0 | 75.0 | 80.0 | 18.0 | 52.0 | 84.0 | 65. 0 | 78. 0 | 78.0 |
| Cu | - | - | - | - | - | - | - | - | - |
| $\mathrm{Ni}^{\text {Ni}}$ | 7. 0 | 10.0 | 15.0 | 18.0 | 10.0 | 23.0 | 21.0 | 17.0 | 15.0 |
| $\mathrm{TiO}_{2}$ | , | - | , | - | - | - | 21.0 | 17.0 | - |
| V | 33. | 90. | 102. | 65. | 48. | 103. | 114. | 111. | 84. |
| Cr | 40. | 75. | 97. | 84. | 85. | 73. | 81. | 81. | 74. |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 42.0 | 58.0 | 60.0 | 40.0 | 52.0 | 57. 0 | 34.0 | 152.0 | 59.0 |
| El | 2.5 | 2.6 | 3. 6 | 3. 6 | 2.6 | 3. 2 | 2.3 | 2.2 | - |
| B | - | - | - | - | - | - | - | - | - |
| F | 810. | 430. |  | 450. | 340. | 490. | 560. | 410. | 410. |
| Cl | - | - | - |  | - | - | - | - | - |
| U | 2. 00 | 3.00 | 1.00 | . 80 | 2.00 | 1.90 | 1.20 | 1. 40 | 2.80 |
| W | 2.00 | , | 1. | , | - | - | - | - | - |
| Sn | 5.0 | 3. 0 | 4.0 | 1.0 | 12.0 | 10.0 | 8. 0 | - | 3.0 |
| Mo | - | - | - | - | - | - | - | - | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical databose.

## NOVA SCOTIA

|  | NBP608 BPAS TONA | NBP610 gPAS TONA | N8P633 BPAS TONA | NBP642 BPAS TONA | NBP652 BPAS TONA | NBP665 BPAS TONA | NBPAO7 BPAS TONA | NBPAO8 BPAS TONA | NBPA13 BPAS TONA |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 65.85 | 67.08 | 70.77 | 67.10 | 69. 74 | 57.08 | 65.40 | 70.13 | 65.80 |
| $\mathrm{TiO}_{2}$ | . 58 | . 84 | . 69 | . 57 | . 57 | 57 | 64 | . 49 | . 56 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 15.98 | 15.80 | 14.19 | 16.15 | 14. 91 | 14.91 | 15.80 | 14.94 | 15.90 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 4.55 | 4. 31 | 4. 60 | 4.75 | 3. 54 | 3.64 | 4. 55 | 3.02 | 4. 28 |
| Feo | - | - | - | - | - | - | - | - | - |
| MnO | 09 | 09 | . 07 | . 09 | . 08 | . 08 | . 05 | . 07 | . 07 |
| MgO | 2. 29 | 2.08 | 1.64 | 2. 53 | 1. 00 | 1.00 | 1.95 | 1.47 | 1.77 |
| CaO | 3.22 | 3.23 | 1.36 | 3. 18 | 1. 98 | 1. 96 | 3. 29 | 2. 45 | 2. 81 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3. 94 | 4.07 | 2. 75 | 3. 66 | 3. 34 | 3. 34 | 4.05 | 3. 90 | 4. 06 |
| $\mathrm{K}_{2} \mathrm{O}$ | 2.06 | 1.89 | 2.57 | 2. 49 | 3.55 | 3.55 | 2. 02 | 2.32 | 2. 70 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 23 | . 20 | . 12 | . 32 | . 33 | . 33 | . 21 | . 17 | . 21 |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | - | - | - | - | - | - | - |  |
| $\mathrm{H}_{2} \mathrm{O}$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | . 63 | 53 | 1.05 | . 65 | . 98 | 98 | . 31 | . 83 | . 85 |
| TOTAL | 99.53 | 99. 90 | 99. 81 | 101.61 | 100. 10 | 97.42 | 99. 28 | 99. 79 | 100. 01 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 4.55 | 4. 31 | 4. 60 | 4. 76 | 3.64 | 3. 64 | 4. 55 | 3.02 | 4. 28 |
| A/CNK | 1.1 | 1.1 | 1.5 | 1.1 | 1.2 | 1.2 | 1.1 | 1.1 | 1.1 |
| DI | 70.9 | 72.4 | 78.4 | 72.3 | 81.2 | 78.5 | 71.9 | 78. 3 | 75.1 |
| Bo | 233. | 225. | 650. | 532. | 156. | - | 380. | - | 380. |
| Rb | 101. | 91. | 85. | 105. | 36. | - | 60. | - | 110. |
| Sr | 239. | 231. | 225. | 278. | 720. | - | 320. | - | 290. |
| Y | 22. | 19. | 25. | 31. | 7. | - | - | - | - |
| Zr | 157. | 156. | 217. | 178. | 125. | - | - | - | - |
| Nb | 15. | 14. | 14. | 13. | 3. | - | - | - | - |
| Th | 1.70 | 3.60 | 3.40 | 3. 50 | 2.70 | - | - | - | - |
| Pb | 12. | 17. | 15. | 13. | 11. | - | - | - | - |
| Ga | 20. | 22. | 18. | 25. | 21. | - | - | - | - |
| Zn | 78.0 | 68.0 | 65.0 | 77. 0 | 44.0 | - | - | - | - |
| Cu | - | - | - | - | - | - | - | - | - |
| $\stackrel{\mathrm{Ni}}{ }$ | 21.0 | 15.0 | 23.0 | 17.0 | 7.0 | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | . | , | - | - | - | - |
| $\checkmark$ | 74. | 65. | 73. | 85. | 33. | - | - | - | - |
| Cr | 73. | 69. | 84. | 76. | 40. | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 59.0 | ถ3. 0 | 117.0 | 54.0 | 12.0 | - | 54.0 | - | 71.0 |
| Be | 3.3 | 2. 8 | 2. 9 | 3.4 | 2.5 | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| F | 110. | 380. | 340. | 430. | 810. | - | 520. | - | 500. |
| Cl | - | - | - | - | - | - | - | - | - |
| U | 2. 40 | 2. 20 | 1.70 | 2. 00 | 2.00 | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | 5.0 | 7.0 | 8. 0 | - | 5. 0 | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure 0.1 (cont.). Geochemical database.

## NOVA SCOTIA

|  | NBPA19 BPAS TONA | NBPA2 1 BPAS TONA | NCH 10 NROS DRMI | NCH1 13 <br> NROS <br> DRMI | NCH1 15 <br> NROS <br> DRMI | NCH 117 <br> NROS <br> DRMI | NCH 12 <br> NROS <br> DRMI | $\mathrm{NCH}_{3}$ <br> NROS <br> DRMI | $\mathrm{NCH1} 40$ <br> NROS <br> DRMI |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 65.10 | 67. 60 | 72.90 | - | 74. 22 | 74.75 | 74. 54 | - | - |
| $\mathrm{TiO}_{2}$ | . 66 | . 71 | . 03 | - | . 06 | . 08 | . 05 | - | - |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 15.10 | 15.50 | 14.77 | - | 14.14 | 13.98 | 13.92 | - | - |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 4.37 | 5.21 | . 22 | - | . 22 | . 22 | . 22 | - | - |
| FeO | - | - | . 67 | - | . 83 | . 87 | . 72 | - | - |
| MnO | . 09 | . 07 | . 03 | - | . 03 | . 02 | . 02 | - | - |
| MgO | 2. 07 | 1.49 | . 08 | - | . 03 | . 15 | . 08 | - | - |
| CaO | 3.42 | 2. 41 | . 70 | - | . 36 | . 43 | . 54 | - | - |
| $\mathrm{Na}_{2} \mathrm{O}$ | 4.04 | 3. 83 | 4. 39 | - | 4. 27 | 4.43 | 4.92 | - | - |
| $\mathrm{K}_{2} \mathrm{O}$ | 2. 29 | 2. 32 | 4. 22 | - | 4.84 | 5. 18 | 4. 25 | - | - |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 22 | . 15 | . 38 | - | . 17 | . 17 | . 23 | - | - |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | - | . 76 | - | . 89 | . 74 | . 76 | - | - |
| $\mathrm{H}_{2} \mathrm{O}$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | 1.00 | . 85 | - | - | - | - | - | - | - |
| TOTAL | 100.36 | 100.15 | 99.15 | - | 100.06 | 101.02 | 100.25 | - | - |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 4.37 | 5. 21 | . 96 | .00 | 1.14 | 1.19 | 1.02 | . 00 | . 00 |
| A/CNK | 1.1 | 1.2 | 1.1 | - | 1.1 | 1.0 | 1.0 | - | - |
| DI | 71.8 | 75.6 | 93. 0 | - | 95. 5 | 96. 7 | 96.0 | - | - |
| Ba | 320. | 540. | 32. | 9. | 64. | 72. | 15. | 17. | 20. |
| Rb | 80. | 90. | 608. | 673. | 684. | 109. | 626. | 366. | 676. |
| Sr | 350. | 400. | 9. | 2. | 8. | 19. | 4. | 8. | 4. |
| $Y$ | - | - | - | - | - | - | - | - | - |
| Zr | 200. | 360. | 51. | 43. | 53. | 41. | 43. | 45. | 31. |
| Nb | - | - | 15. | 15. | 16. | 11. | 16. | 10. | 15. |
| Th | 6. 00 | 12.00 | - | - | - | - | - | - | - |
| Pb | 6. | 8. | - | - | - | - | - | - | - |
| Go | - | - | - | - | - | - | - | - | - |
| Zn | 77.0 | 日1.0 | - | - | - | - | - | - | - |
| Cu | - | - | 5. 0 | 9. 0 | 59.0 | 4.0 | 8.0 | 5. 0 | 6. 0 |
| Ni | 22.0 | 21.0 | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | , | - | - | - | - | - | - | - | - |
| $V$ | - | - | - | - | - | - | - | - | $\rightarrow$ |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 59.0 | 53. 0 | 363.0 | 450.0 | 281.0 | 801.0 | 289.0 | 123.0 | 411.0 |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| $F$ | 400. | 450. | - | - | - | _ | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | 2. 20 | 2. 70 | - | - | - | - | - | - | - |
| W | . | - | - | _ | - | - | - | - | - |
| Sn | 3. 0 | 3. 0 | 24.0 | 39. 0 | 40.0 | 20.0 | 32.0 | 18.0 | 45.0 |
| Mo | - | - | 2. 00 | 1.00 | 1.00 | 2. 00 | 1.00 | 1.00 | 2. 00 |
| Lo Ce | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure 0.1 (cont.). Geochemical database.


Figure D. 1 (cont.). Geochemical database.

NOVA SCOTIA

|  | NCH19 NROS MONG | NCH19A NROS LMON | NCH 2 <br> NROS <br> PORF | NCH2O NROS MONG | NCH 21 <br> NROS <br> DRMI | NCH 22 NROS DRMI | NCH 23 A NROS DRNI | NCH3 NROS DRMI | NCH3A NROS MONG |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 75.18 | - | 76.01 | 74. 21 | 74.87 | 74. 01 | 74.00 | - | 75.93 |
| $\mathrm{TiO}_{2}$ | . 21 | - | . 10 | . 12 | . 08 | . 03 | . 03 | - | . 09 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 12.70 | - | 13. 26 | 13.93 | 13. 52 | 14. 64 | 13.80 | - | 13.81 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 11 | - | . 15 | . 11 | . 22 | . 22 | . 22 | - | . 11 |
| FeO | 1. 53 | - | 1.15 | 1.14 | . 86 | . 77 | 55 | - | . 96 |
| MnO | . 03 | - | . 03 | . 02 | . 02 | . 02 | . 03 | - | . 02 |
| MgO | . 31 | - | 15 | . 21 | . 11 | . 07 | . 08 | - | . 11 |
| CaO | . 55 | - | 15 | 40 | 59 | . 49 | . 74 | - | . 38 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3. 31 | - | 3.27 | 3.58 | 4.00 | 4.70 | 4.13 | - | 4.15 |
| $\mathrm{K}_{2} \mathrm{O}$ | 3.95 | - | 4.98 | 5. 18 | 4.24 | 3. 95 | 4.01 | - | 4.75 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 14 | - | . 20 | . 15 | . 22 | . 28 | . 32 | - | . 20 |
| $\mathrm{H}_{2} \mathrm{O}+$ | . 89 | - | . 83 | . 72 | . 67 | . 82 | . 92 | - | . 72 |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | . | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 98.91 | - | 100.58 | 99.77 | 99.40 | 100.00 | 98.8! | - | 101. 24 |
| $\mathrm{Fe}_{2} \mathrm{O} 3 \mathrm{t}$ | 1.81 | . 00 | 1.43 | 1.38 | 1.17 | 1.07 | . 83 | . 00 | 1.18 |
| A/CNK | 1.2 | - | 1.1 | 1.1 | 1.1 | 1.1 | 1.1 | - | 1.1 |
| DI | 90.6 | - | 94.3 | 93. 6 | 93.5 | 94.5 | 93.0 | - | 98.4 |
| Ba | 180. | 17. | 34. | 145. | 326. | 20. | - | 19. | 85. |
| Rb | 278. | 931. | 723. | 356. | 289. | 820. | - | 640. | 680. |
| Sr | 34. | 8. | 9. | 24. | 48. | 7. | - | 8. | 10. |
| Y | - | - | - | - | - | - | - | - | - |
| Zr | 94. | 43. | 69. | 70. | 72. | 35. | - | 38. | 50. |
| Nb | 10. | 17. | 15. | 11. | 12. | 20. | - | 15. | 14. |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Go | - | - | - | - | - | - | - | - | - |
| Zn | - | - | - | - | - | - | - | - | - |
| Cu | 7.0 | 15.0 | 8. 0 | 10.0 | 7. 0 | 4.0 | - | 6. 0 | 4.0 |
| $\stackrel{\mathrm{Ni}}{ }$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| $\checkmark$ | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 221.0 | 439.0 | 325.0 | 131.0 | 154.0 | 493.0 | - | 353.0 | 138.0 |
| Be | - | - | , | - | - | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| w | - | 1.0 | 1.0 | - | - | - | - | - | - |
| Sn | 35.0 | 43.0 | 25.0 | ¢. 0 | 31.0 | 49.0 | - | 47.0 | 35.0 |
| Mo | 3.00 | 2. 00 | 4.00 | 1.00 | 1.00 | 2.00 | - | 2.00 | 2.00 |
| La |  | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.


Figure 0.1 (cont.). Geochemical database.


Figure D. 1 (cont.). Geochemical database.


Figure D. 1 (cont.). Geochemical databose.


Figure D. 1 (cont.). Geochemical dotabose.

|  | $\forall \circlearrowleft \vee A$ |  |  |  | $S C$ | $T A$ |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | NKAO28 <br> SHER <br> MONG | NKAO29 SHER MONG | NKAO49 <br> SHER <br> MONG | NKAO49A <br> SHER <br> MONG | NKAOS2 SHER MONG | NKAO62 <br> SHER <br> MONG | NKA065 <br> SHER <br> MONG | NKAO69 SHER MONG | NKA069A SHER MONG |
| $\mathrm{SiO}_{2}$ | 71.00 | 73.79 | 75. 06 | 73.83 | 73. 55 | 72.74 | 74.56 | 75.82 | 74. 50 |
| $\mathrm{TiO}_{2}$ | . 29 | . 15 | . 05 | . 02 | . 32 | . 8 | . 24 | . 18 | . 27 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 15.50 | 15.50 | 15.04 | 14.46 | 14.45 | 16.07 | 14. 35 | 13. 33 | 14.08 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 39 | . 19 | . 14 | . 10 | . 30 | . 11 | . 32 | . 25 | . 34 |
| FeO | 1. 30 | . 70 | . 51 | . 38 | 1. 08 | . 41 | 1. 15 | . 89 | 1.24 |
| MnO | . 04 | . 03 | . 03 | . 02 | . 60 | . 02 | . 03 | . 02 | . 45 |
| MgO | . 62 | . 23 | . 08 | . 09 | . 04 | . 11 | . 48 | . 31 | . 03 |
| CaO | . 66 | . 54 | .41 | . 38 | . 60 | . 60 | . 76 | . 50 | . 40 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3. 45 | 3.74 | 4. 12 | 4. 98 | 3. 19 | 3. 81 | 3.06 | 2. 57 | 3.44 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4.52 | 4.82 | 3. 96 | 3. 67 | 4.40 | 5.94 | 4.47 | 4.83 | 4.83 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 27 | . 25 | . 31 | . 32 | . 12 | . 20 | . 23 | . 18 | . 10 |
| $\mathrm{H}_{2} \mathrm{O}+$ | . |  |  | , | - | , | . | - | - |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | 1.15 | . 82 | . 85 | . 64 | 1.20 | . 71 | . 79 | . 69 | . 98 |
| TOTAL | 99.19 | 100.76 | 100.57 | 98. 89 | 99.85 | 100.81 | 100.44 | 99. 57 | 100.67 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3 t}$ | 1.83 | . 97 | .71 | . 52 | 1.50 | . 57 | 1. 60 | 1.24 | 1.72 |
| A/CNK | 1.3 | 1.3 | 1.3 | 1.1 | 1. 3 | 1.2 | 1.3 | 1.3 | 1. 2 |
| DI | 87.2 | 92. 5 | 93.7 | 94.2 | 89.1 | 93.5 | 89.6 | 91.1 | 91.9 |
| Bo | 240. | 227. | 31. | - | 206. | 667. | 312. | 294. | 271. |
| Rb | 248. | 256. | 374. | 415. | 241. | 249. | 216. | 232. | 229. |
| Sr | 67. | 61. | 24. | 30. | 60. | 105. | 90. | 75. | 64. |
| $Y$ |  | - |  | - |  | , | - | . | - |
| Zr | 102. | 83. | 66. | 29. | 74. | 83. | 109. | 97. | 74. |
| Nb | 7. | 10. | 1. | 5. | 10. | - | 12. | 31. | 6. |
| Th | 3. 00 | 5.00 | 9. 00 | 4. 00 | 21.00 | 2. 00 | 8. 00 | 6. 00 | 25.00 |
| Pb | 55. | 37. | 31. | 40. | 48. | 39. | 34. | 31. | 74. |
| Ga | - | - | - | - | - | - | - | - | - |
| Zn | 15.0 | 14.0 | 17.0 | 15.0 | 72.0 | 16.0 | 51.0 | 29.0 | 49.0 |
| Cu | 15 | - | - | - | 2.0 | 18. |  | - | 2.0 |
| Ni | - | 38.0 | 20.0 | 34.0 | 3. 0 | 66.0 | 34.0 | 21.0 | 2. 0 |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| $\checkmark$ | 4. | - | - | - | 2. | - | - | 1. | 2. |
| Cr | 404. | 9. | $\rightarrow$ | 18. | 8. | 6. | 10. | 17. | 2. |
| Hf | - | - | - | - | - | - | - |  | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | 5. 0 | 5. 0 | 1.0 | 5. 0 | 3. 0 | 7.0 | 6.0 | 6.0 |
| Ta | - | S. | 5. | 1. | S. | 3. | 7. | . | . |
| Co | - | - | - | - | - | - | - | _ | - |
| Li | - | - | - | - | - | - | - | - | - |
| Be | $\rightarrow$ | - | - | - | - | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | 25. 00 | 23. 00 | 26. 00 | 35.00 | - | 17.00 | 18.00 | 16.00 | - - |
| W | - | - | - | , | - | 17.00 | 18.00 | 1. 0 | - |
| Sn | - | - | _ | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| La | 22. | 57. | 68. | 37. | 29. | 60. | 62. | 55. | 27. |
| Ce | 35. | 4. | - | 15. | 43. | - | 40. | 16. | 23. |

Figure D. 1 (cont.). Geochemical databose.

NOVA SCOTIA

|  | NKA083 SHER MONG | NLBA05 LBAY TONA | NLHHO1 HCOV MONG | NLHHO2 HCOV MONG | NLHHO3 HCOV MONG | NLHHO4 HCOV MONG | NLHHO5 HCOV MONG | NLHHO6 HCOV MONG | NLHHO 7 HCOV MONG |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 75. 55 | 59.75 | 72.17 | 71.40 | 72.65 | 72.15 | 71.56 | 72.55 | 71.20 |
| TiO2 | . 03 | . 99 | 30 | 21 | . 25 | . 28 | 25 | . 23 | . 18 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14.84 | 17.50 | 14.75 | 14.46 | 14.72 | 15.42 | 15.30 | 14.78 | 14.99 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 87 | 6. 07 | 21 | . 12 | . 02 | . 30 | . 12 | - | - |
| FeO | . 23 | - | 1.64 | 1.12 | 1.38 | 1.27 | 1. 24 | 1.54 | 1.18 |
| MnO | . 01 | . 07 | . 08 | . 04 | . 03 | . 06 | . 06 | . 04 | . 06 |
| MgO | . 04 | 3. 62 | . 50 | . 36 | . 41 | 45 | . 40 | . 47 | . 34 |
| CaO | . 42 | 4. 91 | . 50 | . 54 | . 55 | . 18 | . 52 | . 54 | . 48 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 5.02 | 3.58 | 3.18 | 3.41 | 3. 33 | 3. 39 | 3. 50 | 3. 23 | 3. 61 |
| $\mathrm{K}_{2} \mathrm{O}$ | 3.77 | 2. 45 | 4. 63 | 4.77 | 5. 20 | 4.94 | 4.92 | 5. 24 | 4.75 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 03 | . 35 | . 37 | . 31 | . 32 | . 30 | . 35 | . 33 | . 34 |
| $\mathrm{H}_{2} \mathrm{O}+$ | , | - | 1.08 | 1.00 | . 85 | . 89 | . 88 | . 59 | . 79 |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | . 48 | . 49 | . 36 | . 42 | . 32 | . 42 | . 33 |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - |  |
| LOI | . 30 | . 54 | - | - | - | - | - | - | - |
| TOTAL | 100. 31 | 99. 83 | 99.89 | 98.23 | 100.07 | 100. 36 | 99.42 | 99. 96 | 98. 25 |
| $\mathrm{Fe}_{2} \mathrm{O} 3 \mathrm{t}$ | . 33 | 6.07 | 2.03 | 1. 36 | 1.55 | 1.71 | 1.50 | 1.71 | 1. 31 |
| A/CNK | 1.1 | 1.0 | 1. 3 | 1.2 | 1.2 | 1.3 | 1.3 | 1.2 | 1.3 |
| DI | 95.6 | 59.4 | 89.7 | 90.1 | 91.5 | 91.0 | 90. 5 | 90.8 | 90. 2 |
| Ba | - | 400. | 215. | 274. | 384. | 322. | 310. | 378. | 275. |
| Rb | 228. | 90. | 285. | 240. | 217. | 246. | 247. | 225. | 263. |
| Sr | 38. | 380. | 58. | 70. | 90. | 81. | 79. | 80. | 62. |
| Y | - | - | 14. | 11. | 11. | 11. | 11. | 10. | 11. |
| Zr | 46. | 110. | 81. | 74. | s0. | 84. | 81. | 92. | 63. |
| Nb | 10. | - | 15. | 10. | 9. | 11. | 11. | 11. | 12. |
| Th | 2.00 | 5.00 | 10.00 | 11.00 | 12.00 | 11.00 | 9. 00 | 9. 0 | 8. 00 |
| Pb | 50. | 6. | 22. | 27. | 28. | 28. | 25. | 29. | 27. |
| Ga | - | - | 24. | 20. | 21. | 23. | 24. | 24. | 22. |
| Zn | - | 65.0 | 90. 0 | 60. 0 | 64.0 | 53.0 | 55.0 | 66.0 | 49.0 |
| Cu | - | - | - | - | - | - | - | - |  |
| Ni | - | 17.0 | 10.0 | 4.0 | 4.0 | 4.0 | 3.0 | 10.0 | 10.0 |
| $\mathrm{TiO}_{2}$ | - | - | . 30 | . 21 | . 25 | . 28 | . 25 | . 28 | . 18 |
| $\checkmark$ | - | - | 19. | 3. | 12. | - | 13. | 13. | 12. |
| Cr | 240. | - | 23. | 12. | 10. | 19. | 15. | 19. | 9. |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | - | 40.0 | 164.0 | 108.0 | 76.0 | 95.0 | 108. 0 | - | - |
| Be | - | - | 7.0 | 6. 5 | 4.5 | 4.5 | 6.5 | 3.5 | 4.0 |
| B | - | - | 15.0 | 15.0 | 25.0 | 10.0 | 20.0 | 20.0 | 20.0 |
| F | - | 560. | 690. | 530. | 510. | 540. | 570. | 450. | 530. |
| Cl | - | - | . | . | . | . | . | . | - |
| U | 11.00 | 1.90 | 11.50 | 7. 40 | 6. 20 | 7. 20 | 5.50 | - | - |
| w | , | 1. | 2.0 | 5.0 | 1.0 | 6. 0 | 4.0 | - | - |
| Sn | - | - | 13.0 | 10.0 | 1.0 | 5. 0 | 9. 0 | 9. 0 | 12.0 |
| Mo | - | - | 1.00 | 1.00 | 1.00 | 1.00 | 1.00 | 2. 00 | 2. 00 |
| Lo | 41. | - | - | - | - | - | - | - | - |
| Ce | 20. | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.

|  | $N O V A$ |  |  |  | $S \bigcirc$ | $T 1 A$ |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\begin{aligned} & \text { NLHHL18 } \\ & \text { HCOV } \\ & \text { MONG } \end{aligned}$ | $\begin{aligned} & \text { NLHHL34 } \\ & \text { HCOV } \\ & \text { MONG } \end{aligned}$ | $\begin{aligned} & \text { NLHHL37 } \\ & \text { HCOV } \\ & \text { MONG } \end{aligned}$ | NLHQOO <br> QUEE <br> MONG | NLHQO1 <br> QUEE <br> MONG | NLHQO2 <br> QUEE <br> MONG | NLHQO3 <br> QUEE <br> MONG | NLIHQO4 <br> QUEE <br> MONG | NLHQ05 QUEE MONG |
| $\mathrm{SiO}_{2}$ | 72. 37 | 71.88 | 74.14 | 71.05 | 69.07 | 69. 28 | 70.10 | 70. 55 | 71.11 |
| $\mathrm{TiO}_{2}$ | . 17 | . 16 | . 02 | . 19 | . 39 | . 43 | . 20 | . 26 | . 26 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14.85 | 14. 65 | 15. 32 | 15.21 | 15.57 | 15. 46 | 14.85 | 14.75 | 14.84 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 09 | - | - | - | . 07 | . 55 | . 03 | - | . 14 |
| FeO | 1. 83 | 1.12 | . 74 | 1. 65 | 2. 02 | 1.84 | 1.12 | 1. 55 | 1. 28 |
| MnO | . 04 | . 03 | . 15 | .10 | . 10 | . 11 | . 04 | . 04 | . 04 |
| MgO | . 31 | . 31 | . 07 | . 48 | , 58 | . 88 | . 39 | . 52 | . 49 |
| CaO | . 52 | . 50 | . 39 | . 93 | 1.45 | 1.74 | . 58 | . 61 | . 61 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 2. 73 | 3. 58 | 3. 35 | 3.44 | 3.43 | 3. 52 | 3.54 | 3. 62 | 3.80 |
| $\mathrm{K}_{2} \mathrm{O}$ | 5.02 | 4.73 | 5.26 | 4.68 | 4.14 | 3. 67 | 5. 06 | 4.85 | 4.77 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 31 | . 34 | . 22 | . 22 | . 16 | . 14 | . 22 | . 23 | . 23 |
| $\mathrm{H}_{2} \mathrm{O}+$ | . 90 | 1.07 | . 79 | . 70 | 1.00 | 1.10 | . 82 | . 69 | . 78 |
| $\mathrm{H}_{2} \mathrm{O}-$ | . 44 | . 35 | . 35 | .18 | . 27 | . 23 | . 19 | . 37 | . 22 |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| ${ }_{\mathrm{Cl}}$ | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 98.79 | 98.73 | 100.81 | 98.83 | 98. 25 | 98. 95 | 97.14 | 98. 05 | 98.37 |
| $\mathrm{Fe}_{2} \mathrm{O} 3 \mathrm{t}$ | 1. 23 | 1.24 | .82 | 1.83 | 2. 31 | 2. 59 | 1.27 | 1. 72 | 1.56 |
| A/CNK | 1.4 | 1.2 | 1.3 | 1. 2 | 1.2 | 1. 2 | 1.2 | 1. 2 | 1.2 |
| DI | 89.8 | 91.1 | 94.0 | 87.3 | 82.9 | 81.6 | 88.9 | 88.6 | 89.2 |
| Ba | 248. | 240. | 64. | 310. | 419. | 435. | 318. | 385. | $379 .$ |
| Rb | 255. | 226. | 257. | 227. | 187. | 178. | 245. | 268. | 267. |
| Sr | 65. | 66. | 34. | 94. | 146. | 168. | 79. | 82. | 77. |
| Y | 9. | 9. | 8. | 13. | 17. | 18. | 10. | 9. | 8. |
| Zr | 62. | 63. | 29. | 69. | 101. | 109. | 80. | 95. | 95. |
| Nb | 10. | 11. | 8. | 10. | 11. | 9. | 8. | 9. | 8. |
| Th | 8.00 | 10. 00 | 2. 00 | 6. 00 | 13. 00 | 14.00 | 12.00 | 19.00 | 19.00 |
| Pb | 23. | 65. | 32. | 34. | 31. | 25. | 26. | 26. | 25. |
| Ga | 19. | 21. | 21. | 22. | 19. | 19. | 22. | 21. | 23. |
| Zn | 44.0 | 195.0 | 13.0 | 54.0 | 58.0 | 55. D | 61.0 | 58.0 | 67.0 |
| Cu | d |  | 1 | , | 58. | 5, | , |  |  |
| Ni | 6.0 | 4.0 | 6.0 | 8. 0 | 6.0 | 6. 0 | 6. 0 | 7.0 | 3. 0 |
| $\mathrm{TiO}_{2}$ | . 17 | . 16 | . 02 | . 19 | . 39 | .43 | . 20 | . 26 | . 26 |
| $\checkmark$ | 5. | 5. | - | 11. | - | - | - | - | $\rightarrow$ |
| Cr | 13. | 10. | 7. | 15. | 23. | 29. | 8. | 14. | 12. |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | _ |
| Sc | - | _ | - | - | _ | - | - | - | - |
| Ta | - | - | - | - | - | - | - | - | - |
| Co | - | - | _ | - | - | - | - | - | - |
| Li | 107.0 | 36. 0 | - | - | 93. 0 | - | 68.0 | 96.0 | 93. 0 |
| Be | 7.0 | 6.5 | - | - | 6. 0 | 3.5 | 5.0 | 6. 5 | 5. 5 |
| B | 15.0 | 20.0 | - | - | 20.0 | 15.0 | 15.0 | 10.0 | 10.0 |
| F | 800. | 540. | - | - | 570. | 730. | 630. | 810. | 690. |
| Cl | - | - | - | - | , | , | - | - | - |
| U | 7.70 | 24.60 | - | - | 4.00 | - | 5.70 | 2. 80 | 3.60 |
| W | 1.0 | 5.0 | - | - | 6. 0 | - | 10.0 | 8.0 | 3. 0 |
| Sn | 6.0 | 6.0 | - | - | 3.0 | 6.0 | 9. 0 | 7.0 | 13.0 |
| Mo | 1.00 | 1.00 | - | - | 1.00 | 2.00 | 1.00 | 1.00 | 1.00 |
| La Ce | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical databose.

|  | $N \circlearrowleft V A$ |  |  |  | $S C$ | $T \mid A$ |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | NLHQ06 QUEE MONG | NLHQO7 <br> QUEE <br> MONG | NLHQO8 QUEE MONG | NLHQO9 QUEE MONG | NLHQ10 <br> QUEE <br> MONG | NLHQ1 1 <br> QUEE <br> MONG | NLHQ12 QUEE MONG | NLHQ13 QUEE MONG | NLHQ14 QUEE MONG |
| $\mathrm{SiO}_{2}$ | 72. 25 | 69.61 | 69. 63 | 72. 65 | 71.90 | 69. 00 | 69.84 | 70.90 | 70. 93 |
| $\mathrm{TiO}_{2}$ | . 26 | , 33 | . 39 | . 24 | . 22 | . 41 | . 35 | . 27 | . 31 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 15.12 | 15.88 | 15.21 | 14.99 | 14.74 | 15.09 | 15.29 | 14.61 | 15. 36 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 36 | - | - | . 03 | . 15 | . 54 | . 16 | . 05 | . 28 |
| FeO | 1.27 | 1. 95 | 2.36 | 1. 36 | 1. 05 | 1. 69 | 1.95 | 1. 44 | 1. 52 |
| MnO | . 07 | . 09 | . 10 | . 05 | . 05 | . 10 | . 09 | . 06 | . 06 |
| MgO | . 57 | . 74 | . 90 | . 52 | . 42 | . 86 | . 81 | . 59 | . 69 |
| CaO | 1.03 | 1.79 | 1. 75 | . 55 | . 52 | 1.72 | 1.67 | . 81 | . 86 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3.51 | 3.65 | 3. 48 | 3.65 | 3. 56 | 3.40 | 3.44 | 3. 61 | 3.55 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4. 44 | 4. 20 | 3.86 | 4.83 | 5. 03 | 4.03 | 4. 23 | 4.66 | 4.80 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 18 | . 12 | . 13 | . 25 | . 23 | . 14 | . 13 | . 20 | . 26 |
| $\mathrm{H}_{2} \mathrm{O}+$ | . 85 | . 80 | . 83 | . 94 | 1.14 | . 74 | . 46 | . 65 | . 65 |
| $\mathrm{H}_{2} \mathrm{O}-$ | . 21 | . 25 | . 20 | .14 | . 12 | . 26 | . 32 | . 29 | . 31 |
| $\mathrm{CO}_{2}$ | . 2 | . | . | . | - | . | , | , | , |
| Cl | - | - | - | - | - | - | $\rightarrow$ | - | - |
| $F$ | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 100.12 | 99. 44 | 98.84 | 100.20 | 99. 13 | 98.08 | 98.74 | 98. 14 | 99. 58 |
| $\mathrm{Fe}_{2} \mathrm{O} 3 \mathrm{t}$ | 1.77 | 2. 16 | 2. 62 | 1.54 | 1. 32 | 2. 52 | 2. 32 | 1.65 | 1.97 |
| A/CNK | 1.2 | 1. 2 | 1. 2 | 1.2 | 1.2 | 1.2 | 1.2 | 1.2 | 1.2 |
| DI | 88.4 | 82.8 | 81.2 | 91.1 | 91.2 | 81.7 | 82. 6 | 87.9 | 88. 0 |
| Ba | 342. | 527. | 473. | 392. | 306. | 412. | 405. | 501. | 567. |
| Rb | 209. | 175. | 175. | 268. | 251. | 186. | 182. | 259. | 267. |
| Sr | 113. | 184. | 187. | 75. | 73. | 160. | 162. | 93. | 94. |
| Y | 14. | 14. | 16. | 10. | 11. | 19. | 15. | 11. | 12. |
| Zr | 79. | 90. | 98. | 90. | 87. | 100. | 91. | 105. | 124. |
| Nb | 11. | 9. | 10. | 9. | 9. | 10. | 8. | 9. | 9. |
| Th | 10.00 | 12.00 | 12.00 | 16.00 | 13.00 | 15.00 | 10.00 | 21.00 | 25.00 |
| Pb | 30. | 32. | 30. | 24. | 25. | 30. | 27. | 25. | $26 .$ |
| Go | 20. | 17. | 20. | 21. | 25. | 19. | 20. | 23. | 24. |
| Zn | 54.0 | 46.0 | 54.0 | 58.0 | 74.0 | 54.0 | 47.0 | 72.0 | 82.0 |
| Cu | S | d6. | 51. | 5. | 71.0 | 5 | 4.0 | 12.0 | 82. |
| Ni | 8. 0 | 5.0 | 7. 0 | 6. 0 | 11.0 | 6. 0 | 8. 0 | 11.0 | 12.0 |
| $\mathrm{TiO}_{2}$ | . 26 | . 33 | . 39 | . 24 | . 22 | .41 | . 35 | . 27 | . 31 |
| $\checkmark$ | 13. | - | - | - | 10. | - | - | 19. | 21. |
| Cr Hf | 14. | 21. | 20. | 13. | 8. | 23. | 21. | 16. | 19. |
| Hf Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 111.0 | 79.0 | 80.0 | 102.0 | - | - | - | - | - |
| Be | 7.5 | 5. 0 | 5. 5 | 5. 5 | 3.5 | 3.0 | 3.0 | 2. 5 | 5. 0 |
| B | 15.0 | 10.0 | 20. 0 | 20. 0 | 15.0 | 15.0 | 15.0 | 10.0 | 15.0 |
| $F$ | 660. | 540. | 540. | 790. | 550. | 670. | 470. | 1000. | 960. |
| Cl | , | , | S | - | - | , | - | 1000 | , |
| U | 5.80 | 6.00 | 5. 60 | 2. 90 | 7. 90 | 6. 90 | 4.20 | 3.10 | 3. 70 |
| W | 3.0 | 1.0 | 3. 0 | 5.0 | - | - | - | - | - |
| Sn | 9. 0 | 5. 0 | 1. 0 | g. 0 | 13.0 | 6. 0 | 6. 0 | 8.0 | 9. 1 |
| Mo | 1.00 | 1.00 | 1.00 | 1.00 | 2. 00 | 2. 00 | 2.00 | 2. 00 | 2. 00 |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical databose.

NOVA SCOTIA

|  | NLHQ15 QUEE MONG | NLHQ16 QUEE MONG | NLHQ17 QUEE MONG | NLHQ18 QUEE MONG | NLHQ19 QUEE MONG | NLHQ2O QUEE MONG | NLHQ2 1 QUEE MONG | NLHQ22 <br> QUEE <br> MONG | NLHQL; 4 QUEE MONG |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 71. 29 | 70.91 | 73. 67 | 70. 51 | 72. 51 | 71.63 | 71.78 | 70. 51 | 70.40 |
| $\mathrm{TiO}_{2}$ | . 24 | . 28 | . 26 | . 25 | . 26 | . 24 | . 26 | . 38 | . 31 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14.74 | 14.52 | 14.90 | 15.00 | 14.73 | 15. 05 | 14.90 | 15.12 | 14.90 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 01 | . 03 | . 18 | . 88 | . 01 | - | .12 | . 40 | . 02 |
| FeO | 1.38 | 1. 46 | 1. 25 | 1. 34 | 1.43 | 1.18 | 1.38 | 2. 15 | 1. 55 |
| MnO | . 05 | . 06 | . 04 | . 05 | . 04 | . 05 | . 05 | . 12 | . 05 |
| MgO | . 50 | . 57 | . 49 | . 49 | . 52 | . 51 | 54 | . 98 | 55 |
| CaO | . 57 | . 61 | . 62 | . 74 | . 80 | . 61 | . 77 | 1. 53 | . 77 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3. 58 | 3. 55 | 3. 64 | 3. 66 | 3. 50 | 3. 62 | 3. 56 | 3.28 | 3.54 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4. 62 | 4.43 | 4. 68 | 4.70 | 4.87 | 4. 68 | 4. 78 | 3. 96 | 4.45 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 24 | . 20 | . 24 | . 24 | . 23 | . 23 | . 24 | . 15 | . 20 |
| $\mathrm{H}_{2} \mathrm{O}+$ | . 78 | . 68 | . 94 | 1.02 | . 76 | . 77 | . 89 | 1.03 | 1. 16 |
| $\mathrm{H}_{2} \mathrm{O}-$ | . 35 | . 26 | . 25 | . 28 | . 25 | . 11 | . 21 | . 17 | . 20 |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| $F$ | - | - | - | - | - | - | - |  |  |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 98.35 | 97. 56 | 101.14 | 98. 36 | 100.01 | 98.98 | 99. 48 | 99.78 | 98.10 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 1. 54 | 1.65 | 1.57 | 1. 57 | 1.60 | 1.84 | 1.65 | 2. 79 | 1.74 |
| A/CNK | 1. 2 | 1. 2 | 1.2 | 1.2 | 1.2 | 1. 2 | 1.2 | 1. 2 | 1. 2 |
| DI | 89.1 | 87.7 | 91.9 | 88. 5 | 90.0 | 89. 2 | 89.4 | 82. 8 | 87. 2 |
| Bo | 415. | 443. | 370. | 382. | 441. | 408. | 120. | 355. | 454. |
| Rb | 282. | 245. | 268. | 275. | 256. | 269. | 266. | 188. | 265. |
| Sr | 79. | 79. | 74. | 76. | 85. | 79. | 80. | 137. | 87. |
| Y | 11. | 13. | 9. | 11. | 9. | 10. | 10. | 21. | 11. |
| Zr | 96. | 112. | 97. | 101. | 106. | 95. | 106. | 102. | 113. |
| Nb | 10. | 10. | 10. | 10. | 9. | 8. | 10. | 11. | 9. |
| Th | 20.00 | 22. 00 | 17.00 | 18.00 | 18.00 | 18.00 | 19.00 | 7.00 | 26. 00 |
| Pb | 36. | 25. | 23. | 25. | 27. | 23. | 27. | 21. | 27. |
| Go | 27. | 25. | 26. | 25. | 25. | 27. | 25. | 23. | 21. |
| Zn | 64.0 | 75.0 | 62.0 | 62.0 | 68.0 | 72. 0 | 69.0 | 63.0 | 57.0 |
| Cu | - | - | - | - | - | - | - | - | - |
| Ni | 12.0 | B. 0 | 11.0 | 11.0 | 10.0 | 9. 0 | 7.0 | 11.0 | 9. 0 |
| $\mathrm{TiO}_{2}$ | . 24 | . 28 | . 24 | . 25 | . 26 | . 24 | . 26 | . 38 | . 31 |
| $V$ | 14. | 15. | 14. | 13. | 20. | 12. | 14. | 28. | - |
| Cr | 8. | 14. | 19. | 14. | 17. | 11. | 11. | 26. | 19. |
| Hf | - | - | - | - | 1 | - | - | . | - |
| Cs | - | - | - | - | $\rightarrow$ | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| Ta | - | - | - | _ | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | - | - | - | _ | - | _ | - | - | 81.0 |
| Be | - | 2. 0 | 4.0 | 4. 0 | 3. 5 | 4. 0 | 3.0 | 2.5 | 5. 5 |
| B | - | 15.0 | 150.0 | 10.0 | 15. $\square$ | 20.0 | 15.0 | 20.0 | 20.0 |
| F | - | 930. | 820. | 960. | 730. | 930. | 790. | 780. | 810. |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | 2. 80 | 3. 50 | 5. 90 | 6. 50 | 9. 40 | 3. 30 | 5. 50 | 6. 60 |
| W | - | - | - | - | - | - | - | - | 8.0 |
| Sn | - | 6.0 | 12.0 | 12.0 | 11.0 | 8. 0 | 11.0 | 10.0 | 4. 0 |
| Mo | - | 2. 00 | 2.00 | 2. 00 | 2.00 | 2. 00 | 2. 00 | 3. 00 | 1.00 |
| Lo | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical databose.

|  | $N O V A$ |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | NLHQL32 <br> QUEE <br> MONG | NLHQL38 <br> QUEE <br> MONG | NLI264 <br> LISC <br> GRAD | NLI265 <br> LISC <br> MONG | NLI266 <br> LISC <br> MONG | NLi267 <br> LISC <br> MONG | NLI300 <br> LISC <br> GRAD | NLI305 LISC GRAD | NLI306 LISC GRAD |
| $\mathrm{SiO}_{2}$ | 71.84 | 69. 72 | 66.18 | 77.72 | 72. 38 | 72.98 | 63.93 | 64. 56 | 64. 28 |
| $\mathrm{TiO}_{2}$ | . 33 | . 27 | . 82 | . 08 | . 17 | . 17 | . 82 | . 82 | . 49 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 15.01 | 14. 67 | 16. 30 | 14.56 | 15.75 | 15.71 | 17.70 | 17.47 | 15. 31 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 10 | - | 4. 84 | . 73 | 1.08 | 1. 16 | 5. 35 | 5.39 | 3. 08 |
| FeO | 1. 62 | 1. 69 | - | - | - | - | - | - | - |
| MnO | . 05 | . 07 | . 12 | . 02 | . 02 | . 02 | . 13 | . 11 | . 05 |
| MgO | . 69 | . 56 | 1. 55 | . 21 | . 36 | . 34 | 1. 68 | 1.72 | 1. 15 |
| COO | . 85 | 1. 46 | 2. 04 | . 31 | . 13 | . 42 | 2. 31 | 2.18 | 1.46 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3. 50 | 3.47 | 3. 51 | 3. 67 | 3. 64 | 4. 14 | 3. 78 | 3.89 | 4.07 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4.51 | 4.04 | 3. 71 | 3. 57 | 5. 63 | 4. 48 | 3.55 | 3. 28 | 3.91 |
| $\mathrm{P}_{2} \mathrm{O} 5$ | . 20 | . 07 | . 32 | . 34 | . 37 | . 34 | . 29 | . 31 | . 26 |
| $\mathrm{H}_{2} \mathrm{O}+$ | . 71 | 1.00 | - | - | - | - | - | . |  |
| $\mathrm{H}_{2} \mathrm{O}-$ | . 37 | . 14 | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | . 69 | . 77 | . 77 | . 85 | . 92 | 1.23 | . 85 |
| TOTAL | 99.78 | 97.15 | 100.08 | 101.98 | 100.60 | 100. 61 | 100.46 | 100.96 | 94.92 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 1.90 | 1. B 8 | 4. 84 | . 73 | 1.08 | 1.16 | 5. 35 | 5. 39 | 3.08 |
| A/CNK | 1.2 | 1.2 | 1. 2 | 1.4 | 1. 2 | 1. 3 | 1.2 | 1.3 | 1.1 |
| DI | 88.0 | 83. 3 | 77.4 | 94.5 | 93. 2 | 92.7 | 74.7 | 75.3 | 79.1 |
| Ba | $506 .$ | 293. | 963. | 82. | 346. | 468. | 869. | 819. | 563. |
| Rb | $262 .$ | 163. | 147. | 160. | 223. | 215. | 149. | 139. | 191. |
| Sr | 93. | 140. | 217. | 24. | 58. | 73. | 248. | 228. | 148. |
| Y | 11. | 18. | 30. | 7. | 8. | 8. | 29. | 21. | 17. |
| Zr | 119. | 79. | 232. | 17. | 68. | 60. | 259. | 263. | 155. |
| Nb | 10. | 8. | 14. | 11. | 8. | 9. | 16. | 15. | 11. |
| Th | 25. 80 | 9. 00 | 11.00 | - | 7.00 | 6. 00 | 6. 00 | 13.00 | 11.00 |
| Pb | 25. | 30. | 37. | 19. | 24. | 31. | 25. | 18. | 27. |
| Go | 22. | 17. | 20. | 22. | 21. | 19. | 22. | 23. | 21. |
| Zn | 62.0 | 12.0 | 107.0 | 27.0 | 50.0 | 46. 0 | 106.0 | 79.0 | 77.0 |
| Cu | - | - | 2.0 | - | - | - | 6. 0 | 4. 0 | - |
| Ni | 8.0 | 3.0 | 10.0 | 4.0 | $4.0$ | 5.0 | $8.0$ | 8.0 | $8.0$ |
| $\mathrm{TiO}_{2}$ | . 33 | . 27 | . 84 | . 05 | . 17 | . 14 | $.93$ | $.94$ | $.52$ |
| V | - | - | 51. | 1. | 5. | 4. | 73. | 69. | 33. |
| Cr | 19. | 19. | 22. | 4. | 13. | 2. | 25. | 25. | 27. |
| Hf | - | - | - | - | - | - | - |  | - |
| Cs | - | - | - | - | - | $\rightarrow$ | _ | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| Ta | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 91.0 | 72.0 | - | - | - | - | - | $-$ | - |
| Be | 5.0 | 4. 5 | - | - | - | - | - | - | - |
| B | 15.0 | 15.0 | - | - | - | - | - | - | - |
| F | 1075. | 330. | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | 2. 90 | 3. 40 | - | - | - | - | - | - | - |
| W | 4.0 | 3.0 | - | - | - | - | - | - | - |
| Sn | 1.0 | 2. 0 | - | - | - | - | - | - | - |
| Mo | 1.00 | 1.00 | - | - | - | - | - | - | - |
| Lo | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure 0.1 (cont.). Geochemical database.


Figure D. 1 (cont.). Geochemical databose.


Figure D. 1 (cont.). Geochemical datobase.

|  | $N O V A$ |  |  |  | $S C$ | $T T A$ |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | NMD103 | NM0104 | NMD108 | NMD109 | NMD22 | NMO44 | NM082 | NMD96 | NMD97 |
|  | MUSQ | MUSQ | MUSQ | MUSQ | MUSQ | MUSQ | MUSQ | MUSQ | MUSQ |
|  | MONG | ORMI | DRMI | MONG | MONG | MONG | ORMI | MONG | MONG |
| $\mathrm{SiO}_{2}$ | 71.44 | 73.43 | 72. 56 | 74.23 | 70. 77 | 72. 57 | 74.83 | 72. 26 | 71.43 |
| $\mathrm{TiO}_{2}$ | . 38 | . 06 | . 04 | . 13 | . 28 | . 27 | . 07 | 22 | . 36 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14.76 | 14.63 | 15.21 | 14.44 | 14. 62 | 14. 34 | 14.19 | 14.77 | 14. 52 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | - | - | - | - | - | - | - | - | - |
| FeO | 2. 07 | . 58 | . 51 | 1.00 | 1. 81 | 1. 59 | .70 | 1.40 | 2.12 |
| MnO | . 07 | . 03 | . 17 | . 05 | . 05 | . 04 | . 06 | . 05 | . 08 |
| MgO | . 15 | . 10 | . 08 | . 35 | . 61 | . 54 | . 06 | . 52 | . 78 |
| COO | 1.22 | . 38 | . 28 | . 66 | . 86 | . 88 | . 30 | . 62 | . 92 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3. 45 | 4.11 | 3.82 | 3. 74 | 3. 51 | 3. 54 | 4. 58 | 3.75 | 3.64 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4.58 | 4.15 | 4.80 | 4. 92 | 4. 65 | 4. 32 | 3. 18 | 4.62 | 4. 31 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 07 | . 39 | . 40 | . 30 | . 30 | . 27 | . 24 | . 30 | . 30 |
| $\mathrm{H}_{2} \mathrm{O}+$ | . 68 | 1. 35 | . 90 | . 84 | 1. 52 | 1.13 | . 97 | . 91 | . 92 |
| $\mathrm{H}_{2} \mathrm{O}-$ | . 11 | . 05 | . 04 | . 14 | . 09 | . 07 | . 03 | . 07 | .17 |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | $\rightarrow$ | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - |  |
| LOl | - | - | - | - | - | - | - | - | - |
| TOTAL | 98.98 | 99. 26 | 98.81 | 100.80 | 99.07 | 99. 66 | 99. 21 | 99.49 | 99. 55 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 2. 30 | . 64 | . 57 | 1.11 | 2. 01 | 1.88 | . 78 | 1. 55 | 2. 35 |
| A/CNK | 1. 2 | 1. 2 | 1.3 | 1.1 | 1. 2 | 1.2 | 1, 2 | 1.2 | 1.2 |
| DI | 86. 7 | 93. 5 | 92.9 | 93.1 | 87.9 | 88.8 | 93.8 | 90.4 | 87.1 |
| Ba | 59. | - | 46. | 161. | 365. | 275. | - |  | 278. |
| Rb | 300. | 560. | 510. | 300. | 310. | 250. | 440. | 300. | 320. |
| Sr | 120. | 10. | 10. | 60. | 90. | 70. | 10. | 60. | 130. |
| Y | - | - | - | - | - | - | - | - | $\rightarrow$ |
| Zr | 130. | 30. | 20. | 50. | 90. | 80. | 40. | 70. | 100. |
| Nb | - | - | - | - | - | - |  | \% | , |
| Th | 1.30 | 1. 20 | 1.00 | 2. 50 | 7. 30 | 8. 10 | 1.30 | 5. 00 | 7. 10 |
| Pb | - | - | - | - | - | - | - | - | - |
| Go | - | - | - | - | - | - | - | - | - |
| Zn | 52.8 | 32. 3 | 16.3 | 37.4 | 70. 6 | 49.5 | 25. 5 | 55.8 | 72.1 |
| Cu | 7.0 | 17.4 | 11.0 | 4.3 | 50.4 | 5. 6 | 12.2 | 8. 5 | 16.6 |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| $\checkmark$ | - | - | - | - | - | - | _ | - | - |
| Cr | - | _ | - | - | - | - | - | - | - |
| Hf | 1. 4 | 1.8 | 2.5 | 1. 8 | 2. 7 | 3.0 | 2.1 | 2.1 | 3. 1 |
| Cs | 35.3 | 44.2 | 24.2 | 22.1 | 37. 5 | 19.0 | 15.0 | 20. 2 | 34. 3 |
| Sc | 4. 6 | 3. 7 | 1.7 | 3.1 | 4.7 | 4. 2 | 3. 9 | 4. 1 | 5. 7 |
| To | 7.5 | B. 0 | 9.7 | 5. 3 | 6. 2 | 5. 0 | 8. 0 | 3. 8 | 4. 9 |
| Co | - | - | - | - | - | 5. | 8. | - | - |
| Li | 178.1 | 166.4 | 74.6 | 149.7 | 257. 9 | 159.5 | 50.6 | 191.7 | 197.6 |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | - | - | - | - | - | _ | - |
| F | - | - | - | - | - | - | - | _ | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | 12.80 | 19.90 | 13.10 | 6.10 | 9. 70 | 3.80 | 16. 90 | 2. 70 | 4.00 |
| W | - | - | $\underline{-}$ | - | g. | . 80 | 16. |  | - |
| Sn | - | - | - | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical databose.

NOVA SCOTIA

|  | NMK100A CSMB DRMI | NMK 102 CSMB MONG | NMK114 CSMB MONG | NMK116A CSMB DRMI | NMK119 CSMB MONG | NMK121 CSMB MONG | NMK1 23 CSMB DRMI | NMK 124 CSMB GRAD | NMK 127 CSMB GRAD |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 75.00 | 73.70 | 74.50 | 74.10 | 74. 30 | 73. 50 | 74.80 | 58.80 | 72.20 |
| $\mathrm{TiO}_{2}$ | . 09 | 25 | . 12 | . 09 | . 22 | . 18 | . 18 | . 68 | . 42 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14. 20 | 13.42 | 13.40 | 14.07 | 13.08 | 13. 87 | 13. 20 | 14. 52 | 14.25 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 58 | . 17 | . 16 | . 23 | . 05 | 22 | . 17 | . 45 | 25 |
| FeO | . 62 | 1.69 | 1.51 | . 82 | 1.77 | 1. 55 | 1.09 | 3. 88 | 2.52 |
| MnO | . 03 | . 04 | . 05 | 03 | . 05 | . 83 | . 03 | . 10 | . 09 |
| Mgo | . 08 | . 28 | . 12 | . 10 | . 24 | 24 | . 08 | 1.18 | . 75 |
| CoO | . 53 | 58 | . 45 | . 44 | . 66 | . 60 | . 43 | 2.21 | 1.44 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 4.57 | 3. 32 | 3. 46 | 3.84 | 3. 60 | 3.54 | 4.00 | 3.18 | 3.31 |
| $\mathrm{K}_{2} \mathrm{O}$ | 3.60 | 4.91 | 4.59 | 4.37 | 4. 34 | 4.72 | 4. 31 | 3.45 | 4.04 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 28 | . 07 | . 14 | . 14 | . 07 | . 15 | . 21 | . 10 | . 16 |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1.01 | . 82 | . 58 | . 79 | . 77 | . 69 | . 68 | . 91 | . 88 |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | . 14 | . 13 | . 28 | . 33 | . 10 | . 18 | . 26 | . 05 | . 06 |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 100.73 | 99. 39 | 99. 36 | 99. 35 | 99. 25 | 99. 47 | 98. 64 | 99. 52 | 100. 39 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 1. 27 | 2. 05 | 1.84 | 1.14 | 2.01 | 1.94 | 1. 38 | 4.75 | 3. 06 |
| A/CNK | 1.2 | 1.1 | 1.2 | 1.2 | 1.1 | 1. 2 | 1.1 | 1.1 | 1.2 |
| DI | 94.8 | 90.8 | 91.8 | 92.8 | 90.7 | 91.0 | 93.0 | 75. 2 | 84.6 |
| Ba | 8. | 230. | 66. | 16. | 200. | 164. | 28. | 668. | 429. |
| Rb | 402. | 330. | 532. | 680. | 299. | 411. | 620. | 143. | 174. |
| Sr | - | - | - | - | - | - | - | 155. | 112. |
| Y | - | - | - | - | - | - | - | - | - |
| Zr | - | 91. | 51. | - | 76. | 64. | 20. | 210. | 168. |
| Nb | - | - | - | - | - | - | - | - | - |
| Th | - | - | 10.90 | 2.05 | 11.00 | - | 4. 13 | - | 8.33 |
| Pb | - | - | - | - | - | - | - | - | - |
| Go | - | - | - | - | - | - | - | - | - |
| Zn | 33.0 | 72.0 | 64.0 | 43.0 | 56.0 | 64. 0 | 64.0 | 71.0 | 60.0 |
| Cu | - | - | - | - | - | - | - | - | - |
| Ni | 9.0 | 8.0 | 8.0 | 9. 0 | 10.0 | 9. 0 | 9. 0 | 16.0 | 12.0 |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| $\checkmark$ | - | - | - | - | - | - | - | - | - |
| Cr | 42. | 43. | 45. | 37. | 47. | 47. | 38. | 50. | 42. |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | - | - | - | - | - | - | - | - | - |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | 22.0 | 19.0 | 30.0 | 40.0 | 15.0 | 20.0 | 25.0 | 7.0 | 18.0 |
| Mo | - | - | - | - | - | - | - | - | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.


Figure D. 1 (cont.). Geochemical database.


Figure D. 1 (cont.). Geochemical database.


Figure D. 1 (cont.). Geochemical databose.

NOVA SCOTIA

|  |  | NMO6B MPOI GRAD | NMO6M MPOI MONG | $\begin{aligned} & \text { NMO7 } \\ & \text { MPOI } \\ & \text { GRAD } \end{aligned}$ |  | NOEO42 BREN MONG | NOEO45 BREN MONG | NOEO46 BREN MONG | NOEO47 BREN MONG |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | - | 70.20 | 74.50 | 70.70 | 73.10 | 75.72 | 76. 00 | 76. 42 | 75. 56 |
| $\mathrm{TiO}_{2}$ | - | . 40 | . 16 | . 51 | . 23 | . 12 | . 12 | . 11 | . 15 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | - | 14.70 | 14.10 | 14.90 | 14.70 | 12. 51 | 12. 37 | 12. 23 | 12.14 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | - | 2. 30 | . 90 | 2. 50 | 1.30 | . 32 | . 38 | . 11 | . 26 |
| FeO | - | - | - | - | - | 1. 45 | 1.28 | 1. 32 | 1.35 |
| Mno | - | . 05 | . 03 | . 05 | . 05 | . 03 | . 04 | . 03 | . 04 |
| MgO | - | 1.00 | . 20 | 1. 50 | . 40 | .10 | . 07 | . 07 | . 13 |
| CaO | - | 1.30 | . 40 | 2.00 | . 51 | . 40 | . 33 | . 19 | . 62 |
| $\mathrm{Na}_{2} \mathrm{O}$ | - | 3. 60 | 3. 60 | 4. 20 | 3. 50 | 2. 69 | 2. 97 | 3.16 | 3. 20 |
| $\mathrm{K}_{2} \mathrm{O}$ | - | 4. 40 | 5.10 | 3.00 | 5. 20 | 5. 59 | 5. 58 | 5. 35 | 5.21 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | - | . 45 | . 31 | . 36 | . 38 | . 20 | . 10 | . 03 | . 04 |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | - | - | - | - | . 35 | . 34 | . 23 | . 14 |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | . 06 | . 02 | . 03 | . 06 |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| $F$ | - | - | - | - | - | - | - | - | - |
| LOI | - | 1.00 | . 67 | . 68 | 1.10 | - | - | - | - |
| TOTAL | - | 99.40 | 99.97 | 100.48 | 100.47 | 99. 54 | 99. 50 | 99.28 | 98.91 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | . 00 | 2. 30 | . 90 | 2. 60 | 1.30 | 1.93 | 1.80 | 1.58 | 1.76 |
| A/CNK | - | 1.1 | 1.2 | 1.1 | 1. 2 | 1.1 | 1.1 | 1.1 | 1.0 |
| DI | - | 85.9 | 94.3 | 82.6 | 92. 6 | 93.4 | 94.4 | 94.7 | 92.7 |
| Ba | 820. | 790. | 630. | - | 650. | - | - | - | - |
| Rb | 148. | 150. | 127. | - | 157. | 217. | 234. | 243. | 195. |
| Sr | 190. | 190. | 108. | - | 130. | 40. | 15. | 7. | 46. |
| Y | - | - | - | - | - | - | - | - | - |
| Zr | - | - | - | - | - | 79. | 90. | 93. | 145. |
| Nb | - | - | - | - | - | 23. | 26. | 27. | 26. |
| Th | 10.00 | 7.00 | 2. 00 | - | 3. 00 | - | - | - | - |
| Pb | 28. | 22. | 22. | - | 24. | - | - | - | - |
| Ga | - | - | - | - | - | - | - | - | - |
| Zn | 45.0 | 47.0 | 9. 0 | - | 34.0 | - | - | - | - |
| Cu | 8.0 | 6.0 | 7.0 | - | 6.0 | - | - | - | - |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{THO}_{2}$ | - | - | - | - | - | - | - | - | _ |
| $\checkmark$ | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| Ta | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 79.0 | 78.0 | 19.0 | - | 48.0 | - | - | - | - |
| Be | 8.5 | 14.0 | 8. 0 | - | 20.0 | - | - | - | - |
| B | 10.0 | 8.0 | 7. 0 | - | 8. 0 | - | - | - | - |
| F | 340. | 320. | 200. | - | 320. | - | - | - | - |
| Cl | - | - | - | _ | - | _ | - | - | - |
| U | 3.40 | 3.60 | 3.50 | - | 3.20 | - | - | - | - |
| W |  | - | - | - | - | - | - | - | - |
| Sn | 7. 3 | 7.8 | 5. 5 | - | 7.7 | - | - | - | - |
| Mo | 1.10 | 1.30 | 1.70 | - | . 80 | - | - | - | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical databose.

## NOVA SCOTIA

|  | NOEO50 BREN MONG | NOEO59 BREN MONG | NOE061 BREN MONG | NPLO14 ELAK GRAD | NPLO15 ELAK GRAD | NPLO16 ELAK GRAD | NPLO17 ELAK GRAD | NPLO18 ELAK GRAD | NPLO19 ELAK GRAD |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 71.68 | 74.76 | 75.72 | 68.00 | 65.00 | 68.00 | 67.50 | 57.00 | 67.50 |
| $\mathrm{TiO}_{2}$ | . 30 | . 20 | . 21 | . 60 | . 60 | . 61 | . 60 | . 60 | . 60 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 13.68 | 12.71 | 12. 55 | 14.98 | 14.93 | 14.92 | 14.97 | 15.00 | 14.89 |
| $\mathrm{Fe}_{2} \mathrm{O} 3$ | 1.51 | 12.33 | . 73 | 4.72 | 4. 72 | 4. 76 | 4. 73 | 4. 72 | 4.70 |
| FeO | 2.15 | 1.85 | 1. 56 | - | - | - | - | - | - |
| MnO | . 08 | . 05 | . 06 | . 10 | . 10 | . 09 | . 10 | 09 | 07. |
| MgO | . 27 | . 17 | . 13 | 1. 35 | 1. 33 | 1.30 | 1.42 | 1. 28 | 1.18 |
| CoO | 1.05 | . 57 | . 84 | 1. 67 | 1. 83 | 2.05 | 1. 67 | 1.98 | 1.33 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 2.53 | 2. 88 | 1.52 | 2.86 | 2. 88 | 2. 88 | 2. 98 | 2.86 | 2.74 |
| $\mathrm{K}_{2} \mathrm{O}$ | 5.41 | 5.44 | 5.44 | 4. 38 | 4.56 | 4. 11 | 3. 89 | 4.23 | 4. 30 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 21 | . 15 | . 17 | . 19 | . 10 | . 19 | . 20 | . 20 | . 19 |
| $\mathrm{H}_{2} \mathrm{O}+$ | . 52 | . 61 | . 53 | - | - | - | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}-$ | . 08 | . 03 | . 13 | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | 2. 05 | 2. 80 | . 75 | 1. 69 | 1.10 | 1.50 |
| TOTAL | 99.47 | 99. 76 | 99. 59 | 100.90 | 99. 85 | 99. б6 | 99.75 | 99.06 | 99.00 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3 t}$ | 3.90 | 2. 38 | 2. 46 | 4.72 | 4. 72 | 4. 76 | 4.73 | 4.72 | 4.70 |
| A/CNK | 1.2 | 1.1 | 1.3 | 1.2 | 1.2 | 1.2 | 1.2 | 1.2 | 1.3 |
| DI | 87. 0 | 91.9 | 89.2 | 79.6 | 77.5 | 78.4 | 78.1 | 77. 8 | 79.8 |
| Bo | - | - | - | 850. | 750. | 660. | 760. | 720. | 926. |
| Rb | 209. | 173. | 180. | 154. | 167. | 152. | 139. | 172. | 174. |
| Sr | 129. | 56. | 85. | 210. | 190. | 180. | 245. | 170. | 165. |
| Y | - | - | . | - | - | - | - | - | . |
| Zr | 293. | 180. | 191. | - | - | - | - | - | - |
| Nb | 29. | 28. | 26. | - | - | - | - | - | - |
| Th | - | - | - | 16.00 | 14.00 | 17.00 | 17.00 | 14.00 | 14.00 |
| Pb | - | - | - | 10. | 4. | 10. | 12. | 14. | 6. |
| Ga | - | - | - | - | - | - | - | - | - |
| Zn | - | - | - | 59.0 | 75.0 | 76.0 | 36. 0 | 66.0 | 44.0 |
| Cu | - | - | - | 13.0 | 11.0 | 12.0 | 10.0 | 12.0 | 15.0 |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| V | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | - | - | - | 77. 0 | 62.0 | 82.0 | 86.0 | 71.0 | 78.0 |
| Be | - | - | - | - | , | , | 8. | - | - |
| B | - | - | - | 37. 0 | 20.0 | 51.0 | 42. 0 | 39.0 | 58.0 |
| F | - | - | - | 410. | 540. | 640. | 600. | 490. | 510. |
| Cl | - | - | - | , | - | , | - | - | - |
| U | - | - | - | 3.40 | 2. 90 | 3. 30 | 2.90 | 3. 40 | 6.90 |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | 5.1 | 5.2 | 4. 8 | 4. 1 | 4.2 | 4.8 |
| Mo | - | - | - | 2. 00 | 1.30 | 2.00 | 2.50 | 2.10 | 2. 20 |
| Lo | - | - | - | , | 1. | 2. | 2. | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D .1 (cont.). Geochemical datobase.

NOVA SCOTIA

|  | NPLO2O ELAK GRAD | NPLO21 ELAK GRAD | NPLO22 ELAK GRAD | NPLO23 ELAK GRAD | NPLO24 ELAK GRAD | NPLO25 ELAK GRAD | NPLO26 ELAK GRAD | NPL028 ELAK GPAD | NPM10 PMOU MONG |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 67.00 | 67. 00 | 58.00 | 67. 50 | 67.70 | 68.00 | 68.50 | 68.10 | 73. 99 |
| $\mathrm{TiO}_{2}$ | . 60 | . 61 | 61 | . 60 | . 61 | . 51 | . 51 | 51 | . 14 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 15.01 | 14.92 | 14.90 | 14.95 | 14.93 | 14.92 | 14.93 | 14.97 | 14.76 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 4. 72 | 4. 70 | 4. 72 | 4. 73 | 4. 71 | 4. 72 | 4. 71 | 4. 71 | 1.02 |
| FeO | . | , | . 7 | 1.73 | . | , | - | - | - |
| Mno | . 09 | . 11 | . 09 | . 09 | . 10 | . 09 | . 09 | 09 | . 02 |
| Mgo | 1.32 | 1.40 | 1. 31 | 1. 26 | 1. 34 | 1. 28 | 1.34 | 1. 32 | 28 |
| COO | 2. 05 | 2.12 | 2. 23 | 2. 10 | 2.19 | 1.92 | 2.07 | 2.25 | 1.70 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 2.80 | 2. 90 | 2.76 | 2. 86 | 2. 90 | 2.92 | 2. 89 | 2. 92 | 3. 12 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4.28 | 3. 92 | 4.08 | 4.29 | 4.15 | 4.13 | 4.08 | 4.25 | 4.48 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 19 | . 19 | . 19 | . 19 | .19 | . 19 | . 18 | . 19 | . 06 |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | 1. 35 | 1.31 | . 48 | 1.01 | . 91 | - | 1.12 | . 85 | . 38 |
| total | 99.41 | 99.18 | 99.37 | 99. 58 | 99.73 | 98.78 | 100.50 | 100.27 | 99.95 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 4.72 | 4.70 | 4. 72 | 4.73 | 4.71 | 4. 72 | 4. 71 | 4.71 | 1.02 |
| A/CNK | 1.2 | 1.2 | 1.1 | 1.1 | 1.1 | 1. 2 | 1. 2 | 1.1 | 1.1 |
| DI | 77.5 | 76.7 | 77.6 | 78.2 | 77.9 | 78.8 | 78.7 | 78.4 | 87.7 |
| Ba | 746. | 620. | 830. | 740. | 800. | 950. | 830. | 680. | 972. |
| Rb | 187. | 171. | 144. | 176. | 162. | 147. | 167. | 163. | 77. |
| Sr | 190. | 160. | 190. | 200. | 170. | 170. | 170. | 240. | 229. |
| Y | - | - | - | - | - | - | - | - | 10. |
| Zr | - | - | - | - | - | _ | - | - | 53. |
| Nb | - | - | - | - | - | - | - | - | 6. |
| Th | 13.00 | 18.00 | 14.00 | 15.00 | 15.00 | 17.00 | 16.00 | 14.00 | 1.00 |
| Pb | 15. | 33. | 30. | 33. | 14. | 18. | 16. | 18. | 29. |
| Ga | - | - | - | - | - | - | - | - | 14. |
| Zn | 59. 0 | 83.0 | 59.0 | 69.0 | 80.0 | 77. 0 | 67.0 | 75.0 | 16.0 |
| Cu | 13.0 | 16.0 | 104.0 | 13.0 | 14.0 | 19.0 | 12.0 | 12.0 | , |
| Ni | - | - | - | - | - | - | - | - | 3.0 |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | . 18 |
| $\checkmark$ | - | - | - | - | - | - | - | - | 16. |
| Cr | - | - | - | - | - | - | - | - | 12. |
| Hf | - | - | - | - | - | - | - | - | 1 |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| Ta | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 89.0 | 83.0 | 90.0 | 87.0 | 92.0 | 76. 0 | 80.0 | 74.0 | - |
| Be | - | - | - | 87.0 | - | 18. | - | - | - |
| B | 61.0 | 37.0 | 48.0 | 31.0 | 13.0 | 27. 0 | 37. 0 | 15.0 | - |
| F | 540. | 600. | 580. | 600. | 600. | 520. | 560. | 620. | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | 3.60 | 3.80 | 3. 80 | 3. 90 | 3.20 | 3.30 | 4.60 | 9. 30 | - |
| W |  | , | , |  | , | , | , |  | - |
| Sn | 5.0 | 5. 0 | 4.5 | 4.7 | 4.0 | 4.8 | 4.7 | 4.3 | - |
| Mo | 2. 20 | 1.50 | 2.80 | 1.50 | 2.00 | 1.70 | 2. 00 | 1.40 | - |
| Lo | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.

## NOVA SCOTIA

|  |  | NPM17 <br> PMOU <br> GRAD | NPM2 <br> PMOU <br> TONA | NPM32 <br> PMOU <br> MONG | NPM361 <br> PMOU <br> LMON | NPM368B <br> PMOU <br> GRAD | NPM371 PMOU MONG | NPM440 PMOU GRAD | NPM441 PMOU MONG |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 67.23 | 70.09 | 70.78 | 73.30 | 75. 07 | 72. 96 | 75.72 | 68. 15 | 73. 72 |
| $\mathrm{TiO}_{2}$ | . 55 | . 12 | . 41 | . 14 | . 10 | . 24 | . 26 | . 52 | . 14 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 16.87 | 15.53 | 15.40 | 15.01 | 14.13 | 14.64 | 12. 51 | 15.25 | 14.64 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 3. 86 | 2.79 | 3. 16 | 1. 21 | . 78 | 2. 08 | 1.68 | 3.70 | 1.21 |
| FeO | - | - | - | - | - | - | - | - | - |
| MnO | . 05 | . 05 | . 08 | . 05 | . 02 | .05 | . 04 | . 07 | . 04 |
| MgO | 1. 40 | . 84 | 1.19 | . 30 | . 13 | . 47 | . 39 | 1.40 | . 26 |
| CaO | 3.23 | 2.17 | 2.09 | . 82 | 1.12 | 1.81 | 1.01 | 3. 67 | . 83 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3.62 | 4.18 | 3. 92 | 3.87 | 4. 10 | 4. 35 | 2. 72 | 3. 68 | 3.92 |
| $\mathrm{K}_{2} \mathrm{O}$ | 2. 41 | 3.19 | 2.14 | 4.37 | 3.77 | 2. 82 | 4. 62 | 2. 97 | 4. 28 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 17 | . 30 | . 22 | . 31 | . 16 | . 22 | . 21 | . 20 | . 31 |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | _ | - | - | - |
| F | - | - | - | - | - | - |  |  | - |
| LOO | . 58 | . 41 | . 88 | . 65 | . 44 | . 42 | . 55 | . 52 | . 63 |
| TOTAL | 99.98 | 99.97 | 100.25 | 100.03 | 99. 82 | 100. 05 | 99. 71 | 100.13 | 99.98 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 3.86 | 2.79 | 3.16 | 1. 21 | . 78 | 2. 08 | 1.68 | 3.70 | 1.21 |
| A/CNK | 1.2 | 1.1 | 1.2 | 1. 2 | 1.1 | 1.1 | 1.1 | 1.0 | 1.2 |
| OI | 73.3 | 82.7 | 80.0 | 91.3 | 91.6 | 86.4 | 90.4 | 74.8 | 91.7 |
| Ba | 718. | 645. | 397. | 374. | 316. | 316. | 409. | 494. | 386. |
| Rb | 82. | 124. | 74. | 181. | 125. | 126. | 170. | 134. | 189. |
| Sr | 370. | 184. | 200. | 54. | 85. | 106. | 66. | 207. | 67. |
| Y | 10. | 20. | 14. | 14. | 15. | 20. | 20. | 17. | 14. |
| Zr | 150. | 179. | 127. | 69. | 70. | 137. | 99. | 169. | 73. |
| Nb | 11. | 13. | 10. | 12. | 7. | 10. | 10. | 11. | 11. |
| Th | - | 9.00 | 9. 00 | 6. 00 | 3.00 | 5.00 | 13.00 | 13.00 | 2.00 |
| Pb | 18. | 25. | 22. | 26. | 25. | 16. | 25. | 12. | 24. |
| Go | 18. | 19. | 18. | 20. | 15. | 19. | 19. | 22. | 21. |
| Zn | 59.0 | 57.0 | 47.0 | 43.0 | 26.0 | 60.0 | 56.0 | 74.0 | 49.0 |
| Cu | 16.0 | - | - | - | - | - | 3.0 | 6. 0 | 1.0 |
| Ni | 7.0 | 10.0 | 9. 0 | 4.0 | 10.0 | $16.0$ | 17.0 | $16.0$ | $13.0$ |
| $\mathrm{TiO}_{2}$ | . 73 | . 53 | . 47 | . 14 | $.08$ | $.25$ | $.28$ | $.56$ | $.13$ |
| $V$ | 68. | 38. | 52. | 4. | 4. | 16. | 11. | 55. | 7. |
| Cr | 18. | 20. | 47. | 9. | 9. | 12. | 10. | 42. | 9. |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | _ | - | - | - | - | - |
| Ta | - | - | - | - | - | - | $\rightarrow$ | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | - | - | - | - | - | - | - | - | - |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | _ | - | - | - |
| Cl | - | - | - | - | - | - | _ | - | - |
| U | - | - | - | - | - | _ | _ | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | $\rightarrow$ | - |
| Lo | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical dotabose.


Figure D. 1 (cont.). Geochemical databose.

|  | $N O V A$ |  |  |  | $S C$ | $T \mid A$ |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | NPM533 <br> PMOU <br> TONA | NPM535 <br> PMOU <br> GRAD | NPM536 <br> PMOU <br> GRAD | NPM537 <br> PMOU <br> LMON | NPM538 <br> PMOU <br> MONG | NPM539 <br> PMOU <br> GRAD | $\begin{aligned} & \text { NPM542 } \\ & \text { PMOU } \\ & \text { LMON } \end{aligned}$ | NPM543 <br> FMOU <br> GRAD | NPM544 <br> PMOU <br> TONA |
| $\mathrm{SiO}_{2}$ | 63.65 | 71.33 | 71.77 | 74.56 | 74.50 | 71.32 | 74. 26 | 64.24 | 67. 53 |
| $\mathrm{TiO}_{2}$ | . 58 | . 32 | . 25 | . 09 | . 09 | . 36 | . 11 | . 76 | . 51 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 17. 57 | 15. 27 | 15.20 | 14. 22 | 14.69 | 15.57 | 14.65 | 17.78 | 16. 22 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 4.54 | 2. 19 | 1. 97 | . 77 | . 81 | 2. 41 | . 73 | 3. 69 | 3. 44 |
| FeO | - | - | - | - | - | - | - | - | - |
| Mno | . 09 | . 05 | . 06 | . 02 | . 02 | . 06 | . 02 | . 05 | . 08 |
| MgO | 2. 02 | 1. 04 | . 86 | . 16 | . 34 | . 89 | . 51 | 1.91 | 1. 62 |
| CaO | 4. 02 | 1.61 | 1.59 | 1. 25 | . 92 | 2.14 | . 92 | 3.27 | 2.77 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3.97 | 4.08 | 4. 66 | 3.84 | 3. 35 | 4. 35 | 4. 49 | 4. 42 | 4.10 |
| $\mathrm{K}_{2} \mathrm{O}$ | 2. 12 | 3.45 | 3.13 | 4.44 | 4.75 | 2, 87 | 3. 62 | 2.47 | 2.12 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 19 | . 21 | . 24 | . 11 | . 14 | . 19 | .13 | . 24 | . 19 |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | _ | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | $\rightarrow$ | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | . 30 | .40 | .40 | .27 | 56 | . 37 | . 30 | .50 | . 70 |
| TOTAL | 99.35 | 99. 95 | 100.13 | 99. 73 | 100.17 | 100.53 | 99. 74 | 99, 33 | 99. 28 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 4. 64 | 2. 19 | 1.97 | . 77 | . 81 | 2. 41 | . 73 | 3. 69 | 3. 44 |
| A/CNK | 1.1 | 1.1 | 1.1 | 1.1 | 1. 2 | 1.1 | 1.1 | 1.1 | 1.2 |
| DI | 67.5 | 84.9 | 86. 6 | 91.4 | 91.2 | 83.4 | 91.3 | 71.9 | 75.0 |
| 8 a | 601. | 383. | 316. | 460. | 244. | 399. | 451. | 683. | 484. |
| Rb | 71. | 139. | 137. | 143. | 133. | 128. | 132. | 112. | 128. |
| Sr | 474. | 124. | 102. | 111. | 63. | 151. | 84. | 278. | 337. |
| Y | 17. | 15. | 17. | 15. | 11. | 16. | 10. | 17. | 20. |
| Zr | 182. | 115. | 117. | 76. | 39. | 128. | 47. | 220. | 141. |
| Nb | 12. | 9. | 11. | 5. | $8 .$ | 9. | $6 .$ | 7. | $10$ |
| Th | - | 12.00 | 7.00 | 9. 00 | 2. 00 | 11.00 | 3. 00 | 28.00 | 13.00 |
| Pb | 8. | 17. | 19. | 22. | 24. | 22. | 18. | 11. | 17. |
| Ga | 22. | 22. | 19. | 16. | 21. | 22. | 18. | 26. | 23. |
| Zn | 61.0 | 57.0 | 60.0 | 23.0 | 23.0 | 57.0 | 29.0 | 88.0 | 65.0 |
| Cu | 7.0 | 6.0 | - | - | - | - | - | 4. 0 | 4.0 |
| Ni | 10.0 | 9.0 | 9.0 | B. 0 | $9.0$ | $13.0$ | $8.0$ | $10.0$ | $14.0$ |
| $\mathrm{TiO}_{2}$ | . 76 | . 32 | $.24$ | $.08$ | . DS | $.34$ | $.10$ | $.78$ | $.57$ |
| $\checkmark$ | 77. | 27. | 15. | 3. | 1. | 30. | 3. | 71. | 58. |
| Cr | 21. | 21. | 13. | 3. | 6. | 25. | 8. | 47. | 24. |
| Hf | , | , | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | _ | - | - | - |
| To | - | - | - | - | - | _ | - | _ | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | - | - | - | - | - | - | - | - | - |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| $F$ | - | - | - | - | - | - | - | - | - . |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.

NOVA SCOTIA

|  | NPM545 PMOU GRAD | NPM546 PMOU LTON | NPM548 PMOU MONG | NPM549 PMOU GRAD | NPM551 <br> PMOU <br> MONG | NPM556 <br> PMOU <br> MONG | NPM558 PMOU GRAD | NPM560 <br> PNOU <br> TONA | NPM563 PMOU GRAD |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 69.28 | 63. 53 | 72.73 | 67.10 | 68.49 | 70.33 | 58. 27 | 70.27 | 71.12 |
| $\mathrm{TiO}_{2}$ | . 43 | 36 | 21 | . 65 | 12 | . 45 | 60 | . 35 | . 32 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 15.75 | 20.13 | 15.10 | 16. 62 | 15.98 | 15. 25 | 15.24 | 15.76 | 15.18 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 3.15 | 2. 33 | 1.39 | 3. 31 | 2. 85 | 2. 61 | 3.27 | 2. 54 | 2. 25 |
| FeO | - | - | - | - | - | - | - | - | - |
| MnO | . 05 | . 05 | . 04 | . 05 | . 06 | . 04 | . 04 | . 07 | . 04 |
| Mgo | 1.00 | 1. 25 | . 75 | 1.70 | 1. 43 | 1.02 | 1. 33 | . 77 | . 71 |
| CoO | 2.47 | 4.13 | . 87 | 2. 16 | 2. 35 | 1.74 | 2.27 | 2.81 | 1.92 |
| $\mathrm{No}_{2} \mathrm{O}$ | 4.18 | 5. 82 | 3.99 | 3. 94 | 3. 97 | 3.56 | 3.75 | 4. 44 | 4. 07 |
| $\mathrm{K}_{2} \mathrm{O}$ | 2. 31 | 1.54 | 1. 22 | 3. 42 | 3. 69 | 4.24 | 3.56 | 2. 25 | 3.49 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 32 | . 14 | . 23 | . 21 | . 16 | . 22 | 21 | . 22 | . 23 |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | . 40 | . 60 | . 30 | . 30 | . 30 | . 45 | . 72 | . 42 | . 50 |
| TOTAL | 99.34 | 99.88 | 99.83 | 99. 56 | 99.70 | 99.93 | 100.37 | 99.90 | 99.83 |
| $\mathrm{Fe}_{2} \mathrm{O} 3 \mathrm{t}$ | 3.15 | 2. 33 | 1.39 | 3. 31 | 2. 85 | 2. 61 | 3. 27 | 2.54 | 2. 25 |
| A/CNK | 1.1 | 1.1 | 1.2 | 1.1 | 1.1 | 1.1 | 1.1 | 1.1 | 1.1 |
| DI | 79.3 | 71.8 | 89.7 | 77.4 | 80. 0 | 83. 9 | 79.6 | 80.2 | 84.5 |
| Ba | 318. | 175. | 334. | 929. | 812. | 781. | 901. | 283. | 546. |
| Rb | 112. | 75. | 185. | 139. | 121. | 166. | 137. | 98. | 123. |
| Sr | 130. | 389. | 70. | 223. | 330. | 163. | 225. | 184. | 157. |
| Y | 24. | 11. | 15. | 15. | 18. | 16. | 17. | 15. | 17. |
| Zr | 22 S . | 123. | 88. | 206. | 123. | 151. | 217. | 210. | 146. |
| Nb | 12. | 8. | 9. | 8. | 10. | 7. | 8. | 10. | 10. |
| Th | 10.00 | 10.00 | 12.00 | 32. 00 | 3. 00 | 25.00 | 4. 00 | - | 8. 00 |
| Pb | 13. | 14. | 22. | 15. | 13. | 22. | 21. | 17. | 23. |
| Ga | 19. | 22. | 21. | 23. | 19. | 26. | 25. | 22. | 20. |
| Zn | 83.0 | 50.0 | 51.0 | 91.0 | 47.0 | 83.0 | 90.0 | 60.0 | 59.0 |
| Cu | 3.0 | 7.0 | - | 10.0 | 5. 0 | 4.0 | 8. 0 | 4.0 | 1.0 |
| Ni | 11.0 | 6. 0 | 10.0 | 10.0 | 11.0 | 16.0 | 14.0 | 9. 0 | 11.0 |
| $\mathrm{TiO}_{2}$ | . 42 | . 38 | . 20 | . 64 | . 48 | . 50 | . 61 | . 34 | . 32 |
| $\checkmark$ | 37. | 36. | 14. | 53. | 52. | 38. | 45. | 27. | 21. |
| Cr | 13. | 32. | 16. | 32. | 19. | 20. | 33. | 12. | 21. |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - |  |
| Sc | - | - | - | - | - | - | - |  |  |
| Ta | - | - | - | - | - |  |  |  |  |
| Co | - | - | - | - | - | - | - | - |  |
| Li | - | - | - | - | - | - | - | - | - |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - |  | - | - |
| W | - | - | - | - | - |  |  |  |  |
| Sn | - | - | - | - | - | - | - |  | - |
| Mo | - | - | - | - | - | - | - | - | - |
| Lo | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical databose.

NOVA SCOTIA

|  | NPM565 PMOU GRAD | NPM566 PMOU MONG | NPM578 PMOU MONG | NPM579 PMOU GRAD | NPM580 PMOU LMON | NPM581 PMOU GRAD | NPM582 PMOU TONA | NPM583 PMOU TONA | NPM584 PMOU TONA |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 70. 01 | 72.72 | 70.93 | 72.54 | 72. 79 | 71.45 | 56.24 | 72.85 | 68. 14 |
| $\mathrm{TiO}_{2}$ | 35 | . 22 | . 29 | . 25 | . 14 | 32 | 83 | 35 | 50 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 15.62 | 14.64 | 15.31 | 14.61 | 14.89 | 15. 61 | 17.00 | 15. 02 | 16. 25 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 2. 53 | 1. 47 | 2.16 | 1.47 | . 98 | 2. 29 | 3. 76 | 2.38 | 3. 85 |
| FeO | 2. | 1. | 2. | , | - | - | - | - | - |
| MnO | . 05 | . 03 | . 04 | . 03 | . 03 | . 06 | . 07 | . 04 | . 07 |
| MgO | 1.18 | . 34 | . 97 | . 34 | . 30 | . 95 | 1.56 | . 72 | 1.46 |
| CaO | 2. 05 | . 86 | 1.88 | . 86 | . 85 | 1.76 | 3.94 | 2. 35 | 3.12 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 4. 41 | 2.85 | 4. 21 | 2.85 | 2. 56 | 1. 63 | 3.94 | 3.58 | 4.02 |
| $\mathrm{K}_{2} \mathrm{O}$ | 2.90 | 5.86 | 3.17 | 5. 86 | 6. 30 | 3.09 | 1.55 | 2.19 | 1.73 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 14 | . 18 | . 20 | . 18 | . 32 | . 27 | . 30 | . 15 | . 13 |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - |  |  |  |  |
| LOI | . 40 | . 47 | 40 | .47 | . 61 | . 50 | . 50 | . 54 | 50 |
| TOTAL | 99.64 | 99. 64 | 99. 56 | 99.57 | 99. 75 | 100.93 | 99.49 | 100.27 | 99.78 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 2. 53 | 1.47 | 2.16 | 1.47 | . 96 | 2. 29 | 3. 76 | 2.38 | 3.85 |
| A/CNK | 1.1 | 1.2 | 1.1 | 1.2 | 1.2 | 1.1 | 1.1 | 1.2 | 1.2 |
| DI | 81.9 | 90.6 | 83. 8 | 90.5 | 91.3 | 85.7 | 70.0 | 81.2 | 73.9 |
| Ba | 400. | 963. |  |  | 774. | 451. |  | 657. |  |
| Rb | 112. | 157. | 130. | 150. | 234. | 156. | $59 .$ | $69 .$ | $62 .$ |
| Sr | 225. | 118. | 137. | 152. | 102. | 116. | 213. | 197. | 385. |
| Y | 17. | 16. | 19. | 13. | 19. | 19. | 27. | 22. | 11. |
| Zr | 102. | 112. | 140. | 137. | 60. | 151. | 389. | 209. | 156. |
| Nb | 13. | 10. | g. | 9. | 8. | 10. | 11. | 8. | 9. |
| Th | 5. 0 | 14.00 | 3.00 | 14.00 | 1.00 | 13.00 | 4.00 | 12.00 | 10.00 |
| Pb | 15. | 29. | 17. | 26. | 21. | 18. | 13. | 18. | 15. |
| Ga | 21. | 22. | 20. | 20. | 20. | 21. | 18. | 19. | 18. |
| Zn | 57.0 | 44.0 | 56.0 | 54.0 | 33.0 | 61.0 | 66.0 | 53.0 | 54.0 |
| Cu | 2. 0 | - | - | 1.0 | 3.0 | - | 19.0 | 1.0 | 8. 0 |
| $\mathrm{Ni}^{\mathrm{N}}$ | 9.0 | 13.0 | 12.0 | 13.0 | 18.0 | 13.0 | B. 0 | 1. 0 | 2.0 |
| $\mathrm{TiO}_{2}$ | . 37 | . 21 | . 28 | . 26 | . 13 | . 36 | . 61 | . 37 | . 49 |
| V | 37. | 5. | 24. | 19. | 6. | 24. | 50. | 24. | 37. |
| Cr | 18. | 7. | 14. | 16. | 8. | 20. | 15. | 14. | 14. |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | - | - | - | - | - | - | - | - | - |
| Be | - | - | - | - | _ | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | _ |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| Lo | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical databose.

NOVA SCOTIA

|  | NPM593 PMOU GRAD | NPM8B PMOU LMON | NPS 122 SHER MONG | NPS124 SHER MONG | NPS126 SHER MONG | $\begin{aligned} & \text { NPS200 } \\ & \text { SHER } \\ & \text { MONG } \end{aligned}$ | NPS205 SHER MONG | NSH2O5 SHEL GRAD | $\begin{aligned} & \text { NSH213 } \\ & \text { SHEL } \\ & \text { GRAD } \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 70.20 | 73.09 | 72.21 | 73. 86 | 75.00 | 72.65 | 73. 35 | 70. 66 | 74.78 |
| $\mathrm{TiO}_{2}$ | . 47 | 04 | . 20 | . 14 | . 11 | . 05 | . 13 | . 54 | . 11 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 15.69 | 15.50 | 14. 30 | 15.11 | 14. 53 | 15. 26 | 14.48 | 14.81 | 14.44 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 2.59 | 1. 07 | . 26 | . 22 | . 16 | . 24 | . 17 | 2. 52 | 1.15 |
| FeO | - | - | 1.17 | 1. 01 | . 59 | . 61 | 83 | - | - |
| MnO | . 05 | . 22 | . 02 | . 03 | . 02 | . 02 | . 02 | 05 | . 04 |
| MgO | 1.22 | . 08 | . 51 | . 08 | . 27 | . 16 | . 27 | . 85 | 20 |
| CaO | 1. 30 | . 37 | . 66 | . 53 | . 62 | . 49 | . 84 | 94 | . 63 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3. 94 | 3.10 | 3. 65 | 3.68 | 3.81 | 4.07 | 3.88 | 3.23 | 3. 36 |
| $\mathrm{K}_{2} \mathrm{O}$ | 3.32 | 5.69 | 4.53 | 4. 23 | 5.11 | 4.17 | 1. 55 | 4.28 | 4.39 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 20 | . 15 | . 22 | . 22 | . 16 | . 31 | . 25 | . 22 | . 19 |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | - | . 81 | . 81 | . 43 | . 70 | . 71 | - | - |
| $\mathrm{H}_{2} \mathrm{O}$ | - | - | . 07 | . 01 | . 01 | . 01 | . 01 | - | - |
| $\mathrm{CO}_{2}$ | - | - | . 16 | . 04 | . 11 | . 28 | . 22 | - | - |
| Cl | - | - | - | - | - | - | - | - |  |
| F | - | - | - | - | - | - | - | - |  |
| LOI | . 70 | . 74 | - | - | - | - | - | . 92 | 76 |
| TOTAL | 99.68 | 100.05 | 98.78 | 99. 97 | 100.93 | 99. 02 | 99. 51 | 99.03 | 100.05 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 2. 59 | 1.07 | 1.56 | 1. 34 | . 81 | . 92 | 1.09 | 2.52 | 1.16 |
| A/CNK | 1.3 | 1.3 | 1.2 | 1.3 | 1.1 | 1.3 | 1.2 | 1.3 | 1.3 |
| OI | 83.5 | 92. 5 | 90.0 | 91.9 | 94.6 | 92.2 | 92.3 | 85.4 | 81.7 |
| Bo | 768. | 181. | 390. | 110. | 430. | 12. | 10. | 782. | 592. |
| Rb | 115. | 178. | 242. | 216. | 199. | 305. | 210. | 116. | 82. |
| Sr | 194. | 30. | 75. | 54. | 97. | 11. | 36. | 356. | 308. |
| Y | 13. | 11. | - | - | - | - | - | 10. | 20. |
| Zr | 190. | 37. | - | - | - | - | - | 239. | 183. |
| Nb | 9. | 10. | - | - | - | - | - | 11. | 15. |
| Th | 35.00 | 2. 0 | 5. 00 | 4. 00 | 2. 00 | 2.00 | 4. 00 | 2.60 | 5. 10 |
| Pb | 25. | 36. | 17. | 16. | 30. | 7. | 15. | 15. | 17. |
| Go | 20. | 19. | - | - | - | - | - | 19. | 21. |
| Zn | 58.0 | 18.0 | 54.0 | 49.0 | 26.0 | 40.0 | 38.0 | 66.0 | 74.0 |
| Cu | - | - | 5. 0 | 7.0 | 6.0 | 5. 0 | 5. 0 |  | . |
| Ni | 7. 0 | 5. 0 | - | - | - | - | 5. | 16.0 | 20.0 |
| $\mathrm{TiO}_{2}$ | . 49 | . 03 | - | - | - | - | - | 16. | , |
| V | 36. | - | - | - | - | - | - | 121. | 93. |
| Cr | 19. | 9. | - | - | - | - | - | 102. | 111. |
| Hf | 19. | - | - | - | - | - | - | 1 | 11. |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | - | - | 187.0 | 104.0 | 61.0 | 263. 1 | 103.0 | 85. 0 | 72. 0 |
| Be | - | - | 6. 0 | 11.0 | 8. 3 | 7. 5 | 9.0 | 4.9 | 3. 2 |
| B | - | - | 12.0 | 8. 0 | 5. 0 | 8. 0 | 9. 0 | - | , |
| F | - | - | 500. | 500. | 300. | 700. | 500. | 520. | 180. |
| Cl | - | - | - | - | - | - | - |  | - |
| U | - | - | 3. 60 | 3. 20 | 1.60 | 9. 10 | 2.90 | 6. 30 | 3. 10 |
| W | - | - | 4. 0 | 4. 0 | 4.0 | 4.0 | 4. 0 | - | - |
| Sn | - | - | 14.0 | 9. 0 | 6. 5 | 35.0 | 8. 0 | 10.0 | 6. 0 |
| Mo | - | - | 2.10 | 1. 50 | 1.50 | 2. 20 | 1. 30 | - | . |
| Lo | - | - | . | - |  | 2. | , | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical databose.

|  | $N O V A$ |  |  |  | $S C$ | $\cdots T A$ |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | NSH230 <br> SHEL <br> GRAD | NSH252 <br> SHEL <br> GRAD | NSH255 <br> SHEL <br> TONA | NSH263 <br> SHEL <br> GRAD | NSH269 <br> SHEL <br> GRAD | NSH270 <br> SHEL <br> GRAD | NSH331 <br> SHEL <br> GRAD | NSH422 <br> SHEL <br> GRAD | $\mathrm{NSH} 423$ <br> SHEL <br> GRAD |
| $\mathrm{SiO}_{2}$ | 74.45 | 73. 82 | 67.74 | 73. 61 | 74.43 | 73.10 | 74. 26 | 73. 61 | 72.03 |
| $\mathrm{TiO}_{2}$ | . 10 | . 24 | . 55 | . 25 | . 14 | . 17 | . 25 | 30 | . 27 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14.35 | 14. 27 | 16.51 | 14.40 | 14.47 | 14.35 | 13.83 | 14.47 | 15.37 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 1.16 | 1.14 | 3. 09 | 1.61 | . 83 | 1. 46 | 1. 05 | 1. 54 | 2. 25 |
| FeO | - | - | - | - | - | - | - | $\rightarrow$ | - |
| MnO | . 04 | . 03 | . 05 | . 06 | . 04 | . 03 | . 03 | . 08 | . 06 |
| MgO | . 17 | . 28 | 1. 41 | . 43 | . 18 | . 24 | . 24 | . 47 | 55 |
| CaO | . 59 | . 44 | 3. 32 | 1. 25 | . 36 | 1.98 | . 48 | 1.71 | 1. 95 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3. 35 | 3. 22 | 4.47 | 3. 87 | 3. 81 | 3.44 | 3.21 | 4.22 | 4.19 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4.47 | 5.07 | 1. 81 | 3. 47 | 4. 39 | 4.17 | 4. 83 | 2.47 | 2. 21 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 33 | . 27 | .10 | . 25 | . 36 | . 14 | . 20 | .10 | . 16 |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | _ | - | _ | - | - | - | - | - | - |
| $F$ | - | - | - | - | - | - | - | - | - |
| LOI | . 93 | .92 | . 70 | . 69 | . 73 | . 67 | . 79 | . 73 | . 75 |
| TOTAL | 99. 94 | 99. 70 | S9. 75 | 99. 89 | 99.74 | 99. 75 | 99.17 | 99. 58 | 99.80 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 1.16 | 1.14 | 3.09 | 1.61 | . 83 | 1.46 | 1. 05 | 1. 54 | 2. 25 |
| A/CNK | 1.3 | 1. 2 | 1.1 | 1.2 | 1.2 | 1.0 | 1. 2 | 1.1 | 1. 2 |
| DI | 92.0 | 92. 3 | 74.4 | 88.4 | 93.4 | 86.7 | 92.0 | 85.8 | 83.2 |
| Ba | 475. | 640. | 457. | 351. | 127. | 562. | 728. | 437. | 467. |
| Rb | 127. | 195. | 53. | 147. | 160. | 85. | 161. | 89. | 88. |
| Sr | 53. | 125. | 484. | 85. | 48. | 106. | 81. | 157. | 156. |
| Y | 14. | 13. | 10. | 16. | 12. | 14. | 19. | 14. | 16. |
| Zr | 50. | 190. | 120. | 111. | 34. | 70. | 96. | 111. | 127. |
| Nb | 9. | 5. | 5. | 9. | 10. | 8. | 8. | 7. | 8. |
| Th | 5. 40 | 5. 00 | 1.50 | E. 30 | 6. 50 | 3.70 | 4. 60 | 3.80 | 3. 80 |
| Pb | 27. | 20. | 13. | 22. | 23. | 28. | 29. | 22. | 17. |
| Ga | 17. | 21. | 19. | 17. | 16. | 16. | 16. | 18. | 18. |
| Zn | 42.0 | 91.0 | 55.0 | 56.0 | 38. 0 | 33. 0 | 42. 0 | 41.0 | 54.0 |
| Cu | - | - | - | S. | 8. |  | 12.0 | 1 | 5.0 |
| Ni | 2. 0 | 9.0 | 3. 0 | 4. 0 | 4.0 | 1.0 | 2.0 | 2.0 | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| $\checkmark$ | 1. | 38. | 85. | 10. | 2. | 3. | 5. | 17. | 16. |
| Cr | 36. | 63. | 44. | 43. | 50. | 31. | 22. | 32. | 19. |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| Ta | - | - | - | - | - | _ | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 60.0 | 62.0 | 42.0 | 262.0 | 105.0 | 53. 0 | 57. 0 | 67.0 | 91.0 |
| Be | 4.9 | 3.2 | 2.8 | 2.8 | 5.1 | 2. 0 | 1.5 | 3.7 | 5. 5 |
| B | - | - | - | - | - | - | - | - | - |
| $F$ | 220. | 110. | 380. | 290. | 240. | 150. | 250. | 250. | 220. |
| Cl | 22. | 11. |  | , | 210 | . | S | , | , |
| U | 4.10 | 5. 80 | 1.80 | 9. 90 | 4.70 | 3. 50 | 5.90 | 3.90 | 3.90 |
| W | - | - | - | - | - | - | - | - | - |
| Sn | 3.0 | 31.0 | 3. 0 | 9.0 | 2.0 | 8. 0 | 16.0 | 9. 0 | 8. 0 |
| Mo | - | - | - | - | - | - | - | - | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.

NOVA SCOTIA

|  | NSH427 <br> SHEL GRAD | NSH428 SHEL GRAD | $\begin{aligned} & \text { NSH43O } \\ & \text { SHEL } \\ & \text { TONA } \end{aligned}$ | $\begin{aligned} & \text { NSH5O1 } \\ & \text { SHEL } \\ & \text { GRAD } \end{aligned}$ | NSH508 SHEL GRAD | NSH519 SHEL TONA | NSH521 <br> SHEL GRAD | NSH52.2 SHEL TONA | NSH583 SHEL GRAD |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 73.09 | 72. 54 | 57.17 | 73. 52 | 73. 35 | 67. 11 | 74. 31 | 65.99 | 74. 30 |
| $\mathrm{TiO}_{2}$ | . 32 | 21 | 35 | . 17 | . 06 | . 55 | . 24 | . 60 | . 17 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14.08 | 13.71 | 20.18 | 14. 36 | 15.11 | 16.41 | 13. 81 | 17.09 | 13.94 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 1.80 | 1.74 | 2. 33 | 1.41 | . 90 | 3. 87 | 1. 90 | 3. 91 | 1. 49 |
| FeO | - | - | - | - | - | - | - | - | - |
| MnO | . 05 | . 05 | . 07 | . 05 | . 06 | . 06 | . 04 | . 05 | 04 |
| MgO | . 45 | 65 | 1.02 | . 17 | . 24 | 1.19 | . 28 | 1.45 | . 17 |
| COO | 1. 24 | . 74 | 5.19 | . 65 | . 66 | 3. 25 | . 64 | 3. 56 | . 61 |
| $\mathrm{No}_{2} \mathrm{O}$ | 3.42 | 3. 20 | 4.69 | 3. 38 | 3.94 | 1.62 | 2.65 | 4.47 | 3.17 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4.30 | 5.11 | 1.19 | 4.83 | 3.92 | 1.55 | 5. 29 | 1.69 | 4.68 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 18 | . 23 | . 18 | . 33 | . 15 | . 15 | . 12 | . 14 | . 17 |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - |  |
| F | - | - | - | - | - | - |  |  |  |
| LOI | 58 | . 88 | . 53 | . 68 | . 81 | . 69 | 82 | . 67 | . 79 |
| TOTAL | 99. 51 | 99.08 | 102.90 | 99. 55 | 99. 20 | 99.75 | 100.10 | 99. 64 | 99.53 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3 t}$ | 1.80 | 1.74 | 2. 33 | 1.41 | . 90 | 3.87 | 1.90 | 3. 91 | 1.49 |
| A/CNK | 1.1 | 1.1 | 1.1 | 1.2 | 1.3 | 1.1 | 1.2 | 1.1 | 1.2 |
| 01 | 88.3 | 89.7 | 69.9 | 91.8 | 90.6 | 74.5 | 90.9 | 71.9 | 91.3 |
| Ba | 783. | 776. | 245. | 500. | 287. | 399. | 787. | 456. | 589. |
| Rb | 176. | 204. | 49. | 210. | 138. | 61. | 161. | 56. | 181. |
| Sr | 116. | 74. | 274. | 55. | 58. | 555. | 126. | 565. | 74. |
| Y | 18. | 22. | 13. | 22. | 12. | 8. | 25. | 11. | 15. |
| Zr | 175. | 105. | 159. | 72. | 38. | 124. | 80. | 142. | 76. |
| Nb | 9. | 9. | 7. | 9. | 8. | 6. | 8. | 6. | 10. |
| Th | 4.70 | 6. 00 | 1.80 | 6. 00 | 5.30 | 3. 10 | 4. 00 | 2.10 | 4.80 |
| Pb | 23. | 34. | 11. | 25. | 25. | 9. | 31. | 14. | 29. |
| Go | 18. | 17. | 23. | 15. | 15. |  | 15. | 20. | 18. |
| Zn | 57.0 | 75.0 | 52.0 | 50. 0 | 28.0 | 57.0 | 15.0 | 51.0 | 46. 0 |
| Cu | - | - | - | - | 28.0 | 57. | . | S1.0 | - |
| $\stackrel{\mathrm{Ni}}{ }$ | 5.0 | 11.0 | 2.0 | 6. 0 | 3.0 | 9. 0 | 5. 0 | 9.0 | 3. 0 |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | 5. | - | - |
| V | 19. | 11. | 37. | - | 84. | 54. | 8. | 75. | 5. |
| Cr | 40. | 40. | 37. | 29. | 74. | 57. | 50. | 77. | 40. |
| Hf | - | - | . | - | 7. | 5. | 5. | T. | \%. |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | _ | - | - |
| co | - | - | - | - | - | - | - | - | - |
| Li | 98.0 | 38.0 | 93. 0 | 155.0 | 68. 0 | 81.0 | 30.0 | 68.0 | 73.0 |
| Be | 3.9 | 2.8 | - | 4.1 | - | . 9 | 3.5 | 2. 3 | 3. 4 |
| B | , | - | - | . | - | - | - | - | 3. |
| F | 380. | 260. | 280. | 260. | 180. | 110. | 260. | 430. | 240. |
| Cl | - | - | - | - | - | - | - | - | - |
| U | 6.20 | 5. 70 | 3.10 | 8.50 | 2.90 | 1.50 | 7.70 | 1.10 | 2.80 |
| W | - | - | - | - | - | - | - | - | - |
| Sn | 23.0 | 12.0 | 1.0 | 11.0 | 4.0 | 1.0 | 13.0 | 2. 0 | 14.0 |
| Mo | - | , |  | - | - | - |  | - | - |
| Lo | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical databose.

|  | $N O V A$ |  |  |  | $S C$ | $T 1 A$ |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | NSH729 <br> SHEL <br> GRAD | NSH794 <br> SHEL <br> GRAD | NSH796 SHEL GRAD | NSHA26 <br> SHEL <br> GRAD | NSHA27 <br> SHEL <br> GRAD | NSHA34 <br> SHEL <br> GRAD | NSHA35 <br> SHEL <br> GRAD | NSHA84 <br> SHEL <br> GRAD | NSI74S SISL MONG |
| $\mathrm{SiO}_{2}$ | 72.90 | 73.64 | 74.50 | 72. 30 | 72.74 | 73.10 | 72.70 | 74.50 | 72.65 |
| $\mathrm{TiO}_{2}$ | . 26 | . 11 | . 09 | . 16 | . 37 | . 21 | . 17 | . 06 | . 35 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14. BB | 15.32 | 14.90 | 15.10 | 14.64 | 14.40 | 15.10 | 14.30 | 14.30 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 1. 88 | 1. 61 | 1.37 | 1.48 | 1. 65 | 1. 58 | 1.37 | . 74 | 2.85 |
| FeO | - | - | - | - | - | - | - | - | - |
| MnO | . 05 | . 05 | . 04 | . 04 | . 04 | . 05 | . 03 | - | . 05 |
| MgO | . 54 | . 53 | . 31 | . 56 | . 54 | . 54 | - 39 | . 23 | . 42 |
| CaO | . 85 | 1. 50 | . 57 | 1. 59 | . 70 | 1,64 | 1.80 | . 40 | . 88 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3. 59 | 4. 43 | 3. 48 | 5.11 | 3.27 | 4. 30 | 4.80 | 4. 13 | 3. 41 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4. 24 | 2. 52 | 4.75 | 2. 05 | 4. 53 | 3. 05 | 2. 69 | 4.05 | 4. 48 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 22 | . 22 | . 27 | . 21 | . 28 | . 16 | . 13 | . 17 | .12 |
| $\mathrm{H}_{2} \mathrm{O}+$ | , | - | - | - | - | - | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | _ | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| $F$ | - | - | - | - | - | - | - | - | - |
| LOI | . 91 | .47 | . 57 | . 54 | . 98 | 1.16 | . 77 | 1.15 | . 52 |
| TOTAL | 100.30 | 100.40 | 100.85 | 99.14 | 99.72 | 100.19 | 99.95 | 100.04 | 99. 84 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 1.86 | 1.61 | 1. 37 | 1.48 | 1. 65 | 1. 58 | 1.37 | . 74 | 2.85 |
| A/CNK | 1.2 | 1.2 | 1.3 | 1.1 | 1. 3 | 1.1 | 1.1 | 1.2 | 1.2 |
| DI | 89.2 | 87.2 | 32.7 | 86.4 | 89. 3 | 87.0 | 86.9 | 93.9 | 89.3 |
| 8 B | 388. | 659. | 405. | 250. | - | 360. | 560. | 190. | 712. |
| Rb | 277. | 7日, | 159. | 80. | - | 130. | 90. | 200. | 132. |
| Sr | 82. | 249. | 58. | 120. | - | 190. | 260. | 40. | 195. |
| Y | 11. | 21. | 15. | - | - | - | - | - | 17. |
| Zr | 132. | 127. | 82. | - | - | 100. | 80. | - | 201. |
| Nb | 7. | 7. | 9. | - | - | - | - | - | 17. |
| Th | 4.40 | . 70 | 4.10 | - | - | - | - | 2.00 | 1.80 |
| Pb | 18. | 25. | 29. | - | - | - | 6. | ¢. | 19. |
| Go | 23. | 12. | 16. | - | - | - | - | - | 19. |
| Zn | 74.0 | 29.0 | 38.0 | - | - | 36.0 | 36.0 | 5. 0 | 12.0 |
| Cu | - | - | - | - | _ | \% | , | S. | 1 |
| Ni | 7. 0 | 8.0 | - | - | - | - | . 5 | 3. 0 | 6.0 |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| $V$ | 15. | 14. | 3. | - | - | - | - | - | 27. |
| Cr | 25. | 26. | 27. | - | - | - | - | 110. | 33. |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 118.0 | 43. 0 | 54.0 | 110.0 | - | 72.0 | B9, 0 | 99.0 | 45.0 |
| Be | 1 |  |  | , | - | - | - |  | 3. 9 |
| 8 | - | - | - | - | - | - | - | - | - |
| F | 60. | 220. | 180. | 310. | - | 610. | 330. | 290. | 410. |
| Cl | , | 220. | 180 | 310 | - | d | , | 2 O | 1 |
| U | 5. 60 | 3. 90 | 5.10 | - | - | _ | 3.80 | 3. 60 | 3.50 |
| W | - | - | 5. 10 | - | - | - | - | - | - |
| Sn | 24.0 | 9. 0 | 12.0 | - | - | - | 5. 0 | 7. 0 | 11.0 |
| Mo | - | - | - | - | - | - | - | - | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.

NOVA SCOTIA

|  | NSI751 SISL MONG | NSI753 SISL MONG | NSI758 SISL MONG | NSI761 SISL MONG | NSLOO 1 SLAK MONG | NSLOO3 SLAK MONG | $\begin{aligned} & \text { NSLOO } 4 \\ & \text { SLAK } \\ & \text { MONG } \end{aligned}$ | NSL005 SLAK MONG | NSLOO6 SLAK MONG |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 73.98 | 73. 54 | 74. 35 | 72.11 | 72.62 | 72. 30 | 72. 41 | 72.85 | 72.99 |
| $\mathrm{TiO}_{2}$ | . 23 | . 17 | . 25 | 29 | 11 | . 11 | 15 | . 12 | 17 |
| $\mathrm{Al}_{2} \mathrm{O} 3$ | 13.98 | 14.40 | 14.32 | 14.47 | 15. 05 | 15.31 | 15.09 | 14.92 | 14.99 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 1.76 | 1. 98 | 1.69 | . 47 | . 10 | . 11 | . 44 | . 11 | . 11 |
| FeO | - | - | - | - | 1.02 | 1.03 | 1.05 | 1.02 | 1.33 |
| MnO | . 03 | . 04 | . 30 | . 05 | . 05 | . 05 | . 05 | . 05 | . 04 |
| MgO | . 26 | . 31 | . 21 | . 56 | . 24 | . 25 | . 34 | 25 | . 35 |
| CoO | . 74 | . 57 | . 50 | 1. 45 | . 61 | . 44 | 47 | 50 | 43 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3.69 | 3. 26 | 3.16 | 3. 57 | 4.14 | 4. 06 | 3. 91 | 4. 03 | 3.89 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4. 24 | 4.53 | 4.42 | 4.12 | 3.76 | 3.84 | 3.97 | 3. 69 | 3.94 |
| $\mathrm{F}_{2} \mathrm{O}_{5}$ | . 21 | 22 | . 19 | . 12 | . 62 | . 18 | . 47 | 51 | . 50 |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | - | - | - | 1.23 | 1.15 | . 95 | 1.02 | . 92 |
| $\mathrm{H}_{2} \mathrm{O}$ | - | - | - | - | . 12 | . 20 | . 11 | . 09 | . 14 |
| $\mathrm{CO}_{2}$ | - | - | - | - | . 19 | . 15 | . 19 | . 19 | . 09 |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | . 75 | . 76 | . 77 | . 18 | - | - | - | - | - |
| TOTAL | 99.87 | 99.78 | 100.16 | 97.70 | 99. 86 | 99. 48 | 99. 61 | 99. 35 | 99.89 |
| $\mathrm{Fe}_{2} \mathrm{O} 3 \mathrm{t}$ | 1.76 | 1.98 | 1.69 | 47 | 1.23 | 1.25 | 1.61 | 1.24 | 1.59 |
| A/CNK | 1.2 | 1.3 | 1.3 | 1.1 | 1.3 | 1.3 | 1.3 | 1. 3 | 1.3. |
| DI | 91.2 | 90.5 | 91.1 | 86.5 | 91.7 | 91.3 | 91.0 | 91.3 | 91.0 |
| Bo | 565. | 672. | 516. | 496. | 25. | 17. | 24. | 21. | 26. |
| Rb | 165. | 186. | 185. | 163. | 432. | 391. | 347. | 375. | 293. |
| Sr | 85. | 99. | 73. | 160. | 72. | 32. | 33. | 29. | 17. |
| Y | 20. | 27. | 16. | 25. | - | - | - | - | - |
| Zr | 156. | 149. | 114. | 182. | 42. | 40. | 57. | 48. | 57. |
| Nb | 20. | 18. | 19. | 19. | - | - | - | - | - |
| Th | 5.40 | 5.10 | 5.70 | 4.30 | 3. 00 | 3. 00 | 5. 00 | 4.00 | 4.00 |
| Pb | 15. | 22. | 18. | 22. | 14. | 16. | 19. | 17. | 17. |
| Ga | 25. | 24. | 25. | 20. | - | - | - | - | - |
| Zn | 67. 0 | 52.0 | 18.0 | 47.0 | 43. 1 | 52.0 | 69.0 | 57. 0 | 59.0 |
| Cu | - | - | - | - | - | - | - | - | - |
| Ni | 7. 0 | 13.0 | 5.0 | 8. 0 | - | - | - | - | - |
| $\mathrm{TH}_{2}$ | - | - | - | - | - | - | - | - | - |
| $\checkmark$ | 12. | 8. | 3. | 22. | - | - | - | - | - |
| Cr | 41. | 32. | 40. | 34. | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| Ta | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 34.0 | 92.0 | 69.0 | 77. 0 | 248.0 | 246.0 | 252.0 | 228.0 | 217.0 |
| Be | 4:6 | - | 3. 7 | 3.4 | 12.5 | 9. 0 | 14.0 | 13.0 | 7. 5 |
| B | - | $\rightarrow$ | - | - | 20.0 | 30.0 | 30.0 | 20.0 | 25.0 |
| F | 600. | 870. | 600. | 350. | 760. | 700. | 500. | 635. | 610. |
| Cl | - | - | - | - | - | - | - | - | - |
| U | 3.00 | 8.80 | 5.20 | 4.50 | 16.00 | 13.90 | 13.40 | 13.80 | 10.00 |
| W | - | - | - | - | 4.0 | 10.0 | 5.0 | 3.0 | 4. 0 |
| Sn | 11.0 | 10.0 | 11.0 | 17.0 | 19.0 | 21.0 | 22.0 | 20.0 | 12.0 |
| Mo | - | - | - | - | 2.00 | 2.00 | 1. 00 | 1. 00 | 1.00 |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.

|  | $\sqrt{V} A$ |  |  |  | $S$ | $T \mid A$ |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\begin{aligned} & \text { NSLO07 } \\ & \text { SLAK } \\ & \text { MONG } \end{aligned}$ | NSLOO8 SLAK MONG | NSLOOS <br> SLAK <br> MONG | NSLO10 <br> SLAK <br> MONG | NSLO1 1 <br> SLAK <br> MONG | NSLO12 <br> SLAK <br> MONG | NSLO13 SLAK MONG | NSL.O1 4 SLAK MONG | NSLO15 SLAK MONG |
| $\mathrm{SiO}_{2}$ | 75. 07 | 73. 25 | 73. 55 | 73.06 | 72.19 | 72. 21 | 72. 68 | 71.82 | 73.00 |
| $\mathrm{TiO}_{2}$ | . 17 | . 03 | . 07 | . 15 | . 29 | . 23 | . 21 | . 24 | . 17 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 13.87 | 15.05 | 15.27 | 15.36 | 15.43 | 15. 34 | 15. 27 | 15.83 | 15.11 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 05 | - | . 09 | . 14 | . 41 | . 31 | . 19 | . 69 | - |
| FeO | . 91 | . 45 | . 37 | 1. 25 | 1. 50 | 1.39 | 1.39 | 1.35 | 1. 22 |
| MnO | . 03 | . 03 | . 04 | . 06 | . 05 | . 07 | . 06 | . 07 | . 05 |
| MgO | . 29 | . 05 | . 06 | . 29 | . 49 | . 40 | . 39 | . 48 | . 28 |
| CaO | . 52 | . 47 | . 83 | .48 | . 47 | . 45 | . 46 | . 42 | . 48 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 4. 55 | 4.58 | 4.89 | 4.03 | 3. 59 | 3.89 | 3. 82 | 3. 80 | 4.08 |
| $\mathrm{K}_{2} \mathrm{O}$ | 3. 31 | 3. 58 | 3. 38 | 3.95 | 4.13 | 3. 98 | 4. 03 | 4.33 | 3.81 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 28 | . 62 | . 75 | . 50 | . 34 | . 43 | . 39 | . 32 | . 49 |
| $\mathrm{H}_{2} \mathrm{O}+$ | . 60 | . 60 | . 52 | . 93 | . 79 | . 97 | . 73 | . 80 | . 63 |
| $\mathrm{H}_{2} \mathrm{O}-$ | .10 | .17 | . 12 | .14 | . 19 | . 20 | .10 | .13 | .16 |
| $\mathrm{CO}_{2}$ | . 15 | .19 | . 10 | .17 | . 13 | . 08 | . 13 | .01 | . 07 |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 100.00 | 99.07 | 99.84 | 100.52 | 100.00 | 99.96 | 99.85 | 100.29 | 99. 55. |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 1.06 | 50 | 50 | 1.54 | 2. 08 | 1.85 | 1.73 | 2. 19 | 1. 35 |
| A/CNK | 1.1 | 1.2 | 1. 2 | 1.3 | 1.4 | 1.3 | 1.3 | 1.4 | 1.3 |
| DI | 93.2 | 93.3 | 93.9 | 91.7 | B9. 6 | 90.4 | 90.5 | 90.3 | 91.2 |
| Bo | 261. | 14. | 38. | 72. | 52. | 26. | 34. | 28. | 20. |
| Rb | 191. | 423. | 505. | 385. | 267. | 339. | 294. | 357. | 367. |
| Sr | 79. | 95. | 53. | 57. | 22. | 19. | 17. | 17. | 51. |
| $Y$ | - | - | - | - | - | - | - | - |  |
| Zr | 61. | 20. | 18. | 100. | 78. | 61. | 62. | 94. | 45. |
| Nb | - | - | - | - | - | - | - | - | - |
| Th | 5.00 | 1.00 | 2. 00 | 3. 00 | 7. 00 | 5.00 | 4.00 | 6.00 | 6.00 |
| Pb | 26. | 13. | 11. | 8. | 20. | 19. | 18. | 18. | 17. |
| Go | - | - | - | - | - | - | - | - | - |
| Zn | 39.0 | 16.0 | 25.0 | 75.0 | 80.0 | 81.0 | 72.0 | 94.0 | 63.0 |
| Cu | - | - | - | - | - | - | - | - | - |
| Ni | - | - | - | - | - | - | - | _ | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | _ | _ | - |
| $\checkmark$ | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| Ta | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 95.0 | 64.0 | 60. 0 | 271.0 | 135.0 | 234.0 | 210.0 | 263.0 | 246.0 |
| Be | 3. 0 | 29. 5 | 50.0 | 23.0 | 4. 5 | 5. 0 | 4. 5 | 4. 5 | 7.5 |
| B | 25.0 | 25.0 | 9.0 | 15.0 | 20.0 | 25.0 | 40.0 | 20.0 | 65.0 |
| F | 220. | 236. | 310. | 560. | 570. | 570. | 570. | 700. | 700. |
| Cl | , | - | - | - | 5 | - | - | - | - |
| U | 3.90 | 23.00 | 16.40 | 15.40 | 12. 20 | 17.60 | 10.90 | 20.10 | 15.50 |
| W | 4.0 | 1.0 | 4.0 | 4.0 | 4.0 | 4.0 | 5. 0 | 5.0 | 5.0 |
| Sn | 1.0 | 25.0 | 15.0 | 19.0 | 9. 0 | 14.0 | 11.0 | 15.0 | 20.0 |
| Mo | 1.00 | 1.00 | . 70 | 1.00 | 1.00 | 1.00 | 1.00 | 1. 00 | 1.00 |
| La Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.


Figure D. 1 (cont.). Geochernical database.

|  | $N O V A$ |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\begin{aligned} & \text { NTS814 } \\ & \text { HALI } \\ & \text { MONG } \end{aligned}$ | NTS815 <br> HALI <br> MONG | NTS816 <br> HALI <br> MONG | NTS817 <br> HALI <br> MONG | NTS818 <br> HALI <br> MONG | NTS819 <br> HALI <br> MONG | NTS820 <br> HALI <br> MONG | NTS821 <br> HALI <br> MONG | NTS822 <br> HALI <br> MONG |
| $\mathrm{SiO}_{2}$ | 73.13 | 74.03 | 72. 31 | 72. 64 | 73.49 | 74. 92 | 71.74 | 71.96 | 72. 11 |
| $\mathrm{TiO}_{2}$ | . 34 | . 24 | . 32 | . 30 | . 32 | . 26 | . 34 | . 31 | . 18 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 13. 41 | 13. 32 | 13. 72 | 13. 69 | 13.46 | 12.96 | 13. 65 | 13. 39 | 14.12 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 2. 44 | 1.75 | 2. 28 | 2. 27 | 2. 40 | 1.88 | 2. 44 | 2. 34 | 1.47 |
| FeO | - | - | - | - | - | - | - | - | - |
| Mno | . 04 | . 04 | . 05 | . 04 | . 05 | . 05 | . 04 | . 05 | . 04 |
| MgO | . 46 | . 11 | . 42 | . 31 | . 54 | . 36 | . 45 | . 34 | . 70 |
| CaO | . 87 | . 63 | . 96 | . 85 | . 85 | . 82 | . 85 | . 70 | . 65 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3. 29 | 3. 67 | 3. 68 | 3. 36 | 3.44 | 3. 37 | 3. 49 | 3. 52 | 3.49 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4.33 | 4.47 | 4.48 | 4.10 | 3. 99 | 4.08 | 4. 58 | 4. 46 | 4. 43 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 23 | . 21 | . 22 | . 26 | . 28 | . 25 | . 23 | . 22 | . 32 |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | - | . | - | - | - | - | - | - . |
| $\mathrm{H}_{2} \mathrm{O}$ | - | - | - | - | - | - | - | _ | _ |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| $F$ | - |  | - | - | - | - | - | - | - |
| LOI | . 81 | 1.02 | .77 | 1. 12 | 1.08 | . 93 | . 89 | 1. 04 | 1.13 |
| TOTAL | 99.35 | 99.49 | 99. 21 | 98.94 | 99. 88 | 99.88 | 98. 70 | 98.33 | 98. 94 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 2. 44 | 1.75 | 2. 28 | 2. 27 | 2.40 | 1.88 | 2.44 | 2. 34 | 1.47 |
| A/CNK | 1.2 | 1.1 | 1.1 | 1. 2 | 1.2 | 1.1 | 1.1 | 1.1 | 1. 2 |
| DI | 89.0 | 92.1 | 89.3 | 88. 5 | 89. 0 | 90.7 | 88.7 | 89.2 | 89, 3 |
| Bo | 63. | 240. | $276 .$ | 201. | 204. | 206. | 279. | 244. | 269. |
| Rb | $250$ | 258. | 249. | 241. | 235. | 224. | 240. | 253. | 218. |
| Sr | 27. | 62. | 69. | 59. | 60. | 57. | 76. | 66. | 52. |
| Y | 13. | 17. | 17. | 16. | 17. | 14. | 17. | 17. | 12. |
| Zr | 77. | 130. | 124. | 106. | 120. | 93. | 127. | 115. | 56. |
| Nb | 12. | 15. | 14. | 13. | 15. | 11. | 13. | 13. | 10. |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Ga | - | - | - | - | - | - | - | - | - |
| Zn | 6.0 | 35.0 | 35.0 | 37.0 | 34.0 | 22.0 | 27.0 | 36.0 | 5.0 |
| Cu | - | 4. 0 | - | 1.0 | 2.0 | 3.0 | 34.0 | 23.0 | 13.0 |
| Ni | 1.0 | 4.0 | 1.0 | 1.0 | 3.0 | 3. 0 | 2.0 | 2.0 | 2.0 |
| $\mathrm{TO}_{2}$ | - | $\rightarrow$ | - | - | - | - | - | - | - |
| $\checkmark$ | 16. | 26. | 27. | 23. | 24. | 20. | 27. | 24. | 11. |
| Cr | 1 | . |  |  | 2 |  |  |  | 1 |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | 3. | 1. | 4. | 3. | 4. | 2. | 5. | 3. | 1. |
| Li | 3. | 1. | 4. | , | 4. | 2. | 5. | 3. | 1. |
| Be | - | - | - | - | - | - | - | - | - |
| 8 | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | _ | - | _ | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | 2.0 | 3.0 | - | 3. 0 | 3.0 | 2. 0 | 3. 0 | 4.0 | 6. 0 |
| Mo | - | - | - | - | - | - | - | - | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.

## NOVA SCOTIA

|  | $\begin{aligned} & \text { NTS823 } \\ & \text { HALI } \\ & \text { MONG } \end{aligned}$ | $\begin{aligned} & \text { NTS824 } \\ & \text { HALI } \\ & \text { MONG } \end{aligned}$ | $\begin{aligned} & \text { NTS825 } \\ & \text { HALI } \\ & \text { MONG } \end{aligned}$ | $\begin{aligned} & \text { NTS826 } \\ & \text { HALI } \\ & \text { MONG } \end{aligned}$ | NTS827 HALI MONG | $\begin{aligned} & \text { NTS828 } \\ & \text { HALI } \\ & \text { MONG } \end{aligned}$ | $\begin{aligned} & \text { NTS829 } \\ & \text { HALI } \\ & \text { MONG } \end{aligned}$ | NTS830 HALI ALAS | NTS831 HALI ALAS |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 71.97 | 75. 28 | 72.68 | 75. 28 | 75.47 | 73. 54 | 75.02 | 74.71 | 75.05 |
| $\mathrm{TiO}_{2}$ | . 18 | . 15 | . 18 | . 16 | . 14 | 15 | . 13 | . 11 | . 11 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14.10 | 13.74 | 14.58 | 13.16 | 13.14 | 13. 39 | 13.75 | 13.90 | 13.78 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 1. 54 | 1. 52 | 1. 51 | 1. 81 | 1. 43 | 1.84 | 1.43 | 1. 44 | 1.62 |
| FeO | - | - | - | - | - | - | - | - | - |
| MnO | . 04 | . 04 | . 03 | . 05 | . 04 | . 06 | . 05 | . 04 | . 04 |
| MgO | . 24 | . 04 | . 05 | . 07 | . 15 | . 09 | . 08 | . 05 | . 18 |
| CoO | . 62 | . 47 | . 65 | . 48 | . 56 | . 61 | . 55 | . 44 | . 43 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3.54 | 3.44 | 3. 99 | 3. 54 | 3.11 | 3. 59 | 3. 42 | 3. 68 | 3.72 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4. 48 | 4.07 | 4.66 | 4. 25 | 1. 40 | 4. 25 | 4.43 | 4.08 | 4.12 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 30 | . 25 | . 33 | . 20 | . 20 | . 27 | . 20 | . 24 | . 24 |
| $\mathrm{H}_{2} \mathrm{O}+$ | . 3 | . | . 3 | . 20 | . 20 | . 2 | . | - | - |
| $\mathrm{H}_{2} \mathrm{O}$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | 1.08 | 1.05 | 1.08 | . 82 | . 88 | . 90 | . 87 | 1. 05 | . 99 |
| TOTAL | 98.09 | 100.07 | 99.74 | 99.82 | 100.12 | 98. 81 | 99.96 | 99.74 | 100.28 |
| $\mathrm{Fe}_{2} \mathrm{O} 3 \mathrm{t}$ | 1.54 | 1.52 | 1.51 | 1.81 | 1.43 | 1.84 | 1.43 | 1.44 | 1. 62 |
| A/CNK | 1.2 | 1.3 | 1.1 | 1.2 | 1.2 | 1.2 | 1.2 | 1.2 | 1.2 |
| DI | 89.8 | 92.5 | 92.4 | 92. 9 | 92. 8 | 91.3 | 92.6 | 92. 6 | 93.0 |
| Ba | 270. | 43. | 286. | 52. | 82. | 72. | 53. | 5. | 6. |
| Rb | 214. | 464. | 215. | 384. | 328. | 353. | 342. | 431. | 413. |
| Sr | 55. | 21. | 57. | 23. | 26. | 34. | 39. | 7. | 7. |
| Y | 8. | 3. | 9. | 13. | 13. | 7. | 10. | 7. | 7. |
| Zr | 59. | 43. | 59. | 52. | 48. | 53. | 46. | 41. | 38. |
| Nb | 11. | 15. | 12. | 13. | 9. | 10. | 9. | 12. | 13. |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Go | - | - | - | - | - | - | - | - | - |
| Zn | 17.0 | 7.0 | 9. 0 | 23.0 | 8. 0 | 25.0 | 10.0 | - | - |
| Cu | 5.0 | 8. 0 | 5.0 | 2.0 | 3.0 | 7. 0 | - | 1.0 | 3.0 |
| Ni | 3.0 | 3.0 | 2. 0 | 1.0 | 1.0 | 1.0 | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| $\checkmark$ | 13. | 9. | 13. | 10. | 9. | 11. | 9. | 7. | 9. |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | _ | - | - | - | - |
| Ta | - | - | - | - | - | - | - | - | - |
| Co | 2. | 1. | 1. | 2. | 1. | 2. | 2. | - | - |
| Li | - | - | - | - | - | - | - | - | - |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| w | - | - | - | - | - | - | - | - | - |
| Sn | 6. 0 | - | 6. 0 | 7. 0 | 9. 0 | 5. 0 | 8. 0 | 14.0 | 12.0 |
| Mo | , | - |  | , | , | 5. | - |  | - |
| Lo | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.

NOVA SCOTIA

|  | $\begin{aligned} & \text { NTS832 } \\ & \text { HALI } \end{aligned}$ ALAS | $\begin{aligned} & \text { NTS833 } \\ & \text { HALI } \end{aligned}$ ALAS | NTS834 hall ALAS | $\begin{aligned} & \text { NTSE85 } \\ & \text { HALI } \\ & \text { ALAS } \end{aligned}$ | NTS836 HALI MONG | NTS837 HALI MONG | NTS838 HALI ALAS | NTS839 HALI alas | NTS840 HALI MONG |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 74.54 | 75. 54 | 74. 81 | 75. 58 | 14.90 | 75.80 | 75. 67 | 75. 43 | 74.69 |
| $\mathrm{TiO}_{2}$ | . 11 | . 12 | . 11 | . 10 | 17 | 12 | 11 | . 13 | 23 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 13. 56 | 13.70 | 13.58 | 14.02 | 13.09 | 13.70 | 13.31 | 12.95 | 13.57 |
| $\mathrm{Fe}_{2} \mathrm{O} 3$ | 1.25 | 1.31 | 1. 33 | 1.03 | 1.78 | 1. 57 | 1.21 | 1.60 | 1.94 |
| FeO | - |  | - | - | - |  | - | - |  |
| MnO | . 05 | . 05 | . 04 | . 03 | . 05 | . 04 | . 05 | 04 | . 05 |
| M9O | . 25 | . 05 | . 02 | . 03 | . 14 | . 05 | . 13 | . 08 | 33 |
| CoO | . 44 | . 42 | . 10 | . 12 | . 78 | . 39 | 40 | . 40 | 58 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3. 40 | 3. 45 | 3. 58 | 3. 60 | 3.14 | 3. 18 | 3. 62 | 3. 53 | 3. 54 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4.13 | 4.17 | 4.32 | 3. 79 | 4.22 | 4.32 | 4.53 | 4.76 | 4.35 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 23 | . 21 | . 22 | . 20 | . 20 | . 22 | . 18 | . 14 | . 24 |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{Cl}^{\mathrm{C}}$ | - | - | - | - | - | - |  | - |  |
| F | - | - | - | - | - | - |  | - | - |
| LOI | 1.01 | 1.01 | 1.10 | 1. 20 | 1.08 | 1.13 | . 91 | . 83 | 1.11 |
| TOTAL | 98.97 | 100.04 | 99. 51 | 100.00 | 99.56 | 101.63 | 100.12 | 99. 69 | 100.53 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 1. 25 | 1.31 | 1.33 | 1.03 | 1.78 | 1.57 | 1.21 | 1.40 | 1. 94 |
| A/CNK | 1.3 | 1.3 | 1.2 | 1.3 | 1.2 | 1.2 | 1.2 | 1.1 | 1.2 |
| ${ }^{\circ}$ | 91.5 | 92.9 | 93.0 | 92.6 | 90.7 | 94.5 | 94.1 | 94.1 | 92.0 |
| Bo | 3. | 8. | 6. | - | 58. | 34. | 2. | 10. | 116. |
| Rb | 424. | 417. | 417. | 405. | 320. | 324. | 311. | 279. | 327. |
| Sr | 7. | 8. | 7. | 5. | 23. | 12. | 8. | 15. | 38. |
| Y | 7. | 7. | 5. | 3. | 14. | 14. | 10. | 14. | 15. |
| Zr | 37. | 36. | 32. | 29. | 59. | 4. | 33. | 36. | 80. |
| Nb | 10. | 11. | 11. | 8. | 12. | 12. | 9. | 9. | 13. |
| Tn | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Go | - | - | - | - | - | - |  | - | - |
| Zn | 11.0 | 12.0 | - | - | 28.0 | - | - | 18.0 | 13.0 |
| Cu | 9.0 | 1.0 | 13.0 | 12.0 | 3.0 | 8.0 | 2.0 | 33.0 | 2.0 |
| ${ }^{\mathrm{Ni}}$ | 3.0 | - | 2.0 | - | - | 3.0 |  | - | 2.0 |
| $\mathrm{TiO}_{2}$ | $\varepsilon$ | 6 | 9. | 6 | ${ }^{-}$ | ? | - | - | . |
| $\stackrel{\mathrm{Cr}}{ }$ | 6. | 6. | 9. | 6. | 12. | 7. | 6. | 7. | 9. |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | 2. | - | 3. | - | 3. | 2. | 2. | - | 2. |
| $\mathrm{Li}_{\mathrm{Be}}$ | - | - | - | - | - | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| Sn | 12.0 | 12.0 | 10.0 | 9.0 | 9.0 | 7.0 | 9.0 | 7.10 | 7.0 |
| Mo |  | - | , | . | 9.0 | 7.0 |  | . | . |
| $\stackrel{\mathrm{LO}}{\mathrm{Co}}$ | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical datobose.

## NOVA SCOTIA

|  | NWG627 WEST MONG | NWG636 WEST <br> MONG | NWG771 <br> WEST <br> MONG | NWG779 WEST MONG |
| :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 73.20 | 70.95 | 73.30 | 74.05 |
| $\mathrm{TiO}_{2}$ | . 28 | . 43 | 22 | . 17 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 15.12 | 15. 20 | 15.09 | 14.85 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 1.65 | 2. 86 | 1. 21 | 1. 30 |
| FeO | - | - | - | - |
| Mno | . 04 | . 05 | . 03 | . 04 |
| MgO | . 51 | . 85 | . 36 | . 29 |
| CaO | . 43 | 1. 29 | . 80 | 55 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3.10 | 3. 63 | 3. 76 | 3.37 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4.72 | 3.77 | 4.53 | 4.60 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 27 | . 31 | . 30 | . 16 |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}^{-}$ | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - |
| Cl | - | - | - | - |
| F | - | - | - | - |
| LOI | . 82 | . 61 | 65 | . 61 |
| total | 100.14 | 99.95 | 100.08 | 99.99 |
| $\mathrm{Fe}_{2} \mathrm{O} 3 \mathrm{t}$ | 1.65 | 2. 86 | 1.21 | 1.30 |
| A/CNK | 1.4 | 1.2 | 1.2 | 1.3 |
| DI | 90.3 | 85.2 | 91.7 | 91.4 |
| Ba | 347. | 693. | 596. | 768. |
| Rb | 297. | 118. | 157. | 178. |
| Sr | 67. | 146. | 92. | 84. |
| Y | 13. | 19. | 11. | 23. |
| zr | 119. | 168. | 79. | 73. |
| Nb | 12. | 12. | 9. | 11. |
| Th | 4. 00 | 5. 00 | 5. 50 | 4.00 |
| Pb | 22. | 25. | 30. | 27. |
| Ga | 20. | 21. | 18. | 17. |
| Zn | 54.0 | 58.0 | 58.0 | 48.0 |
| Cu | - | - | - | - |
| Ni | 5.0 | 7.0 | 3.0 | 5. 0 |
| $\mathrm{TiO}_{2}$ | - | - | - | - |
| V | 17. | 23. | 6. | 4. |
| Cr | 28. | 28. | 33. | 26. |
| Hf | - | - |  | - |
| Cs | - | - | - | - |
| Sc | - | - | - | - |
| Ta | - | - | - | - |
| Co | - | - | - | - |
| Li | 125.0 | 33.0 | 55.0 | 30.0 |
| Be | 4.0 | - | 3.0 | - |
| B | - | - | - | - |
| F | 560. | 310. | 210. | 360. |
| Cl | - | - | - | - |
| U | 4. 00 | 3.80 | 5.00 | 3. 60 |
| W | - | - | - | - |
| Sn | 29.0 | 11.0 | 5.0 | 9. 0 |
| Mo | - | - | - | - |
| Lo | - | - | - | - |
| Ce | - | - | - | - |

Figure D. 1 (cont.). Geochemical dotabase.

MOROCCO

|  | MAG1 1 ZAER GRAD | MAG12 <br> ZAER <br> TONA | MAG13 ZAER GRAD | MAG15 ZAER GRAD | MAG18 ZAER GRAD | MAG2 ZAER MONG | MAG2O ZAER GRAD | MAG21 ZAER GRAD | MAG22 <br> ZAER <br> GRAD |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 65.10 | ธ6. 70 | 68. 30 | 70.40 | 86. 70 | 67. 20 | 57. 30 | 67.00 | 72.50 |
| $\mathrm{TiO}_{2}$ | . 50 | 40 | . 40 | . 40 | . 50 | . 65 | . 45 | 55 | 20 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 16.70 | 17. 50 | 15. 10 | 14.90 | 16. 50 | 16. 50 | 15.90 | 15.90 | 14.10 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 3. 90 | 2. 93 | 2. 90 | 1. 73 | 2.11 | 1.55 | 2. 73 | 3. 34 | 1. 33 |
| FeO | - | . 15 | - | 65 | 1. 88 | 2. 17 | . 15 | 14 | 15 |
| Mno | . 87 | . 05 | . 0.5 | . 05 | . 05 | . 05 | . 05 | 05 | . 03 |
| MgO | 1.50 | 1.10 | . 90 | 1.00 | 1.90 | 1.90 | 1.00 | 1.35 | 55 |
| CoO | 2. 30 | 1.99 | 1. 30 | 1.35 | 2. 60 | 2.90 | 1.40 | 1.85 | . 70 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3.50 | 3. 70 | 3. 50 | 3.00 | 3. 65 | 3. 10 | 2. 70 | 3.00 | 3.20 |
| $\mathrm{K}_{2} \mathrm{O}$ | 3. 60 | 3. 40 | 3.70 | 4.75 | 3.25 | 2.80 | 4.30 | 4.00 | 4.80 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | 3. | 3. | , | 1. | 3. | 2. | 1. | 1. | 4.8 |
| $\mathrm{H}_{2} \mathrm{O}+$ | 2.22 | 1.96 | 2. 36 | 1.05 | 1.03 | . 84 | . 43 | 1.86 | 1.55 |
| $\mathrm{H}_{2} \mathrm{O}-$ | . 55 | . 74 | . 40 | . 24 | . 03 | - | . 72 | . 49 | . 85 |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - |  |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 99.94 | 100. 63 | 100. 21 | 99.52 | 100.20 | 99.96 | 98.13 | 99. 53 | 99.96 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 3.90 | 3.10 | 2. 90 | 2. 45 | 4. 20 | 3. 96 | 2. 90 | 3. 50 | 1.50 |
| A/CNK | 1.2 | 1.3 | 1.4 | 1.2 | 1.2 | 1.2 | 1.5 | 1.3 | 1.2 |
| DI | 76.8 | 79.8 | 83.4 | 84.9 | 75.1 | 72.8 | 79.8 | 79. 2 | 90.8 |
| Ba | 480. | 310. | 420. | 458. | 285. | 540. | 358. | 402. | 330. |
| Rb | 120. | 192. | 150. | 260. | 109. | 121. | 241. | 222. | 303. |
| Sr | 280. | 325. | 175. | 250. | 340. | 290. | 245. | 260. | 150. |
| Y | - | - | - | - | - | - | - | - | - |
| Zr | - | - | - | - | - | - | - | - | - |
| Nb | - | - | - | - | - | - | - | - | - |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Ga | - | - | - | - | - | - | - | - | - |
| Zn | - | 34.0 | - | 30.0 | 34.0 | 45.0 | 36.0 | 28.0 | 22.0 |
| Cu | - | 2. 0 | - | 6. 0 | 10.0 | 13.0 | 4.0 | 7. 0 | 7. 0 |
| Ni | - | . | - | G. | . | 13.0 |  | - |  |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| $\checkmark$ | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| Ta | - | - | - | - | - | _ | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 50.0 | 58.0 | 90.0 | 85. 0 | 44.0 | 45.0 | 59.0 | 57.0 | 97. 0 |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| $F$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | _ | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | 10.0 | - | 10.0 | 10.0 | 10.0 | 10.0 | 10.0 | 10.0 |
| Mo | - | - | - | - | 10.0 | 10. | 10.0 | 10.0 | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical databose.

## MOROCCO

|  |  | MAG24 ZAER GRAD |  | MAG28 ZAER GRAD |  | MAG35 ZAER GRAD | $\begin{aligned} & \text { MAGJ6 } \\ & \text { ZAER } \\ & \text { GRAD } \end{aligned}$ |  | MAG40 ZAER MONG |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 70.30 | 65.10 | 71.50 | 66. 88 | 64.70 | 69.20 | 69. 50 | 70.35 | 70.80 |
| $\mathrm{TiO}_{2}$ | . 35 | . 45 | . 15 | . 35 | . 55 | . 45 | . 20 | . 25 | . 15 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 15.20 | 16.10 | 15.30 | 17.15 | 17. 20 | 16.80 | 16.10 | 16.05 | 16. 70 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 2. 42 | 3. 80 | 1.73 | 2. 75 | 3. 44 | 2. 64 | 2. 10 | 1. 46 | 1. 97 |
| FeO | . 07 | - | . 15 | . 14 | . 14 | . 14 | - | . 22 | . 07 |
| MnO | . 04 | . 05 | . 03 | . 02 | . 04 | . 03 | . 03 | . 02 | . 01 |
| MgO | . 95 | 1.30 | . 54 | 1.10 | 1. 55 | 1.20 | . 80 | . 40 | . 34 |
| CaO | 1.05 | 2. 50 | . 48 | 1.40 | 1.80 | 1.60 | 1.30 | . 45 | . 25 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3.20 | 3.80 | 1.90 | 1. 36 | 2. 60 | 2. 10 | 2.90 | 3. 00 | 1.80 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4.30 | 3. 30 | 4. 40 | 5.80 | 3.90 | 3.80 | 4.50 | 5.00 | 4.80 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1.71 | 1.72 | 2. 60 | . 78 | 2. 65 | 2.34 | 2.02 | 2. 25 | 2. 91 |
| $\mathrm{H}_{2} \mathrm{O}-$ | . 18 | . 39 | . 57 | . 20 | . 56 | . 48 | . 34 | . 30 | . 46 |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | _ | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | _ | - | _ | - | - | - |
| TOTAL | 99.77 | 99. 51 | 99.35 | 97. 95 | 99.13 | 100.78 | 99. 59 | 99.75 | 100. 26 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 2. 50 | 3. 80 | 1.90 | 2. 92 | 3. 60 | 2.80 | 2.10 | 1.70 | 2. 05 |
| A/CNK | 1.3 | 1.1 | 1.8 | 1. 6 | 1.5 | 1. 6 | 1.3 | 1.4 | 1.9 |
| DI | 85. 7 | 77.4 | 86. 6 | 78.7 | 76.0 | 79.7 | 84.9 | 89.4 | 87. 5 |
| Ba | 345. | 115. | 235. | 235. | 345. | 338. |  |  |  |
| Rb | 274. | 115. | 330. | 314. | 270. | $262 .$ | $145$ | $452 .$ | $493 .$ |
| Sr | 215. | 305. | 140. | 335. | 290. | 245. | 220. | 100. | 100. |
| $Y$ | - | - | - | - | - | - | - | - | - |
| Zr | - | _ | - | - | - | - | - | - | - |
| Nb | - | - | - | - | - | - | - | - | - |
| Th | - | - | - | _ | - | - | _ | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Go | $\rightarrow$ | - | - | - | - | - | - | - | - |
| Zn | 34.0 | - | 33.0 | 35.0 | 57.0 | 20.0 | - | 37.0 | 71.0 |
| Cu | 5.0 | - | 60.0 | 9. 0 | 15.0 | 5.0 | - | 3. 0 | 2. 0 . |
| Ni | - | - | - | - | - | - | _ | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| $\checkmark$ | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | _ |
| Hf | - | - | - | - | - | _ | - | - | - |
| Cs | - | - | - | - | _ | - | - | - | _ |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | _ | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 89.0 | 45.0 | 99.0 | 66. 0 | 77.0 | 39.0 | 60.0 | 103.0 | 54.0 |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | - | - | - | - | - | _ | - |
| F | - | - | - | _ | - | _ | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | 11.0 | - | 11.0 | 11.0 | 13.0 | 10.0 | - | 20.0 | 21.0 |
| Mo | - | - | - | - | - | - | _ | - | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.

## MOROCCO

|  | MAG41 ZAER GRAD | MAG42 <br> ZAER <br> TONA | MAG43 ZAER MONG | MAG44 ZAER MONG | MAG45 ZAER MONG | MAG46 <br> ZAER <br> GRAD | MAG49 ZAER MONG | MAG5 ZAER TONA | MAG50 ZAER GRAD |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 71.80 | 67. 30 | 71. 50 | 71.90 | 73.00 | 72.00 | 75.00 | 65.20 | 73.50 |
| $\mathrm{TiO}_{2}$ | . 10 | . 50 | . 25 | . 10 | . 15 | . 15 | . 05 | . 50 | . 10 |
| $\mathrm{Al}_{2} \mathrm{O} 3$ | 15.10 | 16.60 | 15.60 | 15. 30 | 15. 40 | 15.10 | 14.70 | 16. 50 | 15. 60 |
| $\mathrm{Fe}_{2} \mathrm{O} 3$ | 1.13 | 1.55 | 1.29 | 1.13 | 1. 29 | 1.37 | . 92 | 3.25 | . 80 |
| Feo | . 15 | 2.03 | . 14 | . 15 | . 14 | . 07 | . 07 | . 22 | - |
| MnO | 02 | . 05 | . 02 | . 02 | - | . 01 | - | . 06 | . 03 |
| Mgo | . 33 | 1.80 | . 33 | . 35 | . 30 | . 12 | . 21 | 1. 40 | 21 |
| CaO | . 47 | 2.55 | . 47 | . 65 | . 40 | . 58 | . 30 | 2.50 | . 50 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 2.80 | 3. 55 | 3. 10 | 3.10 | 2. 90 | 2.80 | 3.10 | 3.60 | 3.00 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4.70 | 3.25 | 5.15 | 5.10 | 4. 80 | 4.80 | 4.80 | 3.10 | 1.50 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . |  | 5.15 | S | d. |  | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1.95 | . 81 | 1.51 | 1.30 | 1.78 | 1. 58 | . 99 | 1.47 | 1.34 |
| $\mathrm{H}_{2} \mathrm{O}-$ | . 45 | - | . 16 | . 40 | . 20 | . 37 | . 05 | . 52 | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| total | 99.00 | 99.99 | 99. 62 | 99. 50 | 100.36 | 99. 25 | 100.19 | 99.33 | 99. 58 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3 t}$ | 1.30 | 3.80 | 1. 45 | 1. 30 | 1. 45 | 1. 45 | 1.00 | 3. 50 | . 80 |
| A/CNK | 1.4 | 1.2 | 1.4 | 1.3 | 1.4 | 1.4 | 1.4 | 1.2 | 1.5 |
| DI | 89.7 | 75.1 | 90.5 | 90.4 | 91.2 | 89.3 | 93.2 | 76.3 | 90.6 |
| Ba | 302. | 405. | 186. | 100. | 204. | 136. | 100. | 330. | 100. |
| Rb | 543. | 245. | 398. | 729. | 560. | 467. | 582. | 150. | 528. |
| Sr | 85. | 358. | 85. | 65. | 75. | 65. | 50. | 315. | 55. |
| Y | - | - | - | - | - | - | - | - | - |
| Zr | - | - | - | - | - | - | - | - | - |
| Nb | - | - | - | - | - | - | - | - | - |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Ga | - | - | - | - | - | - | - | - | - |
| Zn | 35.0 | 47.0 | 25.0 | 48.0 | 30.0 | 28. 0 | 26.0 | 50.0 | 45.0 |
| Cu | 4.0 | 4. 0 | 2.0 | 4.0 | 2. 0 | 4.0 | 7.0 | 8. 0 | 4. 0 |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| V | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 159.0 | 31.0 | 118.0 | 149.0 | 79.0 | 117.0 | 44.0 | 45. 0 | 183.0 |
| Be | 159.0 | , | 11.0 | 19.0 | 19. | 117. | 11. | , | 183. |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | 26.0 | 10.0 | 24.0 | 18.0 | 20.0 | 16.0 | 22.0 | 10.0 | 19.0 |
| Mo | - | - | - | - | - | - | - | - | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical datobase.

| MOROCCO |  |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | MAGS1 ZAER GRAD | $\begin{aligned} & \text { MAGGS2 } \\ & \text { ZAER } \\ & \text { GRAD } \end{aligned}$ | MAG53 ZAER MONG | MAG54 <br> ZAER <br> MONG | MAG55 zAER. MONG | MAG55 ZAER MONG | MAG57 <br> ZAER <br> GRAD | MAG58 ZAER MONG | MAG59 <br> ZAER <br> MONG |
| $\mathrm{SiO}_{2}$ | 71.70 | 73.40 | 71.90 | 57.80 | 71.50 | 72.70 | 72.60 | 74.00 | 74.10 |
| $\mathrm{TiO}_{2}$ | . 05 | . 10 | . 15 | . 30 | . 20 | . 30 | . 30 | . 15 | . 02 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 15. 65 | 16. 30 | 15. 20 | 18. 30 | 15.55 | 15. 30 | 15. 80 | 14.50 | 15.40 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 73 | 1.10 | 1.13 | 1.55 | 1.50 | 1.95 | 1.75 | . 60 | 1.30 |
| FeO | . 15 | - | . 15 | - | - | - | - | 36 | - |
| MnO | . 03 | . 02 | . 02 | . 02 | . 01 | . 02 | . 02 | 03 | 02 |
| MgO | . 23 | . 24 | . 28 | . 38 | . 25 | 45 | 10 | 31 | 23 |
| CoO | 42 | . 40 | . 57 | 1.08 | 30 | . 82 | 13 | 65 | 40 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3.00 | 2. 70 | 3. 55 | 3. 20 | 2. 00 | 3. 10 | 2. 30 | 3. 50 | 2. 35 |
| $\mathrm{K}_{2} \mathrm{O}$ | 5. 30 | 5. 50 | 5. 00 | 5. 40 | 4.20 | 4.30 | 4.50 | 4. 20 | 4.25 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ |  |  | - | - | - | - | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1.55 | 1.47 | 1.47 | 2.04 | 3.00 | 1.41 | 1.93 | 1.28 | 1.38 |
| $\mathrm{H}_{2} \mathrm{O}-$ | . 25 | . 09 | . 35 | . 39 | . 10 | . 23 | . 33 | . 09 | . 18 |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - |  |
| $\stackrel{\text { cl }}{\text { F }}$ | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 99.06 | 101.32 | 99. 77 | 100.56 | 99.02 | 100.58 | 100.36 | 99.87 | 99.63 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3 \mathrm{t}}$ | . 90 | 1.10 | 1. 30 | 1.65 | 1.50 | 1.95 | 1.75 | 1.00 | 1.30 |
| A/CNK | 1.4 | 1. 5 | 1.2 | 1.4 | 1.9 | 1.4 | 1.7 | 1.3 | 1.7 |
| DI | 91.1 | 92.2 | 91.7 | 86.7 | 87.5 | 88.9 | 88.5 | 91.4 | 89.4 |
| Bo | 100. | 100. | 170. | 350. | 100. | 235. | 214. | 204. | 100. |
| Rb | 729. | 513. | 480. | 458. | 443. | 425. | 570. | 455. | 580. |
| Sr | 65. | 70. | 65. | 120. | 50. | 85. | 60. | 75. | 30. |
| Y | - | - | - | - | - | - | - | - | - |
| Zr | - | - | - | - | - | - | - | - | - |
| Nb | - | - | - | - | - | - | - | - | - |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Ga | - | - | - | - | - | - | - | - | - |
| zn | 1.0 | 37.0 | 47.0 | 46.0 | 37.0 | 56.0 | 54.0 | 55.0 | 52.0 |
| Cu | 3.0 | 2.0 | 1.0 | 2.0 | 2. 1 | 1.0 | 9.0 | 2.0 | 1.0 |
| ${ }^{\mathrm{Ni}}$ | - | - | - | - | - | - |  | - |  |
| $\mathrm{V}^{10}$ | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 149.0 | 150.0 | 128.0 | 56.0 | 126.0 | 131.0 | 142.0 | 167. 0 | 162.0 |
| Be | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - |  | - |  |  |
| Cl | - | - | - | - | - | - | - | - | - |
| u | - | - | - | - | - | - | - | - | - |
| w | - | - | - | - | - | - | - | - | 28.0 |
| Sn | 23.0 | 17.0 | 15.0 | 10.0 | 14.0 | 18.0 | 18.0 | 25.0 | - |
| Mo | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.

|  | $N B P O B O$ |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | MAG60 <br> ZAER <br> GRAD | MAG61 ZAER GRAD | MAG62 <br> ZAER <br> MONG | MAG63 <br> ZAER <br> MONG | $\begin{aligned} & \text { MAG64 } \\ & \text { ZAER } \\ & \text { MONG } \end{aligned}$ | MAG65 <br> ZAER <br> MONG | MAG66 <br> ZAER <br> MONG | MAG67 <br> ZAER <br> MONG | MAG68 ZAER GRAD |
| $\mathrm{SiO}_{2}$ | 65.00 | 66.40 | 70.90 | 71.70 | 74.10 | 70.80 | 70.60 | 72.30 | - |
| $\mathrm{TiO}_{2}$ | . 70 | . 55 | . 35 | . 25 | . 25 | . 35 | . 20 | . 10 | - |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 16.70 | 15.70 | 15.70 | 15. 50 | 14.80 | 15.10 | 15.10 | 15.30 | - |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 4.43 | 4. 30 | 1.85 | 1. 30 | 1.50 | 1.90 | 1. 55 | 1.25 | - |
| FeO | . 15 | - | - | - | - | - | - | - | - |
| Mno | . 07 | . 05 | . 02 | .01 | . 02 | . 04 | . 01 | - | - |
| MgO | 1.80 | 1.60 | . 50 | . 25 | . 30 | . 45 | . 51 | . 30 | - |
| CaO | . 29 | 2. 35 | . 70 | . 43 | . 51 | 1.05 | . 51 | . 68 | - |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3.80 | 3. 30 | 2. 10 | 2. 90 | 2.70 | 3. 20 | 1.80 | 3.40 | - |
| $\mathrm{K}_{2} \mathrm{O}$ | 3.00 | 3.80 | 4.30 | 5. 20 | 4.40 | 4.20 | 6. 30 | 4.40 | - |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1.92 | 1.97 | 2.10 | 1. 38 | 1.40 | 2. 42 | 1.85 | 1.39 | - |
| $\mathrm{H}_{2} \mathrm{O}-$ | . 78 | . 46 | . 58 | . 21 | . 12 | . 39 | . 83 | . 41 | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 98.64 | 100.48 | 99.40 | 99.23 | 100.10 | 99.91 | 99. 26 | 99. 53 | - |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 4.60 | 4. 30 | 1.85 | 1. 30 | 1.50 | 1.90 | 1. 55 | 1.25 | . 00 |
| A/CNK | 1.7 | 1.1 | 1.6 | 1.4 | 1.5 | 1.3 | 1.4 | 1. 3 | - |
| DI | 80.2 | 77.5 | 86. 3 | 90.4 | 90.2 | 87. 5 | 89.0 | 90.2 | - |
| Ba | 475. | 675. | 305. | 337. | 355. | 345. | 225. | 240. | 175. |
| Rb | 108. | 120. | 414. | 350. | 462. | 370. | 330. | 344. | 230. |
| Sr | 340. | 325. | 90. | 75. | 90. | 105. | 75. | 50. | 85. |
| Y |  | , |  |  | . | , | . | S | 8. |
| Zr | - | - | - | - | - | - | _ | $\rightarrow$ | - |
| Nb | - | - | - | - | - | - | - | - | - |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Ga | - | - | - | - | - | - | - | - | - |
| Zn | 92.0 | - | 52.0 | 32.0 | 45.0 | 56.0 | - | 45.0 | 55.0 |
| Cu | 9.0 | - | 7.0 | 3.0 | 4.0 | 3. 0 | - | 2. 0 | 2.0 |
| Ni | - | - | - | - | - |  | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| $V$ | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | $\cdots$ | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 65.0 | 60.0 | 71.0 | 65.0 | 100.0 | 128.0 | 85.0 | 56.0 | 40.0 |
| Be | - | - |  |  | - |  |  | - | - |
| 8 | - | - | - | - | - | - | - | - | - |
| $F$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | 10.0 | - | 13.0 | 12.0 | 12.0 | 17.0 | - | 13.0 | 10.0 |
| Mo | - | - | - | - | - | - | - | - | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical databose.

MOROCCO

|  |  | MAG70 ZAER MONG | MAG71 ZAER MONG | MAG72 ZAER MONG | $\begin{aligned} & \text { MAG73 } \\ & \text { ZAER } \\ & \text { MONG } \end{aligned}$ | MAG74 ZAER MONG |  | MAG76 ZAER GRAD | MAG77 ZAER MONG |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 71.70 | 6B. 70 | 73.20 | 71.60 | 75. 60 | 71.70 | 63. 60 | 75.20 | 70. 50 |
| $\mathrm{TiO}_{2}$ | . 15 | . 04 | . 10 | . 30 | . 10 | . 15 | . 80 | . 05 | . 15 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 15.00 | 16.00 | 14.90 | 15.50 | 13.70 | 15.20 | 17.90 | 15.10 | 15.50 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 1.23 | 2.48 | 1. 25 | 1. 50 | . 83 | 1. 65 | 1. 30 | . 75 | 1. 60 |
| FeO | . 15 | . 15 | - | - | . 15 | - | 3. 33 | - | - |
| MnO | . 01 | . 04 | - | . 02 | - | - | . 03 | . 03 | . 02 |
| MgO | . 35 | . 75 | . 27 | 37 | . 25 | . 31 | 2. 00 | . 12 | . 37 |
| CaO | . 70 | 1.70 | . 15 | . 53 | . 35 | .43 | 3. 15 | .40 | . 52 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 2. 50 | 3. 30 | 2. 50 | 2. 60 | 2. 60 | 2. 60 | 3.70 | 3. 60 | 2. 40 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4.90 | 3.80 | 4.85 | 5. 20 | 4.80 | 4.80 | 3.70 | 4.00 | 4.70 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1.71 | 1. 65 | 1.81 | 1. 53 | 1.31 | 1.88 | . 88 | 1.17 | 1. 91 |
| $\mathrm{H}_{2} \mathrm{O}-$ | . 71 | . 48 | . 45 | . 49 | . 25 | . 41 | . 17 | . 13 | . 38 |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - |  |
| F | - | - | - | - | - | - | - | - |  |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 99. 21 | 99.09 | 99. 48 | 99.64 | 99.94 | 98. 93 | 100.56 | 100. 55 | 99.05 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 1.40 | 2. 65 | 1. 25 | 1.50 | 1.00 | 1.65 | 5.00 | . 75 | 1. 60 |
| A/CNK | 1.4 | 1.3 | 1. 6 | 1.4 | 1.4 | 1. 5 | 1.1 | 1.4 | 1.7 |
| Dl | 88.9 | 82.4 | 91. 3 | 89.4 | 92. 8 | 88. 8 | 70.4 | 93. 2 | 87.0 |
| Ba | 174. | 430. | 232. | 298. | 134. | 100. | 775. | 100. | 265. |
| Rb | 422. | 368. | 486. | 314. | 300. | 340. | 110. | 380. | 286. |
| Sr | 90. | 170. | 55. | 75. | 95. | 50. | 520. | - | 65. |
| $Y$ | - | - | - | - | - | - | - | - | - |
| Zr | - | - | - | - | - | - | - | - | - |
| Nb | - | - | - | - | - | - | - | - | - |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Go | - | - | - | - | - | - | - | - | - |
| Zn | 16.0 | 52.0 | 14.0 | 57.0 | 27.0 | 68.0 | - | 19.0 | 61.0 |
| Cu | 9. 0 | 6. 0 | 2. 0 | 1.0 | 4.0 | 2. 0 | - | 1.0 | 4. 0 |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| V | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 58.0 | 118.0 | 54.0 | 58.0 | 142.0 | 52. 0 | 95.0 | 77.0 | 136.0 |
| Be | - | - | - | - | - | - | - | - | - |
| 8 | - | - | - | - | - | - | - | - | - |
| F | - | _ | - | - | - | - | $\rightarrow$ | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | 18.0 | 12.0 | 16.0 | 10.0 | 18.0 | 16.0 | - | 25.0 | 12.0 |
| Mo | - | - | - | - | - | - | - | - | - |
| La | - | $\cdots$ | - | _ | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure 0.1 (cont.). Geochemical databose.

## MOROCCO

|  | MAG78 ZAER GRAD | MAG79 ZAER MONG | MAG80 ZAER MONG | MAG81 ZAER GRAD | MAG82 ZAER <br> MONG | MAG83 ZAER MONG | MAG84 ZAER GRAD | MAG85 ZAER <br> TONA | MAG86 ZAER QUAD |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 72. 30 | 71.50 | 72.40 | 71.90 | 71.40 | 73.10 | 74.40 | 62.70 | 66.80 |
| $\mathrm{TiO}_{2}$ | . 10 | . 30 | . 10 | . 10 | . 05 | . 10 | . 05 | . 75 | . 65 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 15.50 | 15.60 | 15.50 | 15.50 | 15.80 | 15.15 | 14.70 | 16.70 | 15.90 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 73 | 1,80 | . 75 | . 73 | . 73 | . 10 | . 85 | 2. 28 | 2. 39 |
| Feo | . 15 | - | - | . 15 | . 29 | - | - | 2. 90 | 1.81 |
| MnO | - | - | - | - | . 01 | . 01 | - | . 07 | . 05 |
| MgO | . 23 | . 47 | . 19 | . 25 | . 25 | . 23 | . 05 | 2. 50 | 1.90 |
| CaO | . 82 | . 85 | . 92 | . 50 | . 65 | . 45 | - | 3.60 | 2. 80 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 2.90 | 2.70 | 3. 30 | 2. 75 | 3. 50 | 2. 75 | . 20 | 3. 90 | 3. 90 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4.90 | 5.00 | 4.75 | 5. 55 | 4. 75 | 5.70 | 4. 50 | 2. 80 | 2. 95 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ |  | 1.40 | - | - | - | - | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1.68 | . 46 | . 98 | 1. 45 | 1. 26 | 1.24 | 1. 13 | 1.34 | 1.19 |
| $\mathrm{H}_{2} \mathrm{O}-$ | . 37 | - | . 16 | . 25 | . 39 | . 27 | . 22 | . 16 | . 39 |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 99.68 | 100.08 | 99. 05 | 99.13 | 99. 08 | 99.10 | 96.10 | 99. 70 | 100.74 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | . 90 | 1.80 | . 75 | .90 | 1.05 | .10 | . 85 | 5.50 | 4.40 |
| A/CNK | 1.3 | 1.4 | 1.3 | 1.4 | 1.3 | 1.3 | 2.8 | 1.0 | 1.1 |
| DI | 89.9 | 88.8 | 89.8 | 90.8 | 90.3 | 92. 3 | 85.1 | 67.4 | 75.4 |
| Ba | 200. | 525. | 205. | 335. |  |  |  |  |  |
| Rb | 408. | 463. | 463. | 472. | 516. | 580. | 570. | 105. | 105. |
| Sr | 50. | 100. | 75. | 78. | 90. | 35. | 25. | 460. | 345. |
| Y | - | - | - | - | - | - | - | - | - |
| Zr | - | - | - | - | - | - | - | - | - |
| Nb | - | $\rightarrow$ | - | - | - | - | - | - | - |
| Th | - | - | - | - | $\rightarrow$ | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Ga | - | - | - | - | - | - | - | - | - |
| Zn | 12.0 | 73. 0 | 22.0 | 30. 0 | 40.0 | 40. 0 | 35.0 | - | - |
| Cu | 1.0 | 1.0 | 2. 0 | 1.0 | 12.0 | 11.0 | 6. 1 | - | - |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| $V$ | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 56.0 | 48.0 | 53, 0 | 42.0 | 140.0 | 78.0 | 195.0 | E0. 0 | 55.0 |
| Be | - | 18.0 | 5, | 12. | 18.0 | 78.0 | 195.0 | - | 5.0 |
| B | - | - | - | - | - | _ | - | _ | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | 12.0 | 13.0 | 13.0 | 16.0 | 23. 0 | 21.0 | 42.0 | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| Lo | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical databose.

## MOROCCO

|  | MAG87 <br> ZAER <br> TONA | MAG89 ZAER TONA | $\begin{aligned} & \text { MAGGO } \\ & \text { ZAER } \\ & \text { MONG } \end{aligned}$ | MAG91 ZAER MONG | $\begin{aligned} & \text { MAGG2 } \\ & \text { ZAER } \\ & \text { GRAD } \end{aligned}$ | MAG94 ZAER GRAD | MAG95 ZAER TONA | MAG96 ZAER MONG | MAG97 <br> ZAER <br> MONG |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 69. 20 | 65. 40 | 72. 50 | 75. 30 | - | 87. 90 | 65. 80 | - | 72. 60 |
| $\mathrm{TiO}_{2}$ | . 45 | . 60 | 05 | . 15 | - | 55 | . 60 | - | . 10 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 15.40 | 15.60 | 14.80 | 14.30 | - | 15.60 | 16.80 | - | 15.30 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 1.09 | 2. 25 | . 93 | . 95 | - | 2.10 | 1. 42 | - | 1.00 |
| FeO | 2.17 | 1. 89 | . 29 | - | - | 1.17 | 2. 32 | - | - |
| MnO | 05 | . 06 | . 02 | . 03 | - | . 05 | . 06 | - | . 03 |
| MgO | 1.50 | 2. 00 | . 30 | . 19 | - | 1.70 | 2.00 | - | . 21 |
| COO | 3.15 | 2. 80 | . 55 | . 60 | - | 2. 65 | 3.15 | - | 72 |
| $\mathrm{No}_{2} \mathrm{O}$ | 4.00 | 3. 50 | 2. 60 | 3. 30 | - | 3.70 | 3. 75 | - | 3.00 |
| $\mathrm{K}_{2} \mathrm{O}$ | 2.45 | 3.00 | 6. 50 | 4.50 | - | 3. 25 | 2. 90 | - | 5.00 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | - | - |  | - | - | - | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}+$ | . 86 | . 98 | 1.17 | . 88 | - | 1. 29 | . 85 | - | . 97 |
| $\mathrm{H}_{2} \mathrm{O}-$ | . 14 | . 29 | . 17 | . 06 | - | . 33 | . 10 | - | . 11 |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 100.57 | 99. 37 | 99. 68 | 100. 26 | - | 100. 29 | 99. 55 | - | 99.04 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3 t}$ | 3. 50 | 4. 35 | 1.25 | . 95 | . 00 | 3.40 | 4. 00 | . 00 | 1.00 |
| A/CNK | 1.0 | 1.1 | 1.2 | 1.3 | - | 1.1 | 1.1 | - | 1. 3 |
| Ol | 75.4 | 73.5 | 92.0 | 92.7 | - | 77.5 | 71.6 | - | 90.1 |
| Bo | 490. | 850. | 175. | 176. | 1000. | 542. | 735. | 855. | 100. |
| Rb | 100. | 112. | 430. | 532. | 150. | 149. | 100. | 255. | 275. |
| Sr | 395. | 270. | 95. | 35. | 225. | 315. | 270. | 175. | 55. |
| Y | - | - | - | - | - | - | - | - | - |
| Zr | - | - | - | - | - | - | - | - | - |
| Nb | - | - | - | - | - | - | - | - | - |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Ga | - | - | - | - | - | - | - | - | - |
| Zn | - | 60.0 | 54.0 | 41.0 | 100.0 | E2.0 | 60.0 | - | - |
| Cu | - | 11.0 | 9. 0 | 3. 0 | 12.0 | 8.0 | 2.0 | - | - |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| V | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 65. 0 | 65. 0 | 175.0 | 155.0 | 50. 0 | 63. 0 | 78.0 | 100.0 | 90.0 |
| Be | , | , | - | - | , | - | - | - | - |
| 8 | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | 13.0 | 25.0 | 21.0 | 10.0 | 13.0 | 13.0 | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| Lo | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.

## MOROCCO

|  | MAL1 <br> OULM <br> MONG | MALI 1 <br> OULM <br> MONG | MAL12 OULM MONG | MAL13 OULM MONG | MAL2 <br> OULM <br> MONG | MAL3 <br> OULM <br> MONG | MAL5 <br> OULM <br> MONG | MAL6 <br> OULM <br> MONG | MAL7 <br> OULM <br> MONG |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 71.80 | 73.20 | - | - | 72.60 | 73.00 | 72.40 | 73.00 | 74.00 |
| $\mathrm{TiO}_{2}$ | . 10 | . 10 | - | - | . 05 | 05 | . 15 | . 15 | . 15 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 16.10 | 15.20 | - | - | 15.80 | 15.70 | 15.60 | 15.20 | 14.50 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 95 | . 61 | - | - | . 58 | . 42 | . 53 | . 88 | . 72 |
| FeO | - | . 22 | - | - | . 29 | . 43 | . 51 | . 29 | . 43 |
| MnO | . 02 | . 08 | - | - | . 04 | . 06 | . 03 | . 04 | . 05 |
| MgO | . 20 | 08 | - | - | . 18 | . 13 | . 30 | . 25 | . 20 |
| CaO | . 95 | . 35 | - | - | . 70 | . 40 | . 60 | . 45 | . 50 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3. 70 | 4.80 | - | - | 3.70 | 3.90 | 3.20 | 3.10 | 4. 05 |
| $\mathrm{K}_{2} \mathrm{O}$ | 5.05 | 4. 05 | - | - | 4.45 | 4. 45 | 5. 65 | S. 65 | 4. 85 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1.05 | . 80 | - | - | 1.54 | 1.68 | 1.30 | 1. 41 | . 95 |
| $\mathrm{H}_{2} \mathrm{O}-$ | . 20 | .10 | - | - | .14 | . 08 | . 06 | . 25 | .10 |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 100.12 | 99. 59 | - | - | 100.07 | 100.30 | 100.33 | 100.67 | 100.50 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3 t}$ | . 95 | . 85 | . 00 | . 00 | .90 | . 90 | 1.10 | 1.20 | 1. 20 |
| A/CNK | 1.2 | 1.2 | - | - | 1. 3 | 1. 3 | 1.3 | 1. 3 | 1.1 |
| DI | 90.9 | 94.3 | - | - | 91.4 | 93. 0 | 92.0 | 93.2 | 94.4 |
| Bo | 185. | - | 249. | 7. | - | - | 240. | 135. | 120. |
| Rb | 260. | 510. | 369. | 723. | 250. | 535. | 335. | 270. | 410. |
| Sr | 100. | - | 76. | 19. | 90. | 35. | 75. | 70. | 40. |
| Y | - | - | 10. | 8. | - | - | - | - | - |
| Zr | - | - | 75. | 32. | - | - | - | - | - |
| Nb | - | - | 10. | 17. | - | - | - | - | - |
| Th | $-$ | - | 13.00 | 3. 00 | - | - | - | - | - |
| Pb | - | $\rightarrow$ | 34. | 19. | - | $\sim$ | - | - | - |
| Ga | - | - | 19. | 23. | - | _ | - | - | - |
| Zn | - | - | 67.0 | 68. 0 | - | - | - | - | - |
| Cu | - | - | 1.0 | - | - | - | - | - | - |
| Ni | - | - | 6.0 | 10.0 | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | . 16 | . 01 | - | - | - | - | - |
| $V$ | - | - | 5. | 1. | - | - | - | - | - |
| Cr | - | - | 12. | 10. | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| Ta | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 190.0 | 250.0 | - | - | 365.0 | 670.0 | 415.0 | 225.0 | 255.0 |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | $\rightarrow$ | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| Lo | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure 0.1 (cont.). Geochemical database.

MOROCCO

|  | MAL7A <br> OULM <br> MONG | MAL8A <br> OULM <br> MONG | MAM 106 ZAER GRAD | MAM109 ZAER MONG | MAM266 ZAER GRAD | $\begin{aligned} & \text { MAM267 } \\ & \text { ZAER } \\ & \text { MONG } \end{aligned}$ | MAM268 ZAER GRAD | MAM281 <br> ZAER <br> TONA | $\begin{aligned} & \text { MAM283 } \\ & \text { ZAER } \\ & \text { GRAD } \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 75. 20 | 75.40 | 73.80 | 73. 70 | 67.80 | 68.10 | 67.80 | 67. 30 | 67.30 |
| $\mathrm{TiO}_{2}$ | . 15 | . 10 | . 10 | . 10 | . 60 | . 50 | . 50 | . 50 | . 55 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 13.30 | 14. 50 | 15.50 | 14.50 | 15. 50 | 16.10 | 15.70 | 15.00 | 15.60 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 50 | . 62 | . 29 | . 78 | 1.42 | . 96 | 1. 02 | . 94 | 1. 22 |
| FeO | - | . 43 | . 37 | . 29 | 2.05 | 2. 20 | 2. 05 | 2. 35 | 2.05 |
| MnO | . 01 | . 01 | . 02 | . 02 | . 05 | . 05 | . 05 | . 06 | . 05 |
| MgO | . 07 | . 20 | . 10 | . 29 | 1.70 | 1. 60 | 1. 50 | 1.70 | 1.60 |
| CoO | . 30 | . 65 | . 14 | . 68 | 2. 40 | 2. 80 | 2. 45 | 2.60 | 2. 55 |
| $\mathrm{No}_{2} \mathrm{O}$ | 3. 20 | 3. 00 | 3. 70 | 3. 30 | 3.60 | 4. 20 | 3. 85 | 3. 90 | 3. 75 |
| $\mathrm{K}_{2} \mathrm{O}$ | 5.75 | 4. 85 | 3.80 | 4.30 | 3. 10 | 3. 20 | 3. 30 | 3.10 | 3. 35 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | - | 4. | 1.21 | 4. | 3. 10 | , | - | , | - |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1.04 | 1.00 | . 17 | 1.09 | 1.14 | . 39 | . 79 | .42 | . 93 |
| $\mathrm{H}_{2} \mathrm{O}-$ | .11 | .10 | - | . 13 | . 30 | . 10 | . 20 | . 17 | . 20 |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | _ | - | - | - | - | - | - | - |
| $F$ | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | $\cdots$ | - | - |
| TOTAL | 99.63 | 100.96 | 99. 20 | 99. 18 | 99.76 | 100. 20 | 99. 31 | 99. 04 | 99. 15 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | . 50 | 1.10 | . 70 | 1. 10 | 3. 70 | 3.40 | 3.30 | 3. 55 | 3. 50 |
| A/CNK | 1.1 | 1.3 | 1.5 | 1. 3 | 1.1 | 1.0 | 1.1 | 1.1 | 1.1 |
| DI | 96.0 | 92.7 | 91.4 | 90. 6 | 76.1 | 78.7 | 77.0 | 75. 1 | 76.3 |
| Ea | 115. | 200. | 100. | 100. | 535. | 480. | 515, | 495. | 585. |
| Rb | 390. | 290. | 715. | 390. | 120. | $115$ | 120. | 110. | 100. |
| Sr | 35. | 80. | 25. | 40. | 320. | 340. | 335. | 330. | 340. |
| Y | - | - | - | - | - | - | - | - | - |
| Zr | - | - | - | - | - | - | - | - | - |
| Nb | - | - | - | - | - | - | - | - | - |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - | _ |
| Ga | - | - | - | - | _ | - | - | - | - |
| Zn | - | - | 100.0 | 60.0 | 55.0 | 50.0 | 55.0 | 50.0 | 70.0 |
| Cu | - | - | - | - | 5. 0 | - | - | - | 10.0 |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | _ | - | - | - |
| $\checkmark$ | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | _ | - | _ | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | _ | - | - | - | - | - |
| To | - | - | - | _ | - | - | - | _ | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 255.0 | 250.0 | 420.0 | 225.0 | 80.0 | 70.0 | 95.0 | 70.0 | 65.0 |
| Be | - | - | , |  | - | 2.0 | 9. | \%. | 6. |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | $\rightarrow$ | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | 65.0 |
| Mo | - | - | - | - | - | _ | - | _ | S. |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical databose.

MOROCCO

|  | MAM289 <br> ZAER <br> GRAD | $\begin{aligned} & \text { MAM291 } \\ & \text { ZAER } \\ & \text { GRAN } \end{aligned}$ | MAM293 ZAER TONA | $\begin{aligned} & \text { MAM294 } \\ & \text { ZAER } \\ & \text { MONG } \end{aligned}$ | MAM295 ZAER GRAN | MAM3O3 ZAER MONG | MAM320 ZAER GRAN | MAM322 ZAER MONG | MAM324 ZAER GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 73. 30 | 73. 80 | 62.90 | 74.00 | 68. 40 | 72. 60 | - | 72.80 | - |
| $\mathrm{TiO}_{2}$ | . 15 | . 10 | . 80 | - | . 55 | . 05 | - | . 15 | - |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14. 60 | 14.80 | 17.10 | 14.20 | 15.80 | 15.10 | - | 15.00 | - |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 1.20 | . 93 | 1.49 | 1.10 | 1. 05 | . 33 | - | . 62 | - |
| FeO | . 80 | . 15 | 4. 56 | - | 2. 20 | . 15 | - | 43 | - |
| MnO | . 03 | . 04 | . 07 | . 04 | . 05 | - | - | - | - |
| MgO | . 40 | . 22 | 2. 80 | . 15 | 1. 60 | . 25 | - | . 24 | - . |
| CaO | 1.40 | . 50 | 4.00 | . 50 | 2. 50 | . 25 | - | 1.00 | - |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3.90 | 3.40 | 3. 40 | 3. 50 | 3. 80 | 3. 70 | - | 3. 30 | - |
| $\mathrm{K}_{2} \mathrm{O}$ | 4.00 | 4. 40 | 3.00 | 4.80 | 3. 10 | 5.20 | - | 4.90 | - |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1.00 | . 97 | . 71 | 1.12 | . 55 | 1.34 | - | . 77 | - |
| $\mathrm{H}_{2} \mathrm{O}-$ | . 20 | . 24 | - | . 16 | . 13 | . 12 | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| $F$ | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 100.98 | 99. 55 | 100. 63 | 99. 57 | 99.74 | 100.09 | - | 99. 21 | - |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 2.09 | 1.10 | 6. 55 | 1.10 | 3. 50 | . 50 | .00 | 1.10 | . 00 |
| A/CNK | 1.1 | 1.3 | 1.1 | 1.2 | 1.1 | 1.3 | - | 1. 2 | - |
| Dl | 89.3 | 91.7 | 63.4 | 93. 2 | 76.6 | 93.7 | - | 89.9 | - |
| Ba | 705. | 100. | 515. | 170. | 100. | 100. | 515. | 200. | 228. |
| Rb | 125. | 215. | 105. | 200. | 110. | 350. | 112. | 270. | 138. |
| Sr | 205. | 45. | 345. | 10. | 325. | 55. | 116. | 65. | 254. |
| $Y$ | - | - | - | - | - | - | 18. | - | 26. |
| Zr | - | - | - | - | - | - | 166. | - | 199. |
| Nb | - | - | - | - | - | - | 14. | - | 16. |
| Th | - | - | - | - | - | - | 2. 00 | _ | 9. 00 |
| Pb | - | - | - | - | - | - | 14. | - | 21. |
| Go | - | - | - | - | - | - | 22. | - | 23. |
| Zn | - | 40.0 | 70.0 | - | 50.0 | 30. 0 | 70.0 | 50.0 | 57.0 |
| Cu | - | - | 5. 0 | - | 5. 0 | - | 14.0 | - | 1.0 |
| Ni | - | - | - | - | - | - | 34.0 | - | 22.0 |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | . 95 | - | . 79 |
| $V$ | - | - | - | - | - | - | 101. | - | 81. |
| Cr | - | - | - | - | - | - | 61. | - | 54. |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | _ | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| Ta | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 55.0 | 35. 0 | BD. 0 | 35.0 | 80.0 | 60.0 | - | 110.0 | - |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| $F$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| Lo | - | - | - | - | _ | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.

MOROCCO

|  | MAM325 ZAER GRAD | MAM326 ZAER <br> GRAD | MAM329 ZAER TONA | MAM330 ZAER GRAD | MAM332 ZAER MONG | MAM336 ZAER GRAD | MAM337 <br> ZAER <br> GRAN | MAM352 <br> ZAER <br> GRAN | MAM353 <br> ZAER <br> GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 72. 80 | 73. 60 | 66.50 | 69.50 | 71.50 | 73. 30 | 73. 20 | - | - |
| $\mathrm{TiO}_{2}$ | . 20 | . 05 | . 70 | . 45 | . 30 | . 05 | . 05 | - | - |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 15.60 | 15.50 | 16. 50 | 16.20 | 15.50 | 14.30 | 15.80 | - | - |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 1.07 | . 75 | 1. 37 | 1. 04 | 1. 15 | . 95 | . 86 | - | - |
| FeO | . 43 | - | 2. 82 | 1. 59 | . 94 | . 22 | . 22 | - | - |
| MnO | . 01 | . 01 | . 07 | . 05 | . 04 | - | . 05 | - | - |
| MgO | . 25 | . 20 | 2. 10 | 1.10 | . 80 | . 18 | . 08 | - | - |
| CaO | . 95 | . 67 | 3.10 | 2. 00 | 1. 55 | . 37 | . 37 | - | - |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3.20 | 3.60 | 3. 70 | 3. 90 | 3. 70 | 2. 90 | 3.70 | - | - |
| $\mathrm{K}_{2} \mathrm{O}$ | 5.40 | 5.40 | 3.10 | 3. 50 | 4. 30 | 4.75 | 3.90 | - | - |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}+$ | . 54 | . 70 | . 41 | .47 | . 58 | 1.08 | 1. 25 | - | - |
| $\mathrm{H}_{2} \mathrm{O}-$ | . 13 | . 05 | . 12 | . 05 | . 12 | $\rightarrow$ | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| $F$ | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 100.58 | 100.53 | 100.49 | 99. 86 | 100.49 | 98.09 | 99.48 | - | - |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 1. 55 | . 75 | 4. 50 | 2. 80 | 2. 20 | 1.20 | 1.10 | . 00 | . 00 |
| A/CNK | 1. 2 | 1. 2 | 1.1 | 1.2 | 1.1 | 1.4 | 1.4 | - | - |
| DI | 90.7 | 93. 3 | 72.0 | 80.9 | 85.2 | 90.8 | 91.3 | - | - |
| Bo | 355. | 175. | 276. | 495. | 400. | 100. | 100. | 599. | 529. |
| Rb | 355. | 200. | 57. | 140. | 170. | 445. | 900. | 115. | 135. |
| Sr | 110. | 105. | 680. | 275. | 220. | 20. | 55. | 336. | 265. |
| Y | - | - | 20. | - | - | - | - | 25. | 23. |
| Zr | - | - | 101. | - | - | - | - | 154. | 139. |
| Nb | - | - | 7. | - | - | - | - | 12. | 13. |
| Th | - | - | 2. 00 | - | - | - | - | 9.00 | 11.00 |
| Pb | - | - | 6. | - | - | - | - | 20. | 22. |
| Ga | - | - | 17. | - | - | - | - | 22. | 20. |
| Zn | 65.0 | 15.0 | 85.0 | - | 35.0 | 115.0 | 90.0 | 17.0 | 40. 0 |
| Cu |  | S. | 36.0 | - | - | - | - | 4.0 | - |
| Ni | - | - | 94.0 | - | - | - | - | 18.0 | 12.0 |
| $\mathrm{TiO}_{2}$ | - | - | . 97 | - | - | - | - | . 63 | . 48 |
| $V$ | - | - | 154. | - | - | - | - | 61. | 42. |
| Cr | - | - | 265. | - | - | - | - | 61. | 31. |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| Ta | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 115.0 | 200.0 | - | 85.0 | 120.0 | 165.0 | 450.0 | - | - |
| Be | 1 | , | - |  | , | 165 | , | - | - |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| Lo | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical datobose.

MOROCCO

|  | MAM354 ZAER GRAN | MAM355 ZAER GRAN | MAM356 ZAER GRAN | MAM357 ZAER GRAN | MAM358 ZAER GRAN | MAM360 ZAER GRAN | MAM361 ZAER GRAN | MAM362 ZAER GRAN | MAM363 ZAER GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | - | - | - | - | - | - | - | - | - |
| FeO | - | - | - | - | - | - | - | - | - |
| MnO | - | - | - | - | - | - | - | - | - |
| MgO | - | - | - | - | - | - | - | - | - |
| CaO | - | - | - | - | - | - | - | - | - |
| $\mathrm{Na}_{2} \mathrm{O}$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{K}_{2} \mathrm{O}$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | - | - | - | - | - | - | - | - | - |
| $\mathrm{Fe}_{2} \mathrm{O} 3 \mathrm{t}$ | . 00 | . 00 | . 00 | . 00 | . 00 | . 00 | . 00 | . 00 | . 00 |
| A/CNK | - | - | - | - | - | - | - | - | - |
| DI | - | - | - | - | - | - | - | - | - |
| Bo | 151. | 183. | 176. | 600. | 799. | 32. | 61. | 41. | 425. |
| Rb | 384. | 361. | 359. | 103. | 90. | 865. | 543. | 650. | 249. |
| Sr | 48. | 49. | 50. | 391. | 451. | 22. | 11. | 14. | 113. |
| Y | 14. | 12. | 11. | 23. | 18. | 22. | 20. | 21. | 37. |
| Zr | 55. | 57. | 46. | 175. | 193. | 48. | 48. | 49. | 237. |
| Nb | 12. | 11. | 9. | 12. | 13. | 36. | 23. | 20. | 15. |
| Th | 8. 00 | 7.00 | 8. 00 | 7.00 | 4. 00 | 19.00 | 38.00 | 30.00 | 19.00 |
| Pb | 27. | 26. | 24. | 17. | 16. | 24. | 23. | 29. | 25. |
| Ga | 24. | 23. | 21. | 22. | 24. | 33. | 27. | 25. | 20. |
| Zn | 88.0 | 71.0 | 43.0 | 57. 0 | 57.0 | 37. 0 | 51.0 | 80.0 | 71.0 |
| Cu | 14.0 | 26.0 | - | 1. 0 | 11.0 | 13.0 | 112.0 | 31.0 | - |
| Ni | 4.0 | 6.0 | 5.0 | 24.0 | 24.0 | 16.0 | 11.0 | 13.0 | 13. 0 |
| $\mathrm{TiO}_{2}$ | . 12 | . 11 | . 08 | . 76 | . 71 | . 07 | . 07 | . 08 | . 77 |
| $\checkmark$ | 5. | 2. | 1. | 76. | 67. | 3. | 3. | 1. | 51. |
| Cr | 11. | 14. | 14. | 55. | 52. | 6. | 11. | 13. | 23. |
| Hf | . | 1. | , | S. | S2. | . | . | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Se | - | - | - | - | - | - | - | - | - |
| Ta | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | - | - | - | - | - | - | - | - | - |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | - | _ | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - |  |
| Sn | - | - | - | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| Lo | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.

MOROCCO

|  | MAM364 ZAER GRAN | MAM365 <br> ZAER <br> GRAN | MAM366 ZAER GRAN | MAM5 ZAER GRAN | MAM84 ZAER GRAN | MAM9 ZAER GRAN | $\begin{aligned} & \text { MBC13B } \\ & \text { ZAER } \\ & \text { GRAN } \end{aligned}$ | $\begin{aligned} & \text { MBC15 } \\ & \text { ZAER } \\ & \text { GRAN } \end{aligned}$ | $\begin{aligned} & \text { MBC } 18 A \\ & \text { ZAER } \\ & \text { GRAN } \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | - | - | - | - | - | - | 70.80 | 73.11 | 74.43 |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | . 38 | . 04 | . 13 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | - | - | - | - | - | - | 14.58 | 15.68 | 14.40 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | - | - | - | - | - | - | 2. 80 | 90 | 1. 11 |
| FeO | - | - | - | - | - | - | - | - | - |
| MnO | - | - | - | - | - | - | . 06 | . 05 | . 03 |
| MgO | - | _ | - | - | - | - | 1.08 | . 01 | . 18 |
| CaO | - | - | - | - | - | - | 1.87 | . 37 | . 63 |
| $\mathrm{Na}_{2} \mathrm{O}$ | - | - | - | - | _ | - | 3. 53 | 3. 88 | 3.19 |
| $\mathrm{K}_{2} \mathrm{O}$ | - | - | - | _ | - | - | 3.73 | 3.87 | 4.81 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | - | - | _ | - | - | - | . 13 | . 38 | . 19 |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | $-$ | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | . 57 | 1.41 | 93 |
| TOTAL | - | - | - | - | - | - | 99. 63 | 100.00 | 100.03 |
| $\mathrm{Fe}_{2} \mathrm{O} 3 \mathrm{t}$ | . 00 | .00 | . 00 | . 00 | .00 | .00 | 2. 80 | . 90 | 1. 11 |
| A/CNK | - | - | - | - | - | - | 1.1 | 1.4 | 1. 2 |
| DI | - | - | - | - | - | - | 82.7 | 91.7 | 91.8 |
| Ba | 373. | 288, | 837. | 536. | 625. | 899. | 648. | 11. | 181. |
| Rb | 258. | 251. | 217. | 173. | 215. | 173. | 130. | 752. | 403. |
| Sr | 85. | 57. | 87. | 42. | 90. | 92. | 261. | 21. | 45. |
| $Y$ | 49. | 50. | 51. | 31. | 36. | 34. | 23. | 19. | 16. |
| Zr | 204. | 221. | 242. | 149. | 155. | 200. | 131. | 34. | E6. |
| Nb | 14. | 14. | 17. | 20. | 20. | 20. | 14. | 24. | 15. |
| Th | 24. 80 | 27.00 | 22. 00 | 13.00 | 20. 00 | 14.00 | 13. 00 | B. 00 | 9. 00 |
| Pb | 30. | 37. | 34. | 15. | 24. | 24. | 22. | 17. | 21. |
| Ga | 21. | 20. | 21. | 33. | 36. | 31. | 20. | 29. | 21. |
| Zn | 39.0 | 87.0 | 61.0 | 40.0 | 87.0 | 106.0 | 33.0 | 91.0 | 73.0 |
| Cu | - | - | - | 53.0 | 20.0 | 51.0 | 2. 0 | - | 1.0 |
| Ni | 10.0 | 10.0 | 7.0 | 44.0 | 42.0 | 28.0 | 17.0 | 48.0 | 28.0 |
| $\mathrm{TiO}_{2}$ | . 41 | . 35 | . 34 | 1.01 | $1.12$ | $1.03$ | $.53$ | $.02$ | $.15$ |
| $V$ | 25. | 20. | 11. | 148. | 175. | 122. | 47. | 4. | 8. |
| Cr | 9. | 10. | 11. | 137. | 133. | 106. | 35. | 13. | 8. |
| Hf | - | - | - | - | 1 | 10. | , | . | . |
| Cs | - | - | - | - | - | - | - | - | $\rightarrow$ |
| Sc | - | - | - | - | _ | - | - | - | - |
| Ta | - | - | - | _ | - | - | - | - | - |
| Co | - | - | - | - | - | - | _ | - | - |
| Li | - | - | - | - | - | - | - | - | - |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | - | _ | - | - | - | - | - |
| $F$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | - . |
| Mo | - | - | - | - | - | - | - | _ | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure 0.1 (cont.). Geochemical database.

MOROCCO

|  | MBC20A ZAER GRAN | MBC4 ZAER GRAN | MRRR11A TBAM MONG | MBRR12A TBAM MONG | MBRR14 TBAM MONG | MERR16A TBAM MONG | MBRR18 TBAM MONG | MBRR22A TBAM MONG | MBRR7C TBAM MONG |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | - | 67. 83 | 70. 26 | 71.06 | 70.64 | 70.93 | 70.58 | 67.83 | 70. 57 |
| $\mathrm{TiO}_{2}$ | - | . 80 | . 50 | . 45 | . 48 | . 13 | . 49 | . 65 | . 52 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | - | 14.44 | 14.85 | 14.70 | 14. 69 | 14. 28 | 14.69 | 15.58 | 14. 54 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | - | 5. 33 | 3.73 | 3. 32 | 3.51 | 3. 46 | 3. 30 | 4.53 | 3.44 |
| FeO | - | - | - | . | - | - | - | - | - |
| MnO | - | . 10 | . 05 | . 05 | . 04 | . 06 | . 05 | . 07 | 05 |
| MgO | - | 1. 68 | . 85 | . 75 | . 69 | . 79 | . 75 | 1. 28 | 72 |
| COO | - | 1. 95 | 1. 40 | 1.49 | 1.53 | 1. 57 | 1. 36 | . 88 | 1.75 |
| $\mathrm{Na}_{2} \mathrm{O}$ | - | 3.65 | 2. 52 | 2. 65 | 2.72 | 2. 59 | 2.73 | 2. 51 | 2.81 |
| $\mathrm{K}_{2} \mathrm{O}$ | - | 1.77 | 4.72 | 4.97 | 5. 06 | 4.88 | 5. 11 | 4.92 | 4.92 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | - | . 14 | . 11 | . 12 | . 13 | . 12 | .11 | . 13 | . 12 |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | 2. 22 | . 97 | . 72 | . 55 | . 70 | . 69 | 1. 58 | 40 |
| TOTAL | - | 99.91 | 99. 98 | 100.28 | 100.04 | 99.81 | 100. 15 | 99.94 | 99.84 |
| $\mathrm{Fe}_{2} \mathrm{O} 3 \mathrm{t}$ | . 00 | 5.33 | 3.73 | 3. 32 | 3. 51 | 3. 46 | 3. 30 | 4.53 | 3.44 |
| A/CNK | - | 1.3 | 1.3 | 1.2 | 1. 2 | 1.2 | 1.2 | 1.4 | 1.1 |
| DI | - | 74.9 | 82. 8 | 84.4 | 84.4 | 83.7 | 85. 3 | 81.3 | 83.8 |
| Ba | 213. | 354. | 403. | 433. | 390. | 360. | 580. | 539. | 470. |
| Rb | 5. | 54. | 248. | 248. | 252. | 259. | 260. | 230. | 246. |
| Sr | 564. | 236. | 93. | 85. | 84. | 74. | 78. | 136. | 77. |
| Y | 29. | 30. | 46. | 43. | 15. | 13. | 44. | 45. | 45. |
| Zr | 271. | 208. | 190. | 177. | 173. | 168. | 218. | 229. | 222. |
| Nb | 37. | 15. | 12. | 12. | 12. | 11. | 12. | 15. | 12. |
| Th | 1.00 | 7.00 | 25.00 | 19.00 | 20. 00 | 22.00 | 27.00 | 19.00 | 23. 00 |
| Pb | 2. | 11. | 23. | 22. | 21. | 21. | 21. | 19. | 20. |
| Ga | 16. | 17. | 21. | 18. | 18. | 19. | 18. | 21. | 20. |
| Zn | 50.0 | 75.0 | 54.0 | 43.0 | 40.0 | 38.0 | 3 S .0 | 70.0 | 40.0 |
| Cu | 47.0 | 10.0 | 7. 0 | 3. 0 | 9. 0 | 7. 0 | 6.0 | 5.0 | 4. 0 |
| Ni | 46.0 | 23.0 | 32. ${ }^{\text {d }}$ | 32. 0 | 31.0 | 28.0 | 29.0 | 29.0 | 32.0 |
| $\mathrm{TiO}_{2}$ | 2.58 | . 88 | . 55 | . 49 | . 44 | . 45 | . 53 | . 86 | . 55 |
| $V$ | 209. | 117. | 69. | 54. | 46. | 50. | 53. | 117. | 66. |
| Cr | 141. | 91. | 32. | 29. | 25. | 27. | 27. | 50. | 28. |
| Hf | - | - | . | . | S. | , | , | . | 8. |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | - | - | - | - | - | - | - | - | - |
| Be | - | - | - | - | - | - | - | - | - |
| 8 | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| w | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.

MOROCCO

|  |  | MHUT OUOU GRAD | MHU2 QUOU GRAD | MHU3 OUOU GRAD | MHU4 OUOU GRAD | MHU5 OUOU MONG | MJBL10 <br> OUOU <br> GRAD | MJBL12 <br> OUOU <br> GRAD | MJBL 13 OUOU MONG |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 71.94 | 52.80 | 64.29 | 64.96 | 66. 60 | 72. 56 | 57.16 | 66.95 | 68. 48 |
| $\mathrm{TiO}_{2}$ | . 45 | . 50 | . 88 | . 96 | . 53 | . 35 | . 73 | . 65 | . 59 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14.37 | 16.55 | 14.57 | 16. 12 | 14.79 | 13. 45 | 15.83 | 16.13 | 15. 56 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 3.25 | - | 1.17 | . 46 | . 83 | . 92 | 4. 93 | 4. 31 | 4. 08 |
| FeO | - | 4. 32 | 4.88 | 4. 70 | 3. 73 | 1. 29 | - | - | - |
| MnO | .05 | . 06 | . 09 | . 10 | . 07 | . 03 | . 08 | . 07 | . 05 |
| MgO | . 69 | 2. 25 | 2. 28 | 2. 34 | 1. 65 | . 55 | 1.54 | 1. 42 | 1. 13 |
| CaO | 1.43 | 3. 37 | 2. 50 | 3. 20 | 2. 15 | . 84 | 1.88 | 2. 38 | 1.57 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 2. 66 | 2. 58 | 2. 70 | 1.80 | 3. 08 | 2. 53 | 2.84 | 3.18 | 2.74 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4.76 | 5. 23 | 4.50 | 3.81 | 3.93 | 6. 06 | 4. 08 | 3. 95 | 4.60 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 12 | . 30 | . 46 | . 22 | . 21 | .10 | .17 | .16 | . 15 |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | 1. 16 | 1. 06 | . 86 | 1. 59 | 1. 16 | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | . 11 | . 51 | . 33 | . 14 | . 09 | - | - | - |
| $\mathrm{CO}_{2}$ | - | . 55 | .15 | . 06 | . 57 | . 60 | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | . 82 | - | - | - | - | - | . 78 | . 78 | 1.10 |
| TOTAL | 100.34 | 99. 78 | 100. 04 | 100.02 | 99.97 | 100.53 | 100. 02 | 99. 98 | 100.05 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 3.25 | 4.80 | 6.59 | 5.68 | 4. 97 | 2. 35 | 4.93 | 4. 31 | 4. 08 |
| A/CNK | 1. 2 | 1.0 | 1.1 | 1.3 | 1.1 | 1.1 | 1.3 | 1. 2 | 1.3 |
| DI | 85.1 | 68.8 | 71.5 | 65.7 | 75.8 | 90.7 | 77.3 | 78.8 | 80.7 |
| Bo | 368. | - | - | - | - | - | 638. | 529. | 574. |
| Rb | 249. | - | - | - | - | - | 161. | 156. | 180. |
| Sr | 75. | - | - | - | - | - | 225. | 326. | 173. |
| Y | 42. | - | - | - | - | - | 32. | 35. | 34. |
| Zr | 166. | - | - | - | - | - | 216. | 227. | 212. |
| Nb | 12. | - | - | $\cdots$ | - | - | 15. | 14. | 15. |
| Th | 21.00 | - | - | - | - | - | 15.00 | 18.00 | 18.00 |
| Pb | 18. | - | - | - | - | - | 28. | 27. | 24. |
| Ga | 19. | - | - | - | - | - | 21. | 19. | 19. |
| Zn | 40.0 | - | - | - | - | - | 68. 0 | 62.0 | 48.0 |
| Cu | 6.0 | - | - | - | - | - | 17.0 | 17.0 | 8. 0 |
| Ni | 31.0 | - | - | - | - | - | 30. 0 | 26.0 | 28.0 |
| $\mathrm{TiO}_{2}$ | . 49 | - | - | - | - | - | . 79 | . 65 | . 65 |
| $\checkmark$ | 52. | - | - | - | - | - | 77. | 69. | 62. |
| Cr | 31. | - | - | - | - | - | 51. | 45. | 42. |
| Hf | - | - | - | - | - | - |  | - | - |
| Cs | - | - | - | - | - | - | _ | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| Ta | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | - | - | - | - | - | - | - | - | - |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | _ | - | _ | - | - |
| U | - | - | - | - | - | _ | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.), Geochemical database.

MOROCCO

|  | MJEL1B <br> OUOU <br> APLI | MJBL21 <br> OUOU <br> GRAD | MJBL23 QUOU GRAD | MJBL24 <br> OUOU <br> GRAD | MJBL26 OUOU MONG | MJBL2A <br> OUOU <br> GRAD | MJBL 3 OUOU MONG | MJBL6 OUOU GRAD | MJBL8 OUOU GRAD |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 75. 65 | 66. 21 | 68.12 | 66. 20 | 69. 46 | 66. 41 | 68. 98 | 65.18 | - |
| $\mathrm{TiO}_{2}$ | . 05 | . 74 | . 59 | . 64 | . 51 | 71 | . 56 | . 76 | - |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 13.65 | 16.18 | 15. 67 | 16.10 | 15. 18 | 16. 18 | 15.52 | 16.37 | - |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 42 | 5.26 | 4. 10 | 4. 59 | 3. 65 | 4. 92 | 3. 98 | 5. 32 | - |
| FeO | - | - | - | - | - | - | - | - | - |
| MnO | . 02 | . 08 | . 06 | . 07 | . 05 | . 08 | . 05 | . 08 | - |
| MgO | . 06 | 1. 45 | 1. 34 | 1.71 | . 97 | 1.47 | 99 | 1. 55 | - |
| CoO | . 37 | 2. 46 | 2.16 | 2. 17 | 1.70 | 2. 30 | 1. 48 | 1.99 | - |
| $\mathrm{Na}_{2} \mathrm{O}$ | 2. 99 | 2. 94 | 3.14 | 3.22 | 2. 93 | 2.8B | 2. 53 | 2. 69 | - |
| $\mathrm{K}_{2} \mathrm{O}$ | 5.73 | 3. 78 | 4.09 | 4.10 | 4. 54 | 3. 90 | 4. 71 | 4.14 | - |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 12 | . 18 | . 17 | . 14 | . 14 | . 18 | . 20 | . 18 | - |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | $\cdots$ | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | . 62 | . 81 | . 84 | 1.19 | 1.08 | . 86 | 1.11 | 1.07 | - |
| TOTAL | 99.71 | 100.09 | 100.28 | 100.13 | 100.12 | 99. 79 | 100.11 | 100.33 | - |
| $\mathrm{Fe}_{2} \mathrm{O} 3 \mathrm{t}$ | . 42 | 5. 26 | 4.10 | 4. 59 | 3. 66 | 4.92 | 3. 98 | 5.32 | .00 |
| A/CNK | 1. 2 | 1.2 | 1.2 | 1.2 | 1.2 | 1.2 | 1.3 | 1.3 | - |
| DI | 95.0 | 74.9 | 78. 8 | 76. 4 | 82.0 | 75.5 | 81.4 | 75.8 | - |
| Ba | 217. | 537. | 622. | 553. | 508. | 611. | 640. | 589. | 565. |
| Rb | 286. | 153. | 158. | 151. | 180. | 160. | 183. | 188. | 149. |
| Sr | 94. | 231. | 321. | 277. | 157. | 248. | 158. | 217. | 223. |
| Y | 17. | 35. | 30. | 29. | 35. | 37. | 35. | 33. | 35. |
| Zr | 40. | 229. | 179. | 190. | 186. | 230. | 209. | 231. | 238. |
| Nb | 6. | 15. | 14. | 14. | 15. | 15. | 16. | 16. | 17. |
| Th | 3. 00 | 14.00 | 15.00 | 11.00 | 20.00 | 13.00 | 13.00 | 17.00 | 13.00 |
| Pb | 49. | 23. | 22. | 18. | 24. | 25. | 72. | 26. | 25. |
| Ga | 15. | 21. | 18. | 21. | 18. | 19. | 21. | 18. | 20. |
| Zn | 15.0 | 79.0 | 56.0 | 60.0 | 38.0 | 63.0 | 157.0 | 74.0 | 68.0 |
| Cu | - | 20.0 | 18.0 | 12.0 | 7.0 | 9. 0 | 14.0 | 18.0 | 23.0 |
| Ni | 20.0 | 29.0 | 27.0 | 26.0 | 21.0 | 24.0 | 29.0 | 27.0 | 27.0 |
| $\mathrm{TiO}_{2}$ | . 05 | . 79 | . 59 | . 81 | . 58 | . 78 | . 55 | . 88 | . 76 |
| $\checkmark$ | 1. | 85. | 60. | 83. | 55. | 82. | 62. | 91. | 83. |
| Cr | 10. | 50. | 40. | 60. | 35. | 53. | 34. | 59. | 53. |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| Ta | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | - | - | - | - | - | - | - | - | - |
| Be | - | - | - - | - | - | - | - | - | - |
| 8 | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | _ |
| Sn | - | - | _ | _ | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical databose.

## MOROCCO

|  | MuUg18 OUOU MONG | MJUB19 OUOU MONG | MJUB21 OUOU MONG | MJUB24 OUOU MONG | MJUB4 OUOU MONG | MJUE6 OUOU GRAD | MJUB9 OUOU GRAD | MME1 MENT MONG | MME10 MENT MONG |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 67.03 | 67.12 | 67.69 | 66.57 | 66. 18 | 68.57 | 57.84 | 71.80 | 75.60 |
| $\mathrm{TiO}_{2}$ | . 63 | . 67 | . 52 | . 60 | . 71 | . 48 | . 64 | . 40 | - |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 15.74 | 15.80 | 15.58 | 15. 80 | 16.11 | 15.75 | 15.63 | 13.80 | 13.70 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 4. 36 | 4. 82 | 3. 79 | 4.14 | 4.97 | 3. 21 | 4. 37 | 1.89 | 90 |
| FeO | - | - | - | - | - | - | - | 1.09 | 36 |
| MnO | . 07 | . 10 | . 07 | . 06 | . 08 | . 05 | . 07 | . 05 | . 03 |
| MgO | 1. 44 | 1.30 | 1. 12 | 1. 45 | 1. 46 | 1.01 | 1. 41 | . 55 | . 07 |
| CoO | 2. 36 | 2.09 | 1.96 | 2. 25 | 2. 14 | 2. 02 | 1. 75 | . 90 | . 25 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3.09 | 3.06 | 3.23 | 3. 31 | 2. 97 | 3.86 | 3. 08 | 3.10 | 2.90 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4.05 | 3.99 | 4.19 | 4.01 | 3. 86 | 3. 65 | 4.04 | 5. 45 | 5.05 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 16 | . 19 | . 19 | . 16 | . 18 | .13 | . 15 | - | - |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | - | - | . | - | - | - | . 85 | 60 |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | . 05 | . 02 |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| $F$ | - | - | - | - | - | - | - | - | - |
| LOI | .94 | . 83 | 1. 36 | .99 | . 99 | 1.13 | 1.05 | - | - |
| TOTAL | 99.87 | 99. 97 | 99.70 | 99. 35 | 99. 95 | 99.86 | 100.04 | 99. 93 | 100.48 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 4. 36 | 4.82 | 3.79 | 4.14 | 4.97 | 3.21 | 4. 37 | 3.10 | 1.30 |
| A/CNK | 1.1 | 1.2 | 1.2 | 1.1 | 1.2 | 1.1 | 1. 2 | 1.1 | 1. 3 |
| DI | 76.9 | 77.6 | 79. 6 | 77.1 | 75. 2 | 80.9 | 79.0 | 89.4 | 94.7 |
| Bo | 676. | 587. | 545. | 694. | 618. | 778. | 601. | 580. | - |
| Rb | 158. | 186. | 200. | 153. | 153. | 130. | 170. | 230. | 760. |
| Sr | 331. | 215. | 219. | 377. | 232. | 631. | 255. | 80. | - |
| Y | 30. | 35. | 28. | 28. | 36. | 24. | 33. | - | - |
| Zr | 201. | 207. | 165. | 195. | 219. | 160. | 204. | - | - |
| Nb | 13. | 15. | 14. | 13. | 15. | 10. | 14. | - | - |
| Th | 11.00 | 16. 00 | 15.00 | 8. 00 | 15.00 | 2.00 | 13.00 | - | - |
| Pb | 26. | 29. | 37. | 20. | 23. | 20. | 22. | - | - |
| Ga | 22. | 22. | 22. | 23. | 16. | 20. | 19. | - | - |
| Zn | 66.0 | 126.0 | 64.0 | 57. D | 66.0 | 50.0 | 62.0 | - | - |
| Cu | 14.0 | 17.0 | 23. 0 | 10.0 | 16.0 | 12.0 | 11.0 | - | - |
| Ni | 22.0 | 26.0 | 23.0 | 23.0 | 30.0 | 17.0 | 28.0 | - | - |
| $\mathrm{TiO}_{2}$ | . 64 | . 71 | . 54 | . 63 | . 70 | . 53 | . 73 | - | - |
| $\checkmark$ | 73. | 77. | 62. | 73. | 78. | 54. | 88. | - | - |
| Cr | 45. | 46. | 37. | 49. | 53. | 38. | 52. | - | - |
| Hf | - |  |  | - | - |  |  | - | - |
| Cs | - | - | - | - | _ | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | _ | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | - | - | - | - | - | - | - | 80. 0 | 250.0 |
| Be | - | - | - | - | - | - | - | 8. | 250.0 |
| B | - | - | - | - | - | _ | - | _ | - |
| F | - | - | - | - | _ | _ | - | - | _ |
| Cl | - | - | - | - | - | _ | - | - | - |
| U | - | - | - | - | _ | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical dotabase.

MOROCCO

|  | MME10A MENT MONG | MME13 MENT MONG | MME14 MENT SYEG | MME 15 MENT MONG | MME 16 MENT SYEG | MME17 MENT MONG | MME 18 MENT MONG | MME2 MENT SYEG | MME4 MENT MONG |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 75. 60 | 75.30 | 71.90 | 72. 20 | 75. 30 | 70.20 | 73.70 | 74.30 | 76.10 |
| $\mathrm{TiO}_{2}$ | . 25 | - | . 45 | . 40 | . 15 | . 45 | . 35 | 35 | . 20 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 13.80 | 13.60 | 14.00 | 13.50 | 13. 50 | 15.30 | 13.90 | 13. 50 | 13. 50 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 1.20 | . 70 | . 94 | 1. 01 | . 95 | 1.11 | . 93 | 1.03 | . 51 |
| FeO | . 36 | - | 1. 95 | 1. 16 | - | 1.16 | 1. 23 | . 87 | . 58 |
| MnO | . 03 | - | . 04 | . 04 | - | . 04 | . 03 | . 03 | . 02 |
| MgO | . 10 | . 05 | . 40 | . 22 | . 07 | . 10 | . 20 | . 20 | . 08 |
| CoO | . 15 | - | . 65 | . 75 | . 30 | . 85 | . 75 | . 55 | . 40 |
| $\mathrm{No}_{2} \mathrm{O}$ | 2.70 | 2. 90 | 2. 40 | 3.10 | 3.40 | 3.15 | 3. 15 | 3.10 | 3. 20 |
| $\mathrm{K}_{2} \mathrm{O}$ | 5.45 | 5.55 | 5. 65 | 5. 55 | 5. 65 | 5. 85 | 5. 60 | 5.45 | 5.05 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | - | - | - | - | - | - | - | , | - |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1.00 | 1.06 | 1. 50 | 1.10 | . 90 | . 95 | . 70 | .70 | . 70 |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | . 14 | . 24 | - | . 11 | . 06 | . 05 | . 03 | . 05 |
| $\mathrm{CO}_{2}$ | - | - | - | _ | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 100.64 | 100.30 | 100.12 | 99.13 | 100.33 | 99.53 | 100.59 | 100.11 | 100.49 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 1.80 | .70 | 3.10 | 2. 30 | . 95 | 2.40 | 2. 30 | 2.00 | 1. 25 |
| A/CNK | 1,3 | 1.3 | 1.2 | 1.1 | 1.1 | 1.2 | 1.1 | 1.1 | 1.2 |
| DI | 94. 5 | 96.6 | 88. 6 | 90.7 | 96. 3 | 89.0 | 92.0 | 92.9 | 94.7 |
| Ba | 100. | - | 560. | 320. | - | 500. | 430. | 770. | 1540. |
| Rb | 590. | 345. | 230. | 250. | 355. | 260. | 235. | 260. | 305. |
| Sr | 445. | - | 60. | 40. | 185. | 75. | 45. | 45. | 40. |
| $Y$ | - | - | - | - | - | - | - | - | - |
| Zr | - | - | - - | - | - | - | - | - | - |
| N6 | - | - | - | - | - | - | - | - | - |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Ga | - | - | - | _ | - | - | _ | - | _ |
| Zn | - | - | - | - | - | - | - | - | - |
| Cu | - | - | - | - | - | - | - | - | - |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | _ | - |
| $\checkmark$ | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | _ | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| Ta | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 150.0 | 50.0 | 95. 0 | 90. 0 | 50.0 | 150.0 | 80.0 | 115.0 | 130.0 |
| Be | - | - | - | - |  | 15 | - | 1 | - |
| B | - | - | - | - | - | - | _ | - | - |
| F | - | - | - | - | - | - | - | _ | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | _ | - | - | - |
| Sn | - | - | - | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| Lo | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure 0.1 (cont.). Geochemical database.
MOROCCO

|  | MME5 MENT MONG | MME6 MENT MONG | MMG1 MENT GRAN | MMG2 MENT GRAN | MMM15 MENT GRAN | MMM31 MENT GRAN | MMN66 MENT GRAN | MMN72 <br> NENT <br> GRAN | MMQ19 MENT GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 71.90 | 70.20 | 71.17 | 66.73 | 77.29 | 75. 63 | 73. 98 | 74.26 | 76.89 |
| $\mathrm{TiO}_{2}$ | . 25 | . 55 | . 46 | . 78 | . 03 | . 08 | . 04 | . 04 | - |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14.70 | 15.00 | 14.60 | 17.22 | 12.75 | 12.49 | 14. 69 | 14.71 | 12.98 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 93 | 1.12 | 2. 85 | 2.08 | . 76 | . 53 | 1. 25 | . 92 | . 49 |
| FeO | 1. 23 | 2. 32 | - | - | - | - | - | - | - |
| MnO | . 04 | . 07 | 1.07 | . 04 | .03 | . 02 | . 02 | . 01 | . 02 |
| MgO | . 40 | . 75 | 1. 07 | . 72 | - | . 03 | . 09 | . 07 | . 03 |
| COO | . 90 | 1.75 | 1.08 | 1.95 | - | . 09 | . 14 | . 18 | - |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3.20 | 3.50 | 2.95 | 2.76 | 2. 36 | 3.10 | 3. 15 | 3. 69 | 2. 97 |
| $\mathrm{K}_{2} \mathrm{O}$ | 5.85 | 4. 45 | 4.88 | 6.72 | 5.13 | 5.17 | 4.98 | 4. 53 | 5. 68 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | - | - | - | G. |  | - | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}+$ | . 59 | . 55 | . 50 | . 61 | - | - | 1.32 | 1.05 | - |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | 1. 51 | . 73 | - | - | . 79 |
| TOTAL | 99. 99 | 100.25 | 100.45 | 99.61 | 99.86 | 98.87 | 99.64 | 99.44 | 99. 85 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 2. 30 | 3.70 | 2. 86 | 2. 08 | . 76 | . 53 | 1. 25 | . 92 | 49 |
| A/CNK | 1.1 | 1.1 | 1.2 | 1.1 | 1.4 | 1.2 | 1.4 | 1. 3 | 1. 2 |
| DI | 90.0 | 83.1 | 85.3 | 83. 3 | 94. 2 | 95.4 | 93.4 | 94.0 | 96. 5 |
| Ba | 425. | 660. | - | - | 26. | 70. | - | - | 45. |
| Rb | 240. | 245. | - | - | 275. | 233. | - | - | 346. |
| Sr | 80. | 115. | - | - | 19. | 37. | - | - | 15. |
| $Y$ | - | - | - | - | 19. | . | - | - | 15. |
| Zr | - | - | - | - | - | - | - | - | - |
| Nb | - | - | - | - | $\cdots$ | - | - | - | - |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Ga | - | - | - | - | - | - | - | - | - |
| Zn | - | - | - | - | - | - | - | - | - |
| Cu | - | - | - | _ | _ | - | - | - | - |
| Ni | - | - | - | - | - | - | _ | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | _ | - | - | - | - | - |
| $\checkmark$ | - | - | - | - | 14. | 24. | - | - | 10. |
| Cr | - | - | - | - | , | - | _ | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | _ | _ |
| Co | - | - | - | _ | - | - | - | - | - |
| Li | 75.0 | 100.0 | - | - | 45.0 | 23.0 | - | - | - |
| Be | - | - | - | - | - | - | $\cdots$ | - | - |
| B | - | - | - | - | 20.0 | 20. 0 | - | - | - |
| F | - | - | - | - | 150. | 40. | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | 43.0 | 3.7 | - | - | - |
| Sn | - | - | - | - | - | 20.0 | - | _ | - |
| Mo | - | - | - | - | . 30 | . 20 | - | - | - |
| La | - | - | - | - | - |  | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.

## MOROCCO

|  | MMQ2O MENT GRAN | MMT56 MENT <br> GRAN | MMT81 MENT GRAN | MMT82 MENT GRAN |  | MNZ1OA MENT GRAN | MMZ11 <br> MENT <br> GRAN | MMZ12 <br> MENT <br> GRAN | MMZ14 MENT GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 77.61 | 75.38 | 59.71 | 74. 35 | 74.27 | 73. 35 | 75.96 | 76. 64 | 75.83 |
| $\mathrm{TiO}_{2}$ | - | . 08 | . 62 | . 34 | . 05 | . 10 | . 05 | . 07 | . 07 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 12.88 | 12.99 | 14.30 | 12.85 | 14.15 | 14.31 | 13.20 | 13. 09 | 13.62 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 44 | 1. 68 | 4. 05 | 2. 28 | 1.09 | 1. 21 | 1.07 | 1. 50 | 1. 26 |
| FeO | - | - | - | - | - | - | - | - | - |
| MnO | . 02 | . 05 | . 05 | . 05 | . 04 | . 08 | . 03 | . 04 | . 03 |
| MgO | . 03 | . 03 | . 85 | . 31 | . 03 | - | . 03 | . 03 | . 03 |
| COO | . 06 | . 24 | 1.71 | .70 | . 26 | . 01 | . 23 | . 24 | . 23 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3.04 | 3.16 | 3. 18 | 3.07 | 2. 91 | 2. 90 | 3.15 | 1.84 | 2.78 |
| $\mathrm{K}_{2} \mathrm{O}$ | 5.10 | 4.84 | 4.74 | 4.87 | 5.61 | 5.75 | 4.82 | 4.45 | 4. 38 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | - | . 04 | . 13 | . 04 | - | , | 1.82 |  | d |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | - | - | - | - | _ | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| $F$ | - | - | - | - | - | - | - | - | - |
| LOI | . 84 | 1.02 | . 87 | . 87 | 1.14 | 1.06 | 1.13 | 1.78 | 1. 35 |
| TOTAL | 100.02 | 99. 51 | 100.25 | 99.73 | 99. 56 | 98.77 | 99.67 | 99.69 | 99. 58 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | . 44 | 1. 68 | 4.05 | 2. 28 | 1.09 | 1.21 | 1.07 | 1. 50 | 1. 25 |
| A/CNK | 1.2 | 1. 2 | 1.1 | 1.1 | 1.3 | 1. 3 | 1.2 | 1. 5 | 1.4 |
| DI | 96.1 | 93.4 | 83. 5 | 90.8 | 93.0 | 93.0 | 93.8 | 90.2 | 91.8 |
| 80 | 13. | 69. | 471. | 202. | - | 61. | - | - | - |
| Rb | 387. | 354. | 266. | 316. | - | 1115. | - | - | - |
| Sr | 11. | 25. | 151. | 66. | - | 25. | - | - | - |
| Y | 11. | - | - | - | - | - | - | - | - |
| Zr | - | - | - | - | - | - | - | - | - |
| Nb | _ | _ | - | - | - | - | - | - | - |
| Th | - | - | - | - | - | $\rightarrow$ | - | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Ga | - | - | - | - | - | - | - | - | - |
| Zn | - | - | - | - | - | - | - | - | - |
| Cu | - | - | - | - | _ | - | - | _ | - |
| Ni | - | - | - | - | _ | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | _ |
| $\checkmark$ | 10. | 32. | 97. | 55. | - | 12. | - | - | - |
| Cr | - | - | , | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | _ | - |
| Sc | - | - | - | - | _ | - | - | _ | _ |
| Ta | - | - | - | - | - | - | - | - | _ |
| Co | - | - | - | - | - | - | - | - | - |
| Li | - | - | - | - | 678.0 | 478.0 | - | - | - |
| Be | - | - | - | - | - | - | _ | - | - |
| B | - | - | - | - | 20.0 | 20.0 | - | - | - |
| $F$ | - | - | - | - | 650. | 650. | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | 26. 5 | 26.5 | - | _ | - |
| Sn | - | 2.0 | 7. 0 | 11.0 | 6. 0 | - | - | - | - |
| Mo | - | - | - | - | . 10 | .10 | - | - | - |
| Lo | - | - | - | - | - | , | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.

## MOROCCO

|  | MMZ16 MENT GRAN | MMZ17 MENT GRAN | MMZ18 MENT GRAN | MMZ18A MENT GRAN | MMZ19 MENT GRAN | MMZ19A MENT GRAN |  | MMZ21A MENT GRAN | MMZ26 <br> MENT <br> GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 74.12 | 75.44 | 73. 79 | 74.17 | 75. 51 | 75. 56 | 75.14 | 74.96 | 74. 53 |
| $\mathrm{TiO}_{2}$ | . 05 | . 05 | . 17 | . 17 | . 16 | . 07 | . 08 | . 06 | . 03 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14.09 | 13.88 | 14.11 | 13.41 | 13. 12 | 13.17 | 13.64 | 13.85 | 14.31 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 97 | . 70 | 1.89 | 1.70 | 1. 20 | 1. 28 | 1. 36 | 1.31 | 1.33 |
| FeO | - | - | - | - | - | - | - | - | - |
| MnO | . 01 | . 02 | . 03 | . 06 | . 04 | . 02 | . 05 | . 03 | . 03 |
| MgO | . 08 | . 03 | . 03 | . 13 | - | . 04 | - | . 03 | . 03 |
| CaO | .07 | . 11 | . 60 | . 37 | . 04 | . 29 | - | . 19 | . 08 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 1.01 | 2. 97 | 2. 96 | 3.01 | 2. 87 | 2.77 | 3.11 | 3. 14 | 3.15 |
| $\mathrm{K}_{2} \mathrm{O}$ | 7.32 | 5. 26 | 4. 96 | 4.88 | 5. 12 | 5.15 | 4.73 | 4.85 | 4. 68 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | - | - | - | . 06 | .07 | - | . 15 | - | - |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | _ | - | - | - | - | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| $F$ | - | - | - | - | - | - | - | - | - |
| LOI | 1.50 | 1. 22 | 1.18 | 1.89 | 1.15 | 1.13 | 1.18 | 1. 27 | 1.73 |
| TOTAL | 99.23 | 99. 68 | 99.72 | 99. 05 | 99. 58 | 99. 48 | 99.45 | 99. 69 | 99.90 |
| $\mathrm{Fe}_{2} \mathrm{O} 3 \mathrm{t}$ | . 97 | .70 | 1.89 | 1.70 | 1.20 | 1.2B | 1.36 | 1. 31 | 1.33 |
| A/CNK | 1.5 | 1.3 | 1.3 | 1. 2 | 1. 3 | 1.2 | 1.3 | 1.3 | 1.4 |
| DI | 91.8 | 94.0 | 90. 6 | 91. 5 | 93.8 | 92.9 | 93.2 | 92.9 | 92.4 |
| Ba | - | - | 137. | - | 44. | - | 27. | - | - |
| Rb | - | - | 557. | - | 667. | - | 774. | - | - |
| Sr | - | - | 40. | - | 10. | - | 33. | - | - |
| Y | - | - | - | - | - | - | - | - | - |
| Zr | - | - | - | - | - | - | $=$ | - | - |
| Nb | - | - | - | - | - | _ | - | - | _ |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Ga | - | - | - | - | - | - | - | - | - |
| Zn | - | - | - | - | - | - | - | - | - |
| Cu | - | - | - | - | - | _ | - | _ | - |
| Ni | - | - | - | - | _ | - | _ | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| $\checkmark$ | - | - | 22. | - | 10. | - | 18. | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | _ | - | - | - | - |
| Cs | - | - | - | _ | - | - | - | - | _ |
| Sc | - | - | - | - | - | - | - | _ | - |
| To | - | - | - | - | - | - | _ | - | - |
| Co | - | - | - | - | - | - | - | $\sim$ | - |
| Li | - | - | 312.0 | 312.0 | 89. 0 | B9. 0 | 327. 0 | 327.0 | - |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | 22.0 | 22.0 | 26.0 | 26.0 | 20.0 | 20.0 | - |
| F | - | - | 300. | 330. | 220. | 220. | 330. | 330. | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | 12.0 | 12.0 | 12.6 | 12.6 | 11.3 | 11.3 | - |
| Sn | - | - | - | - | 17.0 | 17.0 | - | - | 6. 0 |
| Mo | - | - | . 50 | . 50 | .10 | .10 | . 20 | . 20 | - |
| Lo | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical datobose.

## MOROCCO

|  | MMZ27 MENT GRAN | $\begin{aligned} & \text { MMZ28 } \\ & \text { MENT } \\ & \text { GRAN } \end{aligned}$ | MMZ29 MENT GRAN | MMZ3O <br> MENT <br> GRAN | MMZ31 MENT GRAN | MMZ35 MENT GRAN | MMZ38 MENT GRAN | MMZ4O MENT GRAN |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 74.75 | 76. 28 | 74.72 | 73. 90 | 74.58 | 75.10 | 74.89 | 73. 56 | 73. 14 |
| $\mathrm{TiO}_{2}$ | . 03 | . 03 | . 04 | . 02 | . 03 | . 02 | . 03 | - | . 17 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14.39 | 13.12 | 14.20 | 13.95 | 14.46 | 14.06 | 14.04 | 15.01 | 14.51 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 1.02 | . 93 | 1.09 | 1.12 | . 92 | 1. 11 | 1.07 | 1. 71 | 1. 21 |
| FeO | - | - | - | - | - | - | - | - | - |
| MnO | . 03 | . 02 | . 13 | . 06 | . 02 | . 08 | . 02 | . 12 | . 08 |
| MgO | . 03 | . 03 | . 05 | - | . 03 | - | . 07 | . 88 | - |
| CaO | . 21 | . 27 | .17 | . 02 | . 18 | . 03 | . 14 | - | - |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3.74 | 2.49 | 2.19 | 2.77 | 3.75 | 1. 41 | 3. 39 | 3.86 | 3.55 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4.42 | 4. 61 | 5. 06 | 5. 44 | 4. 23 | 5. 16 | 4.77 | 4.61 | 4. 72 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | d. | 4 | - | . 16 | - | . 22 | - | . 13 | . 12 |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | - | - | , | - | , | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | 1.18 | 1.38 | 1.83 | 1.24 | 1.17 | 1.67 | 1.32 | 1.39 | 1.18 |
| TOTAL | 99.80 | 99.16 | 99. 38 | 98. 68 | 99. 37 | 100.16 | 99.74 | 100.47 | 98.68 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 1.02 | . 93 | 1.09 | 1.12 | . 92 | 1. 11 | 1.07 | 1.71 | 1. 21 |
| A/CNK | 1.3 | 1.4 | 1.5 | 1.3 | 1.3 | 1.8 | 1.3 | 1.3 | 1, 3 |
| DI | 93. 3 | 91.8 | 90. 6 | 92.6 | 92.9 | 90. 9 | 93. 4 | 93. 2 | 92. 4 |
| Ba | - | - | - | 48. | - | 52. | - | 28. | 20. |
| Rb | - | - | - | 905. | - | 751. | - | 1473. | 894. |
| Sr | - | - | - | 22. | - | 66. | - | 25. | 20. |
| $Y$ | - | - | - | - | - | - | - | 25. | 20. |
| Zr | - | - | - | - | - | - | - | - | _ |
| Nb | - | - | _ | - | - | - | - | - | - |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Go | - | - | - | - | - | - | - | - | - |
| Zn | - | _ | _ | - | - | - | - | _ | - |
| Cu | - | - | - | - | _ | - | - | _ | - |
| Ni | - | - | _ | _ | _ | _ | $-$ | - | _ |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | _ | - | - |
| $\checkmark$ | - | - | - | 28. | - | 27. | - | 20. | 21. |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | $\cdots$ | - | _ | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | _ | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | - | - | - | 279.0 | - | 172.0 | - | 1451.0 | 453.0 |
| Be | - | - | - | - | - | - | - | 1.51. | - |
| B | - | - | - | 20.0 | - | 20.0 | - | 20.0 | 22.0 |
| F | - | - | - | 300. | - | 300. | - | 1000. | 520. |
| Cl | - | - | - | , | - |  | - | 1 | . |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | 14.5 | - | 104.0 | - | 42.5 | 17.9 |
| Sn | - | - | - | 44.0 | - | 930.0 | - | 37.0 | 37.0 |
| Mo | - | - | - | . 10 | - | . 20 | - | .10 | . 50 |
| Lo | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical databose.

MOROCCO

|  | MMZ63 MENT GRAN | MMZ64 MENT GRAN | MOU430 OULM GRAN | MOU431 OULM GRAN | MOU432 OULM GRAN | MOU433 <br> OULM <br> GRAN | MOU434 OULM GRAN | MOU435 <br> OULM <br> GRAN | MOU436 <br> OULM <br> GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 75.70 | 77.42 | 73.10 | 73.00 | 72.70 | 72.90 | 72.80 | 74.50 | 74.85 |
| $\mathrm{TiO}_{2}$ | . 06 | - | . 05 | - | . 08 | . 12 | - | . 15 | - |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 12.77 | 12.19 | 15.73 | 15.94 | 17.10 | 14.74 | 14.52 | 14.49 | 14.85 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 76 | .18 | . 12 | . 48 | - | . 29 | . 20 | 24 | . 25 |
| FeO | - | - | 1.35 | . 75 | . 65 | . 91 | . 69 | 50 | 52 |
| Mno | . 04 | . 02 | - | - | - | . 07 | . 07 | - | . 13 |
| MgO | - | - | . 73 | . 73 | . 54 | . 4 | . 77 | . 87 | . 29 |
| CoO | . 03 | . 22 | . 44 | . 27 | . 70 | 1.17 | . 92 | . 82 | . 85 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 2. 95 | 3. 50 | 4. 24 | 4. 46 | 3.29 | 3.65 | 4.05 | 2. 81 | 4.20 |
| $\mathrm{K}_{2} \mathrm{O}$ | 5. 41 | 4.79 | 2. 36 | 2.50 | 3. 52 | 4. 30 | 3. 75 | 3.69 | 3.45 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | - | - | - | - | . 08 | . 37 | . 29 | . 11 | . 10 |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | - | - | - | - | . 50 | . 23 | - | . 36 |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | . 03 | - | - | . 07 |
| $\mathrm{CO}_{2}$ | - | - | - | - | - |  | - | - |  |
| Cl | - | - | - | - | - | - | - | - | - |
| $F$ | - | - | - | - | - | - | - | - | - |
| LOI | 1.03 | . 74 | 2. 00 | 1.60 | 1.80 | - | - | 1. 60 | - |
| TOTAL | 98. 75 | 99. 06 | 100.12 | 99.73 | 100.46 | 99.4B | 98.29 | 99.88 | 99.92 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | . 76 | .18 | 1.62 | 1.31 | .72 | 1.30 | . 97 | . 91 | . 83 |
| A/CNK | 1.2 | 1.1 | 1.5 | 1.5 | 1. 6 | 1. 2 | 1. 2 | 1.4 | 1.2 |
| DI | 94.7 | 98.2 | 86.2 | 87.9 | 86.2 | 89.3 | 88.7 | 86. 6 | 91.1 |
| 80 | 59. | 47. | - | - | - | - | - | - | - |
| Rb | 351. | 211. | - | - | - | - | - | - | - |
| Sr | 15. | 10. | - | - | - | - | - | - | - |
| Y | - | - | - | - | - | - | - | - | - |
| Zr | - | - | - | - | - | - | - | - | - |
| Nb | - | - | - | - | - | - | - | - | - |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Ga | - | - | - | - | - | - | - | - | - |
| Zn | - | - | - | - | - | - | - | - | - |
| Cu | - | - | - | - | - | - | - | - | - |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| $V$ | 10. | 10. | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | $\cdots$ | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 68.0 | 23.0 | - | - | - | - | - | - | - |
| Be | - | - | - | - | - | - | - | - | - |
| B | 20.0 | 20.0 | - | - | - | - | - | - | - |
| F | 44. | 10. | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | 5.1 | 56.0 | - | - | - | - | - | - | - |
| Sn |  | 17.0 | - | - | - | - | - | - | - |
| Mo | . 20 | . 30 | - | - | - | - | - | - | - |
| LO | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.

MOROCCO

|  | MOU437 OULM GRAN | MOU459 OULM GRAN | $\begin{aligned} & \text { MOU460 } \\ & \text { OULM } \\ & \text { GRAN } \end{aligned}$ | MOU461 OULM GRAN | MOU462 <br> OULM <br> GRAN | MOU463 OULM GRAN | MOU464 OULM GRAN | MOU465 OULM GRAN | MOU466 OULM GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 75.00 | 70.05 | 65.10 | 71.85 | 71.20 | 74.35 | 72. 20 | 72. 95 | 73. 42 |
| $\mathrm{TiO}_{2}$ | . 47 | . 30 | . 60 | . 15 | . 21 | . 15 | . 18 | . 15 | . 15 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 15.35 | 17. 10 | 15.10 | 15.45 | 15.65 | 14.77 | 15.05 | 15.02 | 15. 34 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 1.07 | . 70 | 2. 17 | . 33 | . 32 | 1.34 | . 35 | - | . 45 |
| Feo | - | .97 | 2.01 | . 60 | 1.17 | . 67 | . 87 | . 81 | . 75 |
| MnO | - | - | .10 | - | . 06 | - | . 04 | - | - |
| MgO | . 34 | . 80 | 2.17 | . 76 | . 57 | . 70 | . 90 | .71 | . 51 |
| CaO | . 70 | 1.70 | 2. 97 | 1.70 | 1.30 | 1.28 | 1.45 | 1.42 | 1. 22 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3.08 | 2. 76 | 3. 61 | 3. 03 | 2. 90 | 1. 54 | 3.30 | 3. 23 | 2. 49 |
| $\mathrm{K}_{2} \mathrm{O}$ | 3. 25 | 1.90 | 4. 41 | 3.18 | 5.30 | 4. 33 | 4. 35 | 3.72 | 4.60 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 09 | . 32 | . 33 | . 20 | - | - | . 30 | . 20 | . 12 |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | , | 1.02 | - | .41 | - | . 45 | - | - |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | . 20 | - | - | - | .10 | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | _ | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | 1. 05 | 3. 40 | - | 2. 50 | - | 1.00 | - | 2. 10 | 1.80 |
| TOTAL | 100.40 | 100.30 | 99.79 | 99.75 | 99. 09 | 100.13 | 99. 54 | 100. 31 | 100.85 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} t$ | 1. 07 | 1.78 | 1. 40 | 1.00 | 1.62 | 2. 48 | 1.32 | 90 | 1.28 |
| A/CNK | 1.6 | 1.8 | . 9 | 1. 3 | 1. 2 | 1.6 | 1. 2 | 1. 3 | 1.4 |
| DI | 88.2 | 77.1 | 75.7 | 82.0 | 85.9 | 83.7 | 86. 4 | 85.1 | 86.2 |
| Ba | - | - | - | - | - | - | - | - | - |
| Rb | - | - | - | - | - | - | - | - | - |
| Sr | - | - | - | $\square$ | - | - | - | - | - |
| Y | - | - | - | - | - | - | - | - | - |
| Zr | - | - | - | - | - | - | - | - | - |
| Nb | - | - | - | - | - | - | - | - | - |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Go | - | - | - | - | - | - | - | - | - |
| Zn | - | - | - | - | - | - | - | _ | - |
| Cu | - | - | - | - | - | - | - | - | - |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| $V$ | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | - | - | - | - | - | - | - | - | - |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| $F$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| La | - | - | - | - | - | - | - | - | - . |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.

MOROCCO

|  | MOU467 OULM GRAN | MOU468 OULM GRAN | MOU469 OULM GRAN | MOU470 OULM GRAN | MOU471 <br> OULM <br> GRAN | MOU472 OULM GRAN | MOU473 OULM GRAN | MOU474 <br> OULM <br> GRAN | MOU475 OULM GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 71.85 | 73.70 | 71.70 | 65.65 | 72. 25 | 72.90 | 73.70 | 73.10 | 68. 20 |
| $\mathrm{TiO}_{2}$ | . 16 | . 15 | . 05 | . 10 | 11 | . 21 | - | . 15 | . 40 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 15.65 | 14. 59 | 15.18 | 18.90 | 15.45 | 15.00 | 15.60 | 14.12 | 18.24 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 42 | . 05 | - | . 65 | . 47 | . 31 | - | . 18 | . 37 |
| FeO | 1. 04 | 1.12 | . 97 | 1. 20 | . 42 | 1.10 | . 60 | 1.94 | 1.96 |
| MnO | . 07 | - | - | - | . 07 | . 08 | - | - | . 06 |
| MgO | . 57 | . 29 | 1.30 | . 10 | . 70 | , 93 | 54 | . 76 | . 84 |
| CoO | 1. 42 | 1.13 | 1.47 | 1.39 | 1.15 | 1.20 | . 90 | . 81 | 1. 25 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3. 15 | 1. 59 | 3. 02 | 3.40 | 3. 05 | 2. 85 | 3. 82 | 1.75 | 3. 05 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4. 65 | 4.67 | 4. 10 | 6. 66 | 4.05 | 4.90 | 4.06 | 5. 48 | 5. 20 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 30 | . 16 | - | - | . 24 | . 20 | - | - | . 27 |
| $\mathrm{H}_{2} \mathrm{O}+$ | . 63 | - | - | - | . 65 | . 33 | - | - | . 50 |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | . 18 | .10 | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | 2. 00 | 1.90 | 2.00 | - | - | 1.95 | 2.20 | - |
| TOTAL | 99. 91 | 99.45 | 99. 69 | 100.37 | 98. 79 | 100.11 | 101.17 | 100.49 | 100. 34 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3 \mathrm{t}}$ | 1.57 | 1.29 | 1.08 | 1. 99 | . 94 | 1.53 | 67 | 2. 33 | 2. 55 |
| A/CNK | 1.2 | 1.5 | 1.3 | 1.2 | 1.3 | 1.2 | 1.3 | 1.4 | 1.4 |
| DI | 86.7 | 84.4 | 82.4 | 84.3 | B6. 5 | 86.9 | 89. 0 | 84.8 | 83.1 |
| Ba | - | - | - | - | - | - | - | - | - |
| Rb | - | - | - | - | - | - | - | - | - |
| Sr | - | - | - | - | - | - | - | - | - |
| $Y$ | - | - | - | - | - | - | - | - | - |
| Zr | - | - | - | - | - | - | - | - | - |
| Nb | - | - | - | - | - | - | - | - | - |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | _ | _ | - | - |
| Ga | - | - | - | - | - | - | - | - | - |
| Zn | - | - | - | - | - | - | _ | - | - |
| Cu | - | - | - | - | - | _ | - | - | - |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| $V$ | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | _ | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | - | - | - | - | - | - | - | - | - |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | - | - | _ | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical databose.

MOROCCO

|  | MOU476 OULM GRAN | MQR2 AEBA MONG | MQR4 AEBA MONG | MQR6 AEBA MONG | MSDB1 SDBR MONG | MSDB10 SDBR MONG | MSDB11 SDBR MONG | MSDB12 SDBR MONG | MSOB13 SDBR MONG |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 75.65 | 75.64 | 75. 21 | 75.64 | 76.81 | 75. 01 | 75.02 | 76. 24 | 76. 63 |
| $\mathrm{TiO}_{2}$ | . 10 | 07 | . 06 | . 10 | . 02 | 09 | . 20 | . 03 | . 15 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14.85 | 12.43 | 13.17 | 12.39 | 13. 39 | 13. 56 | 13.00 | 12.81 | 12.25 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 1.10 | 1.22 | . 96 | 1. 32 | . 40 | . 97 | 1.49 | . 90 | 1.18 |
| FeO | . 81 | - | - | - | - | - | - | - | - |
| MnO | - | . 03 | . 03 | . 04 | 01 | . 03 | . 02 | . 02 | . 02 |
| MgO | . 49 | . 06 | . 03 | . 10 | . 01 | . 05 | 21 | . 01 | . 14 |
| COO | 1.75 | 1.92 | . 70 | . 71 | . 37 | . 88 | . 79 | 1.21 | 71 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 2.01 | 3.60 | 3.64 | 3.73 | 4. 23 | 3. 91 | 3. 45 | 4.08 | 3.39 |
| $\mathrm{K}_{2} \mathrm{O}$ | 1.15 | 4.54 | 4. 94 | 4.54 | 4. 64 | 4.73 | 4. 89 | 4. 32 | 4.90 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . | . 02 | . 03 | . 02 | . 03 | . 10 | . 06 | . 02 | . 05 |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | $\bigcirc$ | - | - | . | - | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | 2. 00 | . 31 | . 39 | . 38 | . 29 | . 56 | . 11 | . 35 | . 35 |
| TOTAL | 99.91 | 99.94 | 100.16 | 99.97 | 100. 20 | 99.89 | 109. 54 | 99. 99 | 99.77 |
| $\mathrm{Fe}_{2} \mathrm{O} 3 \mathrm{t}$ | 2.00 | 1. 22 | . 96 | 1.32 | . 40 | . 97 | 1.49 | . 90 | 1.18 |
| A/CNK | 1.9 | . 9 | 1.0 | 1.0 | 1.1 | 1.0 | 1.1 | . 9 | 1.0 |
| DI | 78.7 | 93.0 | 94.7 | 94.3 | 96. 9 | 93.5 | 93.5 | 94.3 | 94.2 |
| Ba | - | 51. | 84. | 13. | 36. | 157. | 234. | 7. | 184. |
| Rb | - | 284. | 291. | 289. | 209. | 222. | 180. | 367. | 187. |
| Sr | - | 17. | 2 E . | 10. | 10. | 67. | 75. | 4. | 50. |
| Y | - | 89. | 62. | 79. | 34. | 23. | 25. | 51. | 30. |
| Zr | - | 105. | 80. | 90. | 43. | 74. | 138. | 62. | 117. |
| Nb | - | 33. | 17. | 19. | 20. | 27. | 31. | 36. | $22 .$ |
| Th | - | 45.00 | 34.00 | 53.00 | 15.00 | 29.00 | $36.00$ | 32.00 | 36. 00 |
| Pb | - | 17. | 28. | 18. | 27. | 13. | 8. | 25. | 12. |
| Ga | - | 20. | 18. | 19. | 19. | 20. | 17. | 23. | 17. |
| Zn | - | 18.0 | 15.0 | 18.0 | 13.0 | 13.0 | 20.0 | 17.0 | 14.0 |
| Cu | - | 4.0 | 1.0 | - | - | 5.0 | 1.0 | - | 2. 0 |
| Ni | - | 47. 0 | 36.0 | 45.0 | 21.0 | 20.0 | 15.0 | 40.0 | 17.0 |
| $\mathrm{TiO}_{2}$ | - | . 06 | . 05 | . 06 | - | . 09 | . 23 | . 02 | . 14 |
| V | - | 1. | 4. | 2. | 2. | 6. | 9. | 4. |  |
| Cr | - | 6. | 3. | 12. | 6. | 7. | 8. | 10. | 4. |
| Hf | - | . | . | 12. | b. | \% | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | _ | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | - | - | - | - | - | - | - | - | - |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | _ | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| Lo | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.

MOROCCO

|  | $\begin{aligned} & \text { MSDB14 } \\ & \text { SDBR } \\ & \text { MONG } \end{aligned}$ | MSDB2A SDBR MONG | $\begin{aligned} & \text { MSDB2B } \\ & \text { SDBR } \\ & \text { MONG } \end{aligned}$ | $\begin{aligned} & \text { MSDB3 } \\ & \text { SDBR } \\ & \text { MONG } \end{aligned}$ | $\begin{aligned} & \text { MSDB4 } \\ & \text { SDBR } \\ & \text { MONG } \end{aligned}$ |  |  | MSDB8 SDBR <br> MONG |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 77. 85 | 74. 74 | 77. 93 | 77.39 | 76.96 | 75.48 | 75.57 | 75.98 | 77.69 |
| $\mathrm{TiO}_{2}$ | . 09 | . 17 | . 89 | . 07 | . 07 | . 12 | . 10 | . 09 | . 07 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 12.88 | 13. 29 | 12.91 | 12.62 | 12. 63 | 12.73 | 12. 28 | 12.73 | 12.50 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 91 | 1.41 | . 92 | 1. 01 | . 83 | 1.09 | 1.01 | . 77 | . 44 |
| FeO | - | - | - | - | - | - | - | - | - |
| MnO | . 02 | . 03 | . 01 | . 02 | . 01 | . 02 | . 02 | . 01 | . 01 |
| MgO | . 01 | . 16 | . 01 | . 01 | . 01 | .10 | . 05 | . 01 | . 01 |
| CaO | . 49 | . 80 | . 45 | . 42 | . 33 | . 62 | 1. 36 | . 58 | . 44 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3.29 | 3. 69 | 3.05 | 3. 70 | 3. 93 | 3. 53 | 3. 42 | 3.57 | 3.73 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4.73 | 5. 10 | 5.19 | 4.69 | 4. 69 | 4. 91 | 4.92 | 4.88 | 4.71 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 03 | . 04 | . 02 | . 01 | . 02 | . 06 | . 02 | . 02 | . 01 |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | $\cdots$ |
| $\mathrm{CO}_{2}$ | - | - | _ | _ | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - |  |  |  | - | - |
| LOl | . 27 | . 32 | . 32 | . 33 | . 32 | . 31 | . 29 | . 33 | . 35 |
| TOTAL | 100.57 | 99.75 | 100.90 | 100.27 | 99.80 | 99. 97 | 100.04 | 99.97 | 99. 96 |
| $\mathrm{Fe}_{2} \mathrm{O} 3 \mathrm{t}$ | . 91 | 1.41 | . 92 | 1.01 | . 83 | 1.09 | 1.01 | . 77 | . 44 |
| A/CNK | 1.1 | 1.0 | 1.1 | 1.1 | 1.0 | 1.0 | . 9 | 1. 0 | 1. 0 |
| DI | 95.4 | 93.3 | 95.9 | 96.1 | 96.4 | 94.7 | 94.2 | 95.4 | 96.4 |
| Bo | 57. | 218. | 58. | 4. | 15. | 108. | 78. | $165 .$ | $70 .$ |
| Rb | 279. | 181. | 204. | 338. | 368. | 231. | 150. | 187. | $245 .$ |
| Sr | 15. | 60. | 21. | 4. | 9. | 35. | 18. | 16. | 33. |
| Y | 41. | 30. | 25. | 36. | 53. | 33. | 19. | 25. | 26. |
| Zr | 96. | 136. | 78. | 82. | 134. | 101. | 87. | 84. | 83. |
| Nb | 30. | 27. | 32. | 52. | 67. | 29. | 12. | 37. | 37. |
| Th | 42.00 | 23.00 | 35.00 | 51.00 | 55.00 | 43.00 | 11.00 | 43.00 | 43.00 |
| Pb | 19. | 11. | 18. | 17. | 23. | 21. | 16. | 17. | $18$ |
| Ga | 21. | 20. | 17. | 22. | 22. | 19. | 18. | 18. | 21. |
| Zn | 15.0 | 18.0 | 21.0 | 14.0 | 11.0 | 22.0 | 27.0 | 15.0 | 17.0 |
| Cu | - | 1.0 | 9.0 | - | 1. 0 | 2. 0 | 6.0 | 2.0 | - |
| Ni | 31.0 | 17.0 | 20.0 | 31.0 | 42.0 | 21.0 | 11.0 | 19.0 | 21.0 |
| $\mathrm{TiO}_{2}$ | . 08 | . 20 | . 07 | . 08 | . 06 | . 10 | . 08 | . 09 | . 06 |
| $V$ | 7. | 11. | 4. | - | - | 3. | 1. | 1. | 1. |
| Cr | 4. | 10. | 11. | 4. | 9. | 10. | 22. | 8. | 7. |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | - | - | - | - | - | - | - | - | - |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | - | _ | - | _ | - | - | _ |
| $F$ | - | - | _ | _ | - | - | - | - | - |
| Cl | - | - | _ | - | - | _ | _ | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical databose.

MOROCCO

|  | MTI6 1 TICH GRAN | MTI62 TICH GRAN | MTI710 <br> TICH <br> GRAN | MTI711 <br> TICH <br> GRAN | MTI712 <br> TICH <br> GRAN | MTI719 TICH GRAN | MZA1 ZAER GRAN | MZA102 ZAER GRAN | MZA104 ZAER GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 71.01 | 75.08 | 70.22 | 69.41 | 69.41 | 67. 32 | 59. 96 | 77.07 | 68. 61 |
| $\mathrm{TiO}_{2}$ | . 36 | . 14 | . 48 | . 51 | . 45 | . 43 | . 35 | . 09 | . 58 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 13. 92 | 13.14 | 14.12 | 14.88 | 15.34 | 14.92 | 15.16 | 12.64 | 15.51 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 75 | . 39 | . 98 | 1. 17 | . 49 | . 67 | 2. 28 | 94 | 3.79 |
| FeO | 1.10 | . 50 | 1. 65 | 1.87 | 1.80 | 1. 50 | - | - | - |
| MnO | . 06 | . 02 | . 04 | . 06 | . 09 | . 10 | 05 | . 02 | . 08 |
| MgO | 1. 35 | 1.75 | 2.14 | 1. 50 | . 80 | 1.80 | . 49 | .13 | 1.58 |
| CoO | 1.16 | . 75 | 1.79 | 2.16 | 2.04 | 1. 98 | 1.40 | . 46 | 2. 46 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 4. 27 | 3.39 | 4.44 | 4. 54 | 4. 53 | 4.72 | 3.70 | 3. 26 | 3. 59 |
| $\mathrm{K}_{2} \mathrm{O}$ | 3. 91 | 5. 00 | 3. 35 | 3. 31 | 3. 47 | 3.19 | 3. 94 | 4. 63 | 3. 25 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 10 | - | . 13 | . 15 | . 13 | . 12 | . 25 | . 36 | . 37 |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | - | - |  | - | - | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | $\rightarrow$ | - | $\sim$ | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOH | . 13 | 06 | .16 | . 22 | .19 | . 19 | 1.93 | 61 | . 65 |
| TOTAL | 98.12 | 100. 23 | 99. 50 | 99.68 | 98.54 | 97.04 | 99. 51 | 100.21 | 100.45 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 1.97 | . 95 | 2. 81 | 3.15 | 2.27 | 2. 45 | 2. 28 | . 94 | 3.79 |
| A/CNK | 1.0 | 1.1 | 1. 1 | 1.0 | 1.0 | 1.0 | 1.2 | 1.1 | 1.1 |
| DI | 85.8 | 90.0 | 81. 6 | 81.0 | 82.5 | 79.0 | 84,9 | 95.1 | 78.3 |
| Bo | - | - | - | - | - | - | 355. | 330. | 691. |
| Rb | - | - | - | - | - | - | 287. | $255 .$ | $122 .$ |
| Sr | - | - | - | - | - | - | 130. | 130. | 324. |
| $Y$ | - | - | - | - | - | - |  | - | - |
| Zr | - | - | - | - | - | - | - | - | - |
| Nb | - | - | - | - | - | - | - | - | - |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Go | - | - | - | - | - | - | - | - | - |
| Zn | - | - | - | - | - | - | 73. 0 | 89.0 | 60.0 |
| Cu | - | - | - | - | - | - | 10.0 | 37.0 | 10.0 |
| Ni | - | - | - | - | - | - | 10.0 | 10.0 | 11.0 |
| $\mathrm{TiO}_{2}$ | - | - | _ | _ | - | - | - | - | - |
| V | - | - | - | - | - | - | 34. | 16. | 69. |
| Cr | - | - | - | - | - | - | 10. | 10. | 41. |
| Hf | - | - | - | - | - | - | 10. | 10. | 4. |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | _ | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | 10. | 10. | 10. |
| Li | - | - | - | - | - | - | - | - | - |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| $F$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | _ | - | - | - | - | - |
| U | - | - | - | - | - | - | 1.00 | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | 163.0 | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.

MOROCCO

|  | MZA106 ZAER GRAN | MZA107 ZAER GRAN | MZA108 ZAER GRAN | MZA109 ZAER GRAN | MZA112 ZAER GRAN | MZA1 13 ZAER GRAN | MZA114 ZAER GRAN | MZA115 ZAER GRAN | MZA116 ZAER GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 73. 61 | 73. 38 | 58.24 | 69. 99 | 88. 85 | 71.54 | 73. 72 | 75. 26 | 74. 25 |
| $\mathrm{TiO}_{2}$ | 13 | 24 | . 54 | . 37 | . 51 | 29 | . 02 | . 08 | . 12 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14.14 | 13.94 | 15. 64 | 14.55 | 15.67 | 14.96 | 15.14 | 14.14 | 14.03 |
| $\mathrm{Fe}_{2} \mathrm{O} 3$ | 1.18 | 1. 55 | 3. 56 | 2. 23 | 3.58 | 1. 90 | 84 | . 73 | . 93 |
| FeO | - | - | - | - | - | - | - | - | - |
| MnO | . 04 | . 05 | . 07 | . 05 | . 07 | . 05 | . 07 | . 04 | 04 |
| MgO | . 14 | . 02 | 1.52 | . 52 | 1. 34 | . 44 | . | . 02 | 04 |
| CaO | . 26 | 65 | 3. 01 | 1.13 | 2.26 | 1. 08 | - | . 29 | 08 |
| No2O | 3.19 | 3.31 | 3. 90 | 3. 40 | 4.02 | 3. 39 | 3. 62 | 3. 29 | 2. 92 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4.62 | 4. 40 | 2.40 | 4.01 | 2.89 | 4.60 | 3.83 | 5.00 | 4.89 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 18 | . 15 | . 13 | . 24 | . 20 | . 24 | . 36 | . 24 | . 37 |
| $\mathrm{H}_{2} \mathrm{O}+$ | . 18 | . 15 | . 13 | . 24 | , 20 | . 24 | - | . 2 | , 37 |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - |  |
| LOI | 2. 76 | 2.49 | 1.04 | 2.31 | 1.07 | . 75 | 1.62 | 1. 39 | 1.57 |
| TOTAL | 100. 25 | 100.18 | 100.05 | 99.21 | 100.44 | 99. 24 | 99. 22 | 100.48 | 99. 24 |
| $\mathrm{Fe}_{2} \mathrm{O} 3 \mathrm{t}$ | 1.18 | 1. 55 | 3. 56 | 2. 23 | 3. 58 | 1.90 | . 84 | . 73 | . 93 |
| A/CNK | 1.3 | 1. 2 | 1.1 | 1.2 | 1.1 | 1.2 | 1.5 | 1.3 | 1.4 |
| DI | 91.4 | 90.3 | 75.2 | 84. 2 | 79.3 | 87.8 | 91.3 | 94.4 | 92.1 |
| Ba | 183. | 263. | 675. | 473. | 536. | 412. | 24. | 193. | 275. |
| Rb | 430. | 364. | 96. | 306. | 239. | 358. | 801. | 468. | 408. |
| Sr | 47. | 75. | 400. | 156. | 286. | 142. | 100. | 52. | 51. |
| Y | - | - | - | - | - | - | - | - | - |
| Zr | - | - | - | - | - | - | - | - | - |
| Nb | - | - | - | - | - | - | - | - |  |
| Th | - | - | - | - | - | - | - | - |  |
| Pb | - | - | - | - | - | - | - |  |  |
| Ga | - | - | - | - | - | - | - | - | - |
| Zn | 42.0 | 51.0 | 71.0 | 80.0 | 79.0 | 75.0 | 103.0 | 44.0 | 59.0 |
| Cu | 10.0 | 10.0 | 10.0 | 10.0 | 10.0 | 10.0 | 10.0 | 10.0 | 10.0 |
| Ni | 10.0 | 10.0 | 13.0 | 10.0 | 15.0 | 10.0 | 10.0 | 10.0 | 10.0 |
| $\mathrm{TiO}_{2}$ | . | - | . | - | Is.0 | - | - | - | - |
| V | 11. | 11. | 88. | 52. | 71. | 45. | 24. | 23. | 28. |
| Cr | 10. | 10. | 40. | 16. | 32. | 14. | 17. | 10. | 10. |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | 10. | 10. | 10. | 10. | 10. | 10. | 10. | 10. | 10. |
| Li | - | - | - | 10 | 10. | 10. | - | - | - |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | 2.00 |
| W | - | - | - | - | - | - | 2.9 | - | - |
| Sn | 20.0 | - | - | 12.0 | - | 14.0 | 56.0 | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| Lo | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.

|  | $N O Q O$ |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | MZA117 <br> ZAER <br> GRAN | MZA118 ZAER GRAN | MZA119 <br> ZAER <br> GRAN | $\begin{aligned} & \text { MZA1 } 21 \\ & \text { ZAER } \\ & \text { GRAN } \end{aligned}$ | MZA122 <br> ZAER <br> GRAN | MZA124 <br> ZAER <br> GRAN | MZA13 ZAER GRAN | $\begin{aligned} & \text { MZA134 } \\ & \text { ZAER } \\ & \text { GRAN } \end{aligned}$ | MZA15 ZAER GRAN |
| $\mathrm{SiO}_{2}$ | 68. 84 | 73.18 | 70.47 | 75. 92 | 74.56 | 64.30 | 74.76 | 73. 27 | 74. 52 |
| $\mathrm{TiO}_{2}$ | . 41 | . 04 | . 40 | . 03 | . 13 | . ¢ $B$ | . 03 | . 11 | 01 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 15. 52 | 15. 29 | 15. 21 | 14.05 | 14.21 | 15.82 | 14.43 | 14. 09 | 13.95 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 2. 81 | 1.39 | 2. 71 | . 65 | 1. 03 | 4.82 | . 60 | 1. 23 | . 92 |
| FeO | - | - | - | - | - | - | - | - | - |
| MnO | . 06 | . 05 | . 05 | . 07 | . 13 | . 11 | . 08 | . 04 | . 03 |
| MgO | 1.08 | - | 1.09 | - | . 13 | 2. 11 | - | . 23 | . 01 |
| CaO | 2. 31 | - | 1.82 | - | . 17 | 1.86 | - | . 37 | .12 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3.88 | 3.50 | 3. 52 | 3. 67 | 3. 37 | 3. 04 | 4. 23 | 3.07 | 2. 30 |
| $\mathrm{K}_{2} \mathrm{O}$ | 3.69 | 4.09 | 4. 22 | 4. 06 | 4.88 | 4. 42 | 3. 90 | 5.03 | 5.28 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 70 | . 41 | - | - | . 22 | . 29 | . 03 | . 02 | . 19 |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | $\rightarrow$ | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | 1.04 | 1. 68 | . 19 | . 78 | 1. 52 | 2. 02 | 1.53 | 1.00 | 2.14 |
| TOTAL | 100.34 | 99.63 | 99. 98 | 100.21 | 100. 25 | 100.47 | 99.59 | 98. 16 | 99.48 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3 t}$ | 2. 81 | 1.39 | 2.71 | . 66 | 1.03 | 4. 82 | .60 | 1.23 | . 92 |
| A/CNK | 1.1 | 1.5 | 1.1 | 1.4 | 1. 3 | 1.3 | 1.3 | 1.3 | 1.4 |
| DI | 82.2 | 91.0 | 83.0 | 95.1 | 93.4 | 75.2 | 94.1 | 90.8 | 91.6 |
| Ba | 723. | 34. | 555. | 27. | 76. | 728. | 28. | 144. | 157. |
| Rb | 155. | - | 166. | 418. | 370. | 297. | 423. | 392. | 414. |
| Sr | 319. | 681. | 287. | 16. | 27. | 380. | 15. | 44. | 41. |
| $Y$ | - | - | - | - | - | - | - | - | - |
| Zr | - | - | - | - | - | - | - | - | - |
| Nb | - | - | - | - | - | - | - | - | - |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | _ | - | - | - | - | - |
| Ga | - | - | - | - | - | - | - | _ | - |
| Zn | 56.0 | - | 43.0 | 46. 0 | 58.0 | 122.0 | 10.0 | - | 67.0 |
| Cu | 10.0 | 27.0 | 10.0 | 10.0 | 10.0 | 10.0 | 10.0 | 10.0 | 10.0 |
| Ni | 10.0 | 10.0 | 10.0 | 10.0 | 10.0 | 34.0 | 10.0 | 10.0 | 10.0 |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| V | 64. | 10. | 37. | 10. | 15. | 88. | 10. | 21. | 17. |
| Cr | 30. | 10. | 26. | 10. | 10. | 59. | 10. | 10. | 10. |
| Hf | - | 1. | 2. | 1. | 10. | 5. | 10. | 1. | 10. |
| Cs | - | - | - | - | $\rightarrow$ | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | 10. | 62. | 10. | 10. | 10. | 10. | 10. | 10. | 10. |
| Li | - | - | - | - | - | - | - | - | - |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | _ | - |
| U | 3.10 | - | - | - | - | - | _ | - | - |
| W | , | . 3 | - | 2. 4 | 1.2 | - | - | 1. 2 | - |
| Sn | - | - | - | - | 18.0 | - | 30. 0 | - | 2. 5 |
| Mo | - | - | - | - | - | - | - | - | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.

## MOROCCO

|  | MZA1 64 ZAER GRAN | $\begin{aligned} & \text { MZA22 } \\ & \text { ZAER } \\ & \text { GRAN } \end{aligned}$ | MZA23 ZAER GRAN | MZA28 ZAER GRAN | MZA36 ZAER GRAN | MZA7 ZAER GRAN | MZAB ZAER GRAN | MZA85 ZAER GRAN | MZA88 ZAER GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 72.21 | 73.74 | 74. 15 | 72. 70 | 71.41 | 67.13 | 72. 66 | 76. 09 | 74.11 |
| $\mathrm{TiO}_{2}$ | . 29 | . 12 | . 06 | . 18 | . 04 | . 66 | 24 | . 06 | . 08 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14.58 | 14.34 | 13.83 | 15. 21 | 16. 08 | 15. 52 | 13.90 | 14. 31 | 14. 57 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 1.45 | 1.10 | . 17 | 1.15 | . 52 | 4. 17 | 1.49 | . 14 | . 66 |
| FeO | - | - | - | - | - | - | - | - | - |
| MnO | . 05 | . 01 | . 02 | - | . 03 | . 07 | . 01 | . 07 | . 04 |
| MgO | . 26 | . 01 | . 02 | . 05 | . 03 | 1.80 | . 02 | . | . 01 |
| CoO | . 47 | . 25 | . 32 | . 92 | . 08 | 2. 95 | . 28 | - | . 13 |
| $\mathrm{No}_{2} \mathrm{O}$ | 3.12 | 2. 93 | 4.10 | 3. 70 | 3.08 | 3.74 | 2. 56 | 4. 06 | 3. 30 |
| $\mathrm{K}_{2} \mathrm{O}$ | 5.25 | 5.24 | 3. 51 | 4.31 | 4. 43 | 3.02 | 4.87 | 3.86 | 5, 25 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 21 | . 19 | . 21 | . 19 | 1.45 | . 16 | . 43 | . 22 | . 29 |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | 1. 25 | 2.47 | . 98 | . 84 | 2. 69 | . 60 | 2.50 | . 79 | 1.33 |
| total | 93.14 | 100.10 | 97.35 | 99. 26 | 99. 84 | 99.92 | 98.96 | 99.90 | 99.78 |
| $\mathrm{Fe}_{2} \mathrm{O} 3 t$ | 1.45 | 1.10 | . 17 | 1.16 | . 52 | 4. 17 | 1. 49 | . 44 | . 66 |
| A/CNK | -1. 3 | 1.3 | 1.2 | 1. 2 | 1.6 | 1.1 | 1.4 | 1.3 | 1.3 |
| DI | 90.6 | 92.4 | 92.2 | 90.0 | 88.7 | 74.8 | 89.5 | 94.9 | 93.8 |
| 80 | 571. | 159. | 68. | 414. | 92. | 833. | 169. | 74. | 128. |
| Rb | 445. | 401. | 197. | 266. | 380. | 98. | 364. | 529. | 384. |
| Sr | 93. | 47. | 51. | 130. | 37. | 385. | 87. | 13. | 33. |
| Y | - | - | - | - | - | - | - | - | - |
| Zr | - | - | - | - | - | - | - | - | - |
| No | - | - | - | - | - | - | - | - | _ |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Go | - | - | - | - | - | - | - | - | - |
| Zn | - | 46.0 | 27. 0 | 52.0 | 68.0 | 60.0 | 84.0 | 84.0 | - |
| Cu | 77.0 | 10.0 | 10.0 | 10.0 | 10.0 | 10.0 | 10.0 | 10.0 | 10.0 |
| Ni | 10.0 | 10.0 | 10.0 | 10.0 | 10.0 | 30.0 | 10.0 | 10.0 | 10.0 |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| $\checkmark$ | 39. | 10. | 17. | 24. | 10. | 84. | 14. | 15. | 10. |
| Cr | 12. | 10. | 10. | 12. | 10. | 43. | 10. | 10. | 10. |
| Hf | - | - | - | - | 10. | . | , | I. | , |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | 10. | 10. | 10. | 10. | 10. | 10. | 10. | 10. | 10. |
| Li | - | - | - | - | - | - | - | - | - |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | 2. 35 | - | 3.00 | - | - | 1.65 | - | - |
| W | - | - | - | - | - | . 5 | - | - | - |
| Sn | - | 22.0 | - | - | - | 15.0 | 28.0 | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical databose.

| $M O P O C O$ |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | MZA89 | MZA90 | MZA91 | MZA92 | MZA94 | MZA95 | MZA96 | MZA98 |
|  | ZAER | ZAER | ZAER | ZAER | ZAER | ZAER | ZAER | ZAER |
|  | GRAN | GRAN | GRAN | GRAN | GRAN | GRAN | GRAN | GRAN |
| $\mathrm{SiO}_{2}$ | 74.15 | 72.75 | 64.93 | 68. 85 | 75.19 | 60.48 | 65.70 | 69.07 |
| $\mathrm{TiO}_{2}$ | . 06 | . 14 | . 65 | . 78 | - | 1.13 | . 60 | . 51 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14. 25 | 13. 71 | 15.93 | 12.50 | 14.34 | 17.17 | 16. 11 | 14.98 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 79 | 1. 27 | 4. 15 | 5.47 | . 78 | 5. 28 | 3.73 | 3. 40 |
| FeO | - | - | - | - | - | - | - | - |
| Mno | . 06 | . 02 | . 06 | . 08 | . 04 | . 10 | . 05 | . 06 |
| MgO | - | . 01 | 1.78 | 2.14 | - | 3.17 | 1. 58 | 1. 34 |
| CaO | - | . 16 | 2.88 | 1. 89 | . 23 | 4. 49 | 3. 10 | 2. 25 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 2. 34 | 2. 73 | 3. 57 | 2. 51 | 3. 38 | 3. 37 | 3. 89 | 3. 60 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4. 83 | 5. 19 | 2. 78 | 3.12 | 4.05 | 2. 75 | 2. 81 | 3. 49 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 72 | . 39 | . 52 | .44 | . 46 | . 33 | . 52 | .49 |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | - | - | - | - | - | , | , |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - |
| LOL | 2. 45 | 2.12 | 1.99 | 2. 01 | 1.55 | 1.27 | 1.95 | . 98 |
| TOTAL | 99.65 | 98.39 | 99.24 | 99. 82 | 100.02 | 99. 54 | 100.05 | 100.18 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | . 79 | 1.27 | 4. 15 | 5.47 | . 78 | 5.28 | 3.73 | 3.40 |
| A/CNK | 1.6 | 1.3 | 1.1 | 1.1 | 1.4 | 1.0 | 1.1 | $1.1$ |
| DI | 90.4 | 90.6 | 72.8 | 75.1 | 92. 6 | 61.7 | 74.3 | 80.4 |
| Bo | 27. | 211. | 699. | 534. | $22 .$ | $589 .$ |  | $700 .$ |
| Rb | 549. | 431. | 100. | 156. | 600. | 82. | 85. | 129. |
| Sr | 14. | 50. | 385. | 218. | 11. | 488. | 404. | 356. |
| Y |  | , | - | , | , |  | , | - |
| Zr | - | - | - | - | - | - | - | - |
| Nb | - | - | - | - | - | - | - | - |
| Th | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - |
| Ga | - | - | - | - | - | - | - | - |
| Zn | 64.0 | 65. D | 66.0 | 114.0 | - | 68.0 | 64.0 | 64.0 |
| Cu | 10.0 | 10.0 | 10.0 | 31.0 | 10.0 | 10.0 | 10.0 | 10.0 |
| Ni | 10.0 | 10.0 | 19.0 | 24.0 | 10.0 | 28. 0 | 11.0 | 13.0 |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - |
| $\checkmark$ | 16. | 20. | 79. | 99. | 18. | 144. | 89. | 71. |
| Cr | 10. | 10. | 47. | 54. | 10. | 78. | 39. | 36. |
| Hf | - | 10. | \% | 5. | 10. | \%. | , | . |
| Cs | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - |
| Co | 10. | 10. | 10. | 21. | 10. | 28. | 10. | 10. |
| Li | - | 10. | 1. | 21. | I. | , | 10. | 1 |
| Be | - | - | - | - | - | - | - | - |
| B |  | - | - | - | - | - | - | - |
| $F$ | - | - | - | - | _ | - | - | - |
| Cl | - | - | - | - | - | _ | - | - |
| U | 1.70 | - | - | - | 2. 70 | - | 1.95 | - |
| W | 3.0 | 1.6 | - | - | - | . 8 | . 5 | - |
| Sn | 26.0 | 28.0 | - | - | 39.0 | - | - | - |
| Mo | , | - | - | - | - | - | - | - |
| Lo | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical datobase.

IBERIA

|  | IBJ1 UNKN GRAN | IBJ10 UNKN GRAN | IBJ11 UNKN GRAN | IB.112 UNKN GRAN | I8J13 UNKN GRAN | IEJ14 UNKN GRAN | [BJ15 UNKN GRAN | 18516 UNKN GRAN | IBJ17 UNKN GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 67. 50 | 68.47 | 69. 04 | 67.78 | 67. 91 | 57. 31 | 69.00 | 68. 86 | 70. 63 |
| $\mathrm{TiO}_{2}$ | . 59 | . 71 | . 50 | . 58 | 70 | . 62 | . 59 | . 61 | . 49 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 16.03 | 15.42 | 16. 20 | 15.41 | 14.89 | 14.73 | 15.07 | 14. 36 | 14.50 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 3.92 | 3.72 | 3. 39 | 4. 36 | 3. 69 | 4. 59 | 3. 28 | 4. 11 | 3. 00 |
| FeO | 3. | 3. | 3. | . | 3. | , | 1.17 | , | 3. |
| MnO | . 05 | . 05 | . 06 | . 06 | . 06 | . 07 | . 05 | . 05 | . 05 |
| MgO | 1.30 | 1.28 | 1.08 | 1. 42 | 1. 38 | 1.47 | 1.17 | 1.29 | . 98 |
| CaO | 2.05 | 2.04 | 1.76 | 2.05 | 1.57 | 2. 09 | 1.36 | 2.21 | 1.71 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3.27 | 3.25 | 3. 48 | 3.12 | 2. 83 | 2. 96 | 3.01 | 3.12 | 2.82 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4.69 | 4.28 | 1. 49 | 4. 45 | 4.48 | 4. 40 | 4.62 | 4.26 | 4.96 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 30 | . 27 | . 17 | . 27 | . 19 | . 20 | . 17 | . 26 | . 21 |
| $\mathrm{H}_{2} \mathrm{O}+$ | . 86 | . 84 | . 56 | 1.03 | 1. 36 | 1.62 | 1.53 | 1. 17 | . 98 |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | . | , | , | , | , | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| total | 100.56 | 100.33 | 100.83 | 100.53 | 99.06 | 100.06 | 101.02 | 100.31 | 100. 33 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3 t}$ | 3. 92 | 3. 72 | 3. 39 | 4. 36 | 3. 69 | 4. 59 | 4. 58 | 4.11 | 3.00 |
| A/CNK | 1.1 | 1.1 | 1. 2 | 1.1 | 1.2 | 1.1 | 1.2 | 1.0 | 1.1 |
| DI | 81.1 | 81.1 | 83.2 | 80.4 | 81.0 | 79.5 | 83.7 | 81.0 | 84.7 |
| Ba | 599. | 416. | 397. |  |  |  |  |  |  |
| Rb | 170. | 223. | 277. | 257. | 217. | 238. | 241. | 253. | $300 .$ |
| Sr | 175. | 167. | 140. | 128. | 136. | 133. | 141. | 126. | 132. |
| Y | - | - | - | - | - | - | , | 12. | , |
| Zr | - | - | - | - | - | - | - | - | - |
| Nb | - | - | - | - | - | - | - | - | - |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | 43. | 48. | 55. | 46. | 49. | 41. | 47. | 49. | 52. |
| Ga | - | - | - | - | - | - | - | - | - |
| Zn | 88.0 | 85.0 | 80.0 | 73.0 | 77.0 | 98.0 | 90.0 | 91.0 | 78.0 |
| Cu | 13.0 | 5.0 | 1.0 | 2.0 | 1.0 | 2. 0 | 1.0 | 2. 0 | 3. 0 |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| V | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| co | - | - | - | - | - | - | - | - | - |
| Li | 68.0 | 124.0 | 174.0 | 162.0 | 131.0 | 119.0 | 109.0 | 105.0 | 123.0 |
| Be | - | - | - | - | - | - | - | - | - |
| 8 | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| Lo | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical databose.

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| :--- |
|  |
|  |
|  |
|  |
|  |
|  |

Figure D. 1 (cont.). Geochemical databose.

IBERIA

|  | IBM14 UNKN GRAN | IBM15 UNKN GRAN | IBM16 UNKN GRAN | IBM17 UNKN GRAN | IBM18 UNKN GRAN | IBM19 UNKN GRAN | IBM2 UNKN GRAN | IBM20 UNKN GRAN | I8M21 UNKN GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 70. 36 | 65.89 | ธ7. 26 | 67. 21 | 65. 59 | 86.27 | 68.98 | 65.19 | 65.49 |
| $\mathrm{TiO}_{2}$ | 45 | 1.02 | 65 | . 50 | . 99 | 58 | . 89 | . 65 | . 84 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 15.06 | 16. 59 | 14.72 | 15.17 | 12.83 | 12.50 | 14. 25 | 16. 85 | 16.95 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 3. 30 | 5. 39 | 2.15 | 1.83 | 3.34 | 2.69 | 1.91 | 2. 61 | 1. 90 |
| FeO | - | - | 3. 28 | 3.24 | 5. 01 | 4.12 | 3. 23 | 4. 50 | 3. 36 |
| Mno | . 04 | 07 | . 08 | . 07 | . 12 | 09 | . 05 | . 09 | . 10 |
| MgO | 1.04 | 2.02 | 2. 45 | 2.16 | 2.76 | 2.83 | 1.75 | 1.41 | 3.27 |
| CoO | 1.30 | . 98 | 1.76 | 2.72 | 1.10 | 1. 39 | 1.34 | 93 | 1.63 |
| $\mathrm{No}_{2} \mathrm{O}$ | 3.14 | 2. 50 | 2.35 | 2.42 | 2. 12 | 2. 68 | 2.86 | 2. 08 | 2.62 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4.66 | 4.19 | 3. 89 | 3. 12 | 3.12 | 3.17 | 4.59 | 3.08 | 3.10 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | 20 | . 29 | . 25 | . 41 | . 34 | . 25 | . 10 | . 29 | . 22 |
| $\mathrm{H}_{2} \mathrm{O}+$ | . 78 | 1.07 | 1.13 | 1.08 | 1. 38 | 2. 69 | 1. 29 | . 96 | . 83 |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | - . |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 100.33 | 100. 01 | 99.98 | 100. 03 | 100.00 | 99. 26 | 100.35 | 99.74 | 100. 31 |
| $\mathrm{Fe}_{2} \mathrm{O} 3 \mathrm{t}$ | 3. 30 | 5. 39 | 5. 79 | 5. 43 | 8. 90 | 7. 26 | 5. 50 | 7.72 | 5. 63 |
| A/CNK | 1. 2 | 1.6 | 1.3 | 1. 2 | 1.4 | 1.2 | 1.2 | 2. 0 | 1.6 |
| DI | 85.3 | 77.8 | 74.2 | 71.4 | 72.7 | 73.4 | 79.7 | 73.1 | 70.2 |
| Bo | 737. | 495. | 446. | 636. | 342. | 516. | 623. | 321. | 464. |
| Rb | 141. | 190. | 108. | 105. | 131. | 88. | 171. | 117. | 129. |
| Sr | 175. | 147. | 119. | 171. | 94. | 93. | 138. | 98. | 96. |
| Y | - | - | - | - | - | - | - | - | - |
| Zr | - | - | - | - | - | - | - | - | - |
| Nb | - | - | - | - | - | - | - | - | - |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | 56. | 62. | 42. | 60. | 35. | 60. | 44. | 43. | 37. |
| Go | - | - | - | - | - | - | - | - | - |
| Zn | 73.0 | 115.0 | 100.0 | 107.0 | 135.0 | 122.0 | 97. 0 | 119.0 | 132.0 |
| Cu | 20.0 | 23.0 | 17.0 | 27.0 | 45.0 | 15.0 | 22.0 | 31.0 | 39. 0 |
| Ni | - | - |  | 2 | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| V | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Se | - | - | - | - | - | - | - | - | - |
| Ta | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 64.0 | 120.0 | 94.0 | 79. 0 | 93.0 | 142.0 | 94.0 | 130.0 | 120.0 |
| Be | O. | 120 | 9. | 9.0 | - | 112. | O. | - | 120 |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| Lo | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.), Geochemical database.

BERIA

|  | IBM22 <br> UNKN <br> GRAN | IBM23 <br> UNKN <br> GRAN | IBM24 <br> UNKN GRAN | IBM25 <br> UNKN <br> GRAN | 18M26 <br> UNKN GRAN | IBM27 <br> UNKN GRAN | IEM28 <br> UNKN GRAN | IGM29 <br> UNKN <br> GRAN | IBM3 <br> UNKN <br> GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 66.70 | 86.97 | 66. 53 | 65.70 | 58.78 | 64.96 | 66.70 | 63. 35 | 67.31 |
| $\mathrm{TiO}_{2}$ | . 76 | . 85 | . 85 | . 43 | 1.67 | . 81 | . 76 | 1. 23 | . 59 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 15.92 | 15.70 | 15.35 | 16. 25 | 16. 68 | 15.85 | 15.79 | 17.52 | 14.03 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 1.88 | 1. 56 | 1.75 | 2. 60 | 7. 95 | 5.89 | 5. 21 | 6. 95 | 2. 00 |
| FeO | 2.73 | 2. 82 | 3. 15 | 3. 20 | - | - | - | - | 2. 97 |
| MnO | . 89 | . 09 | . 08 | . 09 | . 09 | . 09 | . 06 | . 09 | . 01 |
| MgO | 2. 83 | 2. 78 | 2. 42 | 1. 92 | 2.78 | 2. 53 | 1.90 | 2. 38 | 1. 51 |
| CaO | 1.29 | 1.82 | 1. 26 | 1. 30 | 4.44 | 1.90 | 1. 50 | . 86 | 1.51 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3. 11 | 3.11 | 2. 99 | 2. 38 | 3. 50 | 3. 33 | 3.20 | 2. 28 | 3.48 |
| $\mathrm{K}_{2} \mathrm{O}$ | 3.10 | 3. 61 | 3. 34 | 3.18 | 2. 38 | 2. 98 | 3. 51 | 3. 33 | 4.22 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 29 | . 18 | . 26 | . 21 | . 31 | . 33 | . 31 | . 26 | . 19 |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1.15 | . 86 | 1.76 | 2. 16 | 1.19 | 1.19 | 1.06 | 1.14 | 1. 56 |
| $\mathrm{H}_{2} \mathrm{O}-$ | 1.15 |  | 1.75 | 2. 16 | 1.1 | 1.19 | 1. | 1.14 | 1.56 |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | $\rightarrow$ | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 99.85 | 100.35 | 99.74 | 99. 42 | 99.78 | 99.96 | 100.03 | 99.40 | 99. 38 |
| $\mathrm{Fe}_{2} \mathrm{O} 3 \mathrm{t}$ | 4. 91 | 4. 69 | 5. 25 | 6.15 | 7.96 | 5.89 | 5.21 | 6. 95 | 5. 30 |
| A/CNK | 1.5 | 1.3 | 1.4 | 1.7 | 1.0 | 1.3 | 1.3 | 2. 0 | 1.1 |
| DI | 75.0 | 74.6 | 75.7 | 73. 7 | 60.9 | 73.8 | 78. 2 | 72.5 | 80.3 |
| Bo | 443. | 365. | 408. | 357. | 501. | 89. | $260 .$ | $415$ | $643 .$ |
| Rb | 163. | 128. | 136. | 120. | 111. | 149. | $152 .$ | 108. | 164. |
| Sr | 85. | 98. | 83. | 54. | 311. | 177. | 90. | 99. | 206. |
| $Y$ | - | - | - | - | - | - | - | - | - |
| Zr | - | - | - | - | - | - | - | - | - |
| Nb | - | - | - | - | - | - | - | - | - |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | 41. | 33. | 42. | 33. | 38. | 39. | 75. | 33. | 40. |
| Ga | , | - | - | - | - | - | - | - | - |
| Zn | 112.0 | 133.0 | 131.0 | 101.0 | 82.0 | 80.0 | 92.0 | 111.0 | 77.0 |
| Cu | 30.0 | 26. 0 | 43.0 | 35.0 | 16.0 | 20.0 | 20.0 | 36.0 | 11.0 |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| $\checkmark$ | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | _ | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| Ta | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 113.0 | 70.0 | 77.0 | 81.0 | 60.0 | 67.0 | 67.0 | 90.0 | 53.0 |
| Be | - | - | - | - | G. | - | \%. | - | - |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | _ | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | $\cdots$ |
| Lo | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical databose.

IBERIA

|  | IBM30 UNKN GRAN | IBM31 UNKN GRAN | IBM4 UNKN GRAN | IBM5 UNKN GRAN | IBM5 UNKN GRAN | IBM7 UNKN GRAN | IBM8 UNKN GRAN | IBM9 <br> UNKN <br> GRAN | IBP 1 UNKN GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 64.95 | 65.53 | 67.12 | 65.50 | 65.16 | 65. 22 | 65. 11 | 66. 97 | 74.15 |
| $\mathrm{TiO}_{2}$ | 1.27 | 1. 03 | 57 | . 46 | . 83 | . 34 | . 70 | . 73 | . 27 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 17.10 | 16. 55 | 14.88 | 15.03 | 15.76 | 17. 66 | 16.02 | 15. 28 | 14.57 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 5.73 | 4. 42 | 2.11 | 1.72 | 2.09 | 1. 43 | 1.95 | 2. 31 | 1. 00 |
| FeO | - | - | 3.11 | 3.10 | 2. 34 | 1.43 | 2. 97 | 3. 02 | - |
| MnO | . 08 | . 04 | . 08 | . 09 | . 12 | 07 | . 13 | . 10 | .01 |
| MgO | 1.97 | 1. 59 | 1.70 | 1.71 | 2. 23 | 2. 82 | 2. 28 | 2. 85 | . 53 |
| CoO | 1.42 | 2. 32 | 1. 26 | 2. 80 | 2. 25 | 1.75 | 1. 26 | 1. 68 | . 50 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 2. 59 | 3. 04 | 2. 34 | 2. 62 | 2. 87 | 2. 04 | 3. 01 | 1.91 | 2. 80 |
| $\mathrm{K}_{2} \mathrm{O}$ | 3. 76 | 4. 56 | 4.64 | 4. 31 | 4. 38 | 4. 37 | 3. 76 | 3. 61 | 4.36 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 33 | . 22 | . 26 | . 26 | . 22 | . 24 | . 16 | . 33 | . 23 |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1.36 | 1. 00 | 1. 43 | 1. 21 | 1. 30 | 2. 11 | 2. 55 | 1.00 | 1. 26 |
| $\mathrm{H}_{2} \mathrm{O}-$ |  |  | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 100.56 | 100.31 | 99. 50 | 99. 81 | 99. 55 | 99. 48 | 100. 20 | 99. 79 | 99. 68 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 5.73 | 4.42 | 5. 56 | 5.16 | 4. 69 | 3. 02 | 5. 25 | 5. 66 | 1.00 |
| A/CNK | 1. 6 | 1.2 | 1.3 | 1.1 | 1.2 | 1. 6 | 1.4 | 1. 5 | 1.4 |
| DI | 75. 6 | 77.1 | 78.2 | 72. 8 | 74.7 | 73. 5 | 76.2 | 72.2 | 90.6 |
| Bo | 531. | 629. | 529. |  |  |  |  |  | 285. |
| Rb | 132. | 159. | 137. | 155. | 175. | 169. | 188. | 212. | 310. |
| Sr | 150. | 139. | 154. | 115. | 140. | 72. | 57. | 172. | 95. |
| $Y$ | - | - | - | - | - | - | - | - | 10. |
| Zr | - | - | - | - | - | - | - | - | 110. |
| Nb | - | - | - | _ | - | - | - | - | 12. |
| Th | - | - | - | - | - | - | - | - | 22.00 |
| Pb | 36. | 39. | 47. | 51. | 44. | 37. | 35. | 33. | 48. |
| Ga | 3. | 3. | 17. | S. | 4. | 37. | 3. | 3 . | 1. |
| Zn | 11.0 | 196.0 | 117.0 | 105.0 | 110.0 | 80.0 | 81.0 | 90. 0 | 95.0 |
| Cu | 32.0 | 41.0 | 5.0 | 13.0 | 23.0 | 14.0 | 18.0 | 14.0 | 5.0 |
| Ni | - | - | . | - | - | - | - |  | - |
| $\mathrm{TiO}_{2}$ | - | - | - | $\rightarrow$ | - | - | - | - | _ |
| $V$ | - | - | - | $\sim$ | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | _ | - | - | - | - | - | - | 38.0 |
| Sc | - | - | - | - | - | - | - | - | - |
| Ta | - | - | - | - | - | - | - | _ | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 128.0 | 82.0 | 113.0 | 79.0 | 91.0 | 60. 0 | 85.0 | 139.0 | 20.0 |
| Be | - | - | - | - | - | - | - | - | 4.0 |
| B | - | - | - | - | - | - | - | - | , |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | 10.0 |
| Mo | - | - | - | - | - | - | - | _ | - |
| Lo | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical dotabose.

IBERIA

|  | 18P10 UNKN GRAN | \|BP11 UNKN GRAN | 1BP12 UNKN GRAN | IBP13 UNKN GRAN | IBP14 UNKN GRAN | IBP15 UNKN GRAN | IBP16 UNKN GRAN | IBP17 UNKN GRAN | IBP18 UNKN GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 73.15 | 73.13 | 73.30 | 72.30 | 73.50 | 74.15 | 73.50 | 74. 15 | 73.20 |
| $\mathrm{TiO}_{2}$ | . 20 | . 39 | . 20 | . 20 | . 24 | . 26 | 20 | . 24 | 31 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14.79 | 14.45 | 14.79 | 14.79 | 14.46 | 14.45 | 15.11 | 14.57 | 15.11 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 82 | 1.48 | . 82 | . 85 | . 92 | 1.00 | . 70 | . 82 | 1.00 |
| FeO | - | - | - | - | - | - | - | - | - |
| MnO | . 01 | . 02 | . 02 | . 01 | . 01 | . 01 | . 01 | 01 | . 01 |
| MgO | . 21 | . 61 | . 21 | . 25 | . 29 | . 40 | . 21 | 40 | 50 |
| CoO | . 36 | . 62 | . 36 | . 50 | . 55 | . 50 | . 43 | . 50 | . 50 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3.63 | 3. 30 | 3. 40 | 3. 30 | 3. 40 | 3.03 | 3.15 | 3.03 | 2.80 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4. 1 B | 4.52 | 5. 05 | 5.05 | 5. 05 | 4.52 | 4.35 | 4.52 | 4.52 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 19 | . 19 | . 18 | . 24 | . 20 | . 19 | . 19 | . 24 | . 22 |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1.51 | . 97 | 1.45 | 1. 95 | 1.37 | 1. 20 | 1.43 | 1.37 | 1.14 |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| total | 99.05 | 99. 69 | 99.78 | 99. 45 | 99.99 | 99.70 | 99. 30 | 99.85 | 99.31 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | . 82 | 1.48 | . 82 | . 86 | . 92 | 1.00 | . 70 | . 82 | 1.00 |
| A/CNK | 1.3 | 1.3 | 1.3 | 1.3 | 1.2 | 1.3 | 1.4 | 1.4 | 1.5 |
| DI | 92.3 | 90.5 | 93. 6 | 92.6 | 93.2 | 91.5 | 91.6 | 91.8 | 89.9 |
| Ba | 160. | 250. | 65. | 230. | 115. | 170. | 90. | 240. | 285. |
| Rb | 310. | 300. | 405. | 305. | 355. | 325. | 385. | 325. | 340. |
| Sr | 60. | 80. | 50. | 80. | 95. | 60. | 60. | 80. | 80. |
| Y | 10. | 10. | 10. | 12. | 12. | 10. | 10. | 10. | 10. |
| Zr | 70. | 120. | 65. | 75. | 75. | 85. | 75. | 85. | 115. |
| Nb | 12. | 16. | 16. | 15. | 14. | 15. | 21. | 16. | 17. |
| Th | 20.00 | 20.00 | 20.00 | 22.00 | 22.00 | 20.00 | 22.00 | 1.00 | 20.00 |
| Pb | 35. | 35. | 29. | 29. | 41. | 41. | 24. | 24. | 29. |
| Go | 1. | 1. | 1. | 1. | 1. | 20. | 1. | 1. | 1. |
| Zn | 67.0 | 83.0 | 45.0 | 75. 0 | 75.0 | 71.0 | 75. 0 | 86. 0 | 85.0 |
| Cu | - | - | - | - | - | - | - | - | - |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| V | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | 38.0 | 38.0 | 1.0 | 1.0 | 1.0 | 38.0 | 1.0 | 1.0 | 45.0 |
| Sc | - | - | - | - | - | - | - | - | - |
| Ta | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 10.0 | 46.0 | 20.0 | 25.0 | 44.0 | 34.0 | 15.0 | 22.0 | 29.0 |
| Be | 6.0 | 4.0 | 6. 0 | 4. 0 | 4. 0 | 3. 0 | 4. 0 | 3. 0 | 4. 0 |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | 10.0 | 13.0 | 15.0 | 10.0 | 10.0 | 1.0 | 10.0 | 10.0 | 10.0 |
| Mo | - | - | 15.0 | - | 10. | . | 10.0 | 10.0 | - |
| Lo | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.

IBERIA

|  | IBP51 UNKN GRAN | 18P6 UNKN GRAN | IEP7 UNKN GRAN | 18P8 UNKN GRAN | IBP9 UNKN GRAN | ICA 1 UNKN GRAN | ICA1O UNKN GRAN | ICA11 UNKN GRAN | ICA12 UNKN GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 71.80 | 73.00 | 73.80 | 73.15 | 71. 15 | 73. 52 | 72. 83 | 73.72 | 73. 96 |
| $\mathrm{TiO}_{2}$ | . 05 | . 24 | . 20 | . 24 | . 16 | . 14 | . 28 | . 25 | . 17 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 15.60 | 14.79 | 14.46 | 14.89 | 14.16 | 14.53 | 13.61 | 14.89 | 13.72 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . E ¢ | . 82 | . 82 | 1.00 | . 82 | . 51 | . 89 | . 79 | . 63 |
| FeO | - | - | - | - | - | 54 | . 97 | . 82 | 56 |
| MnO | . 89 | . 02 | . 01 | . 01 | . 01 | . 02 | . 02 | . 02 | . 02 |
| MgO | . 01 | . 29 | . 33 | . 25 | . 25 | . 25 | . 66 | . 45 | . 35 |
| CaO | . 43 | . 50 | . 62 | . 55 | . 43 | . 79 | . 77 | . 89 | . 70 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 4.42 | 2.80 | 3.16 | 3. 16 | 3. 40 | 3.51 | 3.11 | 3.19 | 3. 35 |
| $\mathrm{K}_{2} \mathrm{O}$ | 3. 45 | 5.50 | 4. 36 | 4. 36 | 4. 36 | 4. 67 | 4.82 | 5. 20 | 4. 96 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 32 | . 20 | . 17 | . 17 | . 19 | . 07 | . 01 | . 14 | . 21 |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1. 61 | 1.04 | 1. 32 | 1.33 | 1.09 | 1.00 | 1.55 | . 52 | 2. 21 |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | $\rightarrow$ | - | - | - | - | - |
| $F$ | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 99.45 | 99. 20 | 99.25 | 99.11 | 99. 32 | 99. 55 | 99. 52 | 100.08 | 100.95 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3 t}$ | . 66 | . 82 | . 82 | 1.00 | . 82 | 1.11 | 1.97 | 1.70 | 1. 36 |
| A/CNK | 1.4 | 1.3 | 1.3 | 1.1 | 1.3 | 1. 2 | 1.2 | 1.1 | 1.1 |
| 01 | 92.1 | 91.9 | 91,1 | 90.7 | 92,5 | 91.4 | 89.6 | 91.1 | 93. 6 |
| Ba | 80. | 230. | 250. | 250. | 205. | 210. | 408. | 262. | 168. |
| Rb | - | 405. | 320. | 310. | 395. | 147. | 310. | 402. | 265. |
| Sr | 195. | 75. | 65. | 80. | 55. | 71. | 108. | 61. | 48. |
| Y | 1. | 10. | 10. | 10. | 10. | - | - | - | - |
| Zr | 25. | 85. | 70. | 80. | 60. | - | - | - | - |
| Nb | 25. | 15. | 17. | 17. | 16. | - | - | - | - |
| Th | 20.00 | 20.00 | 20. 00 | 25.00 | 20.00 | - | - | - | - |
| Pb | 10. | 18. | 29. | 41. | 18. | - | - | - | - |
| Ga | 20. | 1. | 1. | 1. | 1. | - | - | - | - |
| Zn | 145.0 | 71.0 | 71.0 | 83. 0 | 67.0 | - | - | - | - |
| Cu | - | - | - | - | - | - | - | - | - |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| V | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | $\rightarrow$ | - | - | - |
| Cs | 150.0 | 45.0 | 45.0 | 38.0 | 58. 0 | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| Ta | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 275.0 | 25.0 | 12.0 | 51.0 | 35.0 | 46.0 | 60.0 | 106.0 | 84.0 |
| Be | 13. 0 | 4.0 | 4.0 | B. 0 | 6.0 | - | $\rightarrow$ | - | - |
| B | - | - | - | - | - | - | - | - | - |
| $F$ | - | - | - | - | - | - | - | - | _ |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | 124.0 | 13.0 | 15.0 | 10.0 | 25.0 | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| Lo | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.), Geochemical dotabase.

IBERIA

|  | ICA13 UNKN GRAN | ICA14 UNKN GRAN | ICA15 UNKN GRAN | ICA16 UNKN GRAN | ICA17 UNKN GRAN | ICA18 UNKN GRAN | ICA19 UNKN GRAN | ICA2 UNKN GRAN | ICA21 UNKN GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 72.99 | 73.05 | 72.37 | 72.16 | 71.05 | 75.45 | 76. 09 | 73. 14 | 75.93 |
| $\mathrm{TiO}_{2}$ | . 16 | . 05 | 15 | . 27 | . 41 | . 13 | . 07 | . 14 | . 03 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14.98 | 14.89 | 15.01 | 14.92 | 14.93 | 14.01 | 13.51 | 14.85 | 12.92 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 56 | 1. 35 | 1. 23 | 1.32 | 1. 07 | . 50 | . 58 | . 55 | . 34 |
| FeO | . 54 | . 20 | . 70 | . 89 | 1.14 | . 56 | . 56 | . 59 | . 40 |
| MnO | . 03 | . 03 | . 02 | . 03 | . 06 | . 04 | . 06 | . 02 | . 03 |
| Mgo | . 31 | . 29 | . 39 | . 50 | 1.87 | . 25 | . 34 | . 30 | 11 |
| CaO | . 42 | . 54 | . 55 | . 63 | . 79 | 47 | . 53 | . 76 | 29 |
| $\mathrm{No}_{2} \mathrm{O}$ | 2.53 | 3.18 | 2.76 | 2.88 | 2. 43 | 3. 14 | 3. 28 | 3. 25 | 3. 28 |
| $\mathrm{K}_{2} \mathrm{O}$ | 5.08 | 5.02 | 5.41 | 5. 01 | 5. 75 | 4.53 | 4.03 | 4.85 | 4.02 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 28 | . 21 | . 20 | . 21 | . 36 | . 25 | . 28 | . 07 | . 10 |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1.92 | . 87 | 1.17 | 1.16 | . 64 | 1.08 | . 51 | 1.23 | 2.07 |
| $\mathrm{H}_{2} \mathrm{O}-$ | 1. | - | 1. | . | - | 1. | - | . | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | _ | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 99.90 | 99.69 | 99.97 | 100.08 | 100. 53 | 100.31 | 99.92 | 99. 76 | 99.52 |
| $\mathrm{Fe}_{2} \mathrm{O} 3 \mathrm{t}$ | 1.27 | 1.57 | 2. 01 | 2. 31 | 2. 34 | 1.12 | 1.31 | 1. 20 | . 78 |
| A/CNK | 1.4 | 1.3 | 1. 3 | 1.3 | 1.3 | 1.3 | 1.3 | 1.2 | 1.3 |
| DI | 91.2 | 91.7 | 90.8 | 89.5 | 86.3 | 93.2 | 92.4 | 90.9 | 94.2 |
| Bo | 201. | 210. | 220. | 245. | 398. | 221. | 220. | 314. | 103. |
| Rb | 159. | 152. | 155. | 197. | 269. | 142. | 154. | 215. | 228. |
| Sr | 50. | 54. | 57. | 70. | 64. | 40. | 59. | 81. | 4. |
| Y | - | - | - | - | - | - | - | - | - |
| Zr | - | - | - | - | - | - | - | - | - |
| Nb | - | - | - | - | - | - | - | - | - |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Go | - | - | - | - | - | - | - | - | - |
| Zn | - | - | - | - | - | - | - | - | - |
| Cu | - | - | - | - | - | - | - | - | - |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| V | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 81.0 | 52.0 | 95.0 | 91.0 | 168.0 | 55.0 | 66.0 | 98.0 | 96.0 |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| Lo | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical datobase.

|  | IBERIA |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | ICA22 | ICA23 | ICA24 | ICA25 | ICA26 | ICA27 | ICA28 | ICA29 | ICA3 |
|  | UNKN | UNKN | UNKN | UNKN | UNKN | UNKN | UNKN | UNKN | UNKN |
|  | GRAN | GRAN | GRAN | GRAN | GRAN | GRAN | GRAN | GRAN | GRAN |
| $\mathrm{SiO}_{2}$ | 72.17 | 72. 21 | 75.02 | 74.72 | 70.90 | 74.67 | 72. 73 | 73.55 | 74.94 |
| $\mathrm{TiO}_{2}$ | . 24 | . 21 | . 01 | . 08 | . 35 | . 13 | . 19 | . 05 | . 04 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14.82 | 14.81 | 15.07 | 15.04 | 15.14 | 14.33 | 14.45 | 13.98 | 13.90 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 2. 13 | . 78 | . 24 | . 49 | 1. 53 | . 61 | 1. 00 | 1. 03 | . 44 |
| Feo | . 01 | . 88 | . 29 | . 54 | 1.84 | . 67 | . 65 | . 49 | .47 |
| MnO | . 02 | . 01 | . $\square 2$ | . 02 | . 17 | . 03 | . 06 | . 05 | . 01 |
| MgO | . 55 | . 67 | . 29 | . 21 | 1. 27 | . 23 | 1. 00 | . 42 | . 20 |
| COO | . 63 | . 76 | . 49 | .47 | . 61 | . 41 | . 65 | . 54 | . 75 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 2. 67 | 2. 76 | 3.04 | 2. 83 | 2. 72 | 3. 02 | 2. 84 | 2. 79 | 2.79 |
| $\mathrm{K}_{2} \mathrm{O}$ | 5.31 | 4.96 | 4.08 | 4. 96 | 3.24 | 4.79 | 5.13 | S. 41 | 5.39 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 20 | . 24 | . 29 | . 32 | . 10 | . 22 | . 16 | . 15 | . 04 |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1.62 | 1.26 | . 71 | . 67 | 1.77 | . 92 | 1. 15 | 1.54 | 1.03 |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | _ |
| F | - | - | - | - | - | - | _ | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 100.37 | 99. 55 | 99.55 | 100.35 | 99.54 | 100.03 | 100.01 | 100.01 | 100.00 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 2.14 | 1.76 | . 56 | 1.09 | 3. 57 | 1.35 | 1.72 | 1. 57 | .96 |
| A/CNK | 1. 3 | 1. 3 | 1.5 | 1.4 | 1.7 | 1. 3 | 1. 3 | 1.2 | 1. 2 |
| DI | 90.1 | 88.7 | 91.4 | 92. 5 | 82.6 | 92. 6 | 89.4 | 92.2 | 92.5 |
| Ba | 312. | $424 .$ | $178 .$ | 197. | 497. | 283. | 389. | 727. | 300. |
| Rb | 197. | $205 .$ | $194 .$ | 163. | $135$ | 341. | 283. | 208. | 164. |
| Sr | 80. | 97. | 43. | 50. | 117. | 24. | 52. | 200. | 71. |
| $Y$ | - | - | - | - |  | - | - | - | - |
| Zr | - | - | - | - | - | - | - | - | - |
| Nb | - | - | - | - | - | - | - | - | - |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Go | - | - | - | - | - | - | - | - | - |
| Zn | - | - | - | - | - | - | - | - | - |
| Cu | - | - | - | - | - | - | - | $\rightarrow$ | - |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| V | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| Ta | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 105.0 | 73. 0 | 40.0 | 39.0. | 68.0 | 141.0 | 107.0 | 80.0 | 34.0 |
| Be | - | - | - | - | - | 111.0 | - | - | - |
| 8 | - | - | - | - | - | - | - | - | - |
| $F$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | $\cdots$ | - | - | _ | - | - | - |
| Sn | - | - | - | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| Lo | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.

IBERIA

|  | ICAJO UNKN GRAN | ICA31 UNKN GRAN | ICA32 UNKN GRAN | ICA3J UNKN GRAN | ICA34 UNKN GRAN | ICA35 UNKN GRAN | ICA36 UNKN GRAN | ICA37 UNKN GRAN | ICA38 UNKN GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 73.26 | 73.41 | 73.68 | 71.62 | 73. 58 | 73.79 | 73.90 | 71.47 | 75.85 |
| $\mathrm{TiO}_{2}$ | . 11 | . 21 | . 07 | 22 | . 11 | . 19 | . 14 | . 47 | . 31 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14.50 | 13. 63 | 14.96 | 15.15 | 15.07 | 15.13 | 15.06 | 15.11 | 13.80 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 1.11 | 1.08 | . 42 | . 87 | . 45 | . 57 | . 36 | 1.18 | 1.09 |
| FeO | 07 | . 79 | . 46 | . 96 | . 48 | . 53 | . 42 | 1.17 | . 83 |
| MnO | . 05 | . 04 | . 02 | . 02 | . 03 | . 13 | 03 | . 05 | 03 |
| MgO | 34 | . 81 | . 30 | . 14 | . 26 | . 12 | . 31 | . 67 | . 47 |
| CoO | . 67 | . 60 | . 49 | . 60 | . 12 | . 35 | . 57 | . 65 | 54 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3. 22 | 2.84 | 3.10 | 3. 07 | 3.16 | 2. 97 | 2.58 | 2. 70 | 2. 78 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4.40 | 4.98 | 4.87 | 5.14 | 4.83 | 4.76 | 4.78 | 5.05 | 4. 90 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 11 | 21 | . 15 | . 25 | . 32 | 23 | . 31 | . 34 | . 22 |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1.63 | 1.35 | 1. 28 | 1. 28 | 1.61 | 1.08 | 1.65 | 1.19 | 1.00 |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - . |
| TOTAL | 99.47 | 99.93 | 99.80 | 99. 62 | 100.32 | 100.15 | 100.11 | 100.06 | 101.83 |
| $\mathrm{Fe}_{2} \mathrm{O} 3 \mathrm{t}$ | 1.19 | 1.96 | . 93 | 1.94 | . 98 | 1.27 | . 83 | 2. 48 | 2. 01 |
| A/CNK | 1.3 | 1.2 | 1.3 | 1.3 | 1.4 | 1.1 | 1.4 | 1.4 | 1.3 |
| D | 90.8 | 90.4 | 91.9 | 90.0 | 92.9 | 91.6 | 91.2 | 88.5 | 93.0 |
| Bo | 278. | 297. | 237. | 248. | 183. | 150. | 267. | 306. | 285. |
| Rb | 207. | 201. | 181. | 301. | 241. | 277. | 311. | 290. | 371. |
| Sr | 93. | 64. | 63. | 55. | 43. | 31. | 40. | 73. | 13. |
| Y | - | - | - | - | - | - | - | - | - |
| Zr | - | - | - | - | - | - | - | - | - |
| Nb | - | - | - | - | - | - | - | - | - |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Ga | - | - | - | - | - | - | - | - | - |
| Zn | - | - | - | - | - | - | - | - | - |
| Cu | - | - | - | - | - | - | - | - | - |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| $V$ | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 69.0 | 91.0 | 52. 0 | 128.0 | 105.0 | 281.0 | 167.0 | 128.0 | 130.0 |
| Be | - | - | - | , |  |  | 167.0 | - | , |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | _ | - | _ | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical dotabase.

|  | ICA39 <br> UNKN <br> GRAN | ICA4 <br> UNKN GRAN | ICA4O <br> UNKN <br> GRAN | ICA4 1 <br> UNKN <br> GRAN | ICA42 <br> UNKN <br> GRAN | ICA43 UNKN GRAN | ICA44 <br> UNKN <br> GRAN | ICA45 <br> UNKN <br> GRAN | ICA46 <br> UNKN <br> GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 74.06 | 72.96 | 74.72 | 74.60 | 73. 03 | 73.89 | 70.74 | 73.49 | 72.11 |
| $\mathrm{TiO}_{2}$ | . 21 | . 36 | . 12 | . 13 | . 18 | . 09 | . 45 | . 16 | . 19 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14. 27 | 14.41 | 13. 65 | 14.39 | 14.95 | 15.07 | 15.08 | 14.80 | 15.07 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 69 | 1.04 | 1. 05 | . 93 | . 93 | . 51 | 1.47 | . 71 | . 85 |
| FeO | . 72 | 1.14 | . 24 | . 17 | . 15 | . 65 | 1.93 | . 78 | . 91 |
| MnO | . 03 | . 02 | . 04 | . 05 | . 04 | . 02 | . 04 | . 02 | . 02 |
| MgO | . 40 | . 71 | . 45 | . 30 | . 29 | . 25 | . 89 | . 59 | . 41 |
| CaO | . 49 | 1.01 | . 38 | . 52 | . 44 | . 50 | . 85 | . 63 | . 88 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3.21 | 3. 05 | 2.88 | 3. 22 | 3.33 | 2. 89 | 2. 63 | 2.70 | 2. 91 |
| $\mathrm{K}_{2} \mathrm{O}$ | 5.04 | 4.87 | 4. 63 | 4.27 | 4. 90 | 5. 25 | 4. 95 | 5.17 | 5. 31 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 29 | . 22 | . 33 | . 18 | . 18 | . 18 | . 26 | . 25 | . 10 |
| $\mathrm{H}_{2} \mathrm{O}+$ | . 74 | . 86 | 1.69 | 1.33 | 1.78 | 1. 61 | . 98 | . 81 | 1.39 |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | , |
| $\mathrm{CO}_{2}$ | _ | - | - | - | - | - | - | - | _ |
| Cl | - | - | - | - | - | - | - | - | - |
| $F$ | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 100.15 | 100. 65 | 100.18 | 100. 09 | 100. 20 | 100. 22 | 100.25 | 100.11 | 100.15 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 1.49 | 2. 31 | 1. 32 | 1.12 | 1.10 | 1.34 | 3. 61 | 1.58 | 1.86 |
| A/CNK | 1.2 | 1.2 | 1. 3 | 1.3 | 1.3 | 1.3 | 1.3 | 1. 3 | 1. 2 |
| DI | 92.7 | 89.0 | 92.9 | 92. 2 | 92.8 | 91.9 | 85.7 | 90. 3 | 89.6 |
| 8a | 247. | 324. | $354 .$ |  |  | $226$ |  |  |  |
| Rb | 202. | 334. | 344. | 374. | $288 .$ | $437$ | 369. | 311. | $159 .$ |
| Sr | 45. | 101. | 40. | 50. | 45. | 48. | 124. | 76. | 72. |
| Y | - | , | , | - |  |  | , | - | - |
| Zr | - | - | - | - | - | _ | - | - | - |
| Nb | - | - | - | - | - | - | - | - | - |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Ga | - | - | - | - | - | - | - | - | - |
| Zn | - | - | - | - | - | - | - | - | - |
| Cu | - | - | - | - | - | - | - | - | - . |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| $\checkmark$ | - | - | - | - | - | - | - | _ | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | _ |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 164.0 | 110.0 | 36.0 | 93. 0 | 52. | 94.0 | 129.0 | 35.0 | 33.0 |
| Be | - | 110.D | - | - | , | -1. | 129.0 | , | , |
| B | - | - | - | - | - | - | - | - | - |
| $F$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | $-$ | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| Lo | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical databose.

IBERIA

|  | ICA47 UNKN GRAN | ICA48 UNKN GRAN | ICA5 UNKN GRAN | ICAG UNKN GRAN | ICA7 UNKN GRAN | ICAB UNKN GRAN | ICA9 UNKN GRAN | ICCO25 UNKN GRAN | ICC114 UNKN GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 75.58 | 72. 60 | 74.03 | 73.92 | 73.19 | 71.45 | 75.83 | 72. 46 | 73.79 |
| $\mathrm{TiO}_{2}$ | . 04 | . 31 | . 09 | . 11 | . 17 | . 35 | . 02 | . 15 | . 18 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 13.60 | 14.88 | 14. 72 | 14. 29 | 14. 29 | 14.70 | 13.63 | 14. 97 | 12. 59 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 39 | . 93 | . 11 | . 52 | . 60 | . 97 | . 41 | . 33 | 1.85 |
| FeO | . 44 | . 98 | . 44 | . 59 | . 68 | 1.06 | . 47 | - | - |
| MnO | . 09 | . 02 | . 02 | . 02 | . 02 | . 02 | . 02 | . 04 | . 01 |
| MgO | . 14 | . 39 | . 25 | . 34 | . 37 | . 70 | . 24 | . 25 | . 55 |
| CoO | . 65 | . 91 | . 72 | . 78 | . 73 | . 99 | . 70 | . 68 | . 95 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 2. 80 | 3. 21 | 3. 75 | 3. 41 | 2. 98 | 2. 73 | 3. 16 | 3. 35 | 4.31 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4.75 | 5. 11 | 4. 23 | 4.27 | 4.69 | 5.32 | 3. 98 | 4.75 | 3. 67 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 03 | . 12 | . 12 | . 08 | . 05 | . 09 | . 21 | . 40 | . 04 |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1.49 | . 89 | 1.12 | 1,31 | 1. 56 | 1. 36 | 1.61 | 1.42 | 1.70 |
| $\mathrm{H}_{2} \mathrm{O}-$ | 1. | , | 1.12 | 1.3 | 1.5 | 1. | 1. | 1.12 | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 100.00 | 100. 35 | 99.90 | 99. 64 | 99. 33 | 99. 74 | 100.58 | 98.80 | 99.74 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | . 88 | 2. 02 | . 90 | 1.17 | 1.35 | 2. 15 | . 93 | . 33 | 1. 85 |
| A/CNK | 1.2 | 1.2 | 1.2 | 1.2 | 1.3 | 1.2 | 1.2 | 1.3 | 1.0 |
| DI | 92.5 | 90.1 | 92. 1 | 90.8 | 90.1 | 87.8 | 93.3 | 91.8 | 91.7 |
| Ba | 284. | 341. | 106. | 179. | 158. | 352. | 90. | 285. | 350. |
| Rb | 141. | 278. | 397. | 323. | 290. | 289. | 271. | 336. | 174. |
| Sr | 41. | 49. | 40. | 73. | 55. | 86. | 48. | 106. | 182. |
| Y | - | - | - | - | - | - | - | - | - |
| Zr | - | - | - | - | - | - | - | - | - |
| No | - | - | - | - | - | - | - | - | - |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Go | - | - | - | - | - | - | - | - | - |
| Zn | - | - | - | - | - | - | - | - | - |
| Cu | - | - | - | - | - | - | - | - | - |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| V | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | _ | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| Ta | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 29.0 | 87.0 | 15.0 | 147.0 | 74.0 | 82.0 | 141.0 | 94.0 | 102.0 |
| Be | - | - | - | , | - | , | - | - | , |
| E | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | _ | - | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical databose.

BERIA

|  | ICC141 UNKN GRAN | ICC142 UNKN GRAN | ICC144 UNKN GRAN | ICC146 <br> UNKN <br> GRAN | ICC164 UNKN GRAN | ICC166 UNKN GRAN | ICC167 UNKN GRAN | ICC169 UNKN GRAN | ICC170 UNKN GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 71. 28 | 73.00 | 72.82 | 74. 23 | 72. 28 | 72.65 | 74.08 | 74.47 | 74.83 |
| $\mathrm{TiO}_{2}$ | . 36 | . 18 | . 28 | . 20 | . 26 | . 25 | 22 | . 22 | . 18 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 15.47 | 14.75 | 15.15 | 16. 51 | 15. 38 | 14.80 | 14.80 | 14. 54 | 14.95 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 2. 36 | 2.00 | 1.57 | 1.59 | 2.17 | 2. 31 | 1.08 | 1.04 | 1.02 |
| FeO | - | - | - | - | - | - | - | - | - |
| MnO | . 05 | . 03 | . 03 | . 03 | . 03 | . 04 | . 02 | 03 | . 05 |
| MgO | . 56 | . 32 | . 30 | . 27 | . 56 | . 60 | . 26 | . 28 | 20 |
| CaO | . 82 | 49 | 54 | . 33 | . 48 | . 48 | . 39 | . 41 | 49 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 2.45 | 2. 27 | 2. 51 | 1.68 | 2. 17 | 2.39 | 2. 48 | 2.89 | 3.43 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4.69 | 5. 66 | 5.23 | 2. 93 | 4.48 | 4.39 | 5. 33 | 4. 78 | 4.45 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 13 | . 13 | . 12 | . 01 | . 16 | . 21 | . 11 | . 06 | . 08 |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1.25 | 1. 33 | 1. 51 | 1. 86 | 1. 23 | 1.67 | 1.17 | 1. 30 | . 54 |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 99. 40 | 100.15 | 99.87 | 99. 64 | 99.48 | 99.79 | 99.94 | 100.02 | 100. 23 |
| $\mathrm{Fe}_{2} \mathrm{O} 3 \mathrm{t}$ | 2. 36 | 2.00 | 1.57 | 1. 59 | 2.17 | 2. 31 | 1.08 | 1.04 | 1.02 |
| A/CNK | 1.5 | 1.4 | 1.1 | 2.5 | 1.6 | 1.6 | 1.4 | 1.4 | 1.3 |
| DI | 86.4 | 90.8 | 90.2 | 85.3 | 87. 8 | 88.3 | 91. 9 | 92.1 | 92.6 |
| Ba | - | - | - | - | - | - | - | - | - |
| Rb | - | - | - | - | - | - | - | - | - |
| Sr | - | - | - | - | - | - | - | - | - |
| Y | - | - | - | - | - | - | - | - | - |
| Zr | - | - | - | - | - | - | - | - | - |
| Nb | - | - | - | - | - | - | - | - | - |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Ga | - | - | - | - | - | - | - | - | - |
| Zn | - | - | - | - | - | - | - | - | - |
| Cu | - | - | - | - | - | - | - | - | - |
| Ni | - | - | - | - | - | - | - | - | - |
| TiO2 | - | - | - | - | - | - | - | - | - |
| V | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| Ta | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| 4 | - | - | - | - | - | - | - | - | - |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| w | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical databose.

IBERIA

|  | $\operatorname{ICC172}$ UNKN GRAN | $\begin{aligned} & \text { ICC173 } \\ & \text { UNKN } \\ & \text { GRAN } \end{aligned}$ |  | $\begin{aligned} & \text { ICC175 } \\ & \text { UNKN } \\ & \text { GRAN } \end{aligned}$ | ICC190 <br> UNKN <br> GRAN | ICC191 <br> UNKN <br> GRAN | ICC199 UNKN GRAN | ICC200 <br> UNKN <br> GRAN | $\begin{aligned} & \text { ICC2O1 } \\ & \text { UNKN } \\ & \text { GRAN } \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 75.55 | 72. 94 | 73. 75 | 74.00 | 73. 12 | 72. 35 | 72. 05 | 73.16 | 74. 26 |
| $\mathrm{TiO}_{2}$ | . 11 | . 27 | . 15 | . 18 | 20 | . 25 | . 33 | . 24 | . 13 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 13.75 | 14. 56 | 15.49 | 13.69 | 15.42 | 15.03 | 15.32 | 14.75 | 14. 64 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 1.10 | 1.40 | 1. 25 | 1. 55 | 1.25 | 1.99 | 1.96 | 1.33 | 1.04 |
| FeO | - | - | - | - | - | - | - | - | - |
| MnO | . 04 | . 05 | . 04 | . 04 | . 04 | . 03 | . 01 | 05 | . 03 |
| MgO | .40 | . 34 | . 25 | . 37 | 24 | . 29 | . 36 | . 31 | . 22 |
| CaO | . 42 | . 21 | . 59 | . 72 | . 49 | . 40 | . 37 | . 38 | . 40 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 2. 48 | 2. 08 | 3.42 | 3.27 | 2. 28 | 2.08 | 2.46 | 2. 06 | 2. 84 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4.71 | 5. 66 | 4. 23 | 4.32 | 5.41 | 5. 90 | 5.74 | 5. 91 | 4.84 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 05 | . 09 | . 02 | . 08 | . 11 | . 09 | . 13 | . 11 | . 11 |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1. 39 | 1.84 | 1. 23 | 1.45 | 1.05 | 1.41 | 1.01 | 1.42 | 1.36 |
| $\mathrm{H}_{2} \mathrm{O}$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 100.00 | 99.54 | 100.43 | 99.67 | 99. 51 | 99. 82 | 99.74 | 99. 72 | 99.67 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 1.10 | 1.40 | 1.25 | 1. 55 | 1. 25 | 1.99 | 1.96 | 1. 33 | 1.04 |
| A/CNK | 1.4 | 1.5 | 1.4 | 1.2 | 1.5 | 1.4 | 1.4 | 1.4 | 1.4 |
| DI | 91.8 | 91. 2 | 91.1 | 91.1 | 90.3 | 90.4 | 90.5 | 91.2 | 91.8 |
| Bo | - | - | - | - | - | - | - | - | - |
| Rb | - | - | - | - | - | - | - | - | - |
| Sr | - | - | - | - | - | - | - | - | - |
| $Y$ | - | - | - | - | - | - | - | - | - |
| Zr | - | - | - | - | - | - | - | - | - - |
| Nb | - | - | - | - | - | - | - | - | - |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Go | - | - | - | - | - | - | - | - | - |
| Zn | - | - | - | - | - | - | - | - | - |
| Cu | - | - | - | - | - | - | - | - | - |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | _ | - | _ | - | - | - |
| $\checkmark$ | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | $\rightarrow$ |
| To | _ | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | - | - | - | - | - | - | - | - | - |
| Be | - | - | - | $\cdots$ | - | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | _ | - | _ | - | - | - | - |
| Cl | - | - | - | _ | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | $\sim$ | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.

IBERIA

|  | $\begin{aligned} & \text { ICC2O2 } \\ & \text { UNKN } \\ & \text { GRAN } \end{aligned}$ | $\begin{aligned} & \text { ICC203 } \\ & \text { UNKN } \\ & \text { GRAN } \end{aligned}$ | ICC206 <br> UNKN <br> GRAN | ICC208 <br> UNKN GRAN | ICC219 <br> UNKN <br> GRAN | $\begin{aligned} & \text { ICC224 } \\ & \text { UNKN } \\ & \text { GRAN } \end{aligned}$ | ICC308 <br> UNKN <br> GRAN | ICC309 <br> UNKN <br> GRAN |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 73.41 | 74.75 | 73.46 | 73.83 | 72. 66 | 74.03 | 73. 02 | 73.82 | 74.20 |
| $\mathrm{TiO}_{2}$ | . 37 | . 12 | . 11 | . 09 | . 49 | . 16 | . 25 | . 14 | . 09 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14.00 | 14.88 | 14.13 | 13.96 | 15.04 | 14.59 | 14.56 | 14.62 | 14.13 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 1. 93 | 1. 24 | 1.78 | 1.82 | 2. 09 | 1.77 | . 46 | 1.84 | 1. 61 |
| FeO | - | - | - | - | - | - | . 87 | - | - |
| MnO | . 02 | . 02 | . 02 | . 02 | . 02 | . 01 | . 04 | . 01 | . 02 |
| MgO | . 42 | . 18 | . 31 | . 28 | . 31 | . 22 | . 39 | . 38 | . 30 |
| CaO | . 31 | . 67 | . 52 | . 58 | . 89 | . 59 | . 92 | 51 | . 51 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 1. 51 | 2. 78 | 2. 89 | 2. 92 | 3. 36 | 2.80 | 3. 45 | 3. 16 | 2.83 |
| $\mathrm{K}_{2} \mathrm{O}$ | 6.03 | 4.95 | 5.09 | 4.88 | 3. 57 | 5. 13 | 4. 65 | 4. 77 | 4.90 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 13 | .01 | . 07 | . 04 | . 12 | . 01 | . 27 | . 03 | . 06 |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1. 65 | . 22 | 1.30 | 1. 29 | 1.23 | .41 | 1.00 | . 38 | 1. 41 |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | $\rightarrow$ | - |
| $F$ | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 99.78 | 99. 32 | 99. 68 | 99. 71 | 99. 78 | 99.72 | 99. 88 | 99. 65 | 100.06 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 1.93 | 1. 24 | 1.78 | 1.82 | 2.09 | 1.77 | 1. 43 | 1.84 | 1. 61 |
| A/CNK | 1.5 | 1. 3 | 1.3 | 1.3 | 1.1 | 1.3 | 1. 2 | 1.3 | 1.3 |
| DI | 90.5 | 90.9 | 91.5 | 91.3 | 88.0 | 90. 9 | 90. 8 | 90.9 | 91.8 |
| Bo | - | - | 222. | 232. | - | - | 350. | - | 303. |
| Rb | - | - | 439. | 435. | - | - | 320. | - | 322. |
| Sr | - | - | 55. | 39. | - | - | 85. | - | 84. |
| Y | - | - | - | . | - | _ | B. | _ | 8. |
| Zr | - | - | - | - | - | - | - | - | - |
| Nb | - | - | - | - | - | - | - | - | - |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | _ | - | - |
| Go | - | - | - | - | - | - | _ | _ | - |
| Zn | - | - | - | - | - | - | - | - | - |
| Cu | - | - | - | - | - | - | _ | - | - |
| Ni | - | - | - | - | - | _ | - | - | _ |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | _ | - | - |
| $V$ | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | _ | _ | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| Ta | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | _ | _ | - | - | - |
| Li | - | - | 142.0 | 113.0 | - | - | 119.0 | - | 87.0 |
| Be | - | - | 1 | 13. | - | - | 1 | - | - |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | _ | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | _ |
| Sn | - | - | - | - | _ | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical datobase.

## IBERIA

|  | ICC318 UNKN GRAN | ICC327 UNKN GRAN | ICC332 UNKN GRAN | ICC348 UNKN GRAN | ICC357 UNKN GRAN | ICC363 UNKN GRAN | ICC366 UNKN GRAN | ICC408 UNKN GRAN | ICC418 UNKN GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 74.49 | 72.58 | 74.02 | 75.00 | 72.75 | 73. 20 | 71.69 | 73.81 | 71.63 |
| $\mathrm{TiO}_{2}$ | . 08 | 24 | . 24 | 20 | . 12 | . 15 | 33 | . 23 | . 34 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14.47 | 14.84 | 12.78 | 12. 64 | 13.85 | 14.99 | 15.15 | 12.68 | 14. 54 |
| $\mathrm{Fe}_{2} \mathrm{O} 3$ | 1. 56 | 1.54 | 2.01 | 1.61 | 1.85 | 1.97 | 1.83 | 1.85 | 2.28 |
| FeO | 1. | . | 2. | , 6 | . | 1. | - | - | - |
| Mno | . 02 | . 04 | . 01 | . 01 | - | . 02 | . 01 | . 01 | . 01 |
| MgO | . 11 | . 39 | . 45 | . 32 | . 30 | . 39 | . 23 | . 53 | . 75 |
| CaO | . 48 | . 41 | . 73 | . 54 | . 73 | . 54 | . 44 | . 73 | . 68 |
| $\mathrm{No}_{2} \mathrm{O}$ | 4.41 | 3. 48 | 2.40 | 2.64 | 3.17 | 2. 98 | 3.03 | 3. 34 | 3.34 |
| $\mathrm{K}_{2} \mathrm{O}$ | 2.89 | 4. 74 | 4.91 | 5.27 | 5.54 | 5.00 | 5.80 | 4.58 | 4.19 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 05 | . 01 | . 15 | . 06 | . 12 | . 06 | . 08 | . 17 | . 05 |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1. 37 | 2.07 | 2.44 | 1.32 | 1.02 | . 48 | 1.11 | 1. 53 | 1.82 |
| $\mathrm{H}_{2} \mathrm{O}-$ |  |  |  |  | , | - | , | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| total | 99.73 | 100.34 | 100.15 | 99.61 | 99. 46 | 99.78 | 99.71 | 99.46 | 99.63 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 1.56 | 1.54 | 2.01 | 1. 61 | 1.85 | 1.97 | 1.83 | 1.85 | 2.28 |
| A/CNK | 1.3 | 1.3 | 1.2 | 1.2 | 1.1 | 1.3 | 1.3 | 1.1 | 1.3 |
| DI | 92.0 | 92.1 | 91.0 | 92.7 | 91.9 | 90.4 | 91.7 | 91.7 | 88.4 |
| Bo | - | 352. | - | 228. | 338. | - | - | 236. | 311. |
| Rb | - | 300. | - | 301. | 480. | - | - | 286. | 297. |
| Sr | - | 20. | - | 53. | 52. | - | - | 91. | 88. |
| Y | - | - | - | - | - | - | - | - | - |
| Zr | - | - | - | - | - | - | - | - | - |
| No | - | - | - | - | - | - | - | - | - |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Go | - | - | - | - | - | - | - | - | - |
| Zn | - | - | - | - | - | - | - | - | - |
| Cu | - | - | - | - | - | - | - | - | - |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| $\checkmark$ | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | - | 71.0 | - | 75.0 | 152.0 | - | - | 111.0 | 82. 0 |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical dotobase.

IBERIA

|  | ICC419 UNKN GRAN | ICC420 UNKN GRAN | ICC421 UNKN GRAN | ICC429 UNKN GRAN | ICC437 UNKN GRAN | ICC438 UNKN GRAN | ICC482 UNKN GRAN | ICC546 UNKN GRAN | ICC552 UNKN GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 75.12 | 71.80 | 74.08 | 74. 35 | 73. 39 | 74.19 | 72. 82 | 73. 27 | 73.30 |
| $\mathrm{TiO}_{2}$ | . 21 | . 27 | . 24 | . 24 | . 30 | . 11 | . 29 | . 40 | . 04 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 12.67 | 14.82 | 13.83 | 13.05 | 15.18 | 13.95 | 13.83 | 13. 67 | 13.54 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 1.77 | 1. 70 | 1. 20 | 1.29 | 1.83 | 2.05 | 2.15 | 2.05 | 1.83 |
| Fe 0 | - | - | - | - | - | - | - | - | - |
| MnO | . 01 | . 01 | . 03 | . 04 | . 04 | . 02 | . 01 | 01 | . 01 |
| MgO | . 31 | . 44 | . 11 | . 08 | . 56 | . 34 | . 45 | 52 | . 31 |
| CoO | . 41 | 1.18 | . 53 | . 72 | 72 | . 78 | . 42 | . 45 | . 41 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 2. 88 | 2. 28 | 2. 97 | 3. 39 | 3. 37 | 2. 76 | 2. 83 | 2. 33 | 3. 53 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4. 78 | 5. 29 | 5.28 | 4.70 | 3. 81 | 5.67 | 5. 34 | 5. 40 | 4. 95 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 07 | . 03 | . 01 | . 04 | . 02 | . 05 | . 09 | . 17 | . 13 |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1.37 | 1.80 | 2.10 | 1.93 | 1.45 | . 69 | 1.49 | 1.34 | 1.49 |
| $\mathrm{H}_{2} \mathrm{O}$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| total | 99.58 | 99.62 | 100.38 | 99.84 | 100.67 | 100.62 | 99. 73 | 99. 61 | 99. 54 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 1.77 | 1.70 | 1. 20 | 1. 29 | 1.83 | 2. 05 | 2.15 | 2.05 | 1.83 |
| A/CNK | 1. 2 | 1.3 | 1. 2 | 1.1 | 1.6 | 1.2 | 1.2 | 1.3 | 1.1 |
| DI | 92.8 | 87.3 | 93.6 | 93.4 | 89. 2 | \$1. 9 | 91.5 | 90. 6 | 93.1 |
| Ba | 342. | 299. | 314. | 214. | 310. | - | - | - | - |
| Rb | 391. | 303. | 307. | 321. | 253. | - | - | - | - |
| Sr | 70. | 54. | 92. | 54. | 77. | - | - | - | - |
| Y | - | - | , | . | 77. | - | - | - | - |
| Zr | - | - | - | - | - | - | - | - | - |
| Nb | - | - | - | - | - | - | - | - | - |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Go | - | - | - | - | - | - | - | - | - |
| Zn | - | - | - | - | - | - | - | - | - |
| Cu | - | - | - | - | - | - | - | - | - |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| $\checkmark$ | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | _ | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 150.0 | 71.0 | 90.0 | 107.0 | 81.0 | - | - | - | - |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | _ | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical databose.

IBERIA

|  | ICC554 UNKN GRAN | IGA071 UNKN GRAN | IGA187 UNKN GRAN | IGA188 UNKN GRAN | IGA189 UNKN GRAN | IGA204 UNKN GRAN | IGA205 UNKN GRAN | IGA231 UNKN GRAN | IGA232 <br> UNKN <br> GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 73.76 | 75. 25 | 75.25 | 73. 25 | 73. 25 | 15. 80 | 74. 25 | 71.83 | 72.59 |
| $\mathrm{TiO}_{2}$ | . 23 | . 33 | . 22 | 21 | . 39 | . 23 | . 22 | 33 | 24 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 13.58 | 14.36 | 14. 51 | 14.82 | 14. 35 | 14.39 | 14. 15 | 14.78 | 14.54 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 2.04 | . 55 | . 35 | . 32 | . 17 | . 52 | . 39 | 50 | . 57 |
| FeO | - | . 69 | . 59 | . 51 | . 85 | . 14 | . 59 | . 94 | . 83 |
| MnO | . 01 | . 03 | . 03 | . 03 | . 03 | . 03 | . 03 | . 02 | . 03 |
| MgO | . 38 | . 28 | . 04 | . 04 | . 40 | . 07 | . 10 | . 45 | 39 |
| CaO | . 33 | . 63 | 43 | . 54 | . 80 | . 48 | . 60 | . 78 | . 74 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 2.16 | 3. 33 | 3. 27 | 4.17 | 2.98 | 2.91 | 3.98 | 3.21 | 3.17 |
| $\mathrm{K}_{2} \mathrm{O}$ | 5.40 | 3. 41 | 4. 22 | 3.95 | 5. 66 | 4. 27 | 4.17 | 5. 33 | 5.04 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 12 | . 30 | . 39 | . 48 | . 25 | . 48 | . 43 | 38 | . 32 |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1.59 | . 47 | . 44 | 1.97 | . 54 | . 23 | . 98 | 1. 30 | 1.39 |
| $\mathrm{H}_{2} \mathrm{O}$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| total | 99.60 | 99.65 | 99.74 | 100.29 | 99. 69 | 99. 85 | 99.89 | 99. 85 | 99. 85 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 2.04 | 1.32 | 1.00 | . 89 | 1.12 | 1.01 | 1. 04 | 1. 54 | 1.49 |
| A/CNK | 1.4 | 1.4 | 1.1 | 1.2 | 1.2 | 1.4 | 1.2 | 1.2 | 1.2 |
| DI | 91.2 | 90.6 | 92.8 | 94.1 | 91.4 | 92.5 | 94.0 | 91.0 | 91.1 |
| Bo | - | 289. | 162. | 4. | 300. | 35. | 74. | 354. | 340. |
| Rb | - | 981. | 683. | 404. | 327. | 410. | 641. | 446. | 399. |
| Sr | - | 40. | 17. | 17. | 44. | 23. | 25. | 87. | 19. |
| Y | - | - | - | - | - | - | - | . | - |
| Zr | - | - | - | - | _ | - | - | - | - |
| Nb | - | - | - | - | - | - | - | - | - |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Ga | - | - | - | - | - | - | - | - | - |
| Zn | - | - | - | - | - | - | - | - | - |
| Cu | - | - | - | - | - | - | - | - | - |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| V | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| Ta | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | - | 136.0 | 480.0 | 218.0 | 68. 0 | 199.0 | 230.0 | 217.0 | 180.0 |
| Be | - | - | - |  | - | - | - | 217.0 | , |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | _ |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | _ | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical databose.

IBERIA

|  | IGA233 UNKN GRAN | IGA234 UNKN GRAN | IGA235 UNKN GRAN | IGA238 UNKN GRAN | IGA247 UNKN GRAN | IGA248 UNKN GRAN | IGA250 UNKN GRAN | IGA251 UNKN GRAN | IGA275 UNKN GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 73.43 | 71.18 | 72. 71 | 73. 68 | 72. 76 | 72.06 | 75. 11 | 71.98 | 73.15 |
| $\mathrm{TiO}_{2}$ | . 25 | . 31 | 41 | 15 | 36 | 11 | 04 | 28 | . 19 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14.39 | 14.39 | 14. 70 | 14. 05 | 14.44 | 14.39 | 14. 54 | 14.70 | 14.85 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 74 | . 64 | . 53 | . 54 | . 45 | . 63 | . 13 | . 31 | . 60 |
| FeO | . 56 | . 95 | 1.02 | . 44 | 1.12 | . 87 | . 47 | . 59 | . 92 |
| MnO | . 12 | . 02 | . 02 | 02 | . 02 | . 02 | . 07 | . 01 | . 02 |
| MgO | . 36 | . 78 | . 50 | . 32 | . 52 | . 52 | . 10 | . 53 | 40 |
| Coo | . 88 | . 87 | . 74 | . 60 | . 78 | . 74 | . 41 | 1.01 | 48 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3.31 | 3. 00 | 2.87 | 3.35 | 2.63 | 2.83 | 4. 23 | 3. 80 | 2.98 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4.89 | 5.19 | 5.11 | 4.62 | 5.20 | 5. 41 | 3.44 | 4.49 | 4.84 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 31 | . 39 | . 37 | . 34 | . 33 | . 33 | . 86 | . 18 | . 11 |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1.06 | 1.61 | . 89 | 2. 01 | . 99 | 1.40 | . 63 | 1.70 | 1.29 |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | 1. | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 100.00 | 99.33 | 99.87 | 100.12 | 99.61 | 99.61 | 100.09 | 99.56 | 99.84 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3 t}$ | 1.36 | 1.69 | 1.66 | 1.03 | 1. 70 | 1.80 | . 65 | . 96 | 1.62 |
| A/CNK | 1.2 | 1.2 | 1.3 | 1.2 | 1. 3 | 1.2 | 1.3 | 1.1 | 1.4 |
| DI | 92.2 | 89.1 | 90.1 | 93.2 | 89.4 | 90.5 | 93.8 | 90.2 | 90.5 |
| Ba | 276. | 373. | 275. | 429. | 100. | 261. | 64. | 224. | 214. |
| Rb | 405. | 368. | 364. | 375. | 311. | 328. | 922. | 291. | 427. |
| Sr | 64. | 92. | 100. | 82. | 90. | 97. | 57. | 149. | 48. |
| Y | - | - | - | - | - | - | - | - | - |
| Zr | - | - | - | - | - | - | - | - | - |
| Nb | - | - | - | - | - | - | - | - | - |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Go | - | - | - | - | - | - | - | - | - |
| Zn | - | - | - | - | - | - | - | - | - |
| Cu | - | - | - | - | - | - | - | - | - |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| V | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | _ | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 185.0 | 147.0 | 175.0 | 194.0 | 140.0 | 128.0 | 491.0 | 208.0 | 155.0 |
| Be |  | , | - | - | - | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | $\sim$ | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - ' |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.

IBERIA

|  | IGA279 UNKN GRAN | IGAJO6 UNKN GRAN | IGA315 <br> UNKN <br> GRAN | IGA324 <br> UNKN <br> GRAN | IGA325 UNKN GRAN | IGA326 UNKN GRAN | ILC 1 <br> UNKN <br> GRAN | ILC2 <br> UNKN <br> GRAN | ILC3 UNKN GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 71.81 | 70.91 | 73. 62 | 72.06 | 72. 55 | 72, 36 | 70.04 | 67. 94 | 70.20 |
| $\mathrm{TiO}_{2}$ | . 17 | . 31 | . 30 | 26 | . 37 | . 30 | . 70 | . 62 | . 59 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 15.65 | 14.95 | 13. 84 | 15.42 | 15.12 | 15. 13 | 13.01 | 14.68 | 13.70 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 41 | . 61 | . 73 | . 45 | . 62 | 52 | 4.83 | 4. 31 | 3.78 |
| FeO | 1. 03 | 1.19 | 1.03 | . 76 | 1.04 | 1. 21 | - | - | - |
| MnO | . 03 | . 04 | . 03 | . 03 | . 02 | . 03 | . 07 | 05 | . 05 |
| MgO | . 40 | . 59 | . 46 | . 28 | . 38 | . 35 | . 91 | . 84 | . 82 |
| CaO | . 62 | . 99 | . 92 | . 76 | . 82 | . 91 | 1.89 | 1. 99 | 1.62 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3. 06 | 3.10 | 2. 75 | 3.63 | 3. 37 | 2. 97 | 2. 96 | 3. 40 | 3.22 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4.93 | 4.82 | 4.88 | 4. 34 | 4. 59 | 4. 91 | 4. 57 | 5.30 | 5. 10 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 18 | . 26 | . 22 | . 27 | . 27 | . 25 | . 22 | . 22 | . 21 |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1.42 | 1.50 | . 99 | 1.20 | . 72 | . 99 | . 64 | . 60 | . 48 |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | $\rightarrow$ |
| $F$ | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 99.71 | 99. 28 | 99.77 | 99.46 | 99.87 | 99.83 | 99.84 | 99.95 | 99.77 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 1.55 | 1.93 | 1.87 | 1. 29 | 1.77 | 1.86 | 4.83 | 4. 31 | 3.78 |
| A/CNK | 1.4 | 1.2 | 1. 2 | 1.3 | 1. 3 | 1.3 | 1.0 | 1.0 | 1.0 |
| Di | 89.4 | 87. 6 | 89.5 | 90.2 | 89.7 | 88.9 | 83.1 | 83. 6 | 85. 6 |
| Ba | 251. | 357. | 679. | 312. | 275. | 262. | 320. | 550. | 340. |
| Rb | 386. | 460 | 221. | 335. | 328. | 380. | 230. | 230. | 240. |
| Sr | 52. | 90. | 67. | 49. | 36. | 27. | 90. | 120. | 90. |
| $Y$ | . | . |  | , | . | 27. |  | 1 | - |
| Zr | - | - | - | - | - | - | - | - |  |
| Nb | - | - | - | _ | - | - | - | - | - |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Go | - | - | - | - | - | - | - | - | - |
| Zn | - | _ | - | - | - | - | - | - | - |
| Cu | - | - | - | - | - | - | - | _ | _ |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | _ | - | - | - |
| $V$ | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| Ta | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 104.0 | 149.0 | 93. 0 | 170.0 | 167.0 | 161.0 | 180.0 | 150.0 | 190.0 |
| Be | - | - | - | 170.0 | 167.0 | - | 180. | - | - |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | _ | - | _ | - |
| Cl | - | - | - | - | - | _ | - | - | _ - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | _ |
| Mo | - | - | - | - | - | _ | - | - | - |
| La | $\rightarrow$ | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.

IBERIA

|  | ILC4 UNKN GRAN | ILC5 UNKN GRAN | IMB172 UNKN GRAN | IMB176 UNKN GRAN | IMB181 UNKN GRAN | IMC053 UNKN GRAN | IMCO59 UNKN GRAN | IMC06O UNKN GRAN | IMC293 UNKN GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 73.70 | 59. 30 | 75.28 | 75.72 | 73.58 | 70. 54 | ธ9. б5 | 73.29 | 71.87 |
| $\mathrm{TiO}_{2}$ | . 31 | . 85 | 20 | . 18 | . 33 | 47 | . 47 | . 02 | . 02 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 12.84 | 13. 20 | 14.05 | 13.97 | 14. 14 | 15. 71 | 15.67 | 14.45 | 14.95 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 2. 32 | 4. 37 | . 28 | . 29 | . 54 | . 40 | . 23 | . 35 | . 42 |
| FeO | - | - | . 44 | 50 | . 98 | 1.45 | 2.01 | 1.38 | 1.05 |
| MnO | . 04 | . 06 | . 10 | . 03 | . 05 | . 03 | . 03 | . 02 | 02 |
| MgO | . 45 | . 89 | . 02 | . 01 | . 31 | . 65 | . 80 | . 48 | 44 |
| CoO | 1. 12 | 1. 92 | . 11 | . 45 | . 90 | 1.02 | 1. 38 | . 85 | . 67 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3. 20 | 3. 47 | 4. 13 | 4. 06 | 3.76 | 3. 20 | 3.49 | 3.41 | 3. 30 |
| $\mathrm{K}_{2} \mathrm{O}$ | 5.08 | 5. 06 | 4.22 | 3. 84 | 4. 27 | 5. 29 | 4.73 | 5. 05 | 4.82 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 13 | . 25 | . 33 | . 25 | . 33 | . 09 | . 16 | . 22 | 20 |
| $\mathrm{H}_{2} \mathrm{O}+$ | . 59 | . 53 | . 24 | . 44 | . 32 | . 80 | 1.09 | 42 | 1.59 |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 99.79 | 99. 76 | 99.70 | 99.74 | 99. 81 | 99.65 | 99.71 | 99.94 | 99.36 |
| $\mathrm{Fe}_{2} \mathrm{O} 3 \mathrm{t}$ | 2. 32 | 4. 37 | . 77 | . 85 | 1.63 | 2.01 | 2. 46 | 1.88 | 1.59 |
| A/CNK | 1.0 | . 9 | 1.2 | 1.2 | 1.2 | 1.2 | 1.2 | 1.1 | 1.3 |
| DI | 90. 6 | 85.4 | 95.0 | 94.4 | 90.8 | 87.1 | 84.7 | 90. 3 | 89.8 |
| Ba | 180. | 450. | 35. | 35. | 408. | 497. | 538. | 457. | 296. |
| Rb | 250. | 230. | 787. | 683. | 286. | 144. | 218. | 255. | 302. |
| Sr | 60. | 110. | 13. | 15. | 57. | 89. | 122. | 79. | 76. |
| Y | - | - | - | - | - | - | - | - | - |
| Zr | - | - | - | - | - | - | - | - | - |
| Nb | - | - | - | - | - | - | - | - | - |
| Th. | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Ga | - | - | - | - | - | - | - | - | - |
| Zn | - | - | - | - | - | - | - | - | - |
| Cu | - | - | - | - | - | - | - | - |  |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| V | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - |  |
| Ta | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 180.0 | 180.0 | 197.0 | 192.0 | 239.0 | 94.0 | 84. 0 | 128.0 | 103.0 |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - |  |
| Cl | - | - | - | - | - | - | - |  |  |
| U | - | - | - | - | - | - | - | - |  |
| w | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| Lo | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.

IBERIA

|  | IMC294 <br> UNKN <br> GRAN | IMC295 UNKN GRAN | IMC296 <br> UNKN GRAN | IMCJO1 UNKN GRAN | IMC3O3 <br> UNKN <br> GRAN | IMC3O4 UNKN GRAN | IMC3O5 UNKN GRAN | IMM1 UNKN GRAN | IMM1O UNKN GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 72.78 | 75.71 | 70.87 | 72.00 | 72. 81 | 71.06 | 74. 34 | 75.30 | 74.10 |
| $\mathrm{TiO}_{2}$ | . 16 | . 02 | . 21 | . 25 | . 04 | . 07 | . 07 | . 07 | . 07 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14.36 | 13. 63 | 16.14 | 14.60 | 14.78 | 14.96 | 11.03 | 14.25 | 14.55 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 53 | . 16 | . 39 | . 57 | . 35 | . 47 | . 37 | . 40 | 1. 00 |
| FeO | 1. 04 | . 38 | 1.08 | 1.00 | 90 | 1.00 | . 39 | - | - |
| MnO | . 02 | . 03 | . 03 | . 05 | . 04 | . 04 | . 02 | . 09 | . $\square 8$ |
| MgO | . 48 | .10 | . 53 | . 17 | . 35 | . 48 | . 17 | .12 | . 27 |
| CoO | . 64 | . 30 | . 70 | . 51 | . 51 | 1.10 | . 46 | .14 | . 17 |
| $\mathrm{No}_{2} \mathrm{O}$ | 3. 20 | 3.85 | 3. 26 | 3. 16 | 3. 50 | 3.62 | 3. 44 | 4. 60 | 3. 05 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4.77 | 3. 84 | 5. 08 | 5.04 | 4.42 | 4. 13 | 4.56 | 3. 68 | 5. 36 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 24 | . 20 | . 31 | . 28 | . 31 | . 22 | . 27 | - | - |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1. 52 | 1.81 | 1.19 | 1. 66 | 1. 55 | 1.16 | 1. 39 | . 04 | . 06 |
| $\mathrm{H}_{2} \mathrm{O}$ |  | - |  | - |  | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | $\cdots$ | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 99.74 | 100.03 | 99.79 | 99. 59 | 99. 56 | 98. 31 | 99. 51 | 98.69 | 98.81 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 1.88 | . 58 | 1.59 | 1.68 | 1.35 | 1. 58 | . 80 | .40 | 1.00 |
| A/CNK | 1.2 | 1.2 | 1.3 | 1.3 | 1.3 | 1.2 | 1.2 | 1.2 | 1.3 |
| DI | 90.6 | 95.1 | 89.1 | 90.9 | 91.3 | 87.2 | 93. 6 | 94.7 | 92. 6 |
| Bo | 222. | 72. | 439. | 346. | 394. | 581. | 143. | 140. | 40. |
| Rb | 328. | 276. | 324. | 316. | 309. | 309. | 379. | 273. | 455. |
| Sr | 81. | 43. | 124. | 90. | 100. | 190. | 57. | 34. | 42. |
| Y | - | - | - | - | - | - | - | 10. | 10. |
| Zr | - | - | - | - | - | - | - | 30. | 25. |
| Nb | - | - | - | - | - | - | - | 25. | 10. |
| Th | - | - | - | - | - | - | - | 20.00 | 20. 00 |
| Pb | - | - | - | - | - | - | - | 32. | 30. |
| Ga | - | - | - | - | - | - | - | - | - |
| Zn | - | - | - | - | - | - | _ | 22.0 | 55.0 |
| Cu | - | - | - | - | - | - | - | - | - |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| $V$ | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | 50.0 | 35.0 |
| Sc | - | - | - | - | - | - | - | - | - |
| Ta | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 67.0 | 227.0 | 196.0 | 112.0 | 140.0 | 182.0 | 90.0 | 5. 0 | 13.0 |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | - | - | - | - | - | $\rightarrow$ | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | $\rightarrow$ | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | 97.0 | 40.0 |
| Mo | - | - | - | - | - | - | - | 10.00 | 11.00 |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.

IBERIA

|  | IMM1 1 UNKN GRAN | MM12 UNKN GRAN | IMM13 UNKN GRAN | IMM14 UNKN GRAN | IMM15 UNKN GRAN | IMM16 UNKN GRAN | \|MM 17 UNKN GRAN | IMM18 UNKN GRAN | IMM19 UNKN GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 73.90 | 72.10 | 73.00 | 73. 35 | 74.90 | 71.95 | 73. 85 | 61.20 | 58.80 |
| $\mathrm{TiO}_{2}$ | . 03 | . 03 | . 03 | . 11 | . 03 | . 14 | . 07 | . 90 | . 06 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14. 25 | 14.80 | 14.80 | 15.80 | 15.30 | 15. 80 | 15.70 | 18.40 | 20. 50 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 70 | 1.10 | . 90 | . 85 | . 70 | 1.10 | .40 | 6. 50 | 6. 30 |
| FeO | - | - | - | - | - | - | - | - | - |
| MnO | . 09 | . 20 | . 11 | . 10 | . 08 | . 04 | . 05 | . 07 | . 06 |
| MgO | . 17 | . 31 | . 25 | . 23 | . 21 | . 28 | . 13 | 3. 25 | 3.40 |
| CoO | .34 | . 34 | . 20 | . 42 | . 42 | . 44 | . 39 | . 75 | . 43 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 5. 60 | 4.80 | 6. 80 | 3. 48 | 4. 65 | 3. 52 | 5.60 | . 82 | . 50 |
| $\mathrm{K}_{2} \mathrm{O}$ | 2. 58 | 4.76 | 2.18 | 4.15 | 3. 92 | 4. 65 | 2. 06 | 3.85 | 4. 23 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | - | - | - | - | - | - | - | . 25 | . 15 |
| $\mathrm{H}_{2} \mathrm{O}+$ | . 05 | . 04 | . 03 | . 02 | . 04 | . 02 | . 06 | 4.23 | 4. 76 |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 97.71 | 98.48 | 98. 60 | 98. 51 | 100. 25 | 97.96 | 98.31 | 100. 02 | 99.19 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | .70 | 1.10 | .90 | . 85 | .70 | 1.10 | .40 | 6. 50 | 6. 30 |
| A/CNK | 1.1 | 1.1 | 1.1 | 1.4 | 1. 2 | 1.4 | 1.3 | 2.8 | 3.3 |
| DI | 93.1 | 93. 6 | 94.8 | 90.0 | 94.2 | 89.6 | 92.0 | 68.1 | 66.6 |
| Bo | 25. | 25. | 40. | 25. | 110. | 75. | 70. | 750. | 740. |
| Rb | 195. | 380. | 215. | 320. | 272. | 312. | 282. | 170. | 170. |
| Sr | 28. | 22. | 47. | 34. | 51. | 28. | 70. | 125. | 50. |
| $Y$ | 10. | 10. | 10. | 10. | 10. | 10. | 10. | 24. | 25. |
| Zr | 48. | 30. | 60. | 27. | 25. | 40. | 45. | 158. | 160. |
| Nb | 10. | 10. | 18. | 10. | 10. | 13. | 35. | 18. | 10. |
| Th | 30.00 | 20.00 | 30.00 | 20.00 | 20.00 | 20.00 | 20.00 | 23. 00 | 20.00 |
| Pb | 36. | 30. | 28. | 28. | 28. | 40. | 32. |  | , |
| Ga | - | - | - | - | - | - | - | - | - |
| Zn | 44.0 | 44.0 | 19.0 | 52. 0 | 33.0 | 55.0 | 22.0 | 116.0 | 125. D |
| Cu | - | - | - | - | - | - | - | - | - |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| $V$ | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | _ | - | - | - | - | - |
| Cs | 50.0 | 85.0 | 80.0 | 50.0 | 60.0 | 25.0 | 50.0 | 50.0 | 43.0 |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | _ | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 7. 0 | 7.0 | 7.0 | 7. 0 | 6.0 | 19.0 | 4.0 | 110.0 | 90.0 |
| Be | - | - | - | - | - | - | - | 3.0 | 4.0 |
| 8 | - | - | - | - | - | _ | - | - | - |
| $F$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | _ | _ | _ | - | - | _ | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | 10.0 | 10.0 | 48.0 | 10.0 | 10.0 | 10.0 | 125.0 | 10.0 | 10.0 |
| Mo | 13.00 | 11.00 | 11.00 | 10.00 | 13.00 | 10.00 | 10.00 | 5.00 | 5.00 |
| La Ce | - | - | - | - | $\xrightarrow{-}$ | - | - | - | - |

Figure D. 1 (cont.). Geochemical dotabase.

|  | IBERIA |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | IMM2 <br> UNKN GRAN | IMM2O UNKN GRAN | IMM21 <br> UNKN GRAN | IMM22 UNKN GRAN | IMM23 <br> UNKN <br> GRAN | IMM24 UNKN GRAN | IMM25 <br> UNKN GRAN | IMM3 <br> UNKN GRAN | IMM4 <br> UNKN GRAN |
| $\mathrm{SiO}_{2}$ | 72.85 | 62.90 | 59.70 | 59.20 | 50.50 | 58.60 | 52.50 | 73. 60 | 75.00 |
| $\mathrm{TH}_{2}$ | . 07 | 1.10 | . 80 | 1.10 | . 85 | . 80 | . 75 | . 05 | . 03 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14.75 | 17.10 | 19.30 | 17.50 | 16.90 | 20. 30 | 17.50 | 14.75 | 14. 25 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 70 | 6.20 | 6. 30 | 7.20 | 6.90 | 6. 30 | 5. 80 | . 50 | . 50 |
| FeO | - | - | - | - | - | - | $\rightarrow$ | - | - |
| MnO | . 02 | . 04 | . 06 | . 08 | . 09 | . 07 | .10 | .09 | . 04 |
| MgO | . 27 | 3.00 | 3. 21 | 3.75 | 3. 15 | 3.25 | 2. 16 | . 12 | . 20 |
| CaO | . 32 | . 34 | . 45 | . 69 | . 81 | . 45 | . 60 | . 17 | . 45 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 5.40 | . 33 | . 34 | . 20 | . 82 | . 33 | . 24 | 6. 50 | 5.00 |
| $\mathrm{K}_{2} \mathrm{O}$ | 3. 30 | 4. 34 | 4. 22 | 3.40 | 4.32 | 4. 33 | 5. 20 | 1.73 | 3. 68 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | - | . 10 | . 80 | . 24 | . 18 | . 10 | . 15 | - | - |
| $\mathrm{H}_{2} \mathrm{O}+$ | . 02 | 4. 56 | 5.49 | 6.03 | 5.05 | 5. 25 | 4.98 | . 03 | . 05 |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | _ | $\rightarrow$ | _ | - | - | - |
| $F$ | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 97.70 | 100.01 | 100. 57 | 99.40 | 99.97 | 99.78 | 99. 98 | 97. 54 | 99. 21 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3 t}$ | . 70 | 6.20 | 6.30 | 7. 20 | 6.90 | 6. 30 | 5.80 | . 50 | . 50 |
| A/CNK | 1.1 | 2.9 | 3. 3 | 3. 3 | 2.2 | 3.4 | 2. 5 | 1.2 | 1.1 |
| DI | 92.9 | 71.1 | 68. 3 | 64.3 | 69.1 | 66.5 | 73.5 | 93.9 | 94.7 |
| Ba | 240. | 690. | 740 | 400. | 850. | 750. | 750. | $25 .$ | $25 .$ |
| Rb | 165. | 250. | 150. | 150. | 165. | 42. | 85. | 205. | $305 .$ |
| Sr | 112. | 30. | 60. | 42. | 98. | 20. | 30. | 25. | 53. |
| Y | 10. | 26. | 25. | 18. | 20. | 152. | 160. | 10. | 10. |
| Zr | 25. | 178. | 150. | 175. | 150. | 16. | 14. | 50. | 25. |
| Nb | 10. | 12. | 18. | 18. | 10. | 10. | 10. | 18. | 10. |
| Th | 25.00 | 20.00 | 24.00 | 20.00 | 20. 00 | 20. 00 | 23.00 | 25.00 | 20.00 |
| Pb | 34. | - | - | , | - | - | 340. | 28. | 30. |
| Go | - | - | - | - | - | - | 31. | 28. | . |
| Zn | 29.0 | 96.0 | 131.0 | 160.0 | 160.0 | 124.0 | 200.0 | 44.0 | 14.0 |
| Cu | - | - | - | - | - | - | - | - | - |
| Ni | - | - | - | - | _ | - | _ | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| $\checkmark$ | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | _ | - |
| Hf | - | - | _ | - | - | - | - | - | - |
| Cs | 35.0 | - | - | - | - | - | - | 70.0 | 30.0 |
| Sc | , | - | _ | - | - | - | - | 70.0 | 30. |
| To | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 6.0 | 120.0 | 95.0 | 130.0 | 140.0 | 190.0 | 78.0 | 3. 0 | 4.0 |
| Be | - | 3.0 | 4. 0 | 5. 0 | 3.0 | 4. 0 | 6. 0 | , | - |
| B | - | - | - | - | - | - | - | - | - |
| $F$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| 5 n | 10.0 | 10.0 | 10.0 | 10.0 | 10.0 | 45.0 | - | 65.0 | 10.0 |
| Mo | 13.00 | - | - | - | - | - | - | 11.00 | 10.00 |
| Lo | - | - | - | - | - | - | - | - |  |
| Ce | - | - | - | - | - | - | - | - | - |

Figure 0.1 (cont.). Geochemical databose.

IBERIA

|  | IMM5 UNKN GRAN | IMM6 UNKN GRAN | IMM7 UNKN GRAN | IMM8 UNKN GRAN | IMM9 UNKN GRAN | 100241 UNKN GRAN | $10 C 256$ UNKN GRAN | $10 C 257$ <br> UNKN <br> GRAN | $10 C 259$ UNKN GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 75.10 | 75.10 | 75.10 | 75. 35 | 74.30 | 72. 75 | 73. 03 | 72.37 | 72.80 |
| $\mathrm{TiO}_{2}$ | . 03 | . 03 | . 03 | . 03 | . 03 | . 23 | . 32 | 33 | 30 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14. 25 | 13. 75 | 14.00 | 14.00 | 14. 25 | 14.81 | 14.70 | 14. 54 | 14.54 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 0 | . 50 | . 90 | . 50 | . 85 | . 47 | . 52 | . 35 | . 48 |
| FeO | - | - | - | - | - | . 50 | . 78 | . 99 | 79 |
| Mno | . 10 | . 02 | . 09 | . 07 | . 08 | . 01 | . 02 | . 01 | . 02 |
| Mgo | . 16 | . 16 | . 13 | . 14 | . 13 | . 31 | . 11 | . 13 | . 45 |
| CoO | . 44 | . 38 | . 22 | . 28 | . 30 | . 62 | . 58 | . 81 | . 68 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 5.15 | 4.95 | 4. 90 | 4. 65 | 5.75 | 3. 37 | 2. 77 | 3.12 | 3.44 |
| $\mathrm{K}_{2} \mathrm{O}$ | 2. 23 | 3.68 | 2. 73 | 3.46 | 3. 28 | 5.13 | 5. 30 | 5.19 | 4.93 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | - | - | - | - | - | . 36 | . 31 | . 33 | . 31 |
| $\mathrm{H}_{2} \mathrm{O}+$ | . 04 | . 05 | . 04 | . 05 | . 08 | 1. 36 | 1. 00 | 1. 26 | 1.05 |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | . | 1. | 1. | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| total | 98.10 | 98.60 | 98.14 | 98. 53 | 99. 03 | 99.92 | 99. 72 | 99.84 | 99.79 |
| $\mathrm{Fe}_{2} \mathrm{O} 3 \mathrm{t}$ | . 60 | . 50 | . 90 | 50 | . 85 | 1.03 | 1. 36 | 1.46 | 1.36 |
| A/CNK | 1.2 | 1.1 | 1.2 | 1.2 | 1.1 | 1.2 | 1.3 | 1.2 | 1. 2 |
| DI | 92.2 | 94.9 | 93.1 | 94.1 | 95.6 | 92.8 | 91.1 | 90. 7 | 91.6 |
| Ba | 70. | 30. | 40. | 30. | 60. | 369. | 280. | 233. | 250. |
| Rb | 183. | 280. | 268. | 257. | 323. | 395. | 551. | 456. | 412. |
| Sr | 62. | 40. | 15. | 31. | 28. | 36. | 125. | 221. | 215. |
| Y | 10. | 10. | 10. | 10. | 10. | - | - | - | - |
| Zr | 27. | 25. | 35. | 47. | 25. | - | - | - | - |
| Nb | 12. | 10. | 12. | 10. | 22. | - | - | - | - |
| Th | 20.80 | 20.00 | 20.00 | 25. 00 | 25.00 | - | - | - | - |
| Pb | 32. | 34. | 34. | 32. | 32. | - | - | - | - |
| Ga | - | - | - | - | - | - | - | - | - |
| Zn | 37.0 | 70.0 | 59.0 | 40.0 | 40.0 | - | - | - | - |
| Cu | - | - | - | - | - | - | - | - | - |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| $\checkmark$ | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | 85.0 | 70.0 | 80.0 | 30. 0 | 40.0 | - | - | - | - |
| Sc | , | , | , | - | . | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 8.0 | 12.0 | 5. 0 | 6. 0 | 5.0 | 268.0 | 294.0 | 184.0 | 141.0 |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | 10.0 | 10.0 | 10.0 | 10.0 | 10.0 | - | - | - | - |
| No | 11.00 | 10.00 | 15.00 | 15.00 | 12.00 | - | - | - | - |
| Lo | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical datobose.

IBERIA

|  | 10C262 <br> UNKN <br> GRAN | 10 C 263 UNKN GRAN |  | 10 C 267 <br> UNKN <br> GRAN | 10 C 269 <br> UNKN GRAN | 10C309 UNKN GRAN |  |  | ITR 1 <br> UNKN <br> GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 73.11 | 71.39 | 72. 81 | 73.17 | 71.92 | 78. 13 | 75. 00 | 73. 16 | 71.53 |
| $\mathrm{TO}_{2}$ | . 18 | . 29 | . 21 | . 26 | . 24 | . 08 | . 15 | . 32 | . 22 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14.70 | 15. 30 | 15.00 | 14. 23 | 15.15 | 12. 25 | 14.39 | 14. 67 | 15.93 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 62 | . 80 | . 32 | 54 | . 47 | . 11 | . 25 | . 53 | 1. 81 |
| FeO | . 34 | 1. 16 | . 76 | . 80 | . 50 | . 16 | . 36 | . 86 | - |
| MnO | . 03 | . 13 | . 02 | . 03 | . 01 | . 04 | . 04 | . 19 | . 03 |
| MgO | . 55 | . 73 | . 38 | . 55 | . 43 | . 11 | . 18 | . 44 | . 42 |
| CaO | 1.64 | 1.45 | . 72 | 1.04 | . 70 | . 55 | . 62 | . 76 | . 68 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 4. 60 | 4.45 | 3. 34 | 3. 97 | 3. 38 | 3.67 | 3.89 | 2.80 | 3. 81 |
| $\mathrm{K}_{2} \mathrm{O}$ | 2. 93 | 3. 13 | 5. 04 | 3. 91 | 5. 41 | 3. 99 | 4. 46 | 4.74 | 4.58 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 18 | . 16 | . 43 | . 19 | . 42 | . 09 | . 35 | . 35 | . 14 |
| $\mathrm{H}_{2} \mathrm{O}+$ | . 76 | 1.13 | 1.11 | . 99 | 1. 32 | . 85 | . 12 | 1. 09 | 1.14 |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 99. 64 | 100.02 | 100.14 | 99. 68 | 99,95 | 100.03 | 99.81 | 99. 91 | 100.39 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 1.00 | 2. 09 | 1.16 | 1. 43 | 1. 03 | .29 | . 65 | 1.48 | 1. 81 |
| A/CNK | 1.1 | 1.2 | 1.2 | 1.1 | 1.2 | 1.1 | 1.2 | 1. 3 | 1. 3 |
| DI | 88.2 | 86.4 | 91.9 | 89.9 | 92.3 | 95.8 | 94.0 | 89.7 | 90.7 |
| Ba | 400. | 630. | 224. | 875. | 294. | 400. | 315. | 288. | 186. |
| Rb | 152. | 186 , | 351. | 253. | 419. | 313. | 289. | 386. | 382. |
| Sr | 428. | 310. | 46. | 123. | 67. | 66. | 57. | 36. | 40. |
| Y | - | - | - | - | - | - | - | - | - |
| Zr | - | - | - | - | - | - | - | - | - |
| Nb | - | - | - | - | - | - | - | - | - |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - | 44. |
| Ga | - | - | - | - | - | - | - | - | - |
| Zn | - | - | - | - | - | - | - | - | 54.0 |
| Cu | - | - | - | - | - | - | - | - | 3.0 |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - . | - | - | - | - | - | - | - | - |
| V | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 135.0 | 151.0 | 148. 0 | 282.0 | 68. 0 | 209.0 | 279.0 | 163.0 | 270.0 |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | _ |
| Cl | - | - | - | - | _ | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.

IBERIA

|  | ITR10 UNKN GRAN | ITR11 UNKN GRAN | ITR12 UNKN GRAN | ITR13 UNKN GRAN | ITR14 UNKN GRAN | ITR15 UNKN GRAN | ITR16 UNKN GRAN | ITR17 UNKN GRAN | \|TR18 UNKN GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 73. 26 | 72.58 | 74. 32 | 74. 32 | 72. 39 | 72.58 | 72. 68 | 72.69 | 73.45 |
| $\mathrm{TiO}_{2}$ | . 19 | . 12 | . 09 | . 13 | 14 | . 17 | . 21 | . 19 | . 21 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14.51 | 15. 23 | 14. 36 | 14. 06 | 14.50 | 14. 36 | 14.64 | 14.40 | 14.14 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 1.25 | 1. 20 | . 23 | 1.23 | 1. 34 | 1. 45 | 1.39 | 1. 58 | 1.39 |
| FeO | - | - | - | - | - | - | - | - | - |
| MnO | . 02 | . 02 | . 01 | . 02 | . 01 | . 01 | . 02 | . 02 | . 02 |
| MgO | . 35 | . 25 | . 16 | . 31 | . 12 | . 39 | . 35 | . 34 | . 35 |
| CoO | . 52 | . 51 | . 11 | . 66 | . 51 | . 50 | . 62 | . 58 | . 56 |
| $\mathrm{No}_{2} \mathrm{O}$ | 3.75 | 3. 89 | 3. 77 | 4. 44 | 3.88 | 3. BS | 3. 95 | 3. 91 | 3.79 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4.38 | 4.32 | 4.27 | 3. 96 | 4. 86 | 4. $\mathrm{Bl}^{\text {a }}$ | 4.26 | 4. 55 | 4.57 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 12 | . 14 | . 09 | . 16 | . 13 | . 10 | . 17 | . 10 | . 12 |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1.38 | 1. 32 | 1. 44 | . 70 | 1. 35 | 1.35 | 1.23 | 1.34 | 1.43 |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| total | 99. 73 | 99. 58 | 99. 15 | 99. 99 | 99.63 | 99.67 | 99.52 | 99.70 | 100.04 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 1. 25 | 1.20 | . 23 | 1. 23 | 1.34 | 1. 45 | 1.39 | 1. 58 | 1.39 |
| A/CNK | 1.2 | 1.3 | 1.2 | 1.1 | 1.2 | 1.2 | 1.2 | 1.2 | 1. 2 |
| DI | 92.3 | 92.0 | 93.7 | 93.5 | 92.7 | 82. 8 | 91.8 | 92.3 | 92.9 |
| Bo | 151. | 172. | 103. | 242. | 194. | 203. | 133. | 156. | 174. |
| Rb | 388. | 334. | 336. | 283. | 265. | 337. | 266. | 317. | 219. |
| Sr | 34. | 39. | 28. | 50. | 40. | 33. | 31. | 32. | 37. |
| Y | - | - | - | - | - | - | - | - | - |
| Zr | - | - | - | - | - | - | - | - | - |
| Nb | - | - | - | - | - | - | - | - | - |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | 49. | 46. | 31. | 27. | 57. | 61. | 43. | 37. | 56. |
| Ga | , |  | , | 27. | - | - | - | \% | - |
| Zn | 67.0 | 69.0 | 41.0 | 38.0 | 58.0 | 41.0 | 61.0 | 70.0 | 66.0 |
| Cu | 3.0 | 7. 0 | 3.0 | 3.0 | 1.0 | 4.0 | 1.0 | 1.0 | 1.0 |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| V | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| Ta | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| 4 | 337.0 | 286.0 | 290.0 | 196.0 | 170.0 | 295.0 | 195.0 | 289.0 | 272.0 |
| Be | - |  | - |  | - | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| $\mathrm{Cl}^{\text {l }}$ | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.

IBERIA

|  | ITR19 UNKN GRAN | TR2 UNKN GRAN | ITR3 UNKN GRAN | ITR4 UNKN GRAN | ITR5 UNKN GRAN | TR6 UNKN GRAN | ITR7 UNKN GRAN | ITR8 UNKN GRAN | ITR9 UNKN GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 72. 36 | 72. 67 | 72. 31 | 72.15 | 72. 01 | 73.82 | 73.18 | 73.91 | 73. 33 |
| $\mathrm{TiO}_{2}$ | 21 | 30 | . 24 | 21 | . 19 | . 18 | . 19 | 08 | . 10 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 15. 21 | 14.49 | 14. 86 | 14.95 | 15.12 | 13.93 | 14.81 | 15. 33 | 14. 21 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 1.82 | 1. 91 | 1. 51 | 1. 25 | 1. 12 | 1. 39 | 1. 48 | . 85 | 1. 38 |
| FeO | - | - | - | - | - | - | - | - | - |
| MnO | . 04 | . 04 | . 02 | . 03 | . 03 | . 02 | . 02 | . 02 | . 01 |
| MgO | . 38 | . 44 | . 38 | . 35 | . 31 | . 31 | 37 | . 15 | . 38 |
| CoO | . 45 | . 80 | . 38 | . 73 | . 68 | . 54 | . 56 | . 53 | . 74 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 2. 44 | 3. 86 | 3. 58 | 3. 76 | 3. 88 | 3.81 | 3. 54 | 4.08 | 3.81 |
| $\mathrm{K}_{2} \mathrm{O}$ | 5.37 | 4.72 | 4. 70 | 4.51 | 4. 56 | 4.59 | 4.15 | 3. 98 | 4.34 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 12 | . 09 | . 13 | . 16 | . 08 | . 12 | . 11 | . 09 | . 18 |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1.51 | 1.06 | 1. 36 | 1.73 | 1. 46 | 1. 36 | 1. 23 | 1. 28 | 1.19 |
| $\mathrm{H}_{2} \mathrm{O}$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 99.91 | 100.38 | 99. 27 | 99.84 | 99. 74 | 99.87 | 99. 75 | 100.31 | 99.67 |
| $\mathrm{Fe}_{2} \mathrm{O} 3 \mathrm{t}$ | 1.82 | 1.91 | 1. 51 | 1. 26 | 1. 12 | 1. 39 | 1.48 | . 85 | 1.38 |
| A/CNK | 1.4 | 1.1 | 1.3 | 1. 2 | 1.2 | 1.1 | 1.3 | 1.3 | 1.2 |
| DI | 90.2 | 91.6 | 91.8 | 91.4 | 91.4 | 93.2 | 91.2 | 93.1 | 91.9 |
| Ba | 141. | 130. | 129. | 124. | 175. | 211. | 217. | 46. | 189. |
| Rb | 279. | 302. | 412. | 319. | 392. | 342. | 379. | 513. | 388. |
| Sr | 39. | 209. | 102. | 34. | 35. | 37. | 39. | 40. | 36. |
| Y | - | - | - | - | - | - | - | - | - |
| Zr | - | - | - | - | - | - | - | - | - |
| Nb | - | - | - | - | - | - | - | - | - |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | 51. | 64. | 61. | 31. | 40. | 24. | 17. | 34. | 27. |
| Go | S. | 8. | . | , | . | - | - | - | - |
| Zn | 62.0 | 45.0 | 52.0 | 56.0 | 60.0 | 81.0 | 86. 0 | 51.0 | 62.0 |
| Cu | 1.0 | 2. 0 | 1.0 | 1.0 | 1.0 | 2. 0 | 3. 0 | 1.0 | 1.0 |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| $\checkmark$ | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 291.0 | 293.0 | 319.0 | 353.0 | 346.0 | 336.0 | 278.0 | 525.0 | 306.0 |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | $\rightarrow$ | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.

IBERIA

|  | \|VA1 UNKN GRAN | IVA2 <br> UNKN <br> GRAN | NAS UNKN GRAN | IVA4 <br> UNKN <br> GRAN | NA5 UNKN GRAN | NA6 UNKN GRAN | IVA7 UNKN GRAN | IVA8 UNKN GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 68.38 | 71.25 | 71.58 | 72. 15 | 72. 50 | 72.80 | 74.10 | 74.49 |
| $\mathrm{TH}_{2}$ | . 68 | . 65 | . 43 | . 42 | . 43 | . 40 | . 35 | . 31 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 15.40 | 14.30 | 13.30 | 14.80 | 14.40 | 14.00 | 13.30 | 13.30 |
| $\mathrm{Fe}_{2} \mathrm{O} 3$ | - | - | - | - | - | - | - | - |
| FeO | 5. 20 | 4.50 | 2.60 | 2.50 | 2.50 | 2.50 | 2. 20 | 2. 30 |
| Mno | . 05 | . 04 | . 03 | . 02 | . 02 | . 02 | . 03 | . 03 |
| MgO | 1.70 | 1. 50 | . 69 | . 70 | . 50 | . 70 | 2.20 | 49 |
| Coo | . 85 | . 90 | . 83 | . 92 | . 90 | . 54 | . 80 | 66 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3.70 | 3. 30 | 3.60 | 3. 70 | 3. 70 | 3. 60 | 3. 70 | 3. 70 |
| $\mathrm{K}_{2} \mathrm{O}$ | 2. 50 | 2. 20 | 4. 20 | 4.00 | 3.90 | 3.70 | 4.00 | 3. 60 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 13 | . 25 | . 27 | . 33 | . 35 | . 27 | . 30 | . 28 |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1.94 | 1.50 | . 77 | 1.06 | 1.00 | 1. 30 | . 84 | 1.00 |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - |
| total | 100.53 | 100.50 | 98.30 | 100.60 | 100. 30 | 99.83 | 101.82 | 100.16 |
| $\mathrm{Fe}_{2} \mathrm{O} 3 \mathrm{t}$ | 5.77 | 5.11 | 2.89 | 2.78 | 2. 78 | 2.78 | 2. 44 | 2. 55 |
| A/CNK | 1.5 | 1.5 | 1.1 | 1.2 | 1. 2 | 1.3 | 1.1 | 1.2 |
| DI | 77.0 | 78.9 | 86.6 | 87.3 | 87.7 | 87.9 | 87.2 | 89.6 |
| Ba | 485. | 440. | 340. | 300. | 280. | 235. | 225. | 150. |
| Rb | 149. | 133. | 301. | 306. | 331. | 297. | 308. | 321. |
| Sr | 110. | 110. | 68. | 62. | 57. | 54. | 51. | 33. |
| Y | - | - | - | - | - | - | - | , |
| Zr | 54. | 62. | 115. | 114. | 115. | 125. | 103. | 108. |
| Nb | - | - | - | - | - | - | - | - |
| Th | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - |
| Go | - | - | - | - | - | - | - | - |
| Zn | 53.0 | 19.0 | 50.0 | 46.0 | 60.0 | 30.0 | 44.0 | 16.0 |
| Cu | 33.0 | 23. 0 | 9.0 | 10.0 | 9.0 | 12.0 | 7.0 | 11.0 |
| Ni | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - |
| $\checkmark$ | 96. | 78. | 25. | 24. | 23. | 24. | 19. | 19. |
| Cr | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - |
| Li | 115.0 | 87. 0 | 76.0 | 93. 0 | B1. 0 | 89.0 | 93.0 | 12.5 |
| Be | , |  | - |  |  | - | , | . |
| B | - | - | - | - | - | - | - | - |
| $\stackrel{F}{F}$ | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - |
| Lo | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.

## AUSTRALIA

|  | ABE1 1 UNKN GRAN |  | ABE19 UNKN GRAN | ABB20 UNKN GRAN | ABB3 UNKN GRAN |  |  | ABB38 UNKN GRAN | A8863 UNKN GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 70.89 | 67.37 | 67. 62 | 67.03 | 67.71 | 65.86 | 67.50 | 58.07 | 68.79 |
| $\mathrm{TiO}_{2}$ | . 35 | . 58 | . 55 | . 59 | . 58 | . 62 | . 50 | . 48 | . 50 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 13.92 | 14.91 | 14. 31 | 14.23 | 14. 29 | 14.91 | 14. 36 | 13. 53 | 13. 72 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 72 | 1. 40 | 1. 25 | 1. 57 | . 84 | . 95 | . 78 | , 59 | . 81 |
| FeO | 2. 26 | 2. 88 | 3. 06 | 3.23 | 3. 75 | 4. 35 | 3.77 | 3. 65 | 3. 35 |
| MnO | . 06 | . 07 | . 07 | . 08 | . 07 | . 09 | . 07 | . 07 | . 06 |
| MgO | 1.00 | 2. 12 | 2. 35 | 2. 25 | 2. 23 | 2. 64 | 2. 43 | 3.21 | 2. 55 |
| CaO | 2. 31 | 3. 15 | 2. 80 | 3. 21 | 2. 95 | 3.79 | 2. 98 | 3. 08 | 2. 45 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 2.56 | 2. 35 | 2. 11 | 1.95 | 2. 00 | 1.95 | 2.14 | 1.83 | 1. 83 |
| $\mathrm{K}_{2} \mathrm{O}$ | 3. 96 | 3. 32 | 3. 59 | 3. 40 | 3.41 | 3.07 | 3.40 | 3.27 | 3. 39 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . Dg | . 12 | . 13 | . 13 | .14 | .14 | . 12 | . 11 | . 10 |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1.39 | 1. 39 | 1.43 | 1,56 | 1.48 | 1.35 | 1. 45 | 1. 44 | 1. 83 |
| $\mathrm{H}_{2} \mathrm{O}-$ | . 15 | . 37 | . 32 | . 31 | . 21 | . 13 | . 13 | . 18 | . 25 |
| $\mathrm{CO}_{2}$ | . 29 | .14 | . 21 | . 14 | . 12 | . 09 | . 22 | . 21 | . 16 |
| Cl | - | - | - | - | - | - | - | - | - |
| $F$ | - | - | - | - | - | - | - | - | - |
| LOI | . 18 | . 18 | . 18 | .18 | .18 | . 18 | .17 | . 18 | .17 |
| TOTAL | 100.13 | 100. 35 | 99. 98 | 99.86 | 99.96 | 100.28 | 100.12 | 100.00 | 100.05 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 3.23 | 4.60 | 4.65 | 5.16 | 5.00 | 5. 91 | 4. 96 | 4.74 | 4.53 |
| A/CNK | 1.1 | 1.1 | 1. 2 | 1.1 | 1.2 | 1,1 | 1.1 | 1.1 | 1.2 |
| DI | 79.8 | 70.8 | 71.2 | 69.1 | 69.7 | 83. 8 | 69.4 | 67.4 | 71.2 |
| Bo | 625. | 455. | 485. | 455. | 470. | 455. | 465. | 430. | 460. |
| Rb | 178. | 169. | 179. | 173. | 166. | 151. | 155. | 162. | 153. |
| Sr | 198. | 144. | 130. | 132. | 134. | 151. | 133. | 130. | 119. |
| $Y$ | 40. | 31. | 28. | 28. | 29. | 27. | 28. | 27. | 29. |
| Zr | 136. | 179. | 162. | 171. | 169. | 155. | 175. | 127. | 155. |
| Nb | 11. | 11. | 11. | 12. | 11. | 11. | 12. | 12. | 11. |
| Th | 18.00 | 18.60 | 17.40 | 17.60 | 17.80 | 15.00 | 17.80 | 15.80 | 19.60 |
| Pb | 33. | 28. | 26. | 26. | 25. | 23. | 25. | 24. | 28. |
| Ga | 15. | 17. | 16. | 17. | 17. | 17. | 17. | 15. | 15. |
| Zn | 47.0 | 70.0 | 65.0 | 77.0 | 72.0 | 81.0 | 64.0 | 59.0 | 50.0 |
| Cu | 6. 0 | 11.5 | 16.0 | 13.0 | 16.0 | 8. 0 | 14. 5 | 37.0 | 12.0 |
| Ni | 5.5 | 17.0 | 18.0 | 17.5 | 20.5 | 18.0 | 19.0 | 32.5 | 25.5 |
| $\mathrm{TiO}_{2}$ | - | 1 | - | 1 |  | - | - | - | - |
| $\checkmark$ | 42. | 90. | 83. | 96. | 89. | 115. | 89. | 86. | 76. |
| Cr | 13. | 55. | 65. | 57. | 71. | 69. | 63. | 148. | 96. |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | 12.0 | 17.0 | 16.0 | 19.0 | 18.0 | 22.0 | 17.0 | 17.0 | 14.0 |
| To | - | - | - | 1.0 | 18.0 | 22.0 | 17.0 | 1.0 | 14.0 |
| Co | 8. | 13. | 14. | 16. | 17. | 18. | 15. | 21. | 15. |
| Li | - | - | - | - | - | - | - | - | - |
| Be | - | - | - | - | - | _ | - | - | - |
| 日 | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | 4. 20 | 3.20 | 4.20 | 3.60 | 2. 40 | 2. 60 | 3.00 | 3. 60 | 3.80 |
| W | - | - | - | 3. 6 | 2, |  | d | - | - |
| Sn | - | - | - | - | _ | _ | _ | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| Lo | 31. | 33. | 31. | 34. | 33. | 27. | 23. | 35. | 33. |
| Ce | 70. | 71. | 67. | 71. | 69. | 65. | 56. | 62. | 71. |

Figure 0.1 (cont.). Geochemical databose.

AUSTRALIA

|  | ABE64 UNKN GRAN | ABB66 UNKN GRAN | ABB67 UNKN GRAN | ABE68 UNKN UNKN | ABB70 UNKN GRAN | AB881 UNKN GRAN | ABB82 UNKN GRAN | ABB83 UNKN GRAN | ABB84 UNKN GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 65.81 | 68. 81 | 69. 45 | 71.09 | 67. 37 | 67. 22 | 67. 52 | 67.82 | 68.07 |
| $\mathrm{TiO}_{2}$ | . 52 | . 51 | . 44 | . 40 | . 59 | . 64 | . 54 | 51 | . 63 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14. 21 | 13.93 | 13.96 | 13. 66 | 14. 35 | 14.60 | 13.86 | 14.47 | 14.49 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 1. 22 | 1.19 | 1.13 | . 79 | . 84 | . 80 | . 61 | . 54 | . 58 |
| FeO | 3. 52 | 2.71 | 2. 24 | 2. 28 | 3. 67 | 3. 73 | 4.02 | 4.09 | 3. 82 |
| Mno | . 08 | . 06 | . 06 | . 06 | . 07 | 07 | . 07 | . 07 | . 07 |
| MgO | 3. 57 | 1. 55 | 1. 35 | 1. 27 | 2. 37 | 2. 17 | 3.10 | 2. 33 | 2.13 |
| CaO | 3. 72 | 3.02 | 2. 87 | 2. 31 | 2.88 | 2. 91 | 2. 94 | 3.05 | 2. 55 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 1.83 | 2. 21 | 2. 40 | 2. 38 | 1.99 | 2. 24 | 1.83 | 2. 06 | 2. 07 |
| $\mathrm{K}_{2} \mathrm{O}$ | 3.08 | 3.65 | 3. 69 | 4. 25 | 3. 64 | 3. 34 | 3. 13 | 3.40 | 3.30 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 11 | . 12 | . 12 | . 09 | . 13 | . 14 | . 13 | . 14 | . 15 |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1.57 | 1.56 | 1. 64 | 1.13 | 1.57 | 1.50 | 1. 56 | 1.13 | 1.41 |
| $\mathrm{H}_{2} \mathrm{O}-$ | . 24 | . 21 | . 22 | .16 | . 21 | . 26 | . 14 | . 19 | . 14 |
| $\mathrm{CO}_{2}$ | . 11 | . 09 | . 11 | . 07 | . 18 | . 10 | . 23 | . 18 | . 18 |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOl | . 18 | .17 | .16 | .17 | . 18 | .17 | . 18 | .18 | .17 |
| TOTAL | 99.87 | 99.80 | 99.84 | 100.11 | 100.04 | 99.89 | 99.85 | 100. 26 | 99. 78 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 5. 24 | 4. 20 | 3. 62 | 3. 32 | 1. 91 | 4.94 | 5.07 | 5.08 | 4.82 |
| A/CNK | 1. 1 | 1.1 | 1.1 | 1.1 | 1. 2 | 1. 2 | 1.2 | 1.2 | 1. 3 |
| DI | 63.1 | 73.7 | 76.0 | 79.4 | 69.9 | 69. 9 | 65.8 | 69.0 | 70.9 |
| Bo | 415. | 515. | 450. | 565. | 550. | 450. | 420. | 445. | 455. |
| Rb | 155. | 176. | 175. | 207. | 155. | 168. | 164. | 169. | 169. |
| Sr | 134. | 175. | 170. | 140. | 135. | 135. | 130. | 139. | 129. |
| Y | 27. | 31. | 33. | 33. | 28. | 28. | 28. | 28. | 28. |
| Zr | 143. | 158. | 151. | 148. | 158. | 173. | 150. | 174. | 177. |
| Nb | 10. | 11. | 11. | 10. | 11. | 11. | 11. | 12. | 12. |
| Th | 15.50 | 18.60 | 18. 60 | 20.00 | 19.40 | 18.00 | 17.40 | 17.20 | 18.20 |
| Pb | 21. | 25. | 27. | 32. | 17. | 26. | 21. | 25. | 25. |
| Go | 16. | 16. | 16. | 15. | 16. | 17. | 16. | 17. | 17. |
| Zn | 68.0 | 56.0 | 52.0 | 44.0 | 47.0 | 74.0 | 54.0 | 70.0 | 71.0 |
| Cu | 16.5 | 6.0 | 5.0 | 2.5 | 15.0 | 11.5 | 8.0 | 15.5 | 11.0 |
| Ni | 35.0 | 8.5 | 7.5 | 6. 5 | 19.5 | 15.5 | 29.0 | 19.5 | 17.5 |
| $\mathrm{TiO}_{2}$ | - | - | - | G. | 19. | H. 5 | 29,0 | 1.5 | 17. |
| $\checkmark$ | 102. | 77. | 61. | 56. | 89. | 90. | 90. | 88. | 88. |
| Cr | 186. | 28. | 19. | 27. | 68. | 45. | 132. | 65. | 56. |
| Hf | - | - | - | - | - | - | 1 |  | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | 19.0 | 15.0 | 13.0 | 12.0 | 16.0 | 16.0 | 17.0 | 17.0 | 16.0 |
| Ta |  | - | 1 | I | 1 |  | , | 17. | . |
| Co | 21. | 12. | 16. | 9. | 11. | 15. | 18. | 17. | 15. |
| Li | - | - | - | - | - | - | - | - | - |
| Be | - | - | - | - | - | - | _ | _ | - |
| 8 | - | - | - | - | - | - | - | - | - |
| $F$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | 3.40 | 2. 00 | 2. 60 | 4. 60 | 4.00 | 3.20 | 3.00 | 3.80 | 2. 80 |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| La | 30. | 31. | 31. | 30. | 33. | 3. | 31. | 29. | 31. |
| Ce | 62. | 72. | 65. | 70. | 79. | 70. | 67. | 64. | 68. |

Figure D. 1 (cont.). Geochemical databose.

## AUSTRALIA

|  | ABB89 UNKN GRAN | ABB9 UNKN GRAN | AB890 UNKN GRAN | ABB91 UNKN GRAN | ABB92 UNKN GRAN | ABB94 UNKN GRAN | AB895 UNKN GRAN | ABB96 UNKN GRAN | AKB10 UNKN GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 68. 55 | 68. 21 | 57.64 | 87.96 | 68.86 | 68. 36 | 69. 20 | 58. 33 | 71.86 |
| $\mathrm{TiO}_{2}$ | 63 | . 53 | . 51 | . 53 | . 52 | 56 | . 53 | . 54 | 36 |
| $\mathrm{Al}_{2} \mathrm{O} 3$ | 14.10 | 14. 25 | 14. 65 | 14. 21 | 14. 27 | 14.02 | 14.03 | 13.94 | 13.72 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 1.18 | . 93 | 1.58 | 1. 34 | 1.03 | 85 | 90 | 1. 59 | 71 |
| FeO | 3.29 | 3.10 | 2. 24 | 2.79 | 3. 05 | 3. 34 | 2. 99 | 2. 48 | 2.02 |
| MnO | 07 | . 07 | . 07 | . 06 | 07 | . 07 | . 06 | 07 | . 05 |
| MgO | 2.02 | 1.68 | 1.84 | 1. 73 | 1.54 | 1.83 | 1.75 | 1.75 | 1. 08 |
| CaO | 2. 51 | 3.11 | 3. 38 | 3. 32 | 3. 29 | 3.25 | 2. 93 | 3.02 | 1.95 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 2. 11 | 2. 23 | 2. 37 | 2. 17 | 2. 27 | 2. 27 | 2. 23 | 2.19 | 2.55 |
| $\mathrm{K}_{2} \mathrm{O}$ | 3.45 | 3. 77 | 3. 52 | 3. 12 | 3.14 | 3.50 | 3.79 | 3.67 | 4.11 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 14 | . 13 | . 12 | . 12 | . 12 | . 12 | . 11 | 12 | . 12 |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1. 55 | 1. 28 | 1. 34 | 1. 65 | 1.10 | 1.17 | 1.18 | 1. 67 | . 85 |
| $\mathrm{H}_{2} \mathrm{O}-$ | . 30 | . 12 | . 36 | . 20 | . 15 | . 21 | . 13 | . 31 | . 20 |
| $\mathrm{CO}_{2}$ | . 04 | . 29 | . 08 | . 23 | . 14 | . 15 | . 11 | . 14 | . 08 |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | . 18 | . 18 | . 18 | . 17 | . 19 | . 17 | . 18 | . 18 | - |
| TOTAL | 100.22 | 99. 88 | 99. 79 | 99.91 | 100. 05 | 99.87 | 100.12 | 100.00 | 99. 66 |
| $\mathrm{Fe}_{2} \mathrm{O} 3 \mathrm{t}$ | 4. 83 | 4.37 | 4. 17 | 1. 41 | 4. 43 | 4. 56 | 4.22 | 4. 34 | 2. 95 |
| A/CNK | 1. 2 | 1.1 | 1.1 | 1.1 | 1.1 | 1.0 | 1.1 | 1.1 | 1.1 |
| Di | 72.9 | 72.5 | 72. 3 | 71.5 | 72. 2 | 71.3 | 73. 6 | 73. 5 | 81.4 |
| Bo | 470. | 530. | 510. | 475. | 570. | 480. | 500. | 485. | 425. |
| Rb | 175. | 179. | 171. | 163. | 173. | 153. | 185. | 180. | 199. |
| Sr | 140. | 167. | 184. | 174. | 190. | 135. | 156. | 156. | 103. |
| Y | 31. | 30. | 30. | 30. | 33. | 25. | 32. | - | 35. |
| Zr | 187. | 158. | 157. | 165. | 171. | 152. | 175. | - | 134. |
| Nb | 12. | 12. | 11. | 11. | 13. | 10. | 12. | - | - |
| Th | 19. 20 | 15.80 | 18.00 | 17.00 | 19.80 | 14.80 | 18.80 | - | 16.80 |
| Pb | 25. | 24. | 24. | 24. | 23. | 19. | 26. | - | - |
| Ga | 17. | 16. | 17. | 16. | 17. | 14. | 16. | - | - |
| Zn | 70.0 | 57.0 | 59.0 | 56.0 | 56.0 | 55.0 | 65. 0 | - | 44.0 |
| Cu | 14.0 | 11.0 | 4.0 | 12.0 | 3.0 | 5.5 | 6. 0 | - | 5. 0 |
| Ni | 17.5 | 9.5 | 9. 5 | 7.5 | 8.5 | 9. 0 | 10.5 | - | 7. 5 |
| $\mathrm{TiO}_{2}$ |  | - | - | - | - | - | - | - | - |
| V | 84. | 95. | 83. | 84. | 81. | 92. | 90. | - | 44. |
| Cr | 50. | 32. | 31. | 30. | 27. | 38. | 36. | - | 22. |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | 15.0 | 18.0 | 18.0 | 16.0 | 16.0 | 17.0 | 16.0 | - | - |
| Ta | - | - | - | - | - | - | - | - | - |
| Co | 13. | 14. | 10. | 10. | 16. | 13. | 13. | - | 9. |
| Li | - | - | - | - | - | - | - |  | - |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | 3.00 | 2. 60 | 3. 80 | 4.00 | 4. 20 | 1.40 | 4.00 | - | 2.20 |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| La | 34. | 31. | 34. | 32. | 37. | 31. | 33. | - | 24. |
| Ce | 73. | 58. | 72. | 67. | 77. | 71. | 71. | - | 54. |

Figure D. 1 (cont.). Geochemical databose.

## AUSTRALIA

|  | AKB12 UNKN GRAN | AKB13 UNKN GRAN | AKB19 UNKN GRAN | AKB27 UNKN GRAN | AKB3 1 UNKN GRAN | AKB32 UNKN GRAN | AKB33 UNKN GRAN | AKB42 UNKN GRAN |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 68. 94 | 70.77 | 67.65 | 68. 13 | 71.16 | 67.68 | 65. 95 | 75.11 | 75. 58 |
| $\mathrm{TiO}_{2}$ | . 59 | . 45 | . 70 | . 56 | . 42 | . 54 | . 67 | . 13 | . 23 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14.11 | 13. 93 | 14.60 | 14.46 | 13.81 | 14.70 | 14.90 | 13.04 | 12.05 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 87 | . 80 | 1.00 | . 75 | . 70 | . 68 | . 88 | . 16 | 54 |
| FeO | 3.27 | 2. 55 | 3.77 | 3. 45 | 2. 38 | 4. 03 | 4.42 | 1.10 | 1. 28 |
| MnO | . पE | . 05 | . 07 | . 06 | . 05 | . 07 | . 07 | . 03 | . 04 |
| MgO | 1.91 | 1. 34 | 2. 13 | 2. 26 | 1. 25 | 2. 22 | 2. 56 | . 43 | . 71 |
| CaO | 2. 46 | 2. 18 | 2. 65 | 2.45 | 2. 85 | 2. 28 | 2.87 | . 96 | 1.26 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 2. 22 | 2. 46 | 2. 08 | 1.84 | 2. 45 | 1.92 | 1.99 | 2.80 | 2. 62 |
| $\mathrm{K}_{2} \mathrm{O}$ | 3. 53 | 3.71 | 3. 06 | 3. 97 | 3.93 | 3. 60 | 3. 43 | 5.11 | 4.07 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 13 | . 13 | . 15 | . 13 | . 12 | . 15 | . 21 | . 11 | . 10 |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1. 20 | 1. 00 | 1. 43 | 1. 34 | 1.05 | 1.51 | 1. 57 | .80 | 1.05 |
| $\mathrm{H}_{2} \mathrm{O}-$ | . 31 | . 27 | . 30 | . 20 | . 21 | . 22 | . 21 | .10 | . 15 |
| $\mathrm{CO}_{2}$ | . 14 | . 04 | .16 | . 12 | . 09 | . 12 | . 16 | . 07 | . 09 |
| Cl | - | - | - | - | - | - | - | - | - |
| $F$ | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 99.74 | 99. 70 | 99.75 | 99.73 | 99.67 | 99.80 | 99. 89 | 99. 95 | 99.77 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3 t}$ | 4.50 | 3. 64 | 5. 18 | 4.59 | 3. 34 | 5. 15 | 5.79 | 1. 38 | 1. 96 |
| A/CNK | 1.2 | 1. 2 | 1,3 | 1. 2 | 1.2 | 1.3 | 1.2 | 1.1 | 1.1 |
| DI | 73.6 | 78.2 | 70.1 | 72.0 | 79. 5 | 71.1 | 67.4 | 90.9 | 87.9 |
| Bo | 460. | 470. | 510. | 720. | 445. | 475. | 385. | 190. | 280. |
| Rb | 170. | 190. | 149. | 117. | 195. | 183. | 193. | 253. | 198. |
| Sr | 128. | 115. | 139. | 157. | 117. | 139. | 136. | 50. | 91. |
| Y | 31. | 38. | 31. | 29. | 10. | 27. | 29. | 30. | 33. |
| Zr | 194. | 176. | 207. | 170. | 178. | 187. | 161. | 68. | 107. |
| Nb | - | - | - | - | - | - | - | - | - |
| Th | 19. 20 | 20.00 | 20. 20 | 20.00 | 18. 60 | 19. 20 | 14.40 | 10.60 | 14.60 |
| Pb | - | - | - | - | - | - | - | - | - |
| Go | - | - | - | - | - | - | - | $\sim$ | - |
| Zn | 64.0 | 55.0 | 81.0 | 71.0 | 83.0 | 82.0 | 93.0 | 17.0 | 28.0 |
| Cu | 18.5 | 6.0 | 20, 0 | 17.5 | 14.5 | 16. 5 | 20.5 | 9. 5 | 5.0 |
| Ni | 14.5 | 10.5 | 21.0 | 21.0 | 10.0 | 20.0 | 25. 5 | 1. 5 | 4.0 |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| $V$ | 81. | 54. | 88. | 86. | 51. | 87. | 100. | 12. | 24. |
| Cr | 47. | 31. | 54. | 60. | 28. | 61. | 87. | 8. | 9. |
| Hf | . |  | - | - | - | - | - | - | - |
| Cs | - | _ | - | _ | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| Ta | - | - | - | - | - | - | - | - | - |
| Co | 13. | 10. | 16. | 17. | 7. | 17. | 18. | 3. | 6. |
| Li | - | - | - | - | - | - | - | - | - |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | $\rightarrow$ | - | - | - | - | - | - | _ |
| $F$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | 3.80 | 2. 20 | 3. 60 | 3.00 | 4.40 | 4.00 | 2. 40 | 3.80 | 2. 20 |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| Lo | 32. | 30. | 37. | 37. | 29. | 33. | 31. | 12. | 18. |
| Ce | 67. | 64. | 77. | 74. | 60. | 68. | 52. | 25. | 42. |

Figure D. 1 (cont.). Geochemical database.

## AUSTRALIA

|  | AKB44 UNKN GRAN | AKB45 UNKN GRAN | $\begin{aligned} & \text { AKB46 } \\ & \text { UNKN } \\ & \text { GRAN } \end{aligned}$ | AKB47 UNKN GRAN | AKB5 1 UNKN GRAN | AKB52 UNKN GRAN | AK855 UNKN GRAN | AKB56 UNKN GRAN | AKB57 UNKN GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 71. 0 | 73.79 | 69.11 | 68. 72 | 69. 16 | 67. 58 | 72.54 | 67.66 | 72. 70 |
| $\mathrm{TiO}_{2}$ | . 43 | . 24 | . 62 | . 57 | . 48 | . 68 | . 24 | . 56 | . 34 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 13.83 | 13.40 | 14.39 | 14. 36 | 14.16 | 14.76 | 13. 45 | 14.15 | 13. 21 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 76 | . 37 | . 50 | . 81 | . 85 | . 60 | . 47 | 1.03 | . 79 |
| FeO | 2. 33 | 1. 52 | 3.73 | 3. 36 | 2. 88 | 3.85 | 2. 54 | 3.45 | 1. 86 |
| MnO | . 05 | . 05 | . 06 | . 06 | . 87 | . 07 | . 05 | . 07 | .05 |
| MgO | 1.19 | . 75 | 1. 86 | 1. 95 | 1.72 | 2.14 | . 56 | 2. 11 | . 91 |
| CoO | 1.85 | 1. 25 | 2. 21 | 1.97 | 2. 68 | 2. 58 | 2. 16 | 3.41 | 1.81 |
| $\mathrm{No}_{2} \mathrm{O}$ | 2. 54 | 2. 61 | 2. 24 | 1.99 | 2. 16 | 2.14 | 2. 46 | 2. 18 | 2. 62 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4.09 | 4. 53 | 3.62 | 3. 94 | 3. 58 | 3. 81 | 3.72 | 3. 31 | 3. 84 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 12 | . 12 | . 14 | . 16 | . 15 | .17 | . 12 | . 13 | . 12 |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1.32 | . 90 | 1.05 | 1. 65 | 1. 56 | 1. 30 | 1. 23 | 1.78 | 1. 23 |
| $\mathrm{H}_{2} \mathrm{O}-$ | . 16 | . 15 | . 06 | . 23 | . 22 | .13 | . 12 | .15 | . 20 |
| $\mathrm{CO}_{2}$ | . 18 | . 11 | .14 | . 12 | . 14 | . 23 | . 11 | . 07 | .17 |
| Cl | - | - | - | - | - | - | - | - | - |
| $F$ | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 99.85 | 99.90 | 100.03 | 99. 89 | 99. 81 | 99.84 | 99.87 | 100.05 | 99. 85 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 3.35 | 2.17 | 4. 64 | 4. 54 | 4. 05 | 4. 87 | 3. 29 | 4.87 | 2. 85 |
| A/CNK | 1.2 | 1.2 | 1.2 | 1.3 | 1.2 | 1. 2 | 1.1 | 1.1 | 1.1 |
| Dl | 80.8 | 86, 6 | 74.2 | 74.9 | 74.1 | 71.2 | 80.9 | 69.4 | 83.0 |
| Bo | 475. | 250. | 475. | 475. | 495. | 485. | 525. | 435. | 470. |
| Rb | 201. | 259. | 183. | 196. | 185. | 177. | 172. | 155. | 189. |
| Sr | 103. | 65. | 115. | 138. | 163. | 142. | 143. | 178. | 109. |
| Y | 35. | 27. | 28. | 30. | 35. | 27. | 31. | 30. | 34. |
| Zr | 171. | 90. | 187. | 182. | 152. | 190. | 153. | 159. | 163. |
| Nb | - | - | - | - | - | - | - | - | - |
| Th | 19.00 | 11.40 | 16.40 | 19.00 | 17.20 | 18.40 | 18.40 | 13.40 | 19.40 |
| Pb | - | - | - | - |  | - | - | - | - |
| Ga | - | - | - | - | - | - | - | - | - |
| Zn | 48.0 | 33. 0 | 67.0 | 92.0 | 61.0 | 74.0 | 59.0 | 65.0 | 45.0 |
| Cu | 5.0 | 2. 0 | 16.5 | 18.5 | 9. 0 | 21.0 | 4.0 | 9.0 | 4.0 |
| Ni | 9. 5 | 5.0 | 15.5 | 18.5 | 12.5 | 19.5 | 2. 0 | 11.0 | 5.0 |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| V | 49. | 25. | 89. | 74. | 74. | 82. | 24. | 96. | 38. |
| Cr | 25. | 14. | 49. | 54. | 43. | 50. | 5. | 41. | 18. |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| Ta | - | - | - | - | - | - | - | - | - |
| Co | 11. | 5. | 12. | 13. | 13. | 15. | 5. | 13. | 8. |
| Li | 11. | . | 12. | , | 1. | 15. | . | 13. | , |
| Be | - | - | - | - | _ | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| $F$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | 2. 40 | 3.20 | 3.40 | 3.40 | 1.80 | 3. 00 | 1.60 | 2. 60 | 3. 20 |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| La | 28. | 15. | 27. | 34. | 31. | 32. | 37. | 33. | 29. |
| Ce | 59. | 32. | 58. | 70. | 84. | 67. | 75. | 70. | 62. |

Figure D. 1 (cont.). Geochemical database.


Figure D. 1 (cont.). Geochemical database.

## AUSTRALIA

|  | AK023 UNKN GRAN | AKO31 <br> UNKN <br> GRAN | AK036 UNKN GRAN | AKO37 UNKN GRAN | AKO38 UNKN GRAN | AKO39 UNKN GRAN | AKO41 UNKN GRAN | AKO42 UNKN GRAN | AKO43 UNKN GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 67. 64 | 71.16 | 71.53 | 71.01 | 68. 60 | 68.64 | 70.20 | 75.11 | 75. 58 |
| $\mathrm{TiO}_{2}$ | . 66 | . 42 | . 37 | . 42 | . 62 | . 63 | . 49 | . 13 | . 23 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14.68 | 13.81 | 13.99 | 13. 92 | 14. 42 | 14. 52 | 13.87 | 13.04 | 12.05 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 1.08 | . 70 | . 45 | . 56 | . 93 | . 55 | 89 | . 18 | . 54 |
| FeO | 3. 25 | 2. 38 | 2. 34 | 2. 56 | 3. 35 | 3. 50 | 2. 53 | 1.10 | 1. 28 |
| MnO | . 07 | . 05 | . 04 | . 05 | . 07 | . 06 | . 05 | . 03 | . 04 |
| MgO | 2. 00 | 1. 25 | 1.05 | 1. 26 | 2. 13 | 1. 87 | 1. 62 | . 43 | . 71 |
| CoO | 2. 56 | 2. 05 | 1. 93 | 1. 98 | 1.35 | 2.14 | 2. 26 | . 96 | 1.26 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 2. 21 | 2.45 | 2. 56 | 2. 39 | 1.93 | 2. 29 | 2. 63 | 2. 80 | 2. 62 |
| $\mathrm{K}_{2} \mathrm{O}$ | 3.69 | 3. 93 | 4.37 | 4. 08 | 4. 08 | 3.67 | 3. 65 | 5. 11 | 4.07 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 16 | . 12 | . 11 | . 12 | . 16 | . 14 | . 11 | . 11 | . 10 |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1. 35 | 1.05 | . 92 | 1. 20 | 1. 81 | 1. 34 | 1. 17 | . 80 | 1.05 |
| $\mathrm{H}_{2} \mathrm{O}-$ | . 24 | . 21 | . 07 | .14 | . 26 | . 13 | . 20 | .10 | . 15 |
| $\mathrm{CO}_{2}$ | . 17 | . 09 | . 11 | . 23 | . 11 | . 08 | . 20 | . 07 | . 09 |
| Cl | - | - | - | - | - | - | - | - | - |
| $F$ | - | - | - | - | _ | - |  |  |  |
| LOI | .18 | .17 | .16 | . 15 | .19 | .18 | .15 | .09 | .11 |
| TOTAL | 99.95 | 99.84 | 100.01 | 100.07 | 100.01 | 100.04 | 100.02 | 100.04 | 99.88 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 4.70 | 3.34 | 3. 06 | 3. 40 | 4. 65 | 4.44 | 3. 70 | 1. 38 | 1. 96 |
| A/CNK | 1.2 | 1. 2 | 1. 1 | 1.2 | 1.4 | 1.2 | 1.1 | 1.1 | 1.1 |
| DI | 72. 6 | 79.5 | 81.4 | 79.6 | 76.2 | 73.6 | 77.6 | 90.9 | 87.9 |
| Bo | 470. | 445. | 510. | 450. | 560. | 475. | 380. | $190 .$ | $280$ |
| Rb | 187. | 195. | 180. | 191. | 193. | 182. | 182. | 253. | $198 .$ |
| Sr | 145. | 117. | 118. | 108. | 112. | 136. | 120. | 50. | 91. |
| Y | 29. | 40. | 40. | 35. | 30. | 30. | 35. | 30. | 33. |
| Zr | 185. | 178. | 133. | 157. | 200. | 183. | 171. | 68. | 107. |
| Nb | 13. | 11. | 9. | 10. | 13. | 12. | 10. | 6. | 7. |
| Th | 19.20 | 18. 60 | 17.40 | 18.80 | 21. 10 | 20. 00 | 18.00 | 10. 60 | 14.60 |
| Pb | 29. | 59. | 31. | 33. | 29. | 27. | 21. | 35. | 27. |
| Ga | 17. | 16. | 15. | 15. | 18. | 17. | 16. | 12. | 13. |
| Zn | 75.0 | 83. 0 | 40.0 | 53.0 | 97.0 | 66.0 | 44.0 | 17.0 | 28.0 |
| Cu | 19.0 | 14.5 | 33.0 | 7. 5 | 18.0 | 17.5 | 7.0 | 9. 5 | 5. 0 |
| Ni | 19.0 | 10.0 | 6.5 | 8. 5 | 19.5 | 15.5 | 11.0 | 1. 5 | 4. 0 |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| $V$ | 77. | 51. | 42. | 54. | 77. | 72. | 63. | 12. | 24. |
| Cr | 52. | 28. | 22. | 25. | 52. | 40. | 35. | 8. | 9. |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | 15.0 | 10.0 | 10.0 | 11.0 | 14.0 | 14.0 | 12.0 | 5. 0 | 6. 0 |
| To | , | - | - | $\xrightarrow{-}$ | - | - | . | 5. | . |
| Co | 15. | 7. | 8. | 10. | 13. | 13. | 12. | 3. | 6. |
| Li | - | - | - | - | - | - | - | - | - |
| Be | - | - | - | _ | _ | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| $F$ | - | - | - | - | _ | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | 3.40 | 4.40 | 3. 60 | 2. 80 | 4. 80 | 4. 60 | 2.20 | 3, 80 | 2.20 |
| W | - | - | d | - |  | - | - | 3.80 | 2.20 |
| Sn | - | - | - | - | - | - | _ | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| Lo | 34. | 29. | 26. | 29. | 36. | 34. | 29. | 12. | 18. |
| Ce | 70. | 60. | 55. | 59. | 77. | 72. | 63. | 25. | 42. |

Figure D. 1 (cont.). Geochemical datobase.

## AUSTRALIA

|  | AKO44 UNKN GRAN | AKO45 UNKN GRAN | AKO46 UNKN GRAN | AKO52 UNKN GRAN | AKO57 UNKN GRAN | AKO59 UNKN GRAN | AKO9 <br> UNKN <br> GRAN | AM0107 UNKN GRAN | AMO118 UNKN GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 71.00 | 73.79 | 59.41 | 67.58 | 72. 70 | 75.11 | 69.19 | 69. 52 | 73. 98 |
| $\mathrm{TiO}_{2}$ | . 43 | . 24 | . 62 | . 68 | . 34 | . 10 | . 59 | . 55 | . 25 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 13.83 | 13.40 | 14.39 | 14.76 | 13. 21 | 13.17 | 14.04 | 14.45 | 13.51 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 76 | . 37 | . 50 | . 60 | . 75 | . 34 | . 71 | . 45 | . 33 |
| FeO | 2. 33 | 1. 52 | 3.73 | 3. 85 | 1. 86 | 1.02 | 3. 35 | 3. 01 | 1. 43 |
| MnO | . 05 | . 05 | . 06 | . 07 | . 05 | . 03 | . 05 | . 08 | . 05 |
| MgO | 1.19 | . 75 | 1. 85 | 2.14 | . 91 | . 34 | 1.83 | 1.20 | . 36 |
| CaO | 1.85 | 1. 26 | 2. 21 | 2. 58 | 1. 81 | 1. 31 | 2. 14 | 2. 00 | 1.45 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 2. 54 | 2. 61 | 2.24 | 2. 14 | 2. 62 | 3. 48 | 2. 07 | 3.25 | 3.05 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4.09 | 4. 53 | 3. 62 | 3. 61 | 3.84 | 3.97 | 3. 69 | 3. 98 | 4.71 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 12 | . 12 | . 14 | . 17 | . 12 | . 05 | . 14 | . 15 | .15 |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1.32 | . 90 | 1. 05 | 1.30 | 1.23 | . 68 | 1. 48 | . 86 | . 44 |
| $\mathrm{H}_{2} \mathrm{O}-$ | . 16 | . 15 | . 06 | . 13 | . 20 | . 13 | . 31 | . 15 | . 08 |
| $\mathrm{CO}_{2}$ | . 18 | . 11 | .14 | . 23 | .17 | . 12 | . 08 | . 14 | . 11 |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | .16 | .11 | .17 | . 18 | .15 | .12 | .18 | .18 | .15 |
| TOTAL | 100.01 | 100.01 | 100.20 | 100. 02 | 100.00 | 99.97 | 99.87 | 99.97 | 100.06 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 3. 35 | 2.17 | 4.64 | 4.87 | 2.85 | 1.47 | 4. 44 | 3.79 | 1. 92 |
| A/CNK | 1.2 | 1.2 | 1.2 | 1. 2 | 1.1 | 1.1 | 1.2 | 1.1 | 1.1 |
| Cl | 80.8 | 86.6 | 74.2 | 71.2 | 83. 0 | 89. 6 | 74.7 | 79.7 | 88. 3 . |
| 8 c | 475. | 250. | 475. | 485. | 470. | 465. | 520. | 600. | 580. |
| Rb | 201. | 259. | 183. | 177. | 189. | 154. | 176. | 157. | 206. |
| Sr | 103. | 65. | 115. | 142. | 109. | 111. | 129. | 186. | 144. |
| Y | 36. | 27. | 28. | 27. | 34. | 32. | 32. | 33. | 33. |
| Zr | 171. | 90. | 187. | 190. | 183. | 71. | 184. | 181. | 122. |
| Nb | 11. | 8. | 12. | 13. | 9. | 7. | 12. | 8. | 7. |
| Th | 19. 00 | 11.10 | 16.40 | 18.40 | 19.40 | 14.00 | 11.80 | 16.00 | 14.00 |
| Pb | 28. | 32. | 24. | 28. | 28. | 28. | 27. | 22. | 22. |
| Ga | 16. | 14. | 17. | 17. | 15. | 14. | 17. |  | - |
| Zn | 48.0 | 33. 0 | 67.0 | 74.0 | 4.0 | 13.0 | 62.0 | 67.0 | 42.0 |
| Cu | 5.0 | 2.0 | 16. 5 | 21.0 | 4. 0 | . 5 | 17.0 | 14.0 | 8. 0 |
| Ni | 9.5 | 5. 0 | 15. 5 | 19.5 | 5.0 | . 5 | 15.5 | 9.0 | 4.0 |
| $\mathrm{TiO}_{2}$ | - | - | 1 | - | - | - | - | - | - |
| $\checkmark$ | 49. | 25. | 89. | 82. | 38. | 11. | 74. | 51. | 16. |
| Cr | 25. | 14. | 49. | 50. | 18. | 5. | 44. | 19. | 5. |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | 10.0 | 7.0 | 18.0 | 16.0 | 9. 0 | 7.0 | 14.0 | 11.0 | 5.0 |
| To | - | - | - | - | , |  | - | - | - |
| Co | 11. | 5. | 12. | 15. | 8. | 2. | 11. | 15. | 7. |
| Li | - | 5. | 12. | 15. | Q. | 2. | 1. | 15. | . |
| Be | - | - | - | - | _ | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | 2.40 | 3.20 | 3. 40 | 3. 00 | 3.20 | 2. 80 | 3.80 | 4.00 | 4.00 |
| W | - | - | . | - | - | - | 3. 8 | 1.00 | 4.00 |
| Sn | - | - | - | - | - | _ | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| Lo | 28. | 15. | 27. | 32. | 29. | 17. | 32. | 25. | 19. |
| Ce | 59. | 32. | 58. | 67. | 62. | 37. | 68. | 58. | 44. |

Figure 0.1 (cont.). Geochemical databose.

## AUSTRALIA

|  | AMO120 UNKN GRAN | AMO1 22 UNKN GRAN | AMO125 UNKN GRAN | AMUO2 UNKN GRAN | AMUOT UNKN GRAN | AMU1 1 UNKN GRAN | AMU13 UNKN GRAN | ARP2 UNKN GRAN | ARP3 UNKN GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 72. 37 | 73.52 | 74.92 | 75.93 | 71.57 | 70.88 | 65.85 | 70. 25 | 70.79 |
| $\mathrm{TiO}_{2}$ | . 39 | 27 | . 28 | 11 | . 32 | . 51 | . 80 | . 14 | . 50 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14.04 | 13.79 | 12.76 | 12.91 | 14.53 | 14.11 | 15.34 | 14.93 | 14.45 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 38 | . 28 | . 20 | . 10 | . 21 | . 28 | . 81 | . 54 | . 62 |
| FeO | 1.83 | 1.42 | 1.32 | . 86 | 1.88 | 2.87 | 4. 21 | 1.98 | 2.73 |
| Mno | . 05 | . 04 | . 03 | . 01 | . 05 | . 08 | . 10 | . 04 | . 05 |
| Mgo | . 57 | . 35 | . 34 | . 28 | . 53 | . 94 | 1. 76 | 85 | 1. 34 |
| CaO | 1.81 | 1.47 | 1. 22 | 59 | 1.68 | 1.67 | 2.94 | 2. 02 | 1. 40 |
| $\mathrm{No}_{2} \mathrm{O}$ | 3.11 | 3. 10 | 3. 05 | 2. 82 | 3. 14 | 3.11 | 3.39 | 3.07 | 2.25 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4.57 | 4.57 | 4.77 | 5. 51 | 4.19 | 4. 32 | 3. 17 | 4.26 | 4.40 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 14 | . 12 | . 08 | . 13 | . 14 | . 09 | . 17 | . 22 | . 28 |
| $\mathrm{H}_{2} \mathrm{O}+$ | . 59 | . 54 | . 66 | - | - | - | - | . 94 | 1. 10 |
| $\mathrm{H}_{2} \mathrm{O}-$ | . 12 | . 10 | . 17 | - | - | - | - | . 15 | . 06 |
| $\mathrm{CO}_{2}$ | . 20 | . 09 | . 21 | - | - | - | - | . 08 | .14 |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | . 17 | . 15 | . 13 | - | - | - | - | - | - |
| TOTAL | 100.14 | 99.81 | 100.12 | 99. 25 | 98.84 | 98. 66 | 98. 55 | 99. 79 | 100.12 |
| $\mathrm{Fe}_{2} \mathrm{O} 3 \mathrm{t}$ | 2. 41 | 1.86 | 1.67 | 1.05 | 2. 30 | 3. 24 | 5.48 | 2.74 | 3. 65 |
| A/CNK | 1.1 | 1.1 | 1. 0 | 1.1 | 1.1 | 1.1 | 1.1 | 1.1 | 1.3 |
| Di | 85.9 | 87.5 | 90.1 | 92.9 | 84.7 | 81.6 | 70.3 | 82.1 | 81.0 |
| Ba | 630. | 510. | 420. | - | - | - | - | 666. | 518. |
| Rb | 173. | 206. | 203. | - | - | - | - | 217. | 239. |
| Sr | 151. | 185. | 96. | - | - | - | - | 282. | 121. |
| Y | 32. | 35. | 31. | - | - | - | - | 22. | 37. |
| Zr | 172. | 133. | 148. | - | - | - | - | - | - |
| Nb | 6. | 8. | 5. | - | - | - | - | - | - |
| Th | 16.00 | 15.00 | 20. 00 | - | - | - | - | 20. 00 | 20.00 |
| Pb | 25. | 24. | 20. | - | - | - | - | 36. | 46. |
| Go | - | - | - | - | - | - | - | - | - |
| Zn | 58.0 | 44. 0 | 29.0 | - | - | - | - | - | - |
| Cu | 13.0 | 8.0 | 7.0 | - | - | - | - | - | - |
| Ni | 5.0 | 3. 0 | 2.0 | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| V | 28. | 16. | 15. | - | - | - | - | - | - |
| Cr | 9. | 5. | 3. | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | 5. 0 | 5. 0 |
| Cs | - | - | - | - | - | - | - | 9. 0 | 23.0 |
| Sc | 7.0 | 5. 0 | 5. 0 | - | - | - | - | - | - |
| To | 7.0 | 5. | 5.0 | - | - | - | - | - | - |
| Co | 10. | 4. | 5. | - | - | - | - | - | - |
| Li | - | - | . | - | - | - | - | - | - |
| Be | - | - | - | - | - | - | - | - | - |
| 8 | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | 3.00 | 6.00 | 3.00 | - | - | - | - | 7.00 | 4.00 |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | 5. 0 | 11.0 |
| Mo | - | - | - | - | - | - | - | - | - |
| Lo | 24. | 21. | 23. | - | - | - | - | - | - |
| Ce | 56. | 45. | 54. | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.

## AUSTRALIA

|  | ARP4 UNKN GRAN | ARP5 UNKN GRAN | ARP6 UNKN GRAN | ASB717 <br> UNKN <br> GRAN | ASE718 <br> UNKN <br> GRAN | ASB721 UNKN GRAN | ASB726 UNKN GRAN | ASE754 <br> UNKN <br> GRAN | ASB765 <br> UNKN <br> GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 71.42 | 70.88 | 72. 20 | 72. 30 | 72. 73 | 73. 41 | 75.84 | 75.12 | 72.08 |
| $\mathrm{TiO}_{2}$ | . 47 | . 44 | . 26 | . 34 | . 45 | 30 | . 16 | . 09 | . 35 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14.09 | 14.74 | 13.87 | 14.65 | 14.01 | 14.09 | 13. 25 | 13. 75 | 14.30 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 62 | . 63 | 1. 22 | - | - | - | - | - | - |
| FeO | 2. 65 | 2. 55 | 1.23 | 2. 42 | 2. 73 | 2. 03 | 1.50 | 1.05 | 2. 74 |
| Mno | . 06 | . 04 | . 08 | - | - | - | - | - | - |
| MgO | 1.54 | 1. 32 | . 26 | 1.02 | 1. 31 | 1.09 | . 63 | . 59 | 1.04 |
| CaO | . 81 | . 64 | 1.35 | 1.67 | 1. 51 | 1.71 | . 89 | 1. 29 | 2.15 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 1.95 | 1.93 | 3.77 | 2. 86 | 2. 46 | 3. 29 | 2. 95 | 2. 95 | 3.09 |
| $\mathrm{K}_{2} \mathrm{O}$ | 3. 73 | 4. 55 | 4.27 | 4.50 | 4. 49 | 3.88 | 4. 58 | 4. 35 | 3. 94 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 19 | . 21 | . 06 | . 22 | . 27 | . 21 | . 20 | . 20 | . 28 |
| $\mathrm{H}_{2} \mathrm{O}+$ | 1.21 | 1.13 | . 94 | - | - | - | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}-$ | . 14 | .14 | . 25 | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | . 11 | . 11 | . 08 | - | - | - | - | - | - |
| Cl | - | $\rightarrow$ | , | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | - |
| TOTAL | 99.00 | 99. 31 | 99.84 | 99. 99 | 99. 99 | 99.99 | 100.00 | 99.99 | 99.98 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 3. 57 | 3. 46 | 2. 59 | 2. 69 | 3. 03 | 2. 25 | 1.67 | 1.17 | 3.04 |
| A/CNK | 1.8 | 1.6 | 1.0 | 1.2 | 1. 2 | 1.1 | 1. 2 | 1.1 | 1.1 |
| DI | 80.3 | 82.3 | 88. 6 | 83.0 | 82.1 | 84.0 | 89. 8 | 89.3 | 81.1 |
| Bo | 435. | 410. | 164. | - | - | - | - | - | - |
| Rb | 186. | 283. | 370. | - | - | - | - | - | - |
| Sr | 104. | 88. | 73. | - | - | - | - | - | - |
| Y | 45. | 21. | 59. | - | - | - | - | - | - |
| Zr | , | , | - | - | - | - | - | - | - |
| Nb | - | - | - | - | - | - | - | - | - |
| Th | 14.00 | 14.00 | 48. 10 | _ | - | - | _ | - | - |
| Pb | 4 E . | 39. | 40. | - | - | - | - | - | - |
| Ga | - | - | - | - | - | - | - | - | - |
| Zn | - | - | - | - | - | - | - | - | - |
| Cu | - | - | - | _ | - | - | _ | - | - |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| $\checkmark$ | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | 5. 0 | 4.0 | 6. 0 | - | - | - | - | - | _ |
| Cs | 37.0 | 25.0 | 47.0 | - | - | - | - | - | - |
| Sc | , | - | - | - | - | - | - | - | - |
| Ta | - | - | - | _ | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | _ |
| Li | - | - | - | - | - | - | - | - | - |
| Be | - | - | - | - | - | - | - | - | - |
| 8 | - | - | $-$ | - | - | - | - | - | - |
| $F$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | 5.00 | 5.00 | 12.00 | - | - | - | - | - | - |
| W | , | - | - | - | - | _ | - | - | - |
| Sn | 13.0 | 11.0 | 16.0 | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.

AUSTRALIA

|  | ASB793 <br> UNKN <br> GRAN | ASB799 <br> UNKN <br> GRAN | ASB830 <br> UNKN <br> GRAN | ASB831 UNKN GRAN | $\begin{aligned} & \text { ASB837 } \\ & \text { UNKN } \\ & \text { GRAN } \end{aligned}$ | $\begin{aligned} & \text { ASB844 } \\ & \text { UNKN } \\ & \text { GRAN } \end{aligned}$ | $\begin{aligned} & \text { ASB846 } \\ & \text { UNKN } \\ & \text { GRAN } \end{aligned}$ | ASB857 UNKN GRAN | ASB861 <br> UNKN <br> GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 73.37 | 74.20 | 70.91 | 74. 63 | 71.93 | 72.12 | 74.18 | 71.39 | 71.88 |
| $\mathrm{TiO}_{2}$ | . 23 | . 42 | . 52 | . 22 | . 52 | . 42 | . 34 | . 35 | . 31 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14. 26 | 13.15 | 14.37 | 13. 64 | 13.54 | 14.44 | 13.33 | 15.18 | 14. 25 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | - | - | - | - | - | - | - | - | - |
| FeO | 1. 52 | 2. 93 | 3. 31 | 1.83 | 3. 28 | 2. 92 | 2. 19 | 2. 33 | 1.85 |
| MnO | - | - | . 01 | - | - | - | - | - | . 04 |
| MgO | . 84 | 1. 21 | 1. 59 | . 83 | 1.41 | 1.14 | . 82 | 1. 05 | . 67 |
| CaO | 1.72 | 1.68 | 2. 26 | 1.38 | 1.86 | 1.73 | 1. 39 | 1.78 | 1.90 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3.89 | 2.49 | 2.92 | 2. 82 | 2. 19 | 2. 51 | 2. 88 | 2. 59 | 4.26 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4.01 | 3.74 | 3.85 | 4.50 | 4.74 | 4.50 | 4. 66 | 5. 06 | 4.75 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 15 | . 18 | . 27 | . 15 | . 22 | . 21 | . 19 | . 17 | . 07 |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | - | - | - | . 2 | . 21 | 1 | - | - |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | _ | - |
| TOTAL | 99.99 | 100.00 | 100.01 | 100.00 | 99.99 | 99.99 | 99. 98 | 100. 00 | 100.00 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 1.69 | 3. 25 | 3. 67 | 2. 03 | 3.64 | 3. 24 | 2. 43 | 2. 59 | 2. 06 |
| A/CNK | 1.0 | 1.2 | 1.1 | 1.1 | 1.1 | 1.2 | 1.1 | 1. 2 | . 9 |
| Ol | 86.4 | 81.5 | 77.8 | 86.3 | 80.6 | 81.2 | 86. 3 | 82.5 | 87.5 |
| Bo | - | - | - | - | - | - | - | - | - |
| Rb | - | - | - | - | - | - | _ | - | - |
| Sr | - | - | - | - | - | - | - | - | - |
| Y | - | - | - | - | - | - | - | - | - |
| Zr | - | - | - | - | - | - | - | - | - |
| Nb | - | - | - | - | - | - | - | - | - |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Go | - | - | - | - | - | - | - | - | - |
| Zn | - | - | - | - | - | - | - | - | - |
| Cu | - | - | - | - | - | - | - | - | - |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | _ | _ | - |
| $V$ | - | - | - | _ | - | - | - | - | _ |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | - | - | - | - |
| Cs | - | - | - | - | - | - | _ | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | _ | - |
| Li | - | - | - | - | - | - | - | - | - |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - |
| $F$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | _ |
| Mo | - | - | - | - | - | - | - | - | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical databose.


Figure D. 1 (cont.). Geochemical database.

## AUSTRALIA

|  | AST2 UNKN GRAN | AST3 UNKN GRAN | AST4 UNKN GRAN | AWB1 UNKN GRAN | AWB14 UNKN GRAN | AWB15 UNKN GRAN | AWB16 UNKN GRAN | AWB17 UNKN GRAN | AWB2 UNKN GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 72. 21 | 74.83 | 71.83 | 75. 32 | 76. 54 | 74.68 | 74.93 | 74.79 | 75.21 |
| $\mathrm{TiO}_{2}$ | . 34 | . 09 | . 35 | 10 | . 19 | . 31 | . 06 | . 40 | . 10 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14. 64 | 13.66 | 14. 25 | 12.97 | 12.05 | 12.80 | 13.45 | 12.67 | 13.10 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | - | - | - | . 44 | .47 | . 60 | . 44 | . 79 | . 40 |
| FeO | 2. 42 | 1. 05 | 2. 73 | . 95 | 1.10 | 1. 45 | . 87 | 1.60 | . 90 |
| Mno | - | - | - | . 03 | . 03 | . 03 | . 03 | 03 | . 03 |
| MgO | 1.02 | 59 | 1.04 | . 19 | . 38 | . 59 | . 18 | . 69 | . 20 |
| CoO | 1.67 | 1. 29 | 2.14 | 1. 55 | . 66 | 1. 04 | 1.30 | 1.35 | 1.55 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 2. 86 | 2. 94 | 3. 08 | 2. 97 | 2. 33 | 2. 37 | 2. 88 | 2.36 | 3. 02 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4.50 | 4.91 | 3.93 | 4. 30 | 4.90 | 4.79 | 4. 83 | 4. 59 | 4.33 |
| $\mathrm{P}_{2} \mathrm{O} 5$ | . 22 | . 22 | . 28 | . 03 | . 13 | . 13 | . 04 | . 12 | . 03 |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | - | - | .71 | . 80 | 1.03 | . 79 | . 87 | . 72 |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - |  | - | - | - | - | - |
| LOI | . 47 | . 61 | . 50 | - | - | - | - | - | - |
| TOTAL | 100.35 | 99. 99 | 100.14 | 99.57 | 99. 64 | 99.82 | 99. 60 | 100. 26 | 99.59 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 2.69 | 1.17 | 3.03 | 1.49 | 1.69 | 2. 21 | 1.41 | 2. 57 | 1.40 |
| A/CNK | 1.2 | 1.1 | 1.1 | 1.1 | 1. 2 | 1. 2 | 1. 1 | 1.1 | 1.1 |
| DI | 82.9 | 88.8 | 80.8 | 88.6 | 91. 6 | 88.3 | 89. 4 | 87. 0 | 88.8 |
| Ba | - | - | - | - | - | - | - | - | - |
| Rb | - | - | - | - | - | - | - | - | - |
| Sr | - | - | - | - | - | - | - | - | - |
| Y | - | - | - | - | - | - | - | - | - |
| Zr | - | - | - | - | - | - | - | - | - |
| Nb | - | - | - | - | _ | _ | - | _ | - |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Ga | - | - | - | - | - | - | - | _ | - |
| Zn | - | - | - | - | - | - | - | - | - |
| Cu | - | - | - | - | - | - | - | - | - |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| $V$ | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | _ | - | $\cdots$ | _ | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | _ | - | _ | - | - |
| Ta | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | - | - | - | - | - | - | - | _ | _ |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | - | - | - | - | _ | - | - |
| F | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | _ | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | - |
| Mo | - | - | - | - | - | - | - | - | - |
| Lo | - | - | - | - | - | - | _ | _ | _ |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical databose.

AUSTRALIA

|  | AWB21 UNKN GRAN | AWB22 UNKN GRAN | AWB25 UNKN GRAN | AWB27 <br> UNKN <br> GRAN | AWB29 UNKN GRAN | AWB 4 UNKN GRAN | AWE 6 UNKN GRAN | AWB9 UNKN GRAN | TBT13 UNKN GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 74. 40 | 72. 53 | 72. 33 | 72.79 | 74.47 | 75.10 | 73. 28 | 76. 20 | 71.93 |
| $\mathrm{TiO}_{2}$ | . 13 | . 42 | . 49 | . 41 | . 22 | . 06 | . 41 | . 15 | . 49 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 13. 59 | 13. 72 | 12.93 | 13.03 | 12. 82 | 12.90 | 13.52 | 12.47 | 14. 03 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | . 46 | . 81 | . 70 | 1. 06 | . 56 | . 68 | . 87 | . 71 | 3.50 |
| FeO | 1. 48 | 1.93 | 2.15 | 1. 56 | 1.71 | . 46 | 2.18 | . 82 | - |
| MnO | . 02 | . 04 | . 04 | . 03 | . 03 | . 02 | . 04 | . 05 | 08 |
| MgO | . 38 | . 86 | . 89 | . 98 | . 40 | . 17 | 1. 01 | . 32 | . 84 |
| CaO | 1.38 | 1.85 | 1.48 | 1. 31 | 1. 63 | 1.26 | 2. 27 | 1. 20 | 2. 01 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 2.89 | 2. 41 | 2. 33 | 2.15 | 2. 60 | 3.14 | 3.25 | 2. 81 | 2. 95 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4.32 | 4.43 | 4. 32 | 4.69 | 3.90 | 4.43 | 2. 10 | 4. 65 | 4.15 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 10 | . 12 | . 13 | . 11 | . 10 | . 02 | . 07 | . 03 | . 16 |
| $\mathrm{H}_{2} \mathrm{O}+$ | . 99 | . 91 | . 91 | 1. 42 | . 92 | . 57 | . 98 | . 48 | - |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | _ | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | - | - | - | - | - | - | - | - | . 83 |
| TOTAL | 100.14 | 100.03 | 98.70 | 99.54 | 99. 36 | 99. 81 | 99. 98 | 99.89 | 100.95 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{t}$ | 2.10 | 2. 95 | 3.09 | 2.79 | 2. 46 | 1. 19 | 3. 29 | 1.62 | 3.50 |
| A/CNK | 1.1 | 1.1 | 1.2 | 1.2 | 1.1 | 1.1 | 1.2 | 1.1 | 1.1 |
| DI | 87.7 | 83. 0 | 82.9 | 84.9 | 85.4 | 91. 2 | 79.6 | 90.4 | 83.3 |
| Bo | - | - | - | - | - | - | - | - | 538. |
| Rb | - | - | - | - | - | - | - | - | 229. |
| Sr | - | - | - | - | - | - | - | - | 132. |
| Y | - | - | - | - | - | - | - | - | 36. |
| Zr | - | - | - | - | - | - | - | - | 195. |
| Nb | - | - | - | - | - | - | - | - | 12. |
| Th | - | - | - | - | - | - | - | - | 12. |
| Pb | - | - | - | _ | - | - | - | - | - |
| Go | - | - | - | - | - | - | - | _ | 20. |
| Zn | - | - | - | - | - | - | - | _ | - |
| Cu | - | - | - | - | - | - | - | - | _ |
| Ni | - | - | - | - | - | - | - | - | _ |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| $\checkmark$ | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | - | _ | _ | - | - |
| Cs | - | - | - | - | _ | - | _ | - | - |
| Sc | - | _ | - | _ | - | - | - | - | - |
| To | - | - | - | - | _ | - | - | - | - |
| Co | - | - | - | - | - | - | - | $\rightarrow$ | - |
| Li | - | - | - | - | - | - | - | - | - |
| Be | - | - | - | - | - | - | - | - | - |
| B | - | - | - | - | - | - | - | - | - . |
| F | - | - | - | - | - | - | - | - | 770. |
| Cl | - | - | - | - | _ | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | - | - | - | - | - | - | - | - | 29.0 |
| Mo | - | - | - | - | - | - | - | - | - |
| Lo | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical database.

## AUSTRALIA

|  | TBT20 UNKN GRAN | TBT24 UNKN GRAN | TBT26 UNKN GRAN | TBT31 UNKN GRAN | TBT36 UNKN GRAN | TBT39 UNKN GRAN | TBT42 UNKN GRAN | TBT45 UNKN GRAN | TBT47 UNKN GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 74.45 | 72. 24 | 74.00 | 74. 18 | 74. 88 | 75.75 | 77. 30 | 75. 89 | 75.55 |
| $\mathrm{TiO}_{2}$ | 30 | . 26 | 15 | . 50 | . 28 | Q8 | 08 | . 13 | . 10 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 12.70 | 14.45 | 13.90 | 12.93 | 13.10 | 13.45 | 13.10 | 13.44 | 12. 54 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 2.27 | 2.01 | 1.85 | 3. 20 | 2.21 | 1.47 | 1. 68 | . 39 | 1. 56 |
| FeO | - | - | - | - | - | - | - | - | - |
| MnO | . 04 | . 04 | . 02 | . 04 | . 02 | . 02 | . 04 | . 81 | . 04 |
| MgO | . 34 | . 54 | . 38 | . 69 | . 18 | . 13 | . 05 | . 08 | . 07 |
| CoO | 1.47 | 1. 64 | 1.10 | 1.51 | . 97 | 85 | . 49 | . 55 | . 81 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 3.19 | 2. 82 | 2. 90 | 2. 34 | 2. 06 | 3.45 | 2. 90 | 2. 91 | 2.83 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4.15 | 4.65 | 4.80 | 4.03 | 4.83 | 4. 75 | 5.10 | 4.52 | 4.68 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 10 | . 19 | . 11 | . 13 | . 13 | - | . 06 | . 07 | . 04 |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}$ | - | - | - | - | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - | - | - | - | - |
| Cl | - | - | - | - | - | - | - | - | - |
| F | - | - | - | - | - | - | - | - | - |
| LOI | . 62 | . 57 | 1.85 | . 68 | . 77 | . 51 | 1.00 | 1.19 | . 76 |
| TOTAL | 99. 63 | 99.41 | 101.07 | 100.53 | 99.43 | 100.46 | 101.78 | 99.18 | 98. 98 |
| $\mathrm{Fe}_{2} \mathrm{O} 3 \mathrm{t}$ | 2. 27 | 2. 01 | 1. 85 | 3. 20 | 2. 21 | 1.47 | 1.68 | . 39 | 1.85 |
| A/CNK | 1.0 | 1.1 | 1. 2 | 1.2 | 1.3 | 1.1 | 1.2 | 1.3 | 1.2 |
| DI | 88.2 | 85.6 | 89.1 | 85.2 | 88.4 | 92.8 | 94.6 | 91.9 | 91.5 |
| Ba | - | 369. | 298. | 683. | 387. | 267. | 27. | 141. | 148. |
| Rb | 230. | 385. | 383. | 223. | 294. | 232. | 515. | 543. | 393. |
| Sr | 102. | 103. | 93. | 103. | 62. | 62. | 9. | 31. | 31. |
| Y | 33. | 22. | 23. | 33. | 35. | 44. | 58. | 43. | 54. |
| Zr | 131. | 119. | 109. | 257. | 178. | 73. | 68. | 106. | 110. |
| Nb | 14. | 18. | 11. | 12. | 13. | 10. | 14. | 15. | 13. |
| Th | - | - | - | - | - | - | - | - | - |
| Pb | - | - | - | - | - | - | - | - | - |
| Ga | 15. | 19. | 17. | 16. | 20. | 18. | 25. | 23. | 22. |
| Zn | - | - | - | - | - | - | - | - | - |
| Cu | - | - | - | - | - | - | - | - | - |
| Ni | - | - | - | - | - | - | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - | - | - | - | - |
| V | - | - | - | - | - | - | - | - | - |
| Cr | - | - | - | - | - | - | - | - | - |
| Hf | - | - | - | - | _ | - | - | - | - |
| Cs | - | - | - | - | - | - | - | - | - |
| Sc | - | - | - | - | - | - | - | - | - |
| To | - | - | - | - | - | - | - | - | - |
| Co | - | - | - | - | - | - | - | - | - |
| Li | 55.0 | 112.0 | 200.0 | 41.0 | 64.0 | 100.0 | 122.0 | 98.0 | 92.0 |
| Be |  | - |  | 11. | - | , | 1 |  | 92.0 |
| B | - | - | - | - | - | - | - | - | - |
| F | 580. | - | 1500. | 1200. | 1500. | 340. | 2880. | 2820. | 2200. |
| Cl | - | - | - | - | - | - | - | - | - |
| U | - | - | - | - | - | - | - | - | - |
| W | - | - | - | - | - | - | - | - | - |
| Sn | 9. 0 | 11.0 | 15.0 | 5. 0 | 7. 0 | 6. 0 | 32.0 | 87. 0 | 16.0 |
| Mo | . | 11. | - | - | - | - | - | - | - |
| La | - | - | - | - | - | - | - | - | - |
| Ce | - | - | - | - | - | - | - | - | - |

Figure D. 1 (cont.). Geochemical databose.

## AUSTRALIA

|  | TBT49 UNKN GRAN | TBT51 UNKN GRAN | TBT60 UNKN GRAN | TBT61 UNKN GRAN | T8T8 UNKN GRAN |
| :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 75. 50 | 73. 50 | 76.51 | 76.07 | 71.86 |
| $\mathrm{TiO}_{2}$ | - | . 03 | . 01 | 02 | . 33 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14.00 | 15.10 | 13.28 | 13.30 | 14.08 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 1. 50 | . 94 | . 49 | 1. 12 | 2. 42 |
| FeO | - | - | - | - | - |
| MnO | . 05 | . 06 | - | . 02 | . 05 |
| MgO | . 03 | . 04 | . 06 | . 07 | . 57 |
| CaO | . 36 | . 47 | . 37 | . 42 | 1.74 |
| $\mathrm{No}_{2} \mathrm{O}$ | 3.60 | 3.30 | 4. 04 | 4.31 | 3.52 |
| $\mathrm{K}_{2} \mathrm{O}$ | 4. 30 | 4.30 | 4.53 | 4. 39 | 4.46 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | . 09 | . 11 | . 01 | - | .12 |
| $\mathrm{H}_{2} \mathrm{O}+$ | - | - | - | - | - |
| $\mathrm{H}_{2} \mathrm{O}-$ | - | - | - | - | - |
| $\mathrm{CO}_{2}$ | - | - | - | - | - |
| Cl | - | - | - | - | - |
| $F$ | - | - | - | - | - |
| LOI | 1.11 | 81 | . 69 | . 60 | 1. 24 |
| TOTAL | 100.54 | 98. 65 | 99. 99 | 100. 32 | 100.39 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} t$ | 1. 50 | . 94 | .49 | 1. 12 | 2. 42 |
| A/CNK | 1.3 | 1.4 | 1.1 | 1.1 | 1.0 |
| Di | 93.4 | 90.4 | 95.8 | 95.6 | 86.2 |
| Bo | 20. | 22. | 35. | 31. | 371. |
| Rb | 820. | 1562. | 353. | 356. | 270. |
| Sr | Б. | 11. | 7. | 7. | 121. |
| Y | 31. | 19. | 60. | 65. | 36. |
| Zr | 34. | 25. | 63. | 68. | 123. |
| Nb | 20. | 60. | 16. | 17. | 11. |
| Th | - | - | - | - | - |
| Pb | - | - | - | - | - |
| Go | 32. | 41. | 21. | 23. | 16. |
| Zn | , | 1 | 21. | , | 16. |
| Cu | - | - | - | - | - |
| Ni | - | _ | - | - | - |
| $\mathrm{TiO}_{2}$ | - | - | - | - | - |
| $V$ | - | - | - | - | - |
| Cr | - | - | - | - | - |
| Hf | - | - | - | - | - |
| Cs | - | - | - | - | - |
| Sc | - | - | - | - | - |
| Ta | - | - | - | - | - |
| Co | - | - | - | - | - |
| Li | 260.0 | 465. 0 | 25.0 | 63.0 | 77.0 |
| Be | 260. | . | - | 6 | 7 |
| B | - | - | - | - | - |
| $F$ | 4240. | 4020. | - | 740. | 820. |
| Cl |  | - | - | - | - |
| U | - | _ | - | - | - |
| W | - | - | - | - | $\stackrel{ }{ }$ |
| Sn | 30.0 | 35.0 | 10.0 | 11.0 | 12.0 |
| Mo | - | - | - | - | - |
| Lo | - | - | - | - | - |
| Ce | - | - | - | - | - |

Figure 0.1 (cont.). Geochemical database.


Figure E.1. Frequency distributions of major and trace element data from Northern Nova Scotia.


Figure E. 1 (cont.). Frequency distributions of major and trace element data from Northern Nova Scotia.


Figure E. 1 (cont.). Frequency distributions of major and trace element data from Northern Nova Scotia.


Figure E.2. Frequency distributions of major and trace element data from Southern Nova Scotia.


Figure E. 2 (cont.). Frequency distributions of major and trace element data from Southern Nova Scotia.


Figure E. 2 (cont.). Frequency distributions of major and trace element data from Southern Nova Scotia.


Figure E.3. Frequency distributions of major and trace element data from Nova Scotia.


Figure E. 3 (cont.). Frequency distributions of major and trace element data from Nova Scotia.


Figure E. 3 (cont.). Frequency distributions of major and trace element data from Nova Scotia.


Figure E.4. Frequency distributions of major and trace element data from Morocco.


Figure E. 4 (cont.). Frequency distributions of major and trace element data from Morocco.


Figure E. 4 (cont.). Frequency distributions of major and trace element data from Morocco.


Figure E.5. Frequency distributions of major and trace element data from Iberia.


Figure E. 5 (cont.). Frequency distributions of major and trace element data from Iberia.


Figure E.6. Frequency distributions of major and trace element data from Australia.


Figure E. 6 (cont.). Frequency distributions of major and trace element data from Australia.

## APPENDIX F

## ADDITIONAL INFORMATION ON DISCRIMINANT FUNCTION ANALYSIS

## F. 1 Determining the number of discriminant functions

Generally the number of discriminant functions calculated in an analysis is determined by the number of groups (g) and independent (i.e. discriminating) variables ( p ). The maximum number of possible discriminant functions calculated in an analysis will be the smaller of $\mathrm{g}-1$ and p . For example in a four-group and ten-variable discriminant analysis, a maximum of 3 functions could be computed.

The importance of the number of groups stems from the basic geometric principle that the maximum number of dimensions needed to describe a set of points is one less than the number of points. The degenerate situation of three points falling within a line, for example, is an exception to this rule. The last point falls into the space defined by the two first points and does not add a new dimension. The same principles apply to discriminant analysis. Each group (as defined by its centroid) is treated as a point and each discriminant function is a unique dimension describing the location of that group relative to the others. In some cases not all discriminant functions are significant or useful. For example, if in a four-group analysis only two discriminant functions are computed this case is degenerate and the third function is omitted because of its statistical insignificance. That is, although the two first discriminant functions do not contain all of the information in the discriminating variables, the small amount left in the third possible function is ignored because it is judged to be statistically insignificant.

## F. 2 Judging the Importance of a Discriminant Function

The eigenvalue is the measure of the relative importance of a particular discriminant function. The sum of all eigenvalues
represents the total variance in the discriminating variables. Therefore, the relative importance of an eigenvalue (consequently the discriminant function) can be calculated as a percentage of the total sum of eigenvalues.

## F. 3 Discriminant Function Coefficients and Discriminant Scores

A discriminant score is calculated using the discriminant function in which the raw value for each variable is multiplied by its corresponding coefficient and these products are then added together. There is a separate score for each case and each discriminant function. The group mean of a function is calculated by averaging the scores for all cases within a particular group. The group centroid is the mean of all functions in a given group.

## F. 4 Plots of Discriminant Scores

In a two-group model in which one discriminant function is calculated, a histogram is used to depict the discriminant scores. When two discriminant functions are calculated, the discriminant scores are plotted in binary form with each function representing an axis. In addition a territorial map is also produced. Each point on this graph is classified according to its relationship with the nearest group centroid (see Nie et al., 1975 Section 23.2.5 for details on classification rules). Only the borders of regions for each group are shown on the final printout. This is done by plotting only the points not completely surrounded by points classified in the same group. The result is a graph which could identify group membership of a case.

## F. 5 Minimum Tolerance Test

Discriminant analysis determines the tolerance level of each variable in an analysis to avoid difficulties in subsequent calculations (Nie et al., 1975). Essentially a low tolerance level indicates that the program would encounter some difficulty during
matrix inversions, and if such a variable where to be included in the analysis, large rounding errors would result and lead to faulty classifications of the data. Thus, if a variable fails the minimum tolerance test, discriminant function analysis will not include this variable in the analysis.

## F. 6 Wilks' Lambda Criterion

The differences between several group means can generally be tested using the Wilks' Lambda criterion. However, in cases where the data are closed, the Wilks' Lambda is indeterminate and can not be calculated using the standard method. An alternative method exists which can calculate the Wilks' Lambda value by matrix inversion. It is, however, unclear in SPSS how the discriminant procedure would treat this statistic in such a situation. Because of this uncertainty, the Wilks' Lambda values were not considered in this thesis.

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